

RECLAMATION

Managing Water in the West

**Desalination and Water Purification Research
and Development Program Report No. 165**

Demonstration of Zero Discharge Desalination (ZDD)



**U.S. Department of the Interior
Bureau of Reclamation
Technical Service Center
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**Desalination and Water Purification Research
and Development Program Report No. 165**

Demonstration of Zero Discharge Desalination (ZDD)

**Prepared for the Bureau of Reclamation
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**U.S. Department of the Interior
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Contents

	Page
1. Executive Summary.....	1
2. Introduction.....	3
2.1 Project Objectives and Goals	4
2.2 Project Approach	5
3. Review of Concentrate Management Technologies.....	7
3.1. Thermal Zero Liquid Discharge.....	8
3.2 Evaporation Ponds	8
3.3 Deep Well Injection	8
3.4 High Recovery Desalination and Salt Recovery.....	8
3.4.1 Volume Reduction Technologies.....	9
3.4.2 Volume Reduction with Salt Recovery Technologies	10
4. Zero Discharge Desalination	13
4.1 ZDD Technology Development and Funding History.....	13
4.2 Process Description.....	15
4.3 Process Configurations	15
4.4 Pre-Treatment	18
4.5 Reverse Osmosis and Nanofiltration	18
4.6 Electrodialysis Metathesis	20
4.7 DI Water System.....	26
4.8 Salt Recovery	26
5. Demonstration Testing Locations.....	29
5.1 Alamogordo, New Mexico.....	29
5.2 La Junta and Brighton, Colorado	30
5.3 Beverly Hills, California.....	30
5.4 Water Quality Analysis for Testing Sites	30
6. Alamogordo Results and Discussion.....	33
6.1 Feed Water Quality.....	33
6.2 Equipment Description	34
6.2.1 Phase 1 NF Membrane Evaluation	34
6.2.2 Phase 2 Demonstration Activities.....	36
6.3 Phase 1 Operations.....	37
6.4 Phase 2 Operations.....	41
6.4.1 Preliminary Experiments (Before 4/22/13).....	41
6.4.2 Experiment 1a and 1b (4/22/13-4/26/13).....	45
6.4.3 Experiment 2 (4/30/13-5/2/13)	48
6.4.4 NF Membrane Replacement (5/23/13)	50
6.4.5 Experiment 3 (5/29/13-6/4/13)	50
6.4.6 Experiment 4 (6/17/13-6/21/13)	54
6.4.7 Experiment 5 (6/24/13-6/28/13)	55
6.4.8 Experiment 6 (7/8/13-7/14/13)	56
6.4.9 Summary of Phase 2 Results	58
6.3 Evaluation of Current Density, Current Efficiency, Power Consumption – Alamogordo.....	70
6.3.1 Current Density.....	70

Demonstration of Zero Discharge Desalination (ZDD)

6.3.2 EDM Stack Performance Monitoring	72
6.3.3 Energy Consumption	75
7. La Junta Results and Discussion.....	78
7.1 Equipment Description	78
7.2 Summary of Pilot Testing	80
7.3 Evaluation of Current Density, Current Efficiency, Power Consumption – La Junta	88
7.3.1 EDM Stack Performance Monitoring	88
7.3.2 Power Consumption.....	90
8. Brighton Results and Discussion.....	93
8.1 Equipment Description	93
8.2 Pilot System Upgrades and Commissioning.....	95
8.3 Experiment 1 (9/30/13-10/13/13)	96
8.4 Experiment 2 (10/15/13-10/19/13)	97
8.5 Experiment 3 (10/22/13-10/26/13)	97
8.6 Summary of Brighton Pilot Operations	98
8.6.1 Early Problem Discussion.....	98
8.6.2 Summary of Brighton Operations by UTEP (9/30/13-10/26/13) ..	101
8.7 Evaluation of Current Density, Current Efficiency, Power Consumption – Brighton.....	109
8.8 EDM Stack Performance Monitoring.....	109
9. Beverly Hills Pilot	113
10. Cost Analysis.....	116
10.1 Cost Development Approach.....	116
10.2 Technology Cost Estimates – Alamogordo	119
10.2.1 Treatment Process Descriptions – Alamogordo	119
10.2.2 Capital Cost Development – Alamogordo.....	120
10.2.3. Operation and Maintenance Cost Development – Alamogordo	122
10.2.4 ZDD & BWRO Residuals Cost Development – Alamogordo.....	123
10.3 Technology Cost Estimates – La Junta.....	125
10.3.1 Treatment Process Descriptions – La Junta	125
10.3.2 Capital Cost (CapEx) Development – La Junta.....	126
10.3.3 Operation and Maintenance Cost Development	128
10.3.4 ZDD and BWRO Residuals Cost Development – La Junta	130
10.3.5 Comparison of BWRO and ZDD Cost Estimates – La Junta	130
11. Salt Recovery.....	133
11.1 NaCl Purity Requirements	133
11.2 Feasibility of Supernatant Use in EDM for NaCl Supply.....	136
11.3 Feasibility of Using Evaporation Pond to Recover NaCl	139
11.4 High Purity NaCl Recovery Using Electrodialysis and Recovery of Mg(OH) ₂	144
11.5 Voltage Drop in ED Stacks.....	154
12. Conclusions and Recommendations.....	157
12.1 Technical Performance and Cost Evaluation.....	157
12.2 Problems Encountered & Solutions	161

12.3 Recommendations and Future Research..... 162
 Bibliography165

Appendices

Appendix A Grab Sample Data
 Appendix B Online Data
 Appendix C Water Quality Analyses
 Appendix D Mega Evaluations
 Appendix E Cost Calculations

Figures

	<i>Page</i>
Figure 1.—Greenfield ZDD flow diagram (simplified).	16
Figure 2.—Bolt-on ZDD flow diagram (simplified).	17
Figure 3.—EDM only ZDD flow diagram (simplified).....	18
Figure 4.—Flow of ions in electro dialysis.....	21
Figure 5.—Concentration polarization in electro dialysis showing a) concentration profile and b) electrode reactions.....	21
Figure 6.—Limiting current density in electro dialysis.	22
Figure 7.—Decreased temperature leads to increased ED resistance.....	23
Figure 8.—Solubility of dissolved salts (Siedell, 1907).	24
Figure 9.—Flow of ions in electro dialysis metathesis.....	25
Figure 10.—Phase 1 ZDD system installed in Alamogordo (20 gpm).....	34
Figure 11.—Phase 1 equipment flow diagram.	35
Figure 12.—Phase 2 ZDD System Installed in Alamogordo (40 gpm). Showing a) containers, NaCl tanks, and sulfuric acid tanks; b) NF system; c) EDM standpipes and pumps; and d) EDM stacks.	36
Figure 13.—Phase 2 flow diagram.	37
Figure 14.—Estimated EDM stack specific energy at Alamogordo (Phase 1)....	40
Figure 15.—Grab sample data (preliminary experiments).	42
Figure 16.—EDM-2 Stack with corrected D1 (NaCl outlet) spacer installation..	43
Figure 17.—EDM-1 Stack with membrane piece from manufacturer.	44
Figure 18.—EDM-2 center membrane (top) and center divider (bottom).....	44
Figure 19.—EDM-2 stack showing melted anode spacers and adjacent damaged membranes (C2 is mixed Na).	46
Figure 20.—Grab sample data (Experiment 1).....	47
Figure 21.—NF cartridge filters on May 3. Left: 20-micron filters on source water. Right: 5-micron filters on NF feed.....	48
Figure 22.—Grab sample data (Experiment 2).....	49
Figure 23.—Original NF270 membranes removed on May 23.....	50
Figure 24.—Grab sample data (Experiment 3).....	52
Figure 25.—EDM-2 precipitation and damage from mixing at center of stack. ..	53
Figure 26.—Grab sample data (Experiment 4).....	55
Figure 27.—Grab sample data (Experiment 5).....	56
Figure 28.—Grab sample data (Experiment 6).....	58

Demonstration of Zero Discharge Desalination (ZDD)

Figure 29.—Average Alamogordo Phase 2 water quality for ZDD inlet (a) and outlet streams (b-d).	59
Figure 30.—Variability in Alamogordo Phase 2 Water Quality for ZDD inlet and outlet streams.	60
Figure 31.—EDM stack construction at center and electrodes.....	63
Figure 32.—Bicarbonate and sulfate contamination in NaCl.....	64
Figure 33.—Bicarbonate and sulfate increase over time - Mixed Cl.	64
Figure 34.—pH increase over time – Mixed Cl and NaCl.	64
Figure 35.—Comparison of normal and swollen AEMs.	65
Figure 36.—Relationship between conductivity and specific gravity (Alamogordo – Phase 2)	67
Figure 37.—Limiting Current Density – Alamogordo Phase 2.....	71
Figure 38.—Limiting current density and effects on EDM concentrate stream pH (Alamogordo Phase 2).	71
Figure 39.—EDM stack resistance – Phase 2.....	72
Figure 40.—Alamogordo current efficiency (Phase 2).....	74
Figure 41.—EDM stack resistance and estimated specific energy (Phase 2).....	75
Figure 42.—Comparison of Phase 1 and Phase 2 specific energy.	76
Figure 43.—ZDD and NaCl recovery systems installed in La Junta (20-gpm)....	78
Figure 44.—Equipment flow diagram (La Junta).....	79
Figure 45.—Comparison of feed flow estimates and NF permeate flow.	81
Figure 46.—ZDD flow diagram showing mass balance measuring points.	82
Figure 47.—Comparison of recovery calculation method results (La Junta).	83
Figure 48.—NaCl and Mixed Chloride pH at La Junta.	85
Figure 49.—NaCl conductivity in La Junta (8/21 – 8/24).....	86
Figure 50.—Loss of NaCl conductivity in La Junta (9/17 - 9/21).....	86
Figure 51.—NaCl conductivity in La Junta (8/27 -8/31).....	87
Figure 52.—EDM feed flow in La Junta (8/21 – 9/1).	88
Figure 53.—EDM stack resistance (La Junta).....	89
Figure 54.—Estimated power EDM consumption (La Junta).	91
Figure 55.—ZDD system installed in Brighton (20-gpm).....	93
Figure 56.—Equipment flow diagram (Brighton).	94
Figure 57.—Solubility and temperature relationships between a) Na ₂ SO ₄ -NaHCO ₃ (Marion and Farren, 1999) and b) NaCl (Marion, 2001).	99
Figure 58.—Damaged EDM cathode electrical connection. Photo taken on 8/29/13,	100
Figure 59.—EDM cathode inspection. Photo taken on 9/9/13).....	100
Figure 60.—RO concentrate and NF concentrate flow metering (Brighton).	101
Figure 61.—Brighton grab sample data.....	104
Figure 62.—Average Brighton water quality for ZDD Inlet (a) and outlet streams (b-d).....	105
Figure 63.—Variability in Brighton water quality for ZDD inlet and outlet streams.	106
Figure 64.—ZDD inlet and outlet conductivity monitoring (Brighton).	106
Figure 65.—Typical limiting current density – Brighton (7 cm/s velocity).	109

Demonstration of Zero Discharge Desalination (ZDD)

Figure 66.—Brighton EDM performance and outdoor temperature data near Brighton (Denver International Airport (DIA), Denver, Colorado).	110
Figure 67.—Current density and pH difference (Brighton).....	111
Figure 68.—Brighton current efficiency.....	112
Figure 69.—NaCl contamination – Beverly Hills pilot.....	114
Figure 70.—Process flow diagram of a ZDD system.....	118
Figure 71.—Process flow diagram of a BWRO system.....	118
Figure 72.—Capital cost estimates for ZDD, BWRO, and Dual RO (6.6 MGD).	131
Figure 73.—Unit cost estimates for ZDD, BWRO, and Dual RO (6.6 MGD)...	132
Figure 74.—Transport of ions through ion-exchange membranes in EDM stack.	134
Figure 75.—CaSO ₄ solubility in NaCl solution at 28°C and calculated ion ratios in saturated CaSO ₄ solution.	135
Figure 76.— a) Gypsum solubility in NaCl and MgCl ₂ solutions and (b) Gypsum solubility in NaCl solutions with increasing concentrations of MgCl ₂ (Ostroff & Metler, 1966).....	136
Figure 77.—Operations data from Supernatant Study at BGNDRF.....	138
Figure 78.—Setup for lab ED of supernatant.	139
Figure 79.—Full scale salt (NaCl, CaSO ₄ , Mg(OH) ₂) recovery diagram.....	140
Figure 80.—Recovery of NaCl from EDM concentrate streams by precipitation and evaporation.	140
Figure 81.—Comparison of Na/Ca and Cl/SO ratios and stream pH during Phase 2 Operations at BGNDRF. Left: NaCl stream ratios & pH; Right: Mixed Na & Mixed Cl ratios, Mixed Cl pH.....	141
Figure 82.—Evaporation experiment using EDM concentrates from BGNDRF 40 gpm demonstration plant.....	142
Figure 83.—Selective transport of monovalent ions in electrodialysis.	145
Figure 84.—Hydrated lime injection system (Con-V-Air Solutions, n.d.).....	146
Figure 85.—Addition of Ca(OH) ₂ powder into simulated Mixed Cl solution. ..	146
Figure 86.—Stopping the addition of Ca(OH) ₂ powder at the endpoint of the plateau eliminates undissolved lime in the Mg(OH) ₂ product.....	147
Figure 87.—Molar concentrations (based on conductivity measurements) of NaCl in diluate and concentrate of ED stack.	148
Figure 88.—on ratios in feed-and-bleed mode of ED operation.	149
Figure 89.—Composition of solution in feed tank during batch ED experiments.	152
Figure 90.—Voltage drop in TS-2 stack with electrodes and a CMB membrane.	155
Figure 91.—Measured potential and current of the ED stack as it was being shut down on 12/21/2012.	156
Figure 92.—Capital cost estimates for ZLD, ZDD and BWRO for 1 MGD and 4 MGD designs.	157
Figure 93.—Production cost estimates for ZLD, ZDD and BWRO for 1 MGD and 4 MGD designs.	159

Demonstration of Zero Discharge Desalination (ZDD)

Figure 94.—Comparison of 4 MGD ZDD System with and without NaCl Recovery (Alamogordo).	160
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Tables

	<i>Page</i>
Table 1.—Comparison of ZLD, ZDD, and High Recovery Approaches	7
Table 2.—Comparison of Dow Filmtec Membranes to Produce 1 MGD Permeate from Average Snake Tank Brackish Water	19
Table 3.—Snake Tank Water Quality (average of 3 wells).....	27
Table 4.—Selected Feedwater Quality Data from ZDD Testing Sites	31
Table 5.—Comparison of Alamogordo Snake Tank and BGNDRF Brackish Water Chemistry	33
Table 6.—Alamogordo Phase 1 Operational Data and Field Water Quality Analyses	38
Table 7.—Phase 1 Water Quality Analysis Summary	39
Table 8.—Phase 2 Alamogordo Water Quality Summary (mg/L)	62
Table 9.—Anion Membrane Soak – NaCl Solution IC Results	65
Table 10.—Summary of Alamogordo Phase 2 Operational Data.....	68
Table 11.—Average Current Efficiency (Alamogordo-Phase 2)	74
Table 12.—Comparison of Full Scale and Pilot Demonstration Operating Setpoints.....	76
Table 13.—Alamogordo Estimated Specific Energy (4 MGD Design)	77
Table 14.—Comparison of RO and NF Membranes for La Junta Desalination...	80
Table 15.—Comparison of Recovery Results by Experiment (La Junta).	84
Table 16.—WERF 5T10 Report Energy Consumption	90
Table 17.—Estimated Full Scale Unit Energy Consumption (La Junta).....	92
Table 18.—Summary of Brighton Operational Data	102
Table 19.—Brighton IC Analyses (all in mg/L)	107
Table 20.—Brighton General Water Quality Analyses	108
Table 21.—Average Current Efficiency (Brighton).	111
Table 22.—Black & Veatch EDM Study Summary (Bond et al., 2015).....	114
Table 23.—AACE Cost Estimation Classification (AACE, 2000)	117
Table 24.—Major Operational Cost Comparison by Product Flow Rate and System (Alamogordo).....	119
Table 25.—Comparison of Evaporation Pond CapEx Estimation Costs (\$ Million)	124
Table 26.—Comparison of DWI CapEx Estimation Costs (\$ Million).....	125
Table 27.—Major Operational Cost Comparison by Product Flow Rate and System (La Junta, 6.6 MGD)	129
Table 28.—Comparison of Evaporation Pond CapEx Estimation Methods (La Junta, 6.6 MGD full capacity).....	130
Table 29.—Summary of Prepared NaCl Solutions and Circulating NaCl Streams (all in meq/L)	137
Table 30.—Ion Ratios and Relative Transport during Batch ED of Supernatant Prepared with Simulated Alamogordo EDM Concentrate Streams.....	151

Demonstration of Zero Discharge Desalination (ZDD)

Table 31.—Ion Ratios during Batch ED of Supernatant with Neosepta CMS and ACS Membranes at Low Current Density 153

Table 32.—Ion Ratios During Batch ED of Supernatant with Neosepta CMS and ACS Membranes at High Current Density 154

Table 33.—Energy Consumption for NaCl Recovery by ED..... 156

Table 34.—Conceptual NaCl Recovery Costs (4 MGD, Alamogordo) 160

Acronyms and Abbreviations

AACE	American Association of Cost Engineers
AEM	anion-exchange membrane
AFY	acre feet per year
AMTA	American Membrane Technology Association
BGNDRF	Brackish Groundwater National Desalination Research Facility
BWRO	Brackish water reverse osmosis
CapEx	Capital Cost (Expense)
CapEx	Capital Cost
CCD™	Closed circuit desalination
CEM	cation-exchange membrane
CERRO™	Concentrate Enhanced Recovery Reverse Osmosis (),
CIDS	Center for Inland Desalination Systems
CIP	Clean In Place
DI	deionized
DWI	deep well injection
DWPR	Desalination and Water Purification Research
ED	electrodialysis
EDM	electrodialysis metathesis
EDR	electrodialysis reversal
E-Rinse	electrode rinse
EDM	electrode rinse stream
EWM	Enviro Water Minerals Company
F	Faraday's constant
GAO	Government Accountability Office
Hall	Hall Environmental Analysis Laboratory
HDPE	High density polyethylene
HEEPM™	High Efficiency Electro-Pressure Membrane
HERO™	High Efficiency Reverse Osmosis
I/O	input/output
IC	ion chromatography
ICP-OES	inductively coupled plasma atomic emission spectroscopy
IWA	International Water Association (co-publisher of WERF 5T10 report)
MEGA	MEGA Group
NF	nanofiltration
NTU	Nephelometric Turbidity Unit
O&M	operations and maintenance
OpEx	operational cost (expense)
Panoche	Panoche Irrigation District of California ()
PLC	programmable logic controller
quad	Repeating set in an EDM stack with four membranes and four spacers
Reclamation	Bureau of Reclamation
RO	Reverse osmosis
RTN	Relative transport number
Sandia	Sandia National Laboratories
SARI	Santa Ana Regional Interceptor
SCADA	Supervisory control and data acquisition
SMCL	secondary maximum contaminant level
SPARRO	slurry precipitation and recycle reverse osmosis
Stack	EDM system comprised of electrodes, membranes, spacers, and connectors
TDS	total dissolved solids
USC	University of South Carolina
UTEP	The University of Texas at El Paso
VAC	volts alternating current
VDC	volts direct current
Veolia	Veolia Water Technologies
VSEP™	Vibratory shear enhanced process (
WERF	Water Environment Research Foundation (co-publisher of WERF 5T10)
ZDD	Zero Discharge Desalination
ZLD	Zero Liquid Desalination

Demonstration of Zero Discharge Desalination (ZDD)

Chemical Abbreviations

CaCl ₂	calcium chloride,
CaCO ₃	calcium carbonate
Ca ₃ (PO ₄) ₂	calcium phosphate
CaSO ₄	calcium sulfate
CO ₂	carbon dioxide,
HCO ₃	bicarbonate
KCl	potassium chloride
Mg(OH) ₂	magnesium hydroxide
Na/Ca	Ratio of concentration of sodium to calcium (in meq/L)
NaCl	sodium chloride
NaOH	sodium hydroxide,
Na ₂ SO ₄	sodium sulfate,

Measurements

\$/kgal	dollars per thousand gallons
A	amps
A/m ²	amperes per square meter
cm ²	square centimeters ()
cm/s	centimeters per second
eq/L	equivalent per liter
ft	feet
ft/sec	feet per second
GFD	gallon per square foot per day
GPD	gallon of daily capacity
gph	gallon per hour
gpm	gallon per minute
kgal	1000 gallons
kWh	kilowatt hour
kWh/kgal	kilowatt hours per thousand gallons
L	liter
lb	pound
lb/hr	pounds per hour
M	moles per liter
mA/cm²	milliampere per square centimeter
Meq/L	milliequivalents of solute per litre of solvent
MGD	million gallons per day
mg/L	milligrams per liter
mS/cm	millisiemens per centimeter [mS/cm]
mT/m ³	metric tons per cubic meter [
mT/d	metric tons per day
m/yr	meters per year ().
psi	pounds per square inch
V	volts
μS/cm	= microsiemen per centimeter

Metric Conversions

The metric equivalents for non-metric units used in the text are as follows:

Unit	Metric Equivalent
1 gallon	3.785 liters
1 gallon per minute	3.785 liters per minute
1 gallon per square foot of membrane area per day	40.74 liters per square meter per day
1 inch	2.54 centimeters
1 million gallons per day	3,785 cubic meters per day
1 pound per square inch	6.895 kilopascals
1 square foot	0.093 square meters
°F (temperature measurement)	$(^{\circ}\text{F}-32) \times 0.556 = ^{\circ}\text{C}$
1 °F (temperature change or difference)	0.556 °C

1. Executive Summary

Zero Discharge Desalination (ZDD) is a high recovery desalination process capable of extracting up to 98 percent of the water from brackish groundwater that has high levels of calcium sulfate, silica, and other problematic dissolved constituents. ZDD is comprised of a primary desalination process, typically reverse osmosis (RO) or nanofiltration (NF), and a secondary process called electro dialysis metatheses (EDM).

This report summarizes the results obtained from a multi-year research effort demonstrating ZDD. This effort was a collaborative project between the University of Texas (UTEP) at El Paso's Center for Inland Desalination Systems Texas (CIDS), and Veolia Water Technologies (Veolia). Livingston Associates provided operational support and technical assistance. ZDD was demonstrated at the Brackish Groundwater National Desalination Research Facility (BGNDRF) in the city of Alamogordo, New Mexico, and piloted in Colorado and California. High recovery (96-98 percent overall recovery) was demonstrated at each site and data for evaluating capital, and operational and maintenance (O&M) costs were gathered for use in cost models.

Veolia performed a cost analysis of ZDD and four other technologies in Alamogordo. ZDD with 98 percent recovery is estimated to have a treatment cost of \$3.30-3.40 per 1,000 gallons of product water. This ZDD cost is based on a 4 million gallon per day (MGD) desalination facility and includes amortized capital and annual operating expenses for the ZDD and an 18 acre evaporation pond for disposal of all waste (excludes the cost of final solids disposal). Traditional brackish water reverse osmosis (BWRO) (without softening) is estimated to cost \$1.25 to 3.50/1000 gallons of product water, but is limited to 80 percent overall recovery and depends strongly on the cost of concentrate disposal method and plant size. The cost of production for a BWRO plant with 80 percent recovery and thermal Zero Liquid Desalination (ZLD) for disposal of its concentrate is estimated to be at least \$4.70 per 1,000 gallons.

The cost of ZDD can be further reduced if magnesium hydroxide is recovered from the mixed chloride stream as a salable product. If credit is taken for the value of the magnesium hydroxide, $Mg(OH)_2$, the cost of a 4 MGD ZDD system drops by 16 - 24 percent, bringing the cost of ZDD down to \$2.55-2.80 per 1,000 gallons. This cost includes all salt recovery equipment (including clarifiers and evaporation ponds) and final waste disposal of the gypsum solids. Improved EDM system energy efficiency is expected to further reduce the cost of ZDD by an additional 10 to 20 percent.

Demonstration of Zero Discharge Desalination (ZDD)

Future work should focus on demonstrating a cost effective method for sodium chloride recovery such as evaporation on a larger scale, but it appears to be the most attractive method for achieving a truly zero liquid discharge desalination system for Alamogordo. ZDD has been proven to achieve 98 percent recovery and shows potential to achieve ZLD when solar salt recovery is implemented.

2. Introduction

Increasing population, decreasing water supply (volume and quality), and droughts are leading to increased competition between users of freshwater (municipal, irrigation, power, etc.), for both surface water (lakes, rivers) and groundwater. Much of the semiarid southwest has experienced drought in recent years. As annual precipitation decreases, there is less water available in streams, rivers, lakes, and other surface facilities. Many municipalities and industries are considering non-fresh water sources, as this may be the only way to grow.

Many fast growing locations have inadequate freshwater supplies. Desalination can augment the water supply for areas, but conventional approaches like RO are usually limited to 70 - 80 percent recovery. This means that large volumes of desalination waste (called brine) would have to be disposed of using evaporation ponds or deep well injection (DWI). Additionally, acceptable brine disposal options are often limited in these areas, presenting a major obstacle to implementing membrane-based desalination technology. For example, the population that depends on the water resources in the Tularosa Basin in southern New Mexico is subject to water scarcity. The groundwater of the Tularosa Basin is nearly saturated in calcium sulfate. Even when scale inhibitors are used, conventional BWRO can only recover up to 75-85 percent of the source water, so the volume of concentrate requiring disposal remains a considerable problem. An effective concentrate management solution is key to the design and operation of inland brackish desalination systems.

Local and regional water supplies are becoming more stressed due to overuse and drought. In a 2003 Government Accountability Office (GAO) report, 36 States (out of 47 States participating) reported that they anticipate water shortages under normal conditions in the next 10 years (GAO, 2003). In a drought, the number increases to 46. Michigan, California, and New Mexico did not participate in the 2003 study. As annual precipitation decreases, there is less water available in streams, rivers, lakes, and other surface facilities. Many municipalities and industries are considering non-fresh water sources, as this may be the only way to assure they maintain a sustainable water supply.

By 2025, nearly one-third of the population of the developing world will face severe water scarcity, according to the International Water Management Institute (Keller et al., 2000). Many locations that have inadequate freshwater supplies are fast-growing regions, including the American Southwest, Florida, and Asia, and many municipalities will eventually need desalination to supplement their limited supplies of fresh water (Davis and Rayman, 2007). Additionally, acceptable brine disposal options are often limited in these areas, presenting a major obstacle to implementing membrane-based desalination technology. The population that depends on the water resources in the Tularosa Basin in southern New Mexico is

Demonstration of Zero Discharge Desalination (ZDD)

subject to water scarcity, and desalination yields of the existing water supply have been limited (Brady et al. 2005). The groundwater of the Tularosa Basin is almost 100 percent saturated in calcium sulfate. Even when scale inhibitors are used, conventional BWRO can only recover approximately 75 percent of the source water, so the volume of concentrate requiring disposal remains a substantial problem. An effective concentrate management system is a key to the design and operation of inland brackish desalination systems.

ZDD a form of high recovery desalination, has been shown to provide substantially higher recovery than BWRO alone. ZDD has been demonstrated at pilot scales in New Mexico, Colorado, and Texas by UTEP researchers with funding from the Bureau of Reclamation's (Reclamation) Desalination and Water Purification Research (DWPR) Program, Sandia National Laboratories, the Texas Emerging Technology Fund, and other researchers have tested ZDD in California and Florida with other funding (Bond et al., 2011, 2015). ZDD achieves high recovery by converting the dissolved calcium sulfate to a solid byproduct without the need for energy-intensive evaporation that is characteristic of thermal ZLD processes.

UTEP and Veolia have demonstrated the ZDD technology at the BGNDRF with a blend of two wells that simulates Alamogordo's brackish raw water source, called the "Snake Tank" wells. ZDD is best-suited for desalinating brackish groundwater that has high levels of calcium sulfate like the BGNDRF and Snake Tank wells' groundwater. This is because calcium sulfate will limit desalination recovery to 70-75 percent. About 2 percent of the initial volume of brackish water remains in the ZDD concentrate streams.

2.1 Project Objectives and Goals

The primary goal of this project was to move towards commercialization and regulatory acceptance of the ZDD process. This project was done in support of Reclamation goals to (1) augment the supply of drinking water in the United States and (2) develop an approach to desalination that minimizes the environmental impact of desalination.

Another important aspect of this project was the focus on technology transfer to ensure the transfer of knowledge and work toward the commercialization of the ZDD technology by working with water utility(ies) to train them on the ZDD process and to document the operation of ZDD equipment with a user manual.

The demonstration results provided a baseline for design of a ZDD system for Alamogordo as well as other municipalities that have or will face similar challenges in achieving high desalination yields on their source waters.

Specific research goals addressed by this project are:

Demonstration of Zero Discharge Desalination (ZDD)

- Reduce the environmental impacts of desalination:
 - Evaluate potential for implementing ZDD technology in several water chemistries (laboratory studies)
 - Evaluate commercial potential for salable byproducts produced in ZDD process (calcium sulfate, sodium sulfate, etc.)
 - Evaluate potential for NF desalination, which could reduce overall cost by reducing the amount of energy required for both RO and EDM
 - Evaluate actual waste disposal costs and concerns for solid waste stream disposal to landfills
 - Establish cost/benefit analysis of recovering sodium chloride (NaCl) versus purchase
 - Evaluate the potential for incorporation of renewable energy into the ZDD process for direct power and/or solar drying of waste products

This project's objective was to demonstrate technology that has the potential to reduce the cost of desalination with the aim to develop cost-effective approaches for concentrate management that minimize potential environmental impacts.

2.2 Project Approach

This project was a demonstration of the commercial readiness of the ZDD technology. The following activities were conducted:

Phase 1:

- Preliminary study to evaluate NF membranes
- Design and procure equipment

Phase 2:

- Install and operate Phase 2 equipment at BGNDRF
- Install and operate Phase 1 equipment in Colorado
- Support pilot testing in California (with Phase 2 EDM equipment)
- Salt recovery laboratory studies
- Write final report

A 20 gallons per minute (gpm) ZDD system was installed at the BGNDRF in Alamogordo. The brackish feedwater was a mixture of two wells chosen to simulate the composition of groundwater from the Snake Tank wells that are available to Alamogordo. After being commissioned, the 20 gpm ZDD system was operated for five weeks at NF permeate flows ranging from 14 - 25 gpm. This system used 4-inch NF270 membranes and one full-size (commercially available)

Demonstration of Zero Discharge Desalination (ZDD)

EDM stack. The system was only shut down on weekends or when project personnel didn't have access to the site. System recovery was 98 percent, and good product water quality (total dissolved solids [TDS] below 800 mg/L) was achieved. The NF270 membranes performed better than expected in terms of permeate quality and allowed most of the silica to pass to the permeate. This means that a silica removal system will not be required for further evaluations in Alamogordo or in the full-scale design. Details from the pilot study are included in Chapter 4 of this report.

In Phase 2 the ZDD was demonstrated at the BGNDRF for extended periods of time. This equipment was designed to produce 40 gpm of product water at 98 percent recovery and used full size, commercially available NF membranes (8 inches) and two EDM stacks. The system was operated on a daily basis until nighttime temperatures were well above freezing and then continuously for multiple day runs during the week. The longest, mostly continuous, run was 145 hours (137 hours without interruption for clean in place (CIP) and maintenance shutdowns). Similar to Phase 1, 98 percent recovery with good product water quality was achieved. Problems with controls and stack design and construction were identified. Corrections were made to provide reliable conductivity measurements, eliminate internal leaks in the EDM stacks, and improvements were made for controls upgrades for future demonstration activities and full scale designs. Details are included in Chapter 4 of this report.

BGNDRF, UTEP, and Veolia participated with two other project teams to pilot ZDD technology in Colorado and California. HDR, Inc. led an effort evaluating ZLD techniques for Colorado (Brandhuber et al., 2014). The 20 gpm ZDD equipment was piloted in La Junta, Colorado and in Brighton, Colorado. Up to 98 percent recovery with good product water quality was demonstrated at both sites; however, equipment problems, operator errors, and other issues were encountered. Lessons learned from Colorado have led to better training, documentation, and operations of pilot equipment. Details can be found in the Water Environment Research Foundation (WERF) 5T10 report (written and co-published by WERF and International Water Association [IWA]) and in Chapters 5 and 6 of this report.

UTEP and Veolia provided equipment, labor, and supplies for this effort. The success of this pilot led to the design of the demonstration equipment for Phase 2 and future studies. During Phase 2, additional sites in Colorado and California were identified as side projects to expand the scope of the DWPR-funded project.

3. Review of Concentrate Management Technologies

Brackish groundwater often has enough calcium and sulfate ions to limit the amount of fresh water that can be recovered by desalination. The simplest way to prevent precipitation of calcium sulfate, CaSO₄, is to operate the desalination process under conditions that do not allow supersaturation at any point in the process. That could be done by limiting the yield of desalted water, but the high cost of obtaining raw water is an incentive to push for the highest practical yields. The degree of supersaturation is usually highest at the membrane surface in RO and at the heat-exchange surface in evaporative processes. Some increase in turbulence can be achieved by a modification in the design of the permeator or the evaporator, but substantial increases would likely require increased circulation, which means additional pumps and pumping energy, both of which are economically unattractive. In conventional RO without mineral recovery, the formation of CaSO₄ scale can be alleviated by adding scale inhibitors to the feed stream.

Several high recovery desalination techniques are available at various levels of development. Other researchers have summarized thorough evaluations of high recovery desalination and thermal ZLD approaches (Bond et al., 2015; Bond, et al., 2011; Brandhuber et al., 2014; Juby et al., 2008; and Mickley, 2008). A summary of several approaches is provided in the following sections. A comparison of these and ZDD is provided in Table 1. ZDD offers very high recovery, and with salt recovery implemented can achieve ZLD with substantially less energy than thermal ZLD approaches.

Table 1.—Comparison of ZLD, ZDD, and High Recovery Approaches

Technique	Expected Recovery	Capital Cost	Operating Cost	Revenue Potential
Deep Well Injection	80%	Low-Moderate	Low-Moderate	Low
Evaporation ponds	80%	Moderate	Moderate	Low
Volume Reduction	90-95%	Moderate-High	Moderate-High	Low
Volume Reduction + Salt Recovery	90-98%	Moderate-High	Moderate-High	High*
ZDD	95-100%	Moderate-High	Moderate-High	High*
ZLD (thermal)	Up to 100%	High	High	Low

*Market conditions dictate sale price and sale ability of recovered salts.

3.1. Thermal Zero Liquid Discharge

An approach used by industry for management of liquid wastes, namely ZLD, typically consists of thermal brine concentration and crystallization technologies. However, the substantial capital and operating costs that are inherent in ZLD processes often prevent them from becoming feasible solutions for RO desalination facilities. For example, capital costs of \$30-100 per gallon of daily capacity (GPD) for brine concentrators and crystallizers have been reported for RO brine with 6,000-18,000 milligrams per liter (mg/L) TDS (Juby, et al., 2008). Electrical unit consumption on the order of 75-95 kilowatt hours per thousand gallons (kWh/kgal) is typical for thermal ZLD systems such as brine concentrators and evaporators (Mickley, 2008). Thermal ZLD is not reasonable for dealing with the large volume of desalination concentrate from municipal desalination facilities unless some form of volume reduction is included. ZDD accomplishes similar volume-reduction goals of a thermal ZLD system, but differs in that it uses substantially less energy and is able to separate the salts in the water into potentially salable byproducts.

3.2 Evaporation Ponds

Another ZLD strategy uses solar evaporation ponds. However, evaporation ponds are also expensive, usually requiring double liners to prevent leakage. Additionally, evaporation ponds require large land areas, which are not available in many populated inland communities, and the evaporated water is a lost resource. ZDD accomplishes similar volume-reduction goals of a solar evaporation pond system, but differs in that it requires substantially less pond area and is able to separate the salts in the water into potentially salable byproducts.

3.3 Deep Well Injection

DWI, where feasible, can be a cost effective method for inland desalination communities. El Paso Water Utilities, El Paso, Texas, owns and operates a 27.5 MGD desalination plant that has up to 3 MGD of concentrate pumped to a disposal site about 20 miles northwest of El Paso. The target formation for the concentrate requires no additional pressure for disposal. Some sites may not have DWI as an option because of local regulations or lack of suitable geological formations for disposal. High salinity concentrate likely would not be disposed of using DWI because of scale formation in the pipelines or disposal wells.

3.4 High Recovery Desalination and Salt Recovery

Manufacturers and researchers have described and tested techniques using modified RO, ED, and combinations of processes to achieve high recovery

desalination from brackish and other source waters. Alternatives to the ZDD technology are summarized in the following two sections.

3.4.1 Volume Reduction Technologies

Volume reduction techniques generally use RO, NF, or electrodialysis (ED) as the primary desalination technique and add another RO or ED stage to further reduce the concentrate volume prior to final disposal. Two techniques, namely Closed Circuit Desalination (CCD™) and Concentrate Enhanced Recovery Reverse Osmosis (CERRO™), employ a semi-batch process that takes advantage of slow silica scale formation kinetics.

High Efficiency Electro-Pressure Membrane (HEEPM™) by EET Corporation uses a combination of RO and high efficiency ED to achieve high recovery from brackish feedwater (EET Corporation, 2011 and Mickley, 2008). Pre-treatment to remove sparingly soluble salts is included in this process, which likely helps the performance of the ED system. It is interesting to note that the recovery is not affected much by the type of water for the estimates made for RO concentrate at 8,000 mg/L TDS. In his evaluation of HEEPM™, (Mickley, 2008) includes two RO concentrates, one with high levels of calcium, magnesium, and sulfate, and one with mostly sodium, chloride, and bicarbonate. The HEEPM™ recovery is 95.1 percent for the high scaling feed and 95.2 percent for the lower scaling feed.

Vibratory shear enhanced process (VSEPT™) by New Logic uses vibration to mitigate problems from sparingly soluble salts. Flat RO membranes are used and are enclosed in a housing that is vibrated (Mickley, 2008). The combination of high shear at the membrane surface and vibrating action allows for recoveries as high as 80-85 percent without need for pre-treatment chemicals (Mickley, 2008).

High Efficiency Reverse Osmosis (HERO™) by Aquatech and GE is a commercially available process that has been successfully installed in at least 20 non-municipal sites around the world (Mickley, 2008). The process uses a combination of lime softening or ion exchange to remove calcium and carbonate, pH adjustment to mitigate silica scale formation, and staged RO to achieve high recovery. GE states that recovery above 90 percent is achievable (GE, 2013 a and b) and Aquatech suggests recovery “beyond 95 percent” is achievable (Aquatech, 2014). Mickley states that there is a tradeoff between lower energy cost and higher chemical cost with HERO™ (Mickley, 2008). The Aquatech and GE websites provide additional information on the HERO™ technology.

CCD™ by Desalitech uses a novel operation of RO to achieve high recovery up to 98 percent (Desalitech, 2013). The technology functions in a semi-batch mode. Feedwater enters the CCD, outfitted with the proper RO or NF membranes necessary for desalination, and permeate and concentrate are produced. The concentrate is recycled to the feed until a recovery (TDS) setpoint is achieved and the remaining fluid is drained from the process. Substantial energy savings are possible, up to 35-50 percent are reported by Desalitech (Desalitech, 2013) for

Demonstration of Zero Discharge Desalination (ZDD)

brackish water as compared to brackish RO with high recovery (typically two or three-stage RO).

AquaSel™ by GE combines electro dialysis reversal (EDR) with precipitation to treat RO concentrate and has demonstrated 99 percent recovery for a soda manufacturing process (GWI, 2012). GE is quoted as saying they are focusing on smaller (25-100 gpm) systems with moderate salinity (2,000 - 4,000 mg/L TDS) for its initial clients (GWI, 2012). The RO concentrate contained 838 mg/L TDS and was comprised mainly of bicarbonate, sulfate, and sodium chloride (GE Power & Water, 2013).

CERRO™ from UTEP (invented by Dr. Anthony Tarquin) operates similarly to the CCD™ technology, in that it is a single-pass, multi-membrane, batch-treatment system with periodic flushing. The two systems differ in the membrane array used and in how they are piped and controlled. CERRO has been tested at the pilot scale in El Paso and Alamogordo as part of research sponsored by Reclamation's DWPR, WateReuse, and the Texas Water Development Board. El Paso Water Utilities was recently awarded a Reclamation WaterSMART grant to evaluate 70 gpm CERRO systems for wellhead RO systems (Reclamation, 2014).

3.4.2 Volume Reduction with Salt Recovery Technologies

Several techniques have been tested, and sometimes patented, that include volume reduction with salt recovery. Selective salt recovery from desalination concentrate may provide a revenue stream that offsets the cost of desalination. Many of these are chemically intensive and somewhat complex.

SAL-PROC™ by Geo-Processors is a multi-component system that includes precipitation, desalination by RO, brine concentrators, crystallization, and evaporation ponds. The potential for \$5 million in annual revenue for recovered magnesium hydroxide, calcium carbonate, and other mixed salts from 2.5 MGD of concentrate was reported by Mickley (2008).

Selective Salt Recovery from Reverse Osmosis Brine Using Interstage Ion Exchange from the University of New Mexico (invented by Kerry Howe, Josh Goldman, and Bruce Thomson) uses a combination of RO, NF, ion exchange, and chemical precipitation to achieve both high efficiency desalination and recovery of salts from desalination concentrate. The desalination portion functions similar to ZDD except that it uses ion exchange resin columns instead of ion exchange membranes. The technique has been evaluated at laboratory and pilot scale in Colorado and is patented (8,557,119).

In addition to the processes listed, at least two engineering firms have evaluated and piloted high recovery techniques:

Demonstration of Zero Discharge Desalination (ZDD)

- Carollo Engineering performed laboratory studies and desktop estimates for several combinations including softening, RO, slurry precipitation and recycle reverse osmosis (SPARRO), forward osmosis, brine concentrators and SAL-PROC with different final disposal methods, including the Santa Ana Regional Interceptor (SARI) brine line, evaporation ponds, and landfill (Juby et al., 2008). The various techniques were estimated to be able to recover 93-99.75 percent from a brine with 8,000 mg/L TDS.
- Enviro Water Minerals Company (EWM) has piloted a technique that includes softening and alkalinity removal as pre-treatment prior to RO/NF and ED for desalination of El Paso Water Utilities' Kay Bailey Hutchison Desalination Plant (EWM, 2014). In addition, EWM recovers minerals at various points including magnesium chloride, calcium chloride, sodium sulfate, and sodium sulfate. EWM is awaiting approval from the Texas Commission on Environmental Quality to proceed with a design of a full-scale plant.

4. Zero Discharge Desalination

ZDD is comprised of a primary RO or NF desalination process and EDM. In contrast to conventional ED, the EDM has four membrane and solution compartments in a repeating unit. Two feed streams enter to the EDM, including the concentrate from the primary desalination (called EDM feed) and NaCl. The salts in the feed and the NaCl change partners and form two concentrated byproduct streams, one rich in sodium sulfate (the Mixed Na stream) and one rich in calcium chloride (the Mixed Cl stream). The salt-depleted EDM product (called EDM diluate) is then either returned to the feed or blended with the desalted product of the primary process. The EDM concentrate streams contain the remaining 2 percent of volume and all of the salts removed in the ZDD process. Combining these streams results in the precipitation of calcium sulfate and release of the NaCl, which can be recovered and returned to the EDM. Recovering NaCl will achieve even higher recovery, can reduce overall cost of the process, and reduce the environmental impact of the desalination process.

4.1 ZDD Technology Development and Funding History

ZDD was developed by Dr. Thomas A. Davis while at the University of South Carolina (USC). Two U.S. patents (7,459,088 and 7,083,730) protect the technology for both brackish water and seawater desalination. Patent 7,083,730 also describes techniques for salt recovery from several process streams. ZDD differs from other high recovery techniques in that it does not require pre-treatment to remove hardness and usually only employs moderate pH adjustment (to pH 6.8-7.0), antiscalants, and cartridge filters for pre-treatment.

Reclamation support of the ZDD technology began in 2002 with a proposal by Dr. Davis on behalf of USC in response to a solicitation No.02-FC-81-0757 by Reclamation. The research project entitled “Zero-Discharge Desalination of Seawater” was highly successful, and a small company, ZDD, Inc., was formed to commercialize the technology. Results of lab-scale experiments with seawater from the Atlantic Ocean demonstrated that the use of ED to treat the concentrate from RO produced an ED concentrate stream with about 20 percent NaCl. The ED had special membranes with selectivity for transport of monovalent cations and anions. The ED concentrate was processed in an evaporative crystallizer to produce NaCl of high purity. Bromide ions were selectively concentrated by the ED, and the mother liquor from the NaCl crystallization contained enough bromide to justify recovery in a large-scale desalination plant. The ED diluate was treated by NF to reject divalent ions and returned to the RO feed to increase yield of fresh water. The ED diluate contained the magnesium ions from the RO concentrate with a much higher concentration than in seawater. Treating that

Demonstration of Zero Discharge Desalination (ZDD)

stream with sodium hydroxide, NaOH, precipitated $Mg(OH)_2$ of high purity and potential commercial value.

Based on the success of the laboratory work, Reclamation funded construction of a pilot plant to demonstrate the seawater technology. Partway through the construction of the pilot plant, it became apparent that a suitable seawater intake was not available, so the pilot project was redirected. Reclamation identified a need to treat irrigation drainage contaminated with selenium and agreed to allow the pilot plant to be reconfigured to treat that water. During the 2-year interval between completion of the lab program and initiation of the pilot plant construction, Sandia National Laboratories (Sandia) funded laboratory research at USC on concentrate management for water containing high levels of $CaSO_4$, specifically groundwater in the vicinity of Alamogordo, and that research showed that $CaSO_4$ could be recovered as a solid byproduct. Removal of the $CaSO_4$ and recycle of the $CaSO_4$ -depleted solution to the RO feed allowed substantial improvement in the yield of RO permeate. Similarities between the $CaSO_4$ content of Alamogordo water and irrigation drainage in the Panoche Irrigation District of California (Panoche) made it apparent that the same $CaSO_4$ -removal should be applicable to both.

The ZDD process proved successful at Panoche. Treatment of the RO concentrate by EDM, produced two highly concentrated streams, one containing calcium chloride, $CaCl_2$, and the other containing sodium sulfate, Na_2SO_4 . Mixing those two concentrated streams precipitated solid $CaSO_4$. As anticipated, the selenate anions followed the sulfate through the anion-exchange membrane (AEM) of the EDM and appeared in the Na_2SO_4 -rich concentrate stream but did not precipitate with the $CaSO_4$. Thus the selenate was confined to a highly concentrated supernatant from the precipitation step.

Sandia continued to recognize the applicability of the ZDD process to the $CaSO_4$ -rich groundwater of the Tularosa Basin, the aquifer from which Alamogordo would have to draw water, and Sandia contracted with ZDD, Inc., to build a pilot plant to treat water from the Tularosa basin. The ZDD pilot plant, comprising RO, EDM, and a second ED stack for NaCl recovery, was housed in a 24-foot cargo trailer. Reclamation facilitated efforts to demonstrate the ZDD process, and the process was up and running during the opening ceremony of the BGNDRF in August, 2007. Sandia operated the pilot plant at the BGNDRF for about two years and subsequently donated it to UTEP. UTEP refitted the ZDD pilot plant to include data acquisition, feed decarbonation, and other useful features and named it MDU2. The testing showed that the ZDD technology is capable of increasing the desalination yield from a typical level of 75 percent in a conventional BWRO system up to 84 percent without a silica removal system.

Veolia obtained the license for the ZDD technology in February, 2009, and continued piloting efforts at the BGNDRF until 2010. The tests were conducted both with and without the use of a proprietary silica removal system. The testing showed that the ZDD technology is capable of increasing the desalination yield

from a typical level of 75 percent in a conventional BWRO system up to as high as 97 percent with the use of the silica removal system. High recovery processes are essential due to waste disposal costs, fixed water supply, and environmental concerns associated with brine. The ZDD was able to achieve this criterion while reducing the operational and capital cost compared to a conventional system.

Veolia piloted ZDD for the treatment of wastewater at Hilmar Cheese Company in Hilmar, California, in 2010 using both a small scale ZDD system (1-2 gpm product) and a full scale EDM system to treat various waste streams. The waste water contained high levels of calcium, bicarbonate, phosphate and silica (Biagini et al. 2011). High recovery was demonstrated (96-98 percent) and, if implemented, would provide a treated effluent that would be in compliance with surface discharge.

Black and Veatch has evaluated EDM, the key component of ZDD, in Florida and California. Both projects were aimed at comparing ZLD approaches available for utilities. A bench-scale EDM unit fitted with ASTOM Corporation's neosepta ion exchange membranes was leased to Black and Veatch in 2009 for testing in Florida. A full scale pilot demonstration EDM system fitted with MEGA Group RALEX® ion exchange membranes was provided for testing in California in the Fall 2013 through Spring 2014. UTEP and Veolia provided technical assistance and training for both projects. In Florida, four process trains were evaluated with EDM as the final concentrate volume reduction process. High recovery (87-97 percent) was demonstrated and the cost of treatment was deemed reasonable for brackish water with less than 5,000 mg/L TDS (Bond et al., 2011). In California, the EDM was used to treat the RO concentrate from the Beverly Hills Water Treatment Facility, Beverly Hills, California. High recovery was demonstrated (91-92 percent) and several full scale cost evaluations were developed to assess the cost of treatment using EDM (Bond, et al., 2015). If implemented, EDM could raise overall plant recovery from 80 percent to 97-99 percent.

4.2 Process Description

The ZDD process typically includes filtration, RO and/or NF, EDM, chemical feed systems, and a deionizer (DI) water system. A salt recovery system can be installed to recover NaCl, which is needed in the ZDD process, as well as other useful salts such as $Mg(OH)_2$, Na_2SO_4 , and $CaSO_4$. Each major process is discussed briefly in this section and detailed descriptions of the equipment tested at each site are included later in this report, as references, or in appendices.

4.3 Process Configurations

Three types of ZDD systems were evaluated during this project: greenfield, bolt-on, and EDM-only. The greenfield configuration was tested at BGNDRF and La Junta; the bolt-on configuration was tested in Brighton; and the EDM-only

Demonstration of Zero Discharge Desalination (ZDD)

configuration was briefly tested in La Junta and piloted at Beverly Hills, California. The choice of configuration depended on the water being treated by ZDD and the product water quality requirements.

The first ZDD system evaluated simulates a greenfield site (i.e., no existing desalination equipment) and involves a brackish water that is fed to a NF or RO to produce a permeate quality that meets primary drinking water standards. The concentrate from this NF is fed to an EDM system where offending dissolved constituents such as calcium sulfate are removed. This is shown in Figure 1. This configuration also shows how an existing plant could install different membranes in their existing system and install a new EDM to achieve high recovery. RO membranes have high silica rejection, so a silica purge, silica treatment system, or use of silica-permeable membranes are needed where moderate to high levels of silica are present in the groundwater.

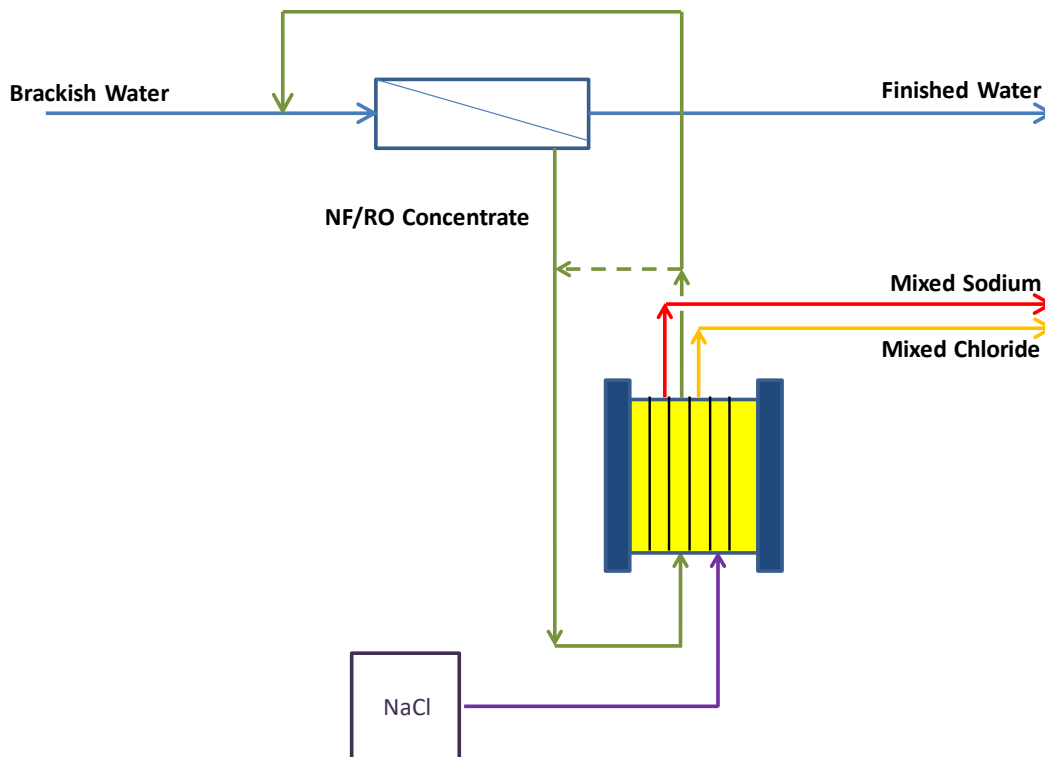


Figure 1.—Greenfield ZDD flow diagram (simplified).

The second type of ZDD system tested simulates a bolt-on approach and involves ZDD desalinating the concentrate from an existing desalination plant. This is shown in Figure 2. EDM is placed at the front of the treatment scheme, because ED operates more efficiently at higher concentration. Additionally, the EDM removes CaSO_4 and other dissolved constituents to a level that allows the NF system to operate efficiently and with a lesser chance for scale formation. A silica purge stream is necessary at sites with high silica in their RO concentrate. If

Demonstration of Zero Discharge Desalination (ZDD)

NF270 membranes are used to desalinate EDM diluate, the rate of silica purge is expected to be 10 percent of the waste, or less than 0.2 percent of the initial feed to the existing plant RO. If RO is used, the waste volume could be 5-10 percent of the initial feed.

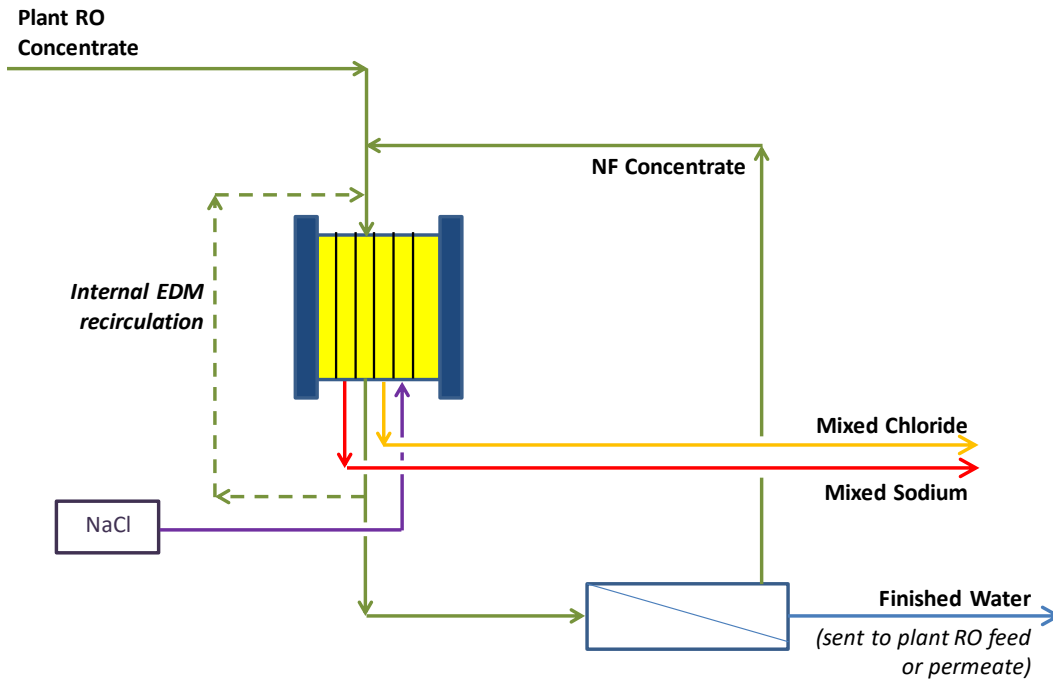


Figure 2.—Bolt-on ZDD flow diagram (simplified).

Some brackish water and plant RO concentrate has a low enough concentration that RO/NF systems are not required in the ZDD process. In this case, the EDM desalinates the feed water or RO/NF concentrate directly and the diluate can either be mixed with the existing plant's feed or permeate. This process is shown in Figure 3.

Demonstration of Zero Discharge Desalination (ZDD)

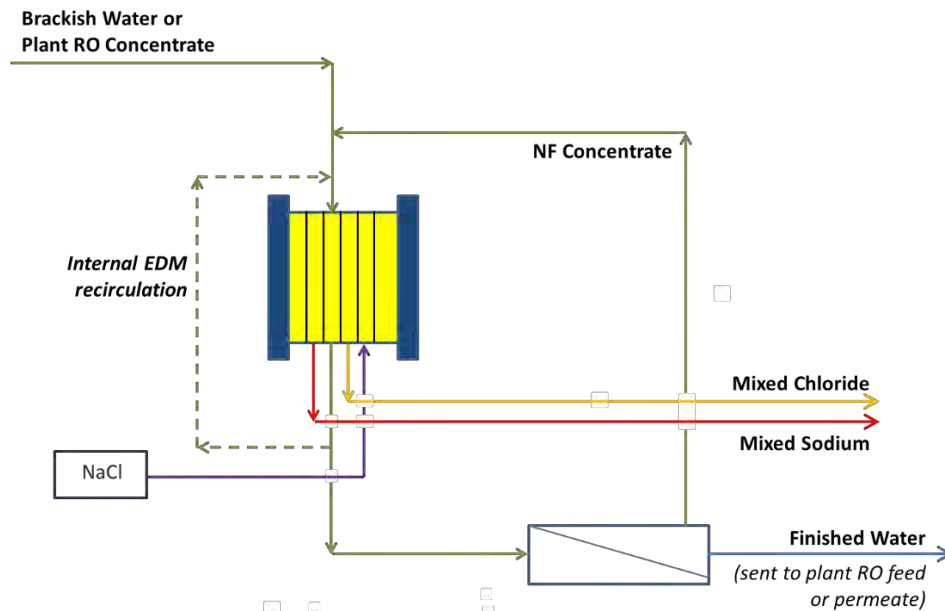


Figure 3.—EDM only ZDD flow diagram (simplified).

4.4 Pre-Treatment

Pressure driven membrane processes such as RO and NF require pretreatment for particles. Cartridge filters were installed on the raw (untreated) feed water and the NF feed water systems to protect the NF membranes; 20-micron and 10-micron cartridge filters were installed on the raw feed and NF feed, respectively.

Bicarbonate alkalinity was reduced by the addition of sulfuric acid to the raw feed at all sites. During the Phase 1 activities at the BGNDRF, a degasifying tower was also used. The addition of sulfuric acid to the raw feed water forms carbonic acid. When the acidified water passes through a degasifying tower, carbon dioxide, CO_2 , is released to the environment and alkalinity is partially reduced in the NF feed. During Phase 2 and in both sites in Colorado, only acid addition was employed to reduce the amount of bicarbonate alkalinity in the NF feed. Hydrex 4101 antiscalant was added to the NF feed at a dosage of 2-3 mg/L at all sites.

4.5 Reverse Osmosis and Nanofiltration

The concept of RO was first studied by Fick in 1855 (Glater, 1998) but pressure-driven RO membranes were not developed and tested until the mid-1950s by researchers at the University of California, Los Angeles, California; University of Florida, Gainesville, Florida; and the U.S. Office of Saline Water, Washington, D.C. (Glater, 1998). RO and NF membrane manufacturers now produce many types of membranes, and different designs are optimized for productivity (high flux), lower energy, high salt rejection, fouling resistance, or other useful properties.

Demonstration of Zero Discharge Desalination (ZDD)

RO and NF produce two streams of water, namely permeate and concentrate. The concentrate contains all of the salts removed by the membrane. Typical brackish RO/NF systems can operate at 75-85 percent recovery and are limited by the sparingly soluble salts such as CaSO₄, calcium carbonate (CaCO₃), and silica. The amount of salts in the concentrate is related to the recovery of the RO/NF—higher recovery systems will have higher concentrations. Disposal or treatment of the concentrate can be a limitation to the installation of RO or NF.

Both RO and NF use pressure to produce permeate from a brackish water source using semi-permeable membranes installed in pressure vessels. RO is able to produce high quality product water. However, it requires higher pressure than NF and can be susceptible to fouling and scaling. Rejection of multi-valent and large molecules in brackish feedwater is similar for RO and NF membranes, but NF membranes have lower rejection of monovalent and uncharged molecules. NF can produce good quality permeate that meets primary drinking water standards and can require substantially lower feed pressure than RO. Table 2 compares the predicted product water and energy consumption for several RO and NF membranes produced by Dow Filmtec. Each of the designs uses the average water quality from Phase 2 operations at the BGNDRF and assumes 1 MGD of permeate. Three brackish water membranes and two nanofiltration membranes were compared with similar staging. Of these, the BW30HR-440i membrane is predicted to produce the best quality permeate and require the most energy for the RO. The NF90-400/34i is predicted to have similar water quality as the XLE-440i membranes, but requires 9 percent less energy. The NF270-400 membranes require the least amount of energy, but also have the highest salinity permeate (note that Phase 2 demonstrated better performance with the same membranes).

Table 2.—Comparison of Dow Filmtec Membranes to Produce 1 MGD Permeate from Average Snake Tank Brackish Water

Membrane	BW30HR-440i	ECO-440i	XLE-440i	NF90-400/34i	NF270-400
Staging	12 x 6 (6M)	12 x 6 (6M)	14 x 7 (6M)	14 x 7 (6M)	14 x 7 (6M)
Average Permeate Flux (gfd)	18.16	16.52	14.16	15.57	15.57
Feed Pressure (psi)	218	137	118	107	101
Permeate Back Pressure (psi)	0	0	0	15	30
Recovery	76%	76%	76%	76%	75%
RO Energy (kWh/kgal)	2.59	1.64	1.41	1.28	1.22
Permeate TDS (mg/L)	27	30	125	162	841

The brackish water in Alamogordo is nearly saturated in CaSO₄ and has moderate silica. Because ZDD operates at 98 percent recovery, silica must be removed, or otherwise dealt with, to protect the RO/NF system in ZDD. Dow Filmtec’s NF270

Demonstration of Zero Discharge Desalination (ZDD)

membranes were used at the BGNDRF site, because they allow for nearly all of the silica in the brackish feedwater to pass to the product water. This eliminates the silica scale problem and produces water suitable for Alamogordo's customers (less than 800 mg/L TDS). A combination of Dow Filmtec's NF270 and NF90 membranes were employed at both La Junta and Brighton, because their water quality requirements were more restrictive.

4.6 Electrodialysis Metathesis

ED has been around nearly as long as RO, with its discovery in 1890, and first installation for the desalination of seawater in the 1960s (Xu, 2005). While ED is a proven method for desalination, it is not as widespread as RO. A recent study estimated that ED accounted for 3.53 percent of the world's desalination capacity while RO accounted for 59.85 percent in 2012 (Ghaffour, Missimer, & Amy, 2013).

ED "stacks" are comprised of alternating cation membranes and AEMs as shown in Figure 4. Commercial stacks include hundreds of cell pairs. An aqueous solution, such as brackish water, is fed into alternating feed compartments. As an electrical potential is applied, anions move towards the positively charged anode and cations move towards the negatively charged cathode. Cations, shown moving to the left in Figure 4, will pass through the cation-exchange membrane (CEM) but will be blocked by the AEM. Similarly, anions will pass through the AEM but will be blocked by the cation exchange membrane. This leads to changes in the salt content of alternating streams: one is a diluted stream (called diluate) and the other is a concentrated stream (called concentrate).

Oxidation and reduction reactions occur and gasses are generated at each of the electrodes. These reactions are summarized in Figure 5. Dashed lines in 5a) represent the concentration in the diluting compartment (center of the figure) and concentrate compartments (left on AEM, and right of cation-exchange membrane, CEM). Although not shown in Figure 4, an electrode rinse (E-Rinse) stream flows across each electrode to provide a path for current to flow from the electrodes into the rest of the stack, as well as to cool both electrodes and sweep gases out of the stack. Gases generated in the E-Rinse compartments are vented to the atmosphere.

Demonstration of Zero Discharge Desalination (ZDD)

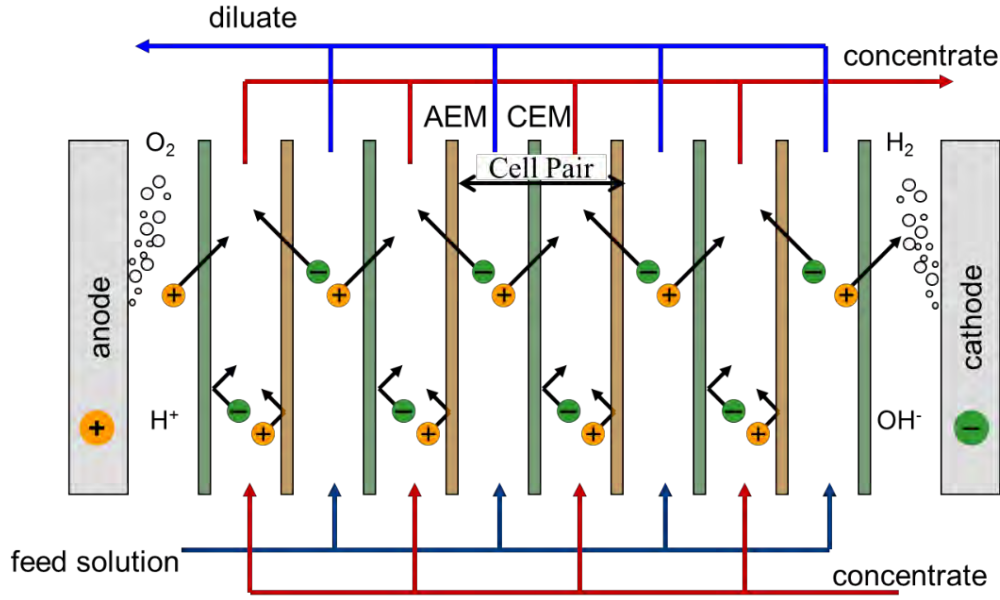


Figure 4.—Flow of ions in electrodialysis.
 (CEM = cation-exchange membrane, AEM= anion-exchange membrane)
 (modified from Strathmann, 2004 and Murray, 1995)

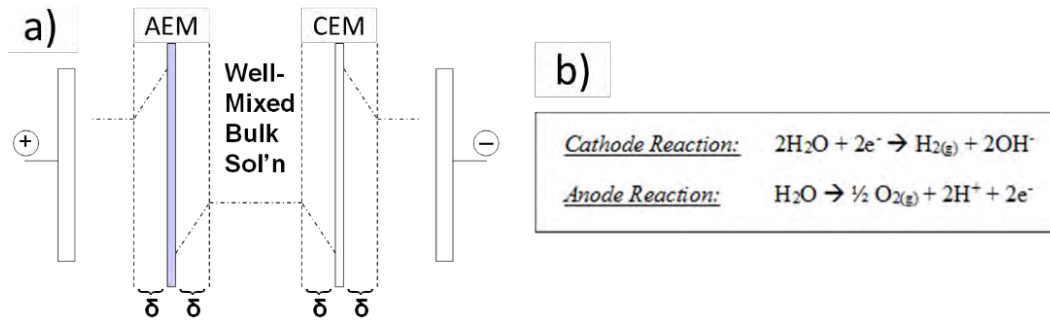


Figure 5.—Concentration polarization in electrodialysis showing a) concentration profile and b) electrode reactions.

Similar to RO and NF, concentration polarization occurs at the boundary layers in the solutions next to the membranes. In ED, concentration boundary layers are on both sides of every ion-exchange membrane. These concentration boundary layers exist because the electric current is carried through the membrane by only one ion type (anion or cation), but in solution the current is carried by both ion types. Consider the AEM in Figure 5. Only a portion of the current is carried by anions migrating through the boundary layer toward the anode, but essentially all of the current is carried by anions migrating through the AEM. Cations migrating through the boundary layer toward the cathode cause depletion of the solution next to the membrane. That depletion is made up by ion pair diffusing toward the membrane, and the driving force for that diffusion is the difference between the salt concentrations in the bulk solution and at the interface. As the current density increases, the rate of removal of anions through the membrane increases, and the

Demonstration of Zero Discharge Desalination (ZDD)

concentration gradient increases. The salt concentration at the interface approaches zero at some current density, because the rate of diffusion cannot keep up with the rate of removal. That condition corresponds to the limiting current density. For potassium chloride, KCl, the mobility of Cl^- ion is the same as the mobility of K^+ ions, so zero concentration is attained at both membranes simultaneously. For NaCl, the Na^+ ions are less mobile than the Cl^- ion, so the zero concentration occurs at the cation-membrane at lower current density than at the AEM. When there are not enough ions available in the solution at the interface, the difference is made up by dissociation of water into H^+ and OH^- ions.

Figure 6 shows the relationship between current density and applied voltage in ED. In the Region I, called the ohmic region, an increase in voltage leads to a proportional increase in current density. In Region II, called the plateau region, there is a much lower increase in current density with increasing voltage. In Region III, called the overlimiting region, current density does increase with increasing voltage; however, at this point water splitting is occurring (other unknown phenomena also occur in this region (Strathmann, 2004). Applying too much voltage can damage the ED equipment and cause operational problems. The limiting current density can be determined by the intersection of the lines for the ohmic and plateau regions (labeled i_{lim} in Figure 6).

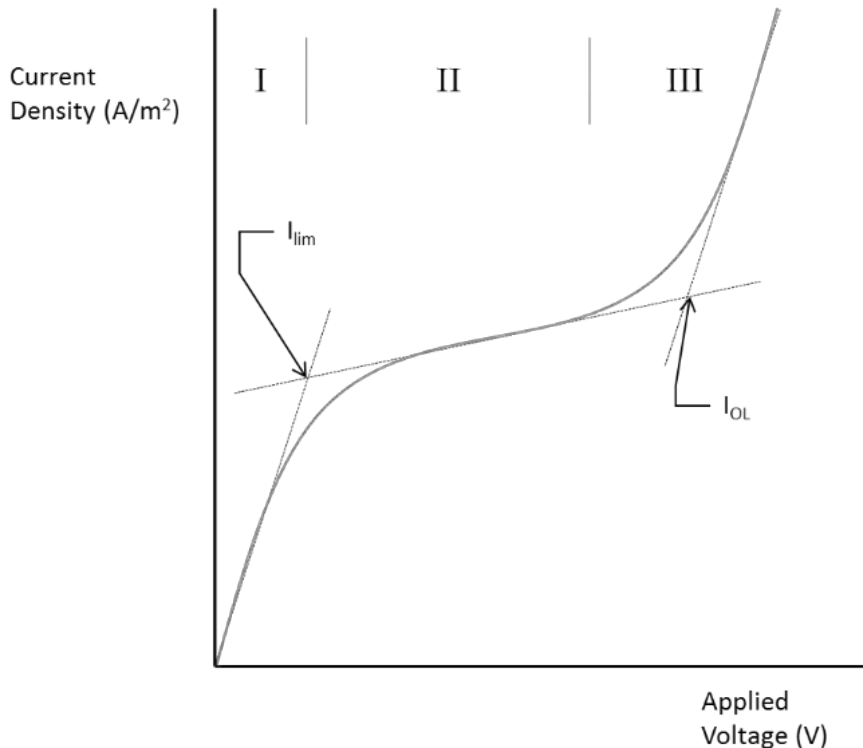


Figure 6.—Limiting current density in electrodialysis.

Demonstration of Zero Discharge Desalination (ZDD)

This method is described by Krol et al (1999). Other methods calculate the point where all ions have been depleted from the boundary layer (labeled i_{OL} in Figure 6) (Bond et al., 2011 and Bard and Faulkner, 2001). At i_{OL} , complete concentration polarization has occurred (Bard and Faulkner, 2001). While an ED system can be operated beyond the i_{lim} , ED is less efficient (with respect to desalination) beyond the i_{lim} , because additional voltage will not lead to a useful increase in ion removal. Additionally, the closer an ED system is operated to the overlimiting region (III in Figure 6), the more likely water splitting and equipment damage will occur.

Current density in general, and limiting current density in particular, is affected by operating conditions, stream concentrations, and hydraulic properties of the stack (Bond et al., 2015). Increased concentration supports increased current density (Krol et al., 1999) and increased stream velocity also leads to increased current density (Kanavova et al., 2013). As temperatures decrease, the effective cell resistance increases, and the current density also declines (Leitz, 1974). Figure 7 shows how decreasing temperature leads to higher cell pair resistance in ED.

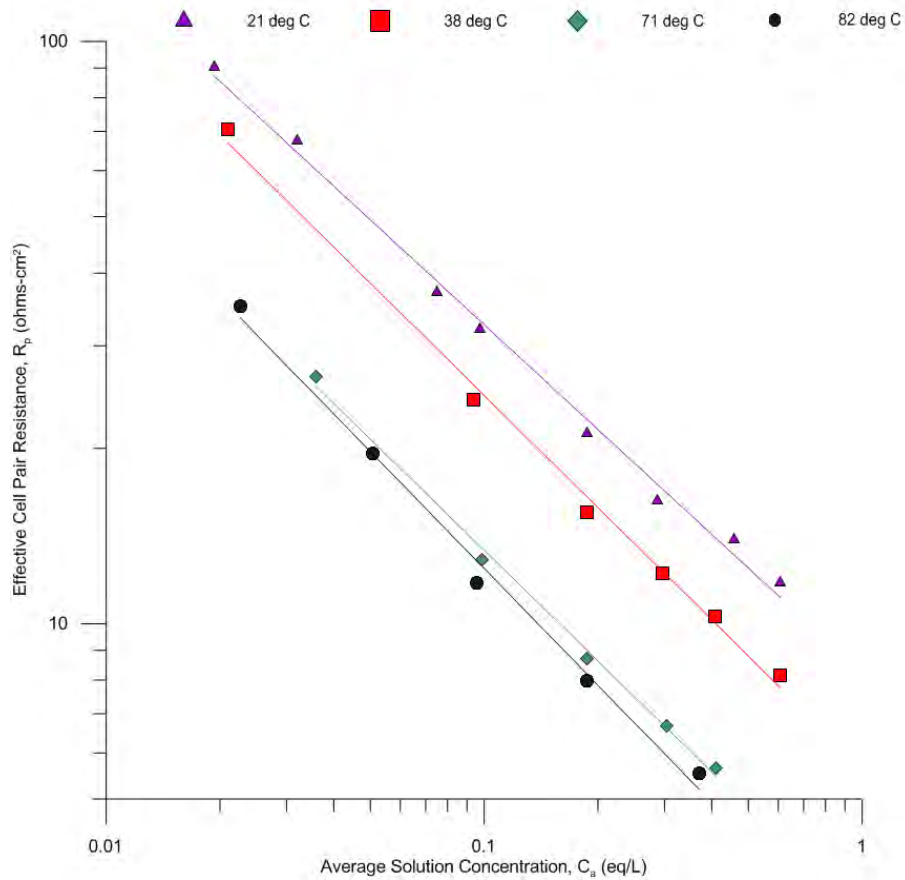


Figure 7.—Decreased temperature leads to increased ED resistance.
(Modified from Leitz, 1974)

Demonstration of Zero Discharge Desalination (ZDD)

Brackish water contains many dissolved salts that, when concentrated, would precipitate in ED concentrate compartments. Electrodialysis reversal (EDR) mitigates this problem by regularly changing the polarity of the applied voltage, which dissolves nascent scale that had formed in compartments that were previously concentrating (Katz, 1979). Figure 8 shows the relative solubility of several common dissolved salts in brackish water and desalination concentrate. In general, chloride salts and sodium salts are very soluble. However, sodium sulfate and sodium carbonate have steep declines in solubility with decreasing temperature and sodium bicarbonate has a relatively low solubility.

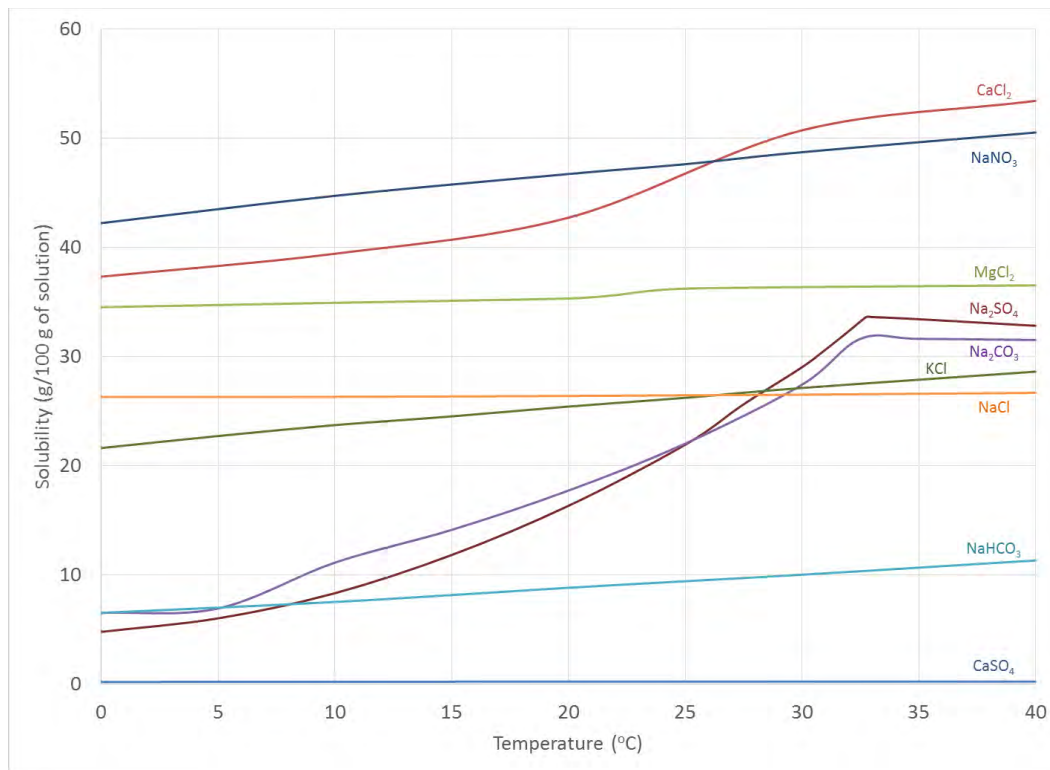


Figure 8.—Solubility of dissolved salts (Siedell, 1907).

Similar to ED, an electrodialysis metathesis (EDM) stack is comprised of alternating cation membranes and AEMs. However, instead of a single diluting and single concentrating stream, EDM has two diluting streams and two concentrating streams. The combination of the four streams and associated membranes is called a quad. Commercial EDM stacks include 100 or more quads.

Figure 9 shows a single quad in detail and how that quad is incorporated in the rest of the EDM stack. Brackish water or desalination concentrate is fed to the EDM feed compartments. The applied voltage causes the cations to move to the right (towards the cathode) and anions to the left (towards the anode). Cations from the EDM feed combine with chloride from the NaCl to produce a mixed chloride salt concentrate stream (called Mixed Cl). Anions from the EDM feed

Demonstration of Zero Discharge Desalination (ZDD)

combine with sodium from the NaCl to produce a mixed sodium salt concentrate stream (called Mixed Na). This metathesis, or changing of partners, produces highly soluble sodium and chloride salts such as calcium chloride and sodium sulfate from sparingly soluble salts. NaCl is added to the EDM at a rate equivalent to the amount of ions removed from the EDM feed. EDM acts like a kidney in the ZDD process by removing troublesome salts from the RO/NF concentrate. The desalinated EDM product (called diluate) can be returned to the RO/NF feed or blended directly with the RO/NF permeate, thus allowing for higher overall water recovery.

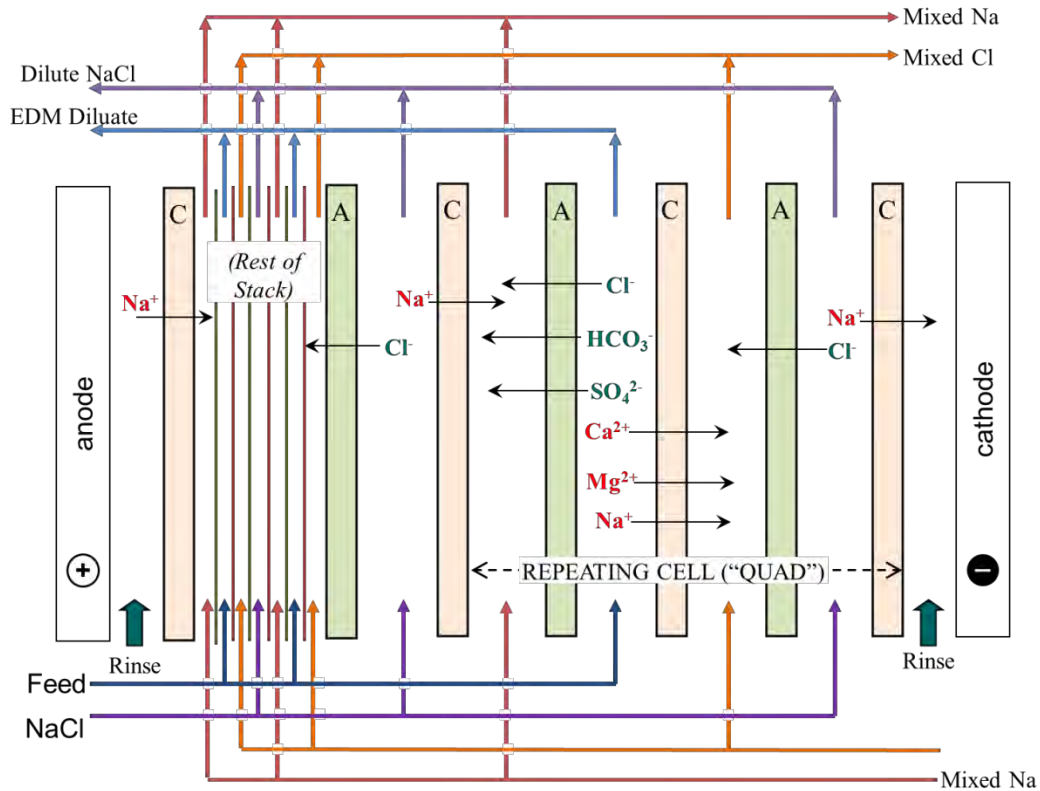


Figure 9.—Flow of ions in electrodiolysis metathesis.
(C=cation-exchange membrane, A=anion-exchange membrane)

The first description of EDM appeared in U.S. Patent 2,721,171, which was issued in 1955 and assigned to DuPont. This patent described production of H_2SO_4 and NaOH from Na_2SO_4 (Monet and Wilfred, 1957). Winger described the use of EDM to generate NaOH from lime and NaCl (Winger, 1957). J. R. Ochoa et al. used an electrolytic cell with a single set of membranes in an EDM arrangement to convert a salt to an acid (Ochoa et al., 1993). Alheritiere et al. (1998) described the use of EDM to convert magnesium chloride and sodium sulfate to magnesium sulfate and sodium chloride. A more recent patent, U.S. Patent 6,712,946, describes the use of EDM to produce 2-keto-L-gulonic acid from its calcium salt (Genders and Hartsough, 2004). In this, HCl is fed to the other depleting compartments, and CaCl_2 is a byproduct. There is no indication that any of the aforementioned examples of EDM was ever practiced on a scale

Demonstration of Zero Discharge Desalination (ZDD)

larger than laboratory scale. Indeed, it is apparent that the ZDD process is the first commercial process to use EDM.

Before the ZDD process created a demand for large EDM stacks, the largest EDM stacks were laboratory scale with about 200 square centimeters (cm^2) of exposed membrane in each solution compartment. For the first pilot plant funded by Reclamation in 2004, Tokuyama Corp designed and built an EDM stack comprising 20 repeating quads with 500 cm^2 of exposed membrane in each solution compartment. For the 20-40 gpm demonstrations funded by Reclamation in 2011-2013, MEGA designed and built an EDM stack with 100 repeating quads with $4,200 \text{ cm}^2$ of exposed membrane in each solution compartment. Further research will use expanded stacks from MEGA with 120 quads.

4.7 DI Water System

Deionized (DI) water (or RO permeate) is used in the ZDD process to prepare NaCl and E-Rinse (Na_2SO_4) solutions as well as to dilute the Mixed Na and Mixed Cl streams to prevent precipitation and minimize back diffusion of ions. It is also used to flush the EDM system during shutdown operations. Virtually all of the DI water used in the ZDD process is captured in the waste streams from the EDM's Mixed Na and Mixed Cl streams. If excessive DI water is used to dilute the concentrate streams, there will be a larger volume of waste measured in the Mixed Na and Mixed Cl outflows. DI water used to prepare the NaCl solution is not considered a loss, because water is transported with the sodium and chloride ions (called water of hydration) that migrate into the Mixed Na and Mixed Cl streams. Finally, there are minimal water losses in the E-Rinse stream, so the DI water used to prepare the initial solution is not considered a significant loss.

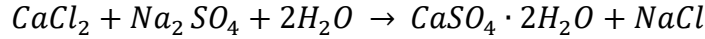
4.8 Salt Recovery

Cost analysis of the ZDD process indicates that the purchase of NaCl represents a substantial portion of the operating cost of the process (87 percent of the chemical cost and 33 percent of the total cost of production). The EDM waste streams contain all of the salts removed by the NF/RO and EDM systems and offer the potential for salt recovery. When these two concentrate streams are combined, calcium sulfate precipitates and can be recovered as a solid product. The remaining liquid contains NaCl (and other chloride salts). By employing NaCl recovery, ZDD can achieve water recoveries as high as 99 percent and reduce NaCl purchases by 50 percent to 90 percent, depending on the method of salt recovery.

The NaCl that is introduced as the NaCl feed appears as Na^+ ions in the Mixed Na stream and as Cl^- ions in the Mixed Cl stream. Moreover, when these two streams are combined to precipitate CaSO_4 , the NaCl is released and appears in the supernatant, the liquid that remains after the solid CaSO_4 settles. (It should be noted that calcium sulfate actually precipitates as the dihydrate $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$,

Demonstration of Zero Discharge Desalination (ZDD)

commonly called gypsum, but the shortened version, CaSO₄, will be used in most of the text for brevity.) The following reaction shows how the primary constituents of the Mixed Cl (CaCl₂) and Mixed Na (Na₂SO₄) streams will combine to form CaSO₄ and release NaCl:



When the entire volumes of the EDM Mixed Na and Mixed Cl streams are combined to precipitate CaSO₄, essentially all of the NaCl fed to the EDM stack appears in the supernatant. The challenge is to recover that NaCl from the supernatant at a concentration and level of purity that would allow its reuse in the EDM stack. The use of NaCl recovered from the supernatant is dictated by the composition of the source water. For the discussion that follows here, the exemplary source water is the average of the Snake Tank wells that are to supply the desalination plant planned by the Alamogordo.

Table 3 shows the major ions in the source water based on the average of analyses of the three wells supplied by Alamogordo. Data in Table 3 indicate that there is more sulfate than calcium in the well water. However, if the magnesium is replaced with calcium, which can be achieved by adding lime to precipitate Mg(OH)₂, then the proportions of calcium and sulfate would be nearly the same, and almost all of the volume of both concentrate streams could be fed to the precipitation process, and all of the supernatant is available for recovery of NaCl.

Therefore, except for small amounts of solutions that would be discarded to achieve process control, all of the NaCl that enters the EDM is potentially recoverable.

NaCl can be recovered using ED with monovalent-selective membranes or other approaches. Research was performed using three techniques during this project, all of which produced NaCl of suitable quality for use in EDM (see Section 3.1 and Appendix E for a cost comparison and Chapter 9 for a detailed description of each process and research performed during this project).

Table 3.—Snake Tank Water Quality (average of 3 wells)

Ion	Mg/L	Meq/L
K	4.7	0.1
Na	164.3	7.1
Mg	87.7	7.2
Ca	413.3	20.6
HCO ₃	131.8	2.2
Cl	132.7	3.7
SO ₄	1,433	29.8
TDS	2,500	

5. Demonstration Testing Locations

ZDD was tested in several different configurations at four different sites during this project. The primary focus of this project is Alamogordo, New Mexico, but, as mentioned previously, the project team participated in studies led by HDR, Inc. in Colorado and Black and Veatch in California.

5.1 Alamogordo, New Mexico

The city of Alamogordo is in the initial steps of permitting and building a desalination facility. They have plans to eventually install a 3-6 MGD desalination plant and are designing a temporary 1 MGD peaking plant to replace the loss of an important surface water supply, Bonito Lake (approximately ten miles north of the village of Ruidoso, New Mexico, off of State Highway 37), while it is being rehabilitated following a forest fire in 2012. Water scarcity is a reality in the city, and the groundwater there has challenging characteristics. The City has rights to 4,000 acre-feet per year (AFY) (3.6 MGD) of brackish water annually. In a drought year, the city is allowed to withdraw up to 5,000 AFY, but the total amount of brackish water extracted cannot exceed 20,000 AFY over a five-year period.

Some of the major challenges that the city of Alamogordo is facing are:

- The city of Alamogordo's 40-year Water Development Plan states that the need for new water supplies is urgent (Livingston Associates, P.C. and John Shomaker and Associates, Inc., 2006).
- Historically, more than 70 percent of the city of Alamogordo's water supply is derived from surface water that is affected by drought, which creates a variable supply that is often not favorable (Livingston Associates, P.C. and John Shomaker and Associates, Inc., 2006).
- The brackish source water in Alamogordo, primarily originating from the Tularosa Underground Water Basin, is nearly saturated in calcium sulfate, and also contains high levels of TDS and silica, which significantly limits the potential desalination yields (~75 percent). The resulting issues are twofold: nearly 30 percent of an already scarce water supply could be wasted if the water were desalted by conventional technology, and without an alternative technology that can provide higher yields, the city of Alamogordo will need to identify and implement an acceptable brine disposal solution.

The BGNDRF is an ideal location to demonstrate the ZDD technology. Preliminary work (supported by Reclamation and Sandia) was successfully performed at BGNDRF, so this location provided a very logical base case in

Demonstration of Zero Discharge Desalination (ZDD)

which to pilot ZDD. The facility has four wells, which can be blended to produce brackish water similar to the water available at Alamogordo's Snake Tank wells, which are the source for the city's future desalination plant.

5.2 La Junta and Brighton, Colorado

According to the American Membrane Technology Association (AMTA) database, the State of Colorado currently has 87 desalination facilities (AMTA, 2014). Nine of these facilities are for municipalities and there were 10 proposed municipal facilities as of 2007 (Brandhuber et al., 2007). Most desalination facilities discharge their concentrate to sewers or surface water discharge in Colorado, and there is only one permitted deep-well injection site (Archuleta et al., 2014). Many water utilities are looking towards brackish water for future demand and desire long-term solutions for concentrate management. HDR, Inc. led an effort to evaluate concentrate management strategies, including ZLD, for the State of Colorado (Brandhuber et al., 2014).

One portion of this project included pilot testing ZDD at the two Colorado sites, La Junta and Brighton, with existing RO facilities. Piloting at these sites offered the chance to train new operators and evaluate ZDD for the desalination of different feed water quality and product water quality requirements.

5.3 Beverly Hills, California

AMTA's database shows nearly 300 desalination facilities in California (AMTA, 2014) with 45 being for municipalities (Archuleta et al., 2014). California desalination plants use a variety of methods for disposal of its concentrate including surface and sewer disposal, DWI, land application, and recycling. Some facilities have access to a "brine line" that allows for disposal into the Pacific Ocean after treatment (Juby et al., 2008). Similar to New Mexico, Colorado, and other States, the State of California is interested in more sustainable concentrate management approaches. With funding from the California Energy Commission, Black and Veatch led an effort to evaluate EDM, the main part of ZDD. The PI, Rick Bond, has also evaluated EDM for treatment of brackish water in Florida (Bond et al., 2015).

Piloting at the Beverly Hills RO Plant allowed for training of other Black and Veatch personnel and for further evaluation of other water quality and treatment schemes.

5.4 Water Quality Analysis for Testing Sites

Selected feedwater quality data is presented in Table 4. ZDD is most efficient in areas with relatively high proportions of divalent ions such as calcium, magnesium, and sulfate. The water at the BGNDRF had the highest proportion of all of these ions, with around 60 percent of the cations present as calcium and

Demonstration of Zero Discharge Desalination (ZDD)

magnesium and around 70 percent of the anions present as sulfate (concentrations in meq/L). Of the sites tested in Colorado, La Junta is a better fit than Brighton because of the relatively high proportion of calcium and sulfate.

Table 4.—Selected Feedwater Quality Data from ZDD Testing Sites

Site Source	Beverly Hills, CA Conc	Brighton, CO Conc	La Junta, CO BW	BGNDRF- Yr1 BW	BGNDRF- Yr2 BW
Ca	157	553	173	250	288
Mg	119	116	63	94	121
Na	537	555	115	370	379
K	16	15	3.8	4.5	2.6
SO ₄	470	870	581	1,300	1,281
Cl	511	638	44	290	342
HCO ₃	791	1,138	263	205	248
NO ₃	0.4	210	4.3	1.6	N/A
SiO ₂	118	99	16	24	19.8
pH	7.6	7.6	7.8	7.4	7.6
TDS	3,190	4,065	1,115	2,300	2,910
Conductivity (μS/cm)	4,126	5,666	1,639	2,857	3,739
% of Ca, Mg to cations*	43%	60%	67%	56%	61%
% SO ₄ to anions*	25%	31%	90%	70%	67%

All in mg/L unless otherwise noted.

*Calculation is made using concentrations in meq/L.

Beverly Hills data from (Bond, et al., 2015 (in progress))

BW=brackish water, Conc=RO concentrate, N/A=not analyzed

6. Alamogordo Results and Discussion

Research in Alamogordo was performed in two phases with two sets of equipment. This section describes the feed water quality, the equipment used, and the results of the testing.

6.1 Feed Water Quality

The BGNDRF has four brackish water sources available for desalination technology evaluations. This project used a blend of Wells 1 and 4 (about 40 percent of Well 1 and 60 percent of Well 4) in order to simulate the chemistry of water from the Snake Tank wells available to the city of Alamogordo (Livingston Associates; Reclamation, 2013). Table 5 compares the well chemistry at both locations. Most of the contribution to the salinity of the Snake Tank water is from calcium, magnesium and sulfate. The targeted mix aimed to match the average sulfate concentration of the Snake Tank wells and was monitored using conductivity. Compared to the Snake Tank wells, the brackish wells at the BGNDRF have more bicarbonate, sodium, and chloride, relative to the overall TDS, and this difference resulted in higher TDS when the waters from Wells 1 and 4 were blended to achieve the sulfate target. Acid and a de-gas column was used to reduce alkalinity to a level close to the Snake Tank water during Phase 1 and acid alone for Phase 2.

Table 5.—Comparison of Alamogordo Snake Tank and BGNDRF Brackish Water Chemistry

Well/Source Name	Snake Tank Wells				BGNDRF
	4	5	8	Average	Well 1+4 Targeted Mix
K	5.9	4.3	3.9	4.7	3.5
Na	320	85	88	164	382
Mg	66	98	99	88	143
Ca	390	410	440	413	334
Fe	0.63	0.14	0.27	0.35	0.24
Mn	0.44	0.021	0.029	0.16	0.012
Ba	0.02	ND	ND	0.02	0.02
Sr	N/A	7.15	7.4	7.3	5.3
HCO ₃	78.08	146.4	170.8	132	228
F	0.59	0.69	0.59	0.62	0.99
Cl	68	170	160	133	410
SO ₄	1600	1400	1300	1433	1455
SiO ₂	23	31	30	28	21

Demonstration of Zero Discharge Desalination (ZDD)

Well/Source Name	Snake Tank Wells				BGNDRF
	4	5	8	Average	Well 1+4 Targeted Mix
TDS	2700	2400	2400	2500	3019
Cond. ($\mu\text{S}/\text{cm}$)	3200	2800	2900	2967	3560
pH	7.56	7.43	7.74	7.58	7.9
Turbidity (NTU)	2.1	ND	0.67	1.39	1.12

(concentrations in mg/L as ion unless otherwise noted)

$\mu\text{S}/\text{cm}$ = microsiemen per centimeter

NTU = Nephelometric Turbidity Unit

6.2 Equipment Description

6.2.1 Phase 1 NF Membrane Evaluation

The ZDD equipment used in Phase 1 is capable of producing 20-25 gpm of permeate and is comprised of NF (housed in a 40-foot container, owned by UTEP) and EDM (housed in a 20-foot container, owned by Veolia). Figure 10 includes pictures of the containers and equipment and Figure 11 shows the process flows for the equipment. The NF was a 4x2x1 array with four Dow Filmtec NF270 4-inch-by-40-inch membranes in each pressure vessel. The NF270 membranes were chosen for their low silica rejection. Hydrex 4101 antiscalant was dosed at 3 mg/L to the NF system. The BGNDRF source water has higher alkalinity than the Snake Tank water, so alkalinity was reduced by adding sulfuric acid to the feed water before de-gassing. The EDM system includes the EDM stack, pumps, chemical dosing systems, and DI tanks. The EDM stack is designed to include 100 quads. NaCl is the only regular EDM chemical input. The only waste streams in the ZDD process are the mixed sodium and chloride salts. The permeate used to produce DI water remains in the system and is not considered a waste.



Figure 10.—Phase 1 ZDD system installed in Alamogordo (20 gpm).

Demonstration of Zero Discharge Desalination (ZDD)

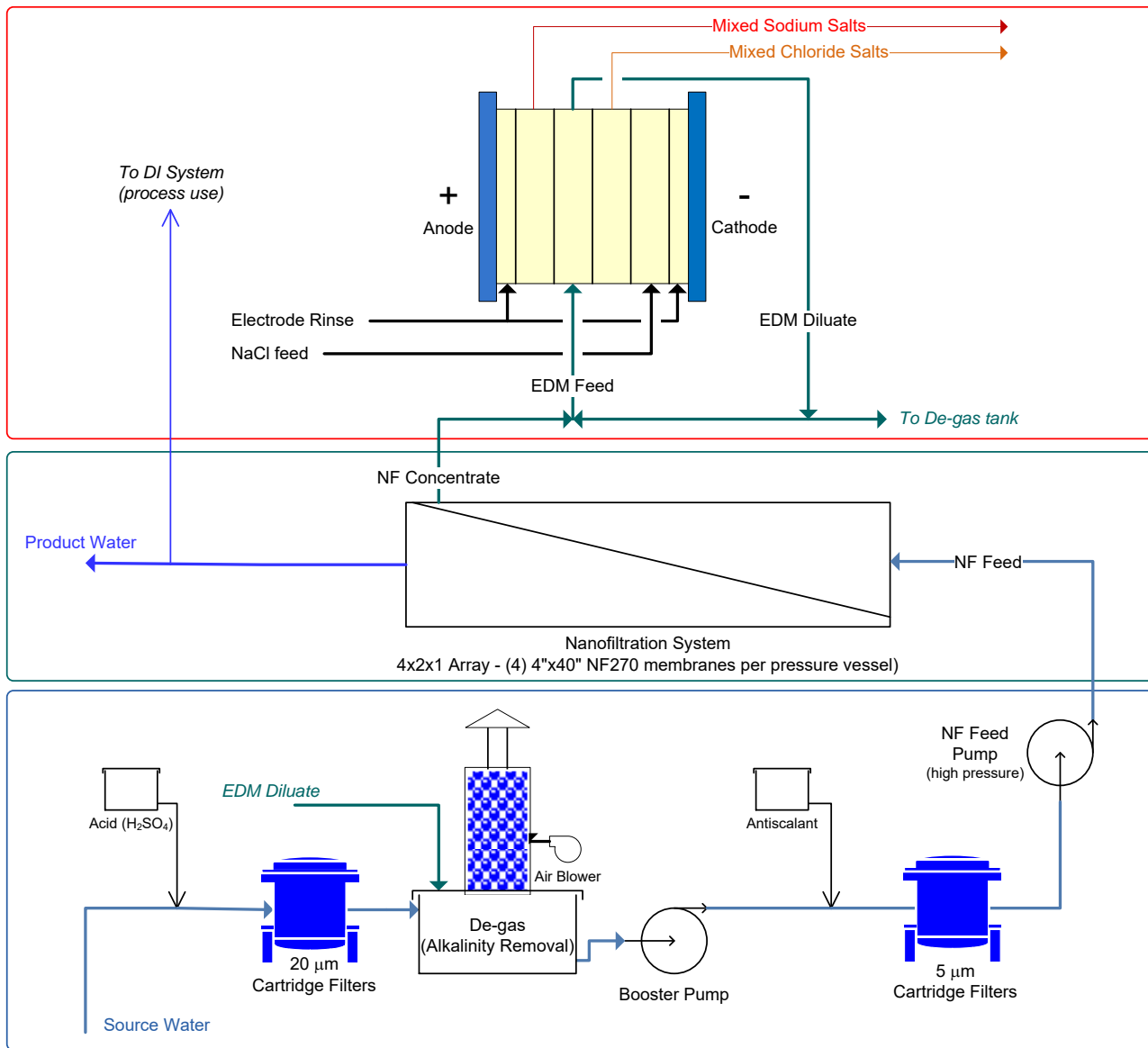


Figure 11.—Phase 1 equipment flow diagram.

6.2.2 Phase 2 Demonstration Activities

The ZDD equipment used in Phase 2 is capable of producing 20-40 gpm of permeate and is comprised of a NF and an EDM, each housed in a 40-foot container. Figure 12 includes pictures of the containers and equipment installed during Phase 2 in Alamogordo and Figure 13 shows the process flows for the equipment. The NF array was a 4x4 array with four Dow Filmtec NF270 8-inch-by-40-inch membranes in each pressure vessel. Hydrex 4101 antiscalant was dosed at 2-3 mg/L to the NF system. Alkalinity was reduced by adding sulfuric acid to the feed water before the 20-micron cartridge filter. The EDM system includes two EDM stacks, pumps, chemical dosing systems, and DI tanks. The stacks are designed to include 100 quads. NaCl is the only regular EDM chemical input. The only waste streams in the ZDD process are the mixed sodium and chloride salts. The NF permeate used to produce RO permeate remains in the system and is not considered a waste.



Figure 12.—Phase 2 ZDD System Installed in Alamogordo (40 gpm). Showing a) containers, NaCl tanks, and sulfuric acid tanks; b) NF system; c) EDM standpipes and pumps; and d) EDM stacks.

Demonstration of Zero Discharge Desalination (ZDD)

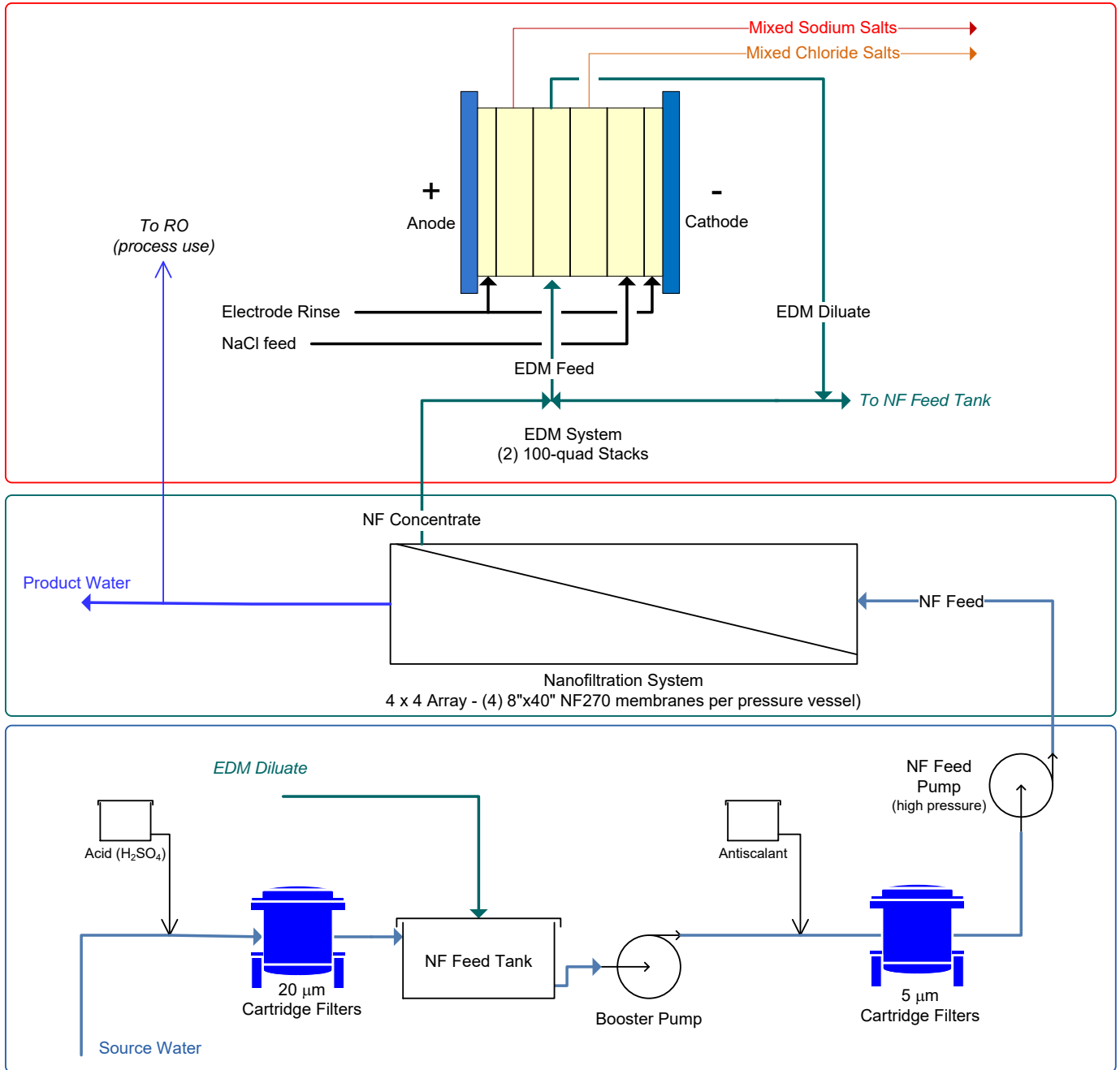


Figure 13.—Phase 2 flow diagram.

6.3 Phase 1 Operations

A 20 GPD ZDD system was installed at the BGNDRF. The brackish feedwater was a mixture of two wells. Pilot system commissioning activities began in January 2011, but the system was shut down from February 1-7 because of a winter storm that caused sub-zero temperatures and rolling blackouts in much of New Mexico and Texas.

Demonstration of Zero Discharge Desalination (ZDD)

Commissioning restarted in late February 2011. Severe scaling in one of the EDM stack compartments required a shutdown and inspection of the EDM stack. The stack had been installed at location in CA prior to being shipped to Alamogordo and some membranes were removed in between operations. The stack was not reassembled correctly, and this allowed the Mixed Cl stream and E-Rinse (i.e., sodium sulfate) streams to mix and resulted in the formation of calcium sulfate. Six severely fouled quads were removed from the stack, and the system was rebuilt. Another potential problem was thought to be from sulfate in the solar salt used in the EDM process. The solar salt was replaced with pelletized NaCl (water softener salt) purchased at the local home improvement store. No further scaling issues were experienced during Phase 1.

After being commissioned, the ZDD system was operated for five weeks at flow rates ranging from 14-25 gpm. The system was only shut down on weekends when project personnel didn't have access to the site. System recovery was 98 percent and product water quality of TDS below 800 mg/L was achieved.

Table 6 summarizes the operational data (average values for the entire period with standard deviation shown in parentheses) and field water quality analyses for Phase 1 operations in Alamogordo. As the NF permeate flow was increased from 14 to 20 gpm, the proportion of the NF reject flow to the EDM feed-diluate recirculating flow increased. This is because the EDM feed-diluate flow was kept constant at around 25 gpm. More voltage would be required to achieve the same ion removal with a fixed recirculation rate. The conductivity was relatively constant for the 14 gpm and 20 gpm runs, but increased dramatically during the 25 gpm run (especially on the final day). This increase is possibly due to internal leakage or back diffusion from the concentrate streams to the feed streams in the EDM. Visible leakage was noted near the center divider during Phase 1 operations. There is some indication that the NaCl stream was contaminated with sulfate and possibly calcium and bicarbonate. The NaCl stream was sampled and sulfate was analyzed in the field regularly; sulfate increased from levels below 200 mg/L to as high as 1,325 mg/L at the end of each run. Similar internal leakage is described in the Phase 2 operation (see Section 4.2). Online data can be found in Appendix A1 and logsheet data in Appendix B1.

Table 6.—Alamogordo Phase 1 Operational Data and Field Water Quality Analyses

NF Permeate Flow		14-gpm Avg (SD)	20-gpm Avg (SD)	25-gpm Avg (SD)
NF Removal (NF feed→NF permeate)				
conductivity	%	60.1 (2.7)	66.2 (1.6)	68.8 (0.8)
total hardness	%	86.2 (2.8)	90.3 (1.3)	91.8 (2.1)
alkalinity	%	14.9 (6.2)	21.6 (12.2)	21.8 (17.3)
sulfate	%	98.5 (0.3)	98.9 (0.2)	99.2 (0.2)
silica	%	-4.6 (22)	2.2 (11.9)	-3.5 (10.2)
NF Permeate	μS/cm	955 (201)	916 (120)	912 (77)

Demonstration of Zero Discharge Desalination (ZDD)

NF Permeate Flow		14-gpm Avg (SD)	20-gpm Avg (SD)	25-gpm Avg (SD)
EDM feed-diluate flow	gpm	24.8 (0.9)	24.9 (1.1)	23.6 (0.5)
EDM feed pressure	psi	10.2 (0.7)	13.5 (0.8)	14.9 (0.4)
EDM recycle ratio	range	1.7-1.9	1.0-1.2	0.6-0.7
Brackish SW conductivity	μS/cm	3,012 (326)	3,020 (215)	2,917 (56)
EDM Stack Voltage	V _{DC}	100 (14)	138 (13)	180 (0.3)
EDM Stack Current	amp	35 (2.4)	51 (2.5)	64 (1.9)
ZDD Recovery	%	98.0 (0.1)	98.2 (0.2)	98.3 (0.05)
EDM Removal (NF Reject→EDM Diluate)				
conductivity	%	68.5 (2.7)	56.3 (2.8)	46.7 (7.8)
sulfate	%	80.1 (7.1)	68.7 (3.2)	51.9 (13)
ZDD Removal (SW→NF Permeate)	%	68.6 (3.3)	69.7 (2)	68.7 (2.1)
Longest Continuous Run	Date Hours	3/15-3/25 239	3/25-4/1* 166*	4/5-4/8 76

*The system was operated for a total of 188 hours at 20-gpm NF permeate flow.

Avg = Average

SD = Standard Deviation

V_{DC} = volts direct current

A set of samples was taken on April 5, 2011, and sent for analysis by Hall Environmental Analysis Laboratory (Hall) in Albuquerque, New Mexico. The results are summarized in Table 7. The original results from Hall indicated poor ion balance for the Mixed Cl and Mixed Na streams. UTEP results are provided for these and are an average of three samples of Mixed Cl and six samples of Mixed Na (gathered in 5-gallon buckets).

Table 7.—Phase 1 Water Quality Analysis Summary

	Source	NF Feed	NF Perm	NF Conc	EDM Feed	EDM Diluate	Mixed Cl	Mixed Na
F	1.7	1.4	0.9	2.5	1.8	0.8	ND	61.3
Cl	290	210	260	160	140	100	74,767	12,694
SO ₄	1,300	1,300	9.50	3,600	2,600	1,300	927	110,161
Ca	250	220	19	NT	380	160	13,838	412
Mg	94	94	7.50	NT	190	86	7,623	75
Na	370	350	150	NT	580	350	18,588	61,172
SiO ₂	24.0	27.0	23.0	NT	33.0	33.0	5.7	11.4
pH	7.4	7.0	7.0	7.0	6.9	6.6	1.8	7.9
HCO ₃	205	110	70	135	128	43	NT	4,931
Cond.	2,857	2,724	872	5,021	4,271	2,404	124,800	109,000
TDS	2,300	2,180	471	4,980	3,840	1,940	160,553	184,387

Demonstration of Zero Discharge Desalination (ZDD)

	Source	NF Feed	NF Perm	NF Conc	EDM Feed	EDM Diluate	Mixed Cl	Mixed Na
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Concentrations in mg/L, conductivity in $\mu\text{S}/\text{cm}$

The EDM specific energy (kilowatt hour [kWh] per unit of treated water) was estimated for the three NF permeate production flows during Phase 1. Figure 14 includes the energy required for desalination and excludes the circulating pumps (EDM stack power is about 70 percent of the total EDM system energy requirement). A factor of 82 percent is assumed for converting alternating current voltage (VAC) to direct current voltage (VDC) voltage. Figure 14 displays the range of values obtained at three recycle ratios. The recycle ratio is the ratio of the EDM diluate flow returned to the EDM feed divided by the NF concentrate flow (Bond et al., 2015). A recycle ratio of one would indicate that the EDM feed is comprised of equal parts EDM diluate and NF concentrate. As the NF permeate flow increased, so did the NF concentrate flow. Since the EDM recirculating rate was kept constant, the recycle ratio decreased, and more energy was required to remove the ions in the NF concentrate.

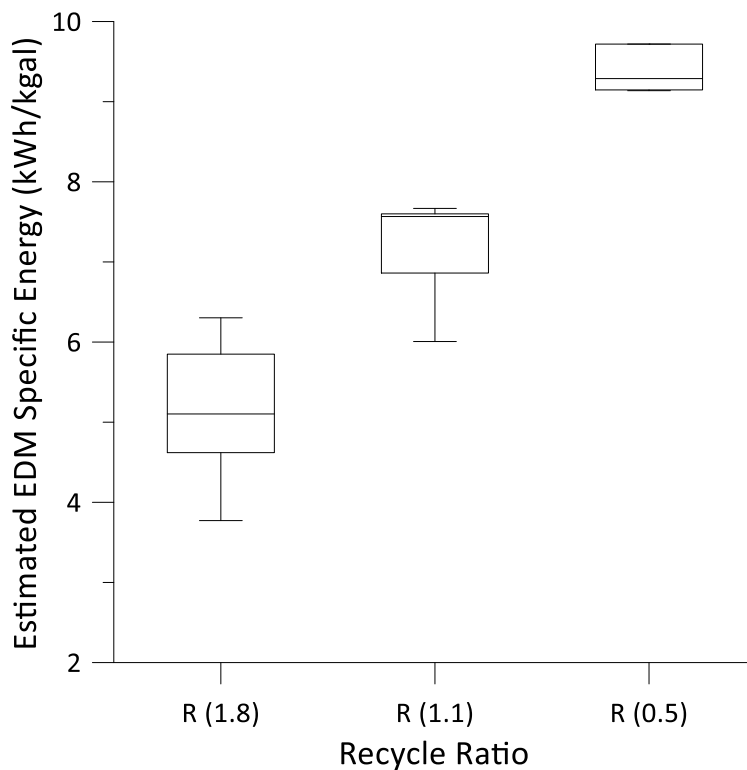


Figure 14.—Estimated EDM stack specific energy at Alamogordo (Phase 1).

6.4 Phase 2 Operations

6.4.1 Preliminary Experiments (Before 4/22/13)

The Phase 2 demonstration equipment arrived in January 2013, and Veolia engineers spent January and February making piping connections, improving processes, and implementing controls for both systems. Several problems were encountered early in the start-up. The equipment was shipped from the integrator, which installed the NF and EDM equipment inside the 40 foot containers, without draining all fluid from the piping. Several online probes and sections of pipe froze and had to be replaced. The NF feed pump arrived damaged from the factory and was repaired and re-installed by local Grundfos technicians. The NF membranes were initially installed incorrectly. The membranes were removed and installed correctly on March 7, 2013.

The system was not operated overnight until nighttime temperatures were high enough to minimize freezing potential and the control system had proper automatic shutdowns and allowed remote viewing and control. Several short experiments were performed between March 20 and 22. Longer experiments took place during March 22-28 and April 8-11. Automated shutdowns, automated DI addition to EDM streams, and remote control and viewing were not in place until April 5. The E-Rinse flow meter stopped working on March 28 (which caused an automatic shutdown). Later operations used E-Rinse pressure to ensure flow (if E-Rinse pressure was less than 3 pounds per square inch (psi), it would trigger an automatic shutdown).

Miscommunication led to higher-than-necessary voltage being applied until April 17. The team intended to apply 1.2-1.3 volts/quad to each stack. The team mistakenly thought the stacks contained 125 quads (instead of the correct 100 quads), so the actual voltage was closer to 1.6 volts/quad. It is likely that the EDM systems were operated at or beyond the limiting current density and water splitting occurred. One way to identify whether water splitting is occurring is to monitor the difference in pH between the Mixed Na and Mixed Cl streams. When water splitting is occurring, the pH of the Mixed Cl stream will decrease and the pH of the Mixed Na stream will increase (see Figure 34).

Grab samples and data were taken at least once per day to check online probe calibration, verify operational setpoints, and act as a backup for loss of online data. Conductivity, pH, specific gravity, and temperature were measured in the field.

Figure 15 summarizes the conductivity, pH, temperature, and specific gravity measurements taken during the preliminary experiments. The online conductivity data before April is not representative of actual conductivity. The programmable logic controller (PLC) code was not fixed to include proper scaling until the first week of April due to personnel availability. The Mixed Cl and Mixed Na online conductivity probes' data are not valid for this most of this time period as well. The conductivity probes had interference from stray current. The probes were

Demonstration of Zero Discharge Desalination (ZDD)

initially installed in the feed piping, and it is possible to have current travel from the stack to the probes through the liquid streams. This stray current tended to cause the transmitter to drift and/or not calibrate properly. Logsheet data can be found in Appendix A2 and online data can be found in Appendix B2 of this report.

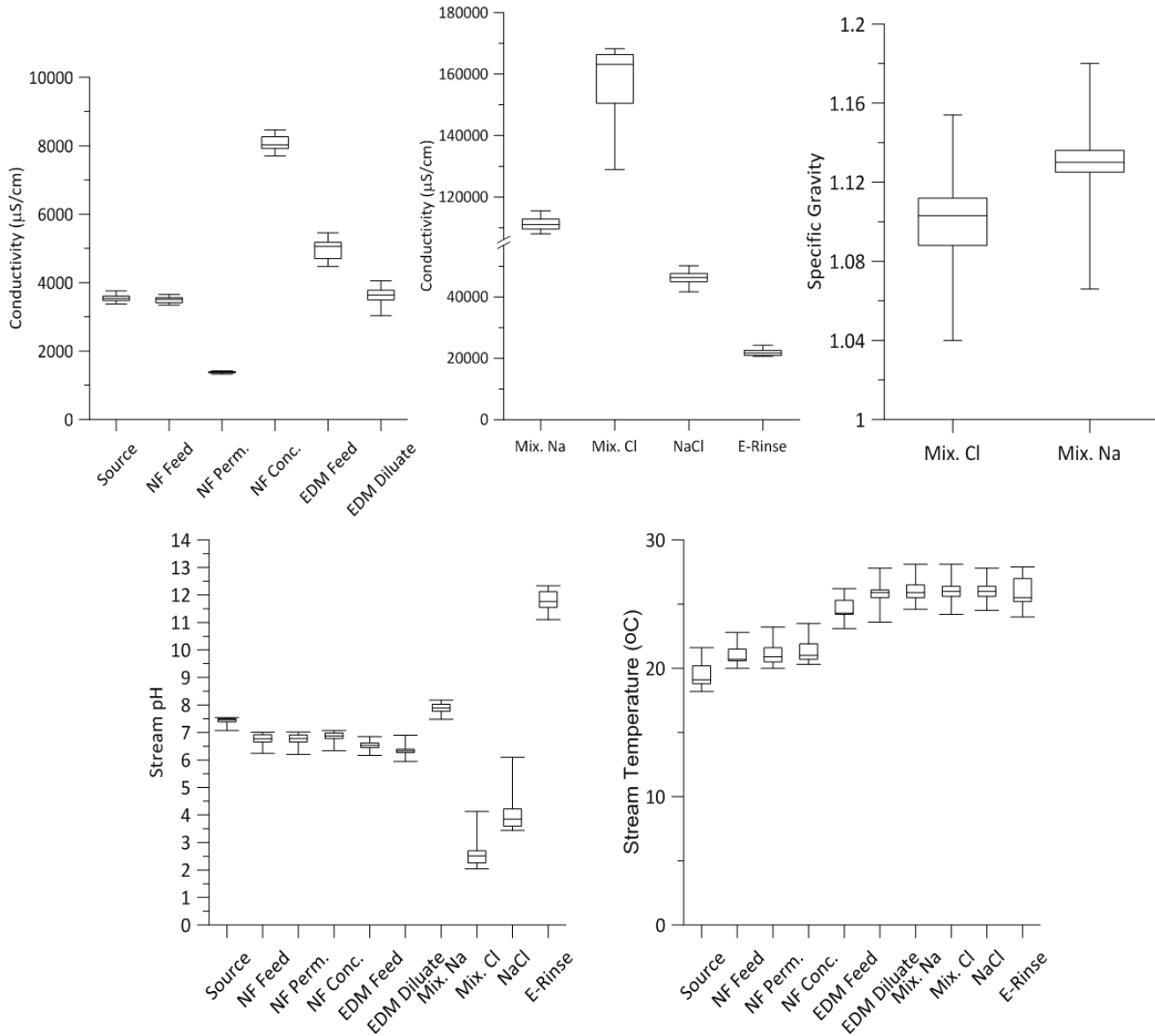


Figure 15.—Grab sample data (preliminary experiments).

Demonstration of Zero Discharge Desalination (ZDD)

The NF and EDM systems operated smoothly with no signs of scale formation until April 11. The datalogger did not record data for April 8-11. Logsheet data does have several data for this period. The system was shut down after there was indication of scale formation (EDM pressures increased and EDM conductivity reduction decreased). A leak test identified major internal leakage, so the stacks were taken apart. One stack (called EDM-1) had a spacer in the incorrect configuration (see Figure 16). The spacer was in D2 (EDM diluate) position, allowing mixing of EDM diluate to the NaCl. The red staining in Figure 16 is typical for the anion membrane in the D2 compartment and is thought to be from iron and/or biological fouling. The other stack (called EDM-2) had a bent spacer due to a piece of membrane leftover from the manufacturer (see Figure 17). Additionally, the center dividers in each stack had holes and were folded. This bent membrane piece allowed mixing of C1 (Mixed Cl) and C2 (Mixed Na) streams. Substantial scale was found at the center of each stack and near the spacers with problems (see **Figure 18**). The scale shown is believed to be caused by misalignment of the black divider. This allowed mixing of all streams. The five quads closest to the anode of EDM-2 were stuck together but seemed to be functional enough for operation.



Figure 16.—EDM-2 Stack with corrected D1 (NaCl outlet) spacer installation.

Demonstration of Zero Discharge Desalination (ZDD)



Figure 17.—EDM-1 Stack with membrane piece from manufacturer.



Figure 18.—EDM-2 center membrane (top) and center divider (bottom).

Demonstration of Zero Discharge Desalination (ZDD)

The ZDD system was re-started on April 17 and two daytime runs were used to test operations. An overnight run with no malfunctions occurred on April 19-20, and the system was considered fixed.

The relationship between conductivity and solution concentration is not reliable for high strength solutions such as the Mixed Na and Mixed Cl streams in the EDM system, because the change in conductivity (related to the change in concentration) diminishes at elevated concentrations of all electrolytes. For this reason, grab samples were taken and the specific gravity and conductivity were measured. Dilution water (RO permeate) was added automatically on a schedule to maintain a specific gravity of 1.13 in the Mixed Na stream. This setpoint was chosen to mitigate sodium sulfate precipitation; a concern in March and April when nighttime temperatures were at or below freezing and the Mixed Na streams were at or below 25 degrees Celsius (see Figure 8 for sodium sulfate solubility vs temperature).

It is important to note that, while there was a larger variation in the NF concentrate, EDM feed, and EDM diluate streams, the NF permeate remained relatively consistent (see

Figure 15 and Appendix B2). The ZDD system reduced the conductivity of the NF reject by an average of 61 percent during this time period. All of the field sample analyses are summarized in

Figure 15. Colder overnight temperatures and a desire to minimize both sodium sulfate and sodium bicarbonate scale formation in the Mixed Na stream led to increased dilution of the Mixed Na stream. This resulted in lower ZDD recovery, ranging from 95.8 percent-97.7 percent.

6.4.2 Experiment 1a and 1b (4/22/13-4/26/13)

The system was operated for two periods during April 22-26. The datalogger failed to record data; however, logsheets from in-person and remote login checks are available to describe operations. Appendix B2 includes data from these sources. The system operated fairly stably from April 22 to 24, but the EDM diluate conductivity was increasing and EDM pressures were increasing slightly. A burning smell on April 24 led to a rapid shutdown and inspection of the stacks. During shutdown, the EDM feed stream turned a reddish-brown color. The stream was rinsed with DI water and the problem did not reappear upon restarting.

EDM-2 had been dripping, and dried salts formed at the bottom of the stack that resembled stalactites towards the cathode and on the cathode side of the center of the stack (Figure 19).

Demonstration of Zero Discharge Desalination (ZDD)

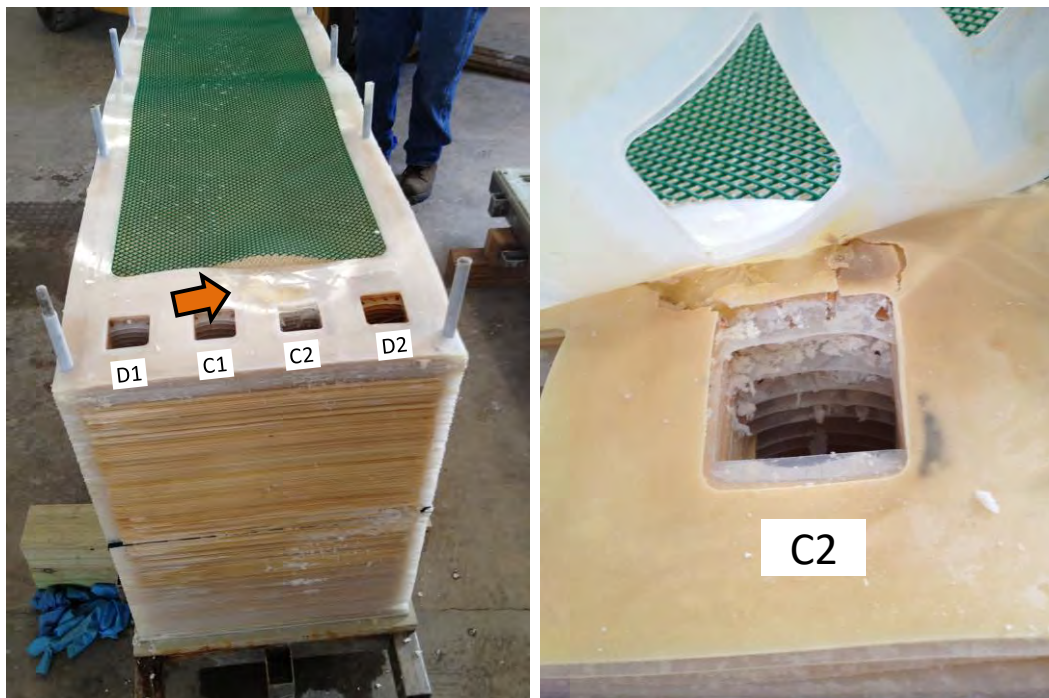


Figure 19.—EDM-2 stack showing melted anode spacers and adjacent damaged membranes (C2 is mixed Na).

Upon removal, the five quads in EDM-2 closest to the anode were found to be much more damaged than what was seen during the previous inspection. The manufacturer, MEGA, limits the applied voltage to less than 2 volts/quad. The team thought that 1.75 volts/quad was necessary to achieve the desired water quality objectives at the time. However, in hindsight, too much voltage was applied on April 23-24 (the stack was operated well above the limiting current density). This, combined with the salt growths outside the stack, allowed shunt currents to travel down the membranes to the manifolds. This shunt current allowed localized heat increases that melted the membranes and spacers.

The system was restarted with a single EDM stack (EDM-1) and a lower NF permeate production rate setpoint (20-gpm). The system was stable for nearly 24 hours and was shut down on Friday, April 26. The BGNDRF is not usually open on weekends and personnel were not available for weekend operation.

The grab sample analyses are summarized in Figure 20. The E-Rinse had a low pH on the final day of operation; this is either an analytical error or could be an indication of contamination of the electrolyte stream. The sample analyzed at UTEP also had a lower pH. Samples taken between April 20-26 had elevated bicarbonate and the samples on April 25 and 26 had higher sodium as well. The stack resistance and the permeate quality were stable for the entire run (see Appendix B2 of this report).

Demonstration of Zero Discharge Desalination (ZDD)

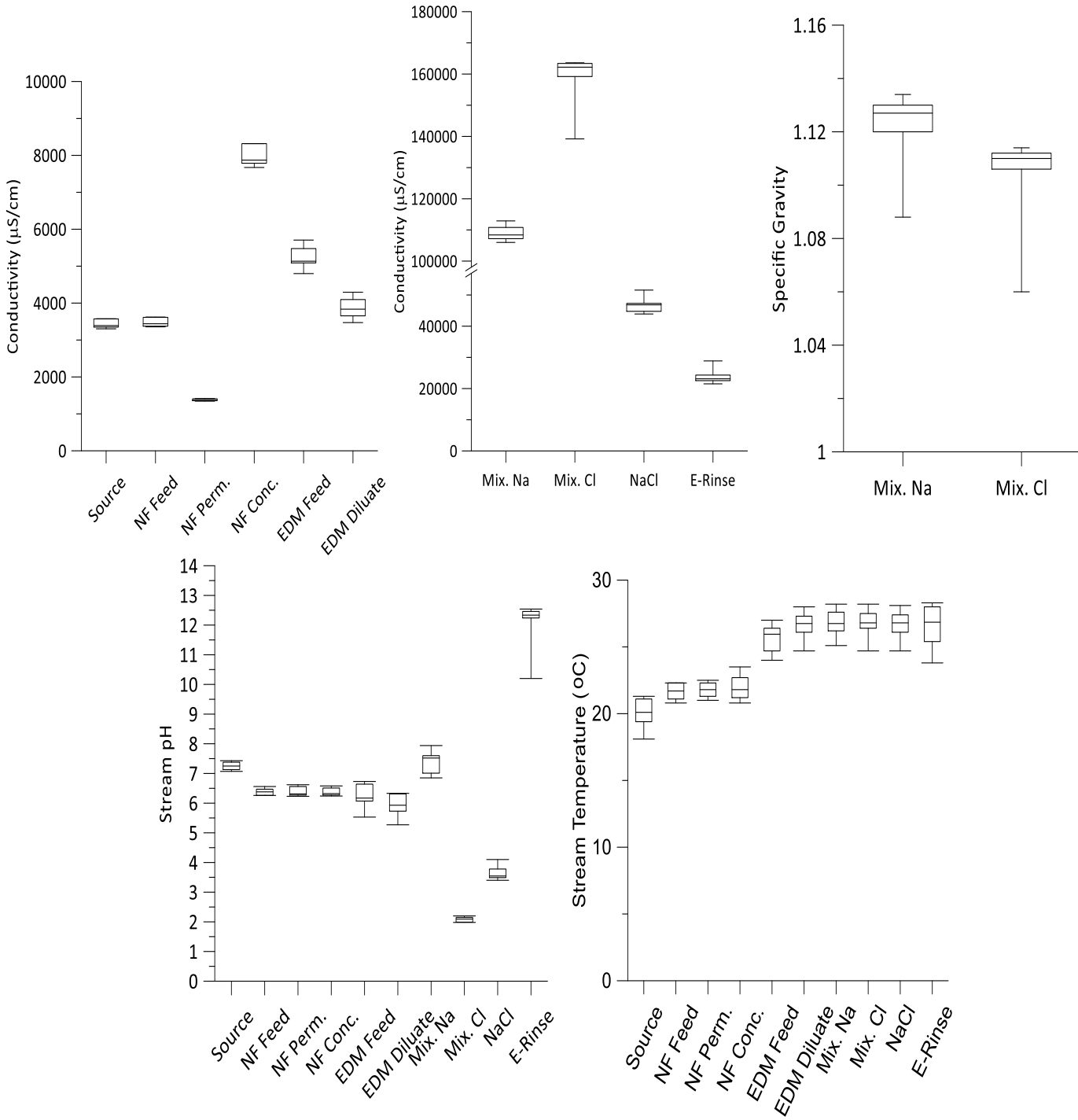


Figure 20.—Grab sample data (Experiment 1).

6.4.3 Experiment 2 (4/30/13-5/2/13)

The system was restarted with a single EDM stack and 20 gpm of NF permeate flow on April 3 and operated continuously for 58 hours. The antiscalant tank went empty at some point, and there was not enough sulfuric acid remaining to operate after May 2.

Isolation transformers were installed on the power supply for the Mixed Cl and Mixed Na conductivity transmitters. This seemed to allow for proper calibration of the conductivity probes and reliable conductivity data were obtained for the first time by the online data logger.

New 5-micron and 20-micron cartridge filters were installed on the source water and NF feed lines as well as CIP for iron, sulfate, and carbonate scale. However, the NF seemed permanently scaled and new membranes were ordered. The source of the iron in the NF and EDM was unclear at the time. Pictures taken on May 3 of the cartridge filters suggest there was some iron (or red particles) in the BGNDRF source water (Figure 21). Analytical results did not indicate regular presence of iron in the source water or NF streams; however, analyses by inductively coupled plasma atomic emission spectroscopy (ICP-OES) require filtration, so it is possible the iron was filtered out.

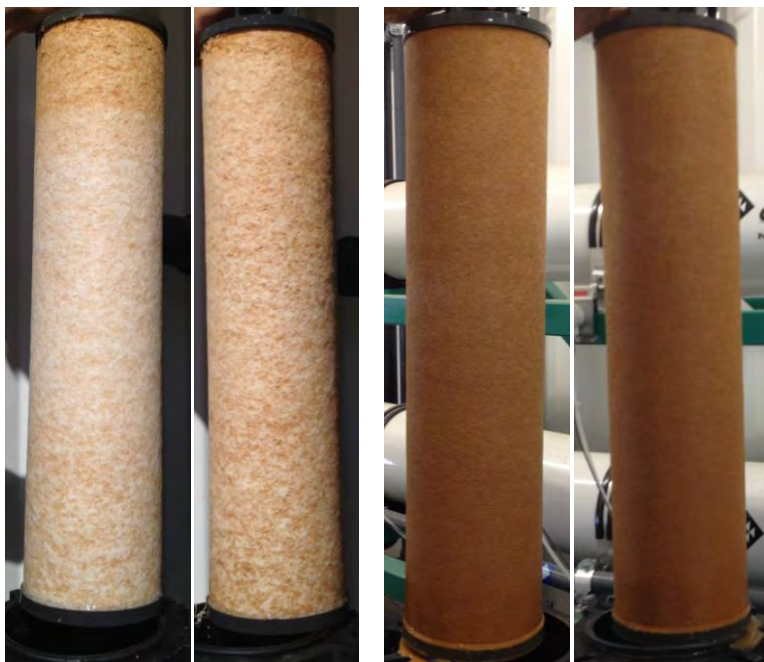


Figure 21.—NF cartridge filters on May 3. Left: 20-micron filters on source water. Right: 5-micron filters on NF feed.

Grab sample analyses are summarized in Figure 22 and online data are summarized in in Appendix B2 of this report.

Demonstration of Zero Discharge Desalination (ZDD)

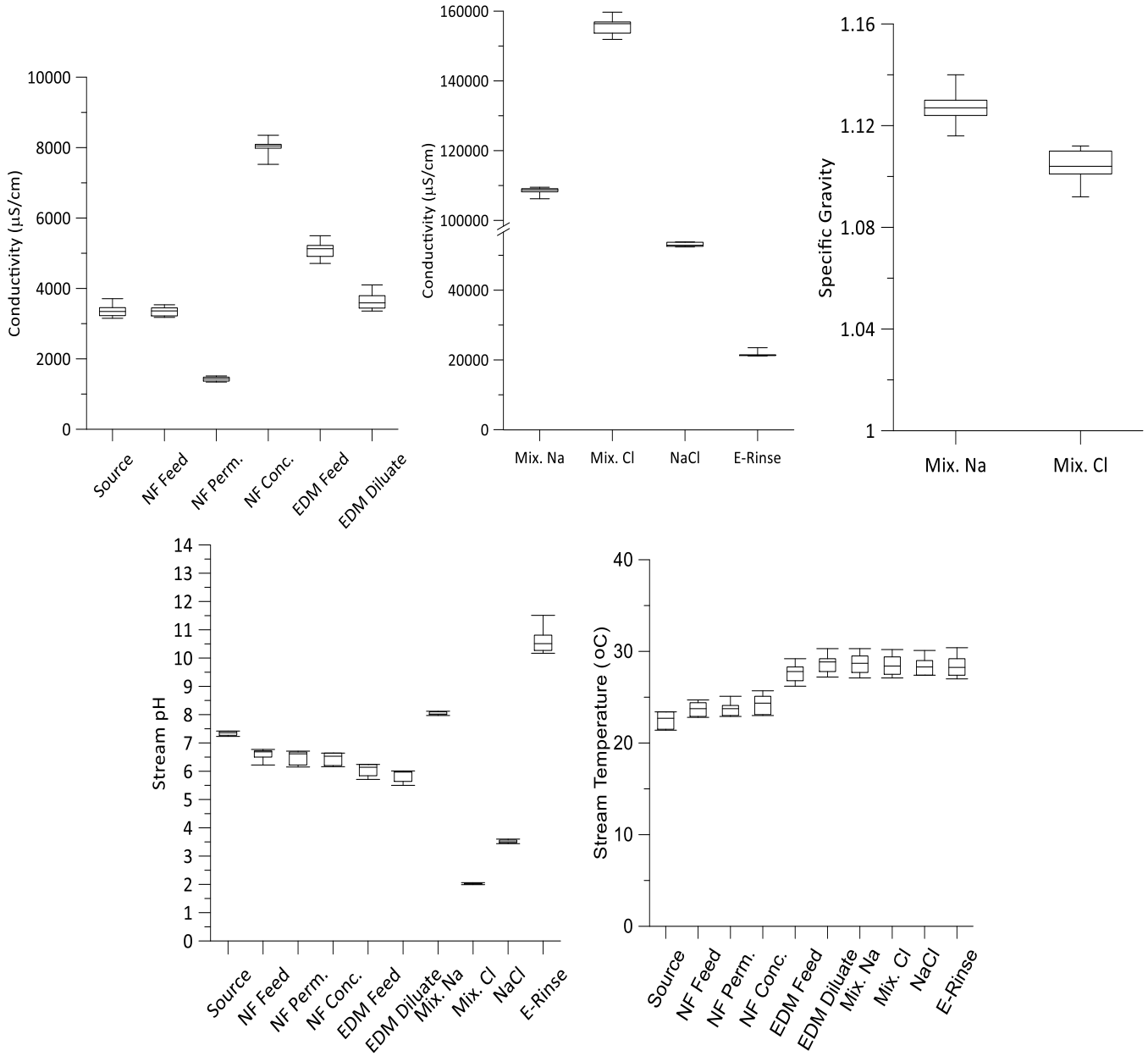


Figure 22.—Grab sample data (Experiment 2).

6.4.4 NF Membrane Replacement (5/23/13)

The original NF270 membranes were removed and replaced with new membranes on May 23. The membrane elements showed signs of particulate fouling by what was suspected to be iron. Additionally, there was scale outside several of the elements and in the end caps of the pressure vessels (Figure 23).

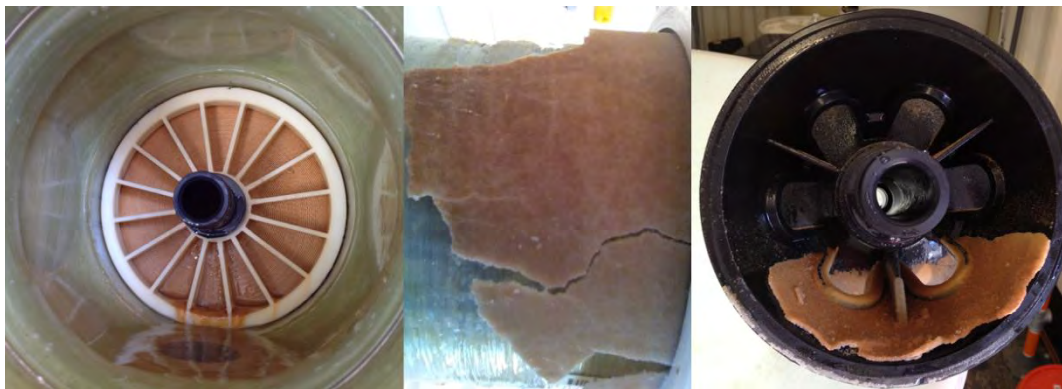


Figure 23.—Original NF270 membranes removed on May 23.

Samples were not gathered for analysis, but the scale was suspected to be a mixture of calcium sulfate (from the loss of antiscalant) and iron particulates (later determined to be from corrosion in the acid feed line).

6.4.5 Experiment 3 (5/29/13-6/4/13)

The system was tested on May 29 and June 3 to determine whether controls, the new acid, and new NF membranes were working properly. Operators were not available to keep the system operating overnight on June 3 so the system was shut down. The system was operated continuously for nearly 21 hours between June 4 and 5. The new sulfuric acid was more concentrated (93 percent) than the original strength (50 percent). This, combined with the controls, led to a much wider pH span measured by the pH meter in the acidified source water pipe (see Appendix B2).

The grab sample analyses are summarized in Figure 24, and the online data are summarized in Appendix B2 of this report. Similar to earlier experiments, there was more variation in the EDM feed, EDM diluate, and NF concentrate than in the NF permeate. The overnight run started on June 4 was more stable than the earlier runs, in terms of all streams' conductivities. The Mixed Na probe began to read incorrectly and would not calibrate properly beginning on June 4.

The system was shut down on June 5 so that replacement membranes could be installed in EDM-2 and EDM-1 could be inspected. Several issues were found in

Demonstration of Zero Discharge Desalination (ZDD)

both stacks and are summarized below. EDM-1 had 91 quads and EDM-2 had 94 quads; both stacks were reinstalled in the EDM container.

- A hole was found in the top of the cation membrane between the EDM feed and Mixed Cl compartments in the second quad of EDM-1 on the cathode end of the stack. This hole allowed mixing and a substantial amount of scale was found on the cathode and on the anode side of the membranes in the first two quads. The quad closest to the cathode was removed; the remaining membranes were cleaned and four membranes were replaced.
- The center divider was damaged and six quads that were stuck together were removed (see Figure 25). Two quads on each side of the center divider were stuck together but were separated and left in the stack.
- Five quads were removed from EDM-2 (the original five quads that were found melted on April 25). Similar to EDM-1, the center divider was damaged and four quads (two on each side of the center divider) were stuck together. The membranes and spacers were separated and remained in the stack.
- New center dividers were made using 6 mil high density polyethylene (HDPE) purchased at the local hardware store.

Demonstration of Zero Discharge Desalination (ZDD)

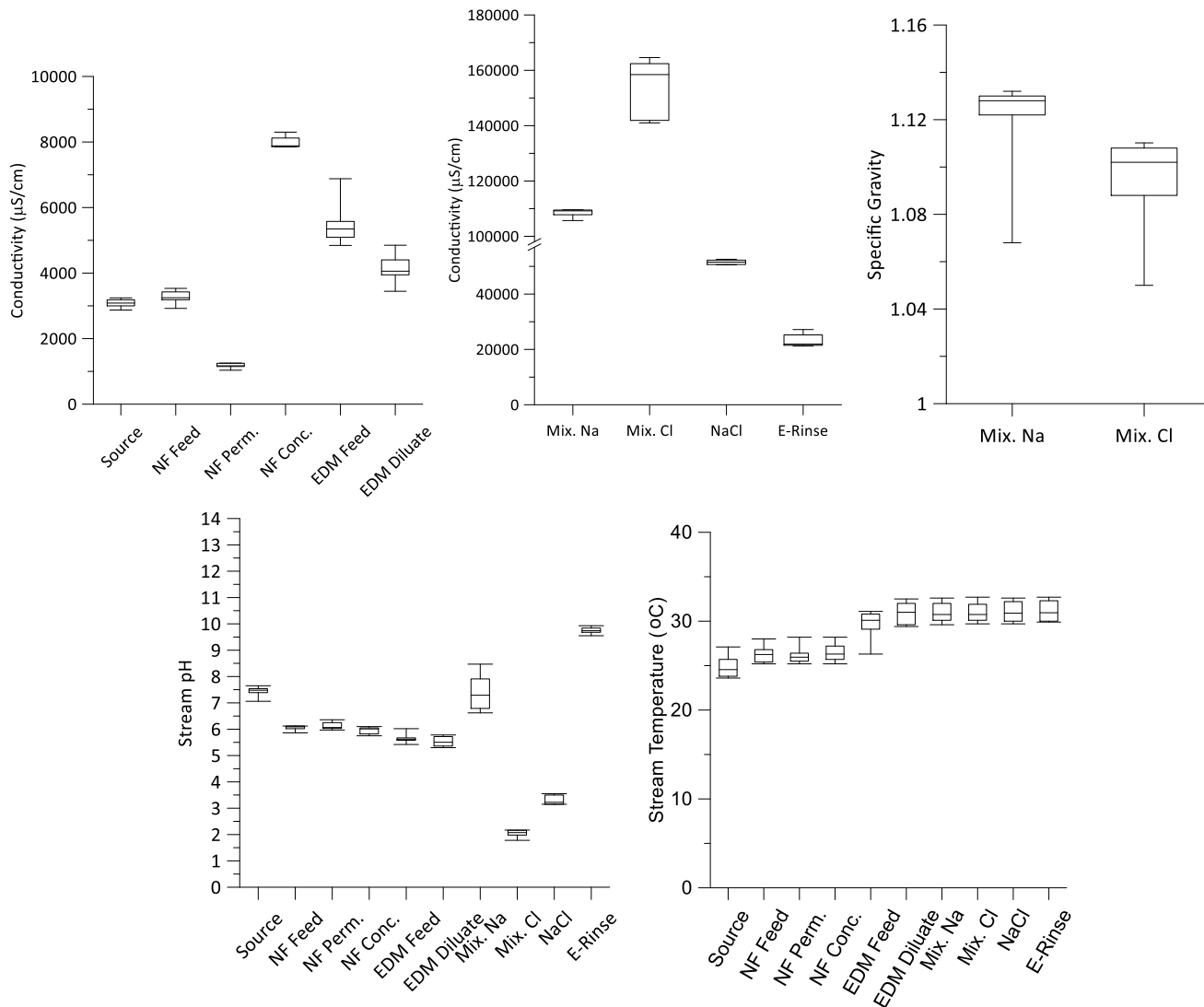


Figure 24.—Grab sample data (Experiment 3).

The system was shut down on June 5 so that replacement membranes could be installed in EDM-2, and so EDM-1 could be inspected. Several issues were found in both stacks and are summarized below. EDM-1 had 91 quads and EDM-2 had 94 quads; both stacks were reinstalled in the EDM container.

- A hole was found in the top of the cation membrane between the EDM feed and Mixed Cl compartments in the second quad of EDM-1 on the cathode end of the stack. This hole allowed mixing and a substantial amount of scale was found on the cathode and on the anode side of the membranes in the first two quads. The quad closest to the cathode was removed; the remaining membranes were cleaned and four membranes were replaced.

Demonstration of Zero Discharge Desalination (ZDD)

- The center divider was damaged and six quads that were stuck together were removed (see Figure 25). Two quads on each side of the center divider were stuck together, but were separated and left in the stack.
- Five quads were removed from EDM-2 (the original five quads that were found melted on April 25). Similar to EDM-1, the center divider was damaged and four quads (two on each side of the center divider) were stuck together. The membranes and spacers were separated and remained in the stack.
- New center dividers were made using 6-mil HDPE purchased at the local hardware store.



Figure 25.—EDM-2 precipitation and damage from mixing at center of stack.

6.4.6 Experiment 4 (6/17/13-6/21/13)

The system was offline for most of the week of June 10 for additional control modifications and was restarted on June 17. The system was shut down on June 17 for a remote update to the controls and restarted again on June 18. The data logger did not record data for part of the day on June 19, but log sheet data and grab samples filled in the gaps. The system ran continuously for 79 hours.

The NF permeate production was intentionally reduced to 35 gpm on June 19 to reduce the percentage of NF reject to EDM feed to 50 percent or lower. This percentage was identified as an operational setpoint during Phase 1. This change also reduced the average flux from 18 gallons per square foot per day (GFD) to 15.75 GFD.

The EDM's operations were fairly stable throughout the week. The flows, pressures, and stack resistance were constant and showed no signs of scale formation. Towards the end of the week the EDM feed and NF feed and concentrate streams showed signs of iron. The NF feed tank and EDM feed systems were drained and refilled while operational to reduce the level of iron visible.

The grab sample analyses are summarized in Figure 26 and the online data are summarized in Appendix B2 of this report. The EDM stack resistance increased when the particles were visible, and this increase in stack resistance caused the EDM diluate conductivity to increase due to the consequent decrease in stack current. That said, the NF permeate conductivity remained stable for most of the week. Lower voltage was applied and the difference in pH between the Mixed Na and Mixed Cl was lower than in earlier experiments. The pH of the NaCl was also higher than the earlier experiments. Subsequent water quality analyses suggest that the NaCl stream could have been contaminated with sulfate and bicarbonate, HCO_3 , from the Mixed Na stream. Beginning on June 14, NaCl stream's bicarbonate alkalinity went from being below detection levels to 91-220 mg/L as HCO_3 (see Figure 32 and Appendix C2). Sulfate also increased at this time. Unfortunately, these analyses were not available for operators and no changes were made to the system.

Demonstration of Zero Discharge Desalination (ZDD)

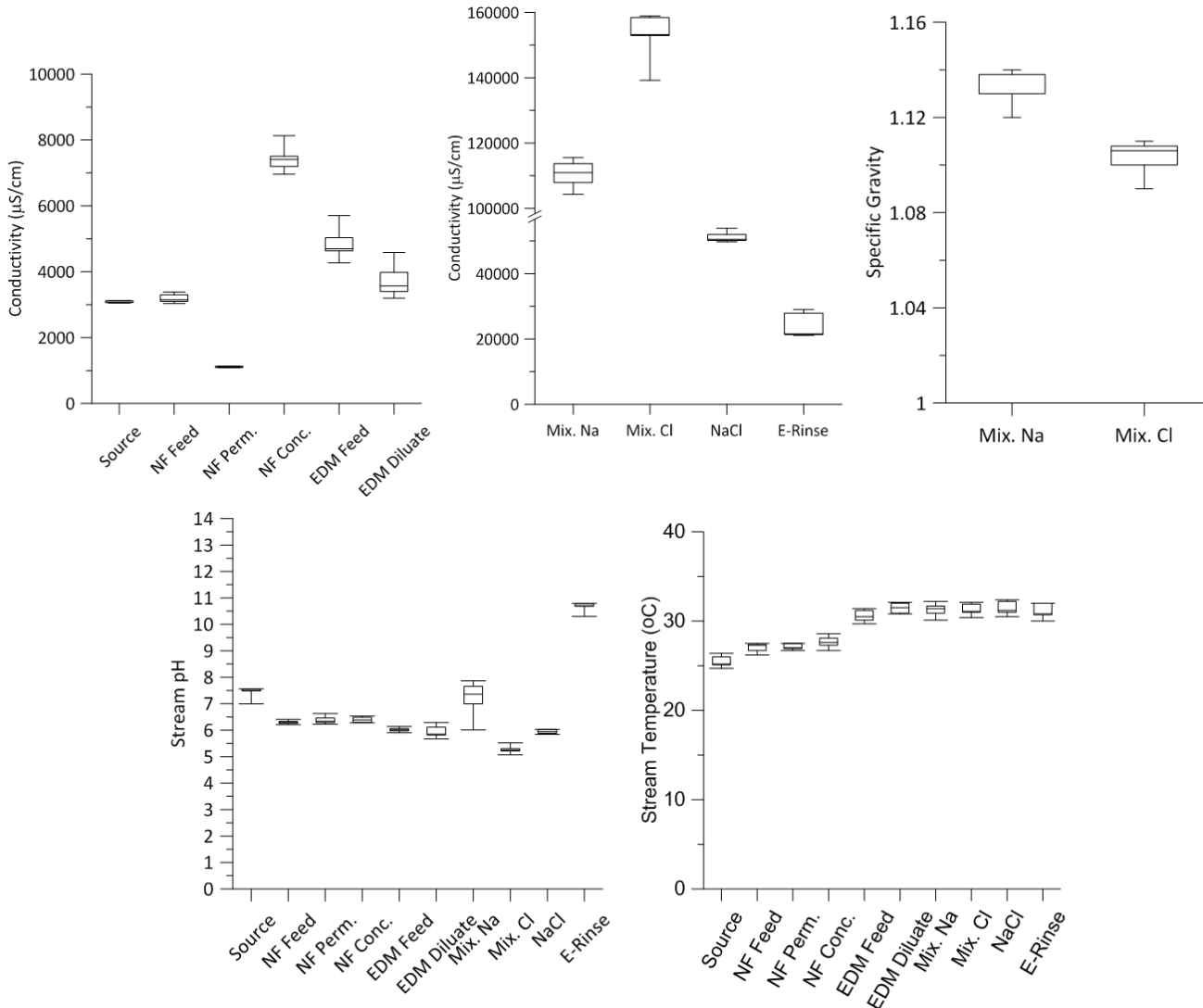


Figure 26.—Grab sample data (Experiment 4).

6.4.7 Experiment 5 (6/24/13-6/28/13)

The system was restarted on June 24 and operated for 24 hours. The system was shut down, and a CIP was performed to remove iron in the EDM feed and NF systems. The CIP was moderately successful, but the improvement was temporary. The system was restarted on June 26 and operated continuously (with occasional purges of the EDM feed and NF feed tanks to remove visible iron) for 54 hours.

The grab sample analyses are summarized in Figure 27 and the online data are summarized in Appendix B2 of this report. The NF system was stable for the week’s operation, but the EDM showed signs of potential scale. The pressures increased throughout the week. As mentioned previously, there may have been contamination in the NaCl stream that caused bicarbonate and sulfate to be transferred to the Mixed Cl stream. The combination of iron particles in the EDM

Demonstration of Zero Discharge Desalination (ZDD)

feed and bicarbonate and sulfate in the NaCl stream likely led to blockages inside the EDM feed and Mixed Cl streams. These blockages would cause increased pressure and reduced flow.

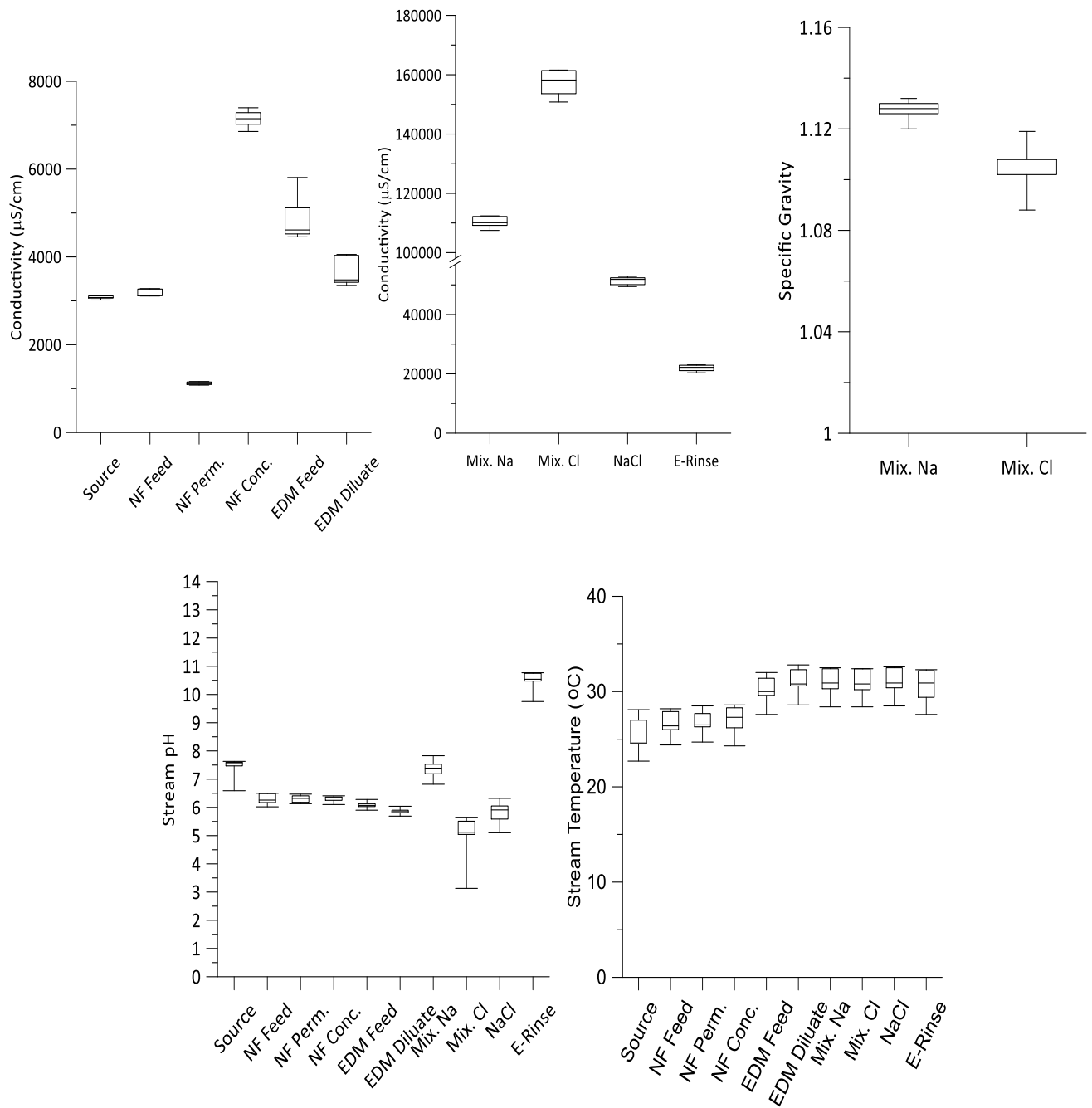


Figure 27.—Grab sample data (Experiment 5).

6.4.8 Experiment 6 (7/8/13-7/14/13)

The final experiment at BGNDRF began on July 8 and, except for two short shut downs, operated continuously for a total of 137 hours (75 hours from July 8

Demonstration of Zero Discharge Desalination (ZDD)

through 11, and 62 hours from July 11 through 14). The system was shut down on July 11 for a CIP on the EDM and a brief (less than one hour) shutdown on July 12 to fix a leak on the E-Rinse piping and install one micron cartridge filters on the NF feed to remove suspected iron particles.

The one micron cartridge filters on the NF feed nearly eliminated the visible iron in the EDM feed and NF feed tanks. However, the EDM resistance continued to increase, especially in EDM-1. The higher resistance before July 12 is believed to be caused by iron and possibly internal leakage. The higher resistance on July 12 is believed to be caused by lower conductivity in the E-Rinse and internal leakage. The final days' operation had a marked decrease in the Mixed Cl flow and an increase in conductivity in the EDM feed and diluate, E-Rinse, and all NF streams. These problems were likely caused by internal leakage and precipitation in the Mixed Chloride stream.

The EDM stacks were removed and inspected after the system was shut down. The EDM-1 stack had a substantial amount of scale at the center of the stack. Both stacks had similarly damaged or misaligned center dividers again. New center dividers were made using a thicker HDPE sheet and installed in EDM-1 (the local supplier didn't have enough material for both stacks). New dividers made from the 6-mil plastic sheets were made and installed in EDM-2.

The grab sample analyses are summarized in Figure 28 and the online data are summarized in Appendix B2 of this report. Similar to all previous experiments, even with reasonably large variation in the EDM salt removal, the NF permeate quality was relatively stable.

Demonstration of Zero Discharge Desalination (ZDD)

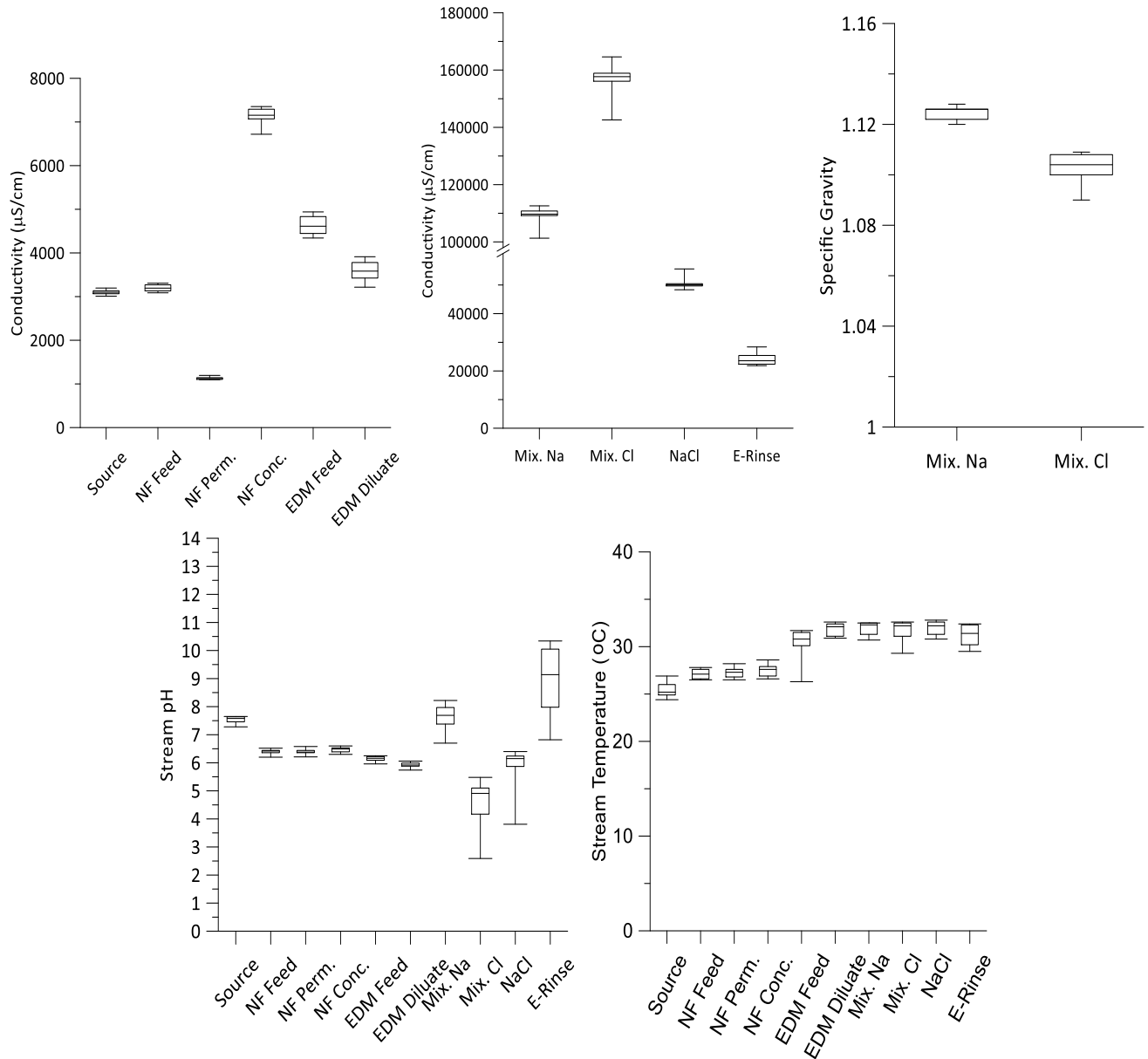


Figure 28.—Grab sample data (Experiment 6).

6.4.9 Summary of Phase 2 Results

Alamogordo’s water is made up mostly of calcium, magnesium, and sulfate. On an equivalent basis, the sulfate is nearly equal to the sum of the calcium and magnesium. The brackish water, finished water (NF permeate), and EDM concentrate streams’ chemical makeup (Mixed Na and Mixed Cl) is shown using Stiff Diagrams in Figure 29. Stiff Diagrams are one way to visualize the composition of water. Distance to the left of the origin on the left represents the concentration of cations (milliequivalents of solute per litre of solvent [meq/L]), and distance to the right represents concentration of anions. The vertical position has no significance. One can see how the Mixed Na stream is composed primarily

Demonstration of Zero Discharge Desalination (ZDD)

of sodium and sulfate and the Mixed Cl is composed primarily of chloride salts of calcium, magnesium, and sodium in nearly equal concentrations.

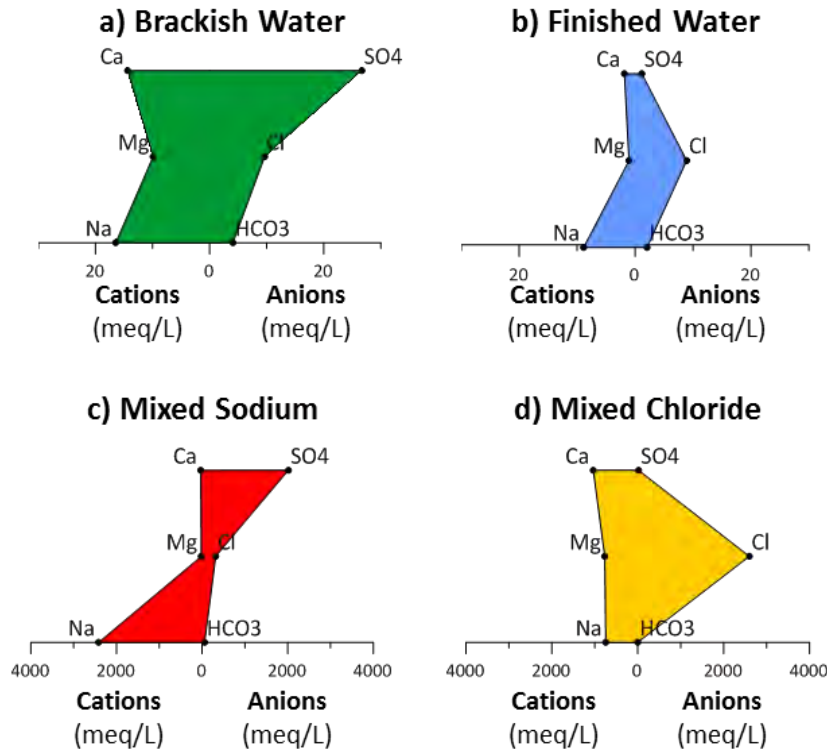


Figure 29.—Average Alamogordo Phase 2 water quality for ZDD inlet (a) and outlet streams (b-d).

The variability in the incoming and exiting streams to the ZDD demonstration system are shown in Figure 30. Since the brackish water feed is a mixture of two wells, some variation is expected in both the brackish feed and NF permeate, although the variation is not large. Also, after nearly 10 quads of membranes were removed, it seemed that there was not enough membrane area in the EDM stacks to achieve the water quality targets, so the BGNDRF Well 1/Well 2 blend was adjusted to achieve a lower conductivity setpoint (around 3,100 $\mu\text{S}/\text{cm}$ instead of around 3,500 $\mu\text{S}/\text{cm}$). There is a fairly wide range of values for the major constituents of the Mixed Na and Mixed Cl streams. The lower values correspond with earlier runs where steady state conditions had not yet been reached. It can take 8 to 12 hours for the Mixed Na and Mixed Cl streams to reach their steady state composition. The high points, especially the outliers, on the Mixed Na graph represent either early runs where the setpoint was not properly controlled or the final days of operation when the conductivity setpoint was increased or eliminated.

Samples were taken nearly every day of operation for all ZDD streams and analyzed at UTEP's water quality laboratory. Several quality assurance and quality control (QA/QC) methods were in place to ensure that results would be

Demonstration of Zero Discharge Desalination (ZDD)

trustworthy. This included using sample blanks, known standards, and rigorous calibration procedures during the analyses steps. After the samples were analyzed, the output was checked for ion balance, and confirmation of TDS by gravimetric and sum of ions, and ion ratios. If one or more of the QA/QC checks failed, a sample was re-analyzed. Less than 10 percent of the remaining samples were excluded from statistical analysis because they failed more than one of the QA/QC checks after being re-analyzed.

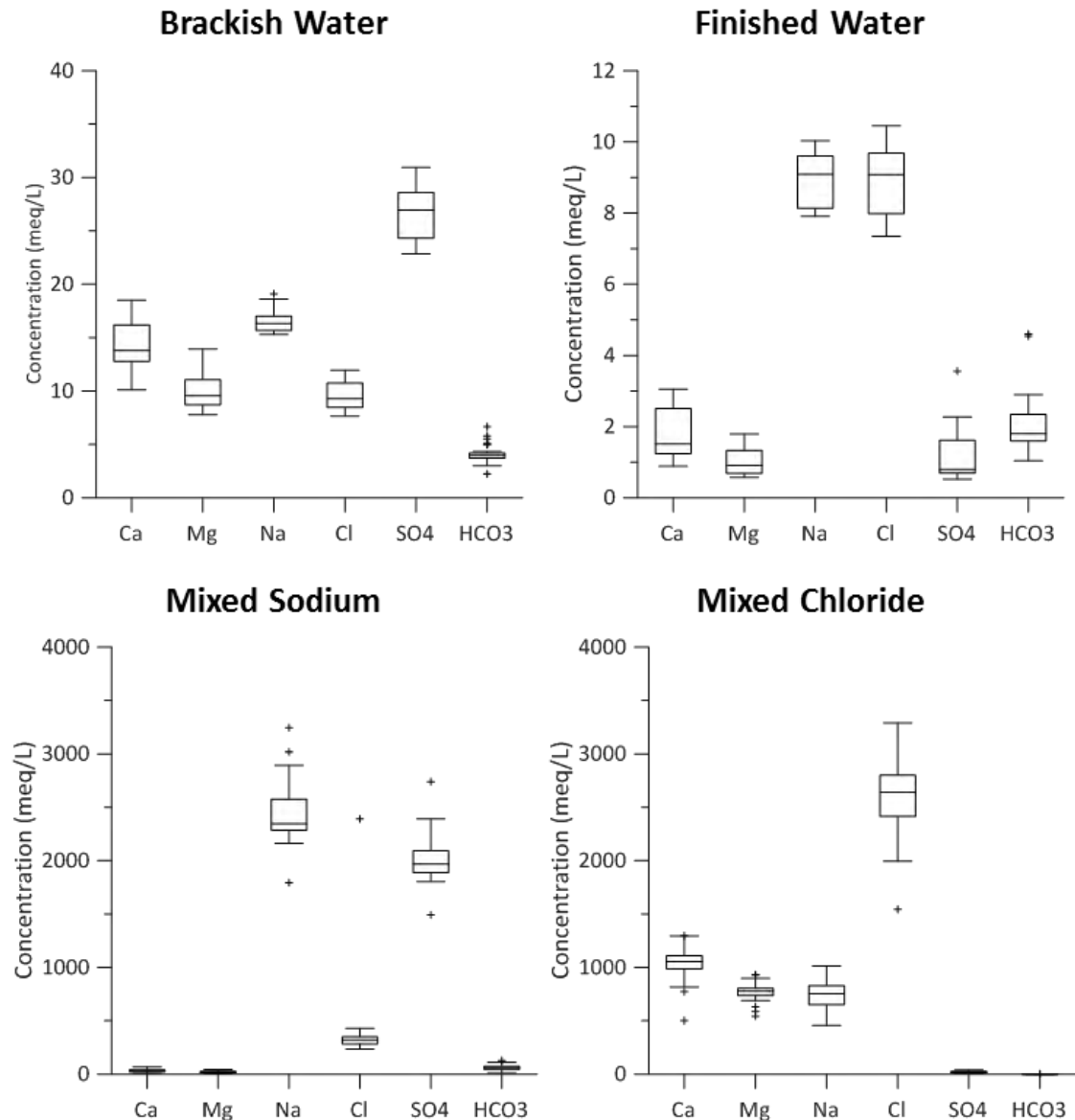


Figure 30.—Variability in Alamogordo Phase 2 Water Quality for ZDD inlet and outlet streams.

Table 8 summarizes the average, standard, and number of samples analyzed for all of the Phase 2 water quality analyses. The NF permeate TDS was, on average, less than the 850 mg/L target. After new membranes were installed and the

Demonstration of Zero Discharge Desalination (ZDD)

brackish water feed TDS decreased in May, the TDS remained below 800 mg/L TDS. It is important to note that the Snake Tank well water has a TDS at or below 2,700 mg/L, so the later setpoint could be viewed as more representative. All sample analyses are provided in Appendix C2.

As mentioned in the experiment summaries, scale was found on several occasions when the EDM stacks were opened. Early on, this scale is thought to have been caused solely by problems with stack construction. After mid-April, scale was typically only found near the center of the stack. The EDM stacks are built in a way that makes them two stacks hydraulically (two sets of inlets and outlets for each stream) and one stack electrically (one set of electrodes). At the center of the stack are two dividers that are intended to prevent mixing from the two sides of the stack but allow ions to flow between the two sides. This configuration is meant to mitigate shunt currents. The center dividers were found to be problematic during both Phase 1 and Phase 2 at Alamogordo. In Figure 31, a NaCl compartment (anode side) and a Mixed Na compartment (cathode side) are on either side of the dividers (shown with gray dashed lines). If the integrity of the dividers is compromised, mixing between the Mixed Na and NaCl streams will occur. This would allow sulfate and bicarbonate to contaminate the Mixed Cl stream by first entering the NaCl stream and then migrating through the AEMs into the Mixed Cl stream. Figure 32 shows the increase in both bicarbonate and sulfate in the NaCl stream over time. Figure 33 shows the increase of bicarbonate and sulfate in the Mixed Cl stream.

The pH of both the Mixed Cl and NaCl streams is typically low (around pH 4.0 or lower), so increased pH levels in these streams can be a proxy for indication of contamination. Figure 34 shows the increase in these streams' pH measurements over time. As the system was being operated, the pH in both the NaCl and Mixed Cl streams increase until the system was shut down. During most of these shutdown periods, the center dividers were replaced. The center dividers were not replaced between June 8 and July 14, and the contamination seems to have been worst during this period of time..

Demonstration of Zero Discharge Desalination (ZDD)

Table 8.—Phase 2 Alamogordo Water Quality Summary (mg/L)

Avg=Average, SD=standard deviation

	TDS*		HCO ₃		Ca		Mg		Na		Cl		SO ₄		SiO ₂	
	Avg (SD)	N	Avg (SD)	N	Avg (SD)	N	Avg (SD)	N	Avg (SD)	N	Avg (SD)	N	Avg (SD)	N	Avg (SD)	N
RO (DI)	364 (188)	32	54 (40.7)	29	2 (2.3)	28	1 (3.4)	28	21 (22.6)	28	22 (33.4)	29	4 (7.4)	29	1.9 (2.8)	32
raw	2,910 (524)	38	248 (47.6)	37	288 (40.1)	35	121 (17.1)	33	379 (21.4)	34	342 (45.5)	34	1,281 (111.8)	34	19.8 (1.6)	39
NF Feed	3,045 (480)	34	147 (41.8)	33	276 (49.3)	32	130 (49.8)	32	402 (23.1)	31	275 (49.2)	32	1,557 (592.2)	31	22.8 (2)	34
NF Perm	767 (262.9)	36	124 (44.6)	37	37 (13.4)	34	13 (4.8)	34	204 (17)	34	316 (33.6)	34	56 (32.9)	35	20.2 (4.2)	38
NF conc	8,344 (676)	33	212 (76.4)	32	674 (183.3)	30	442 (135)	29	940 (71.8)	29	279 (550)	30	4,717 (510.8)	29	25.3 (2.2)	33
EDM feed	4,686 (641.7)	40	106 (45.2)	40	413 (79.1)	37	202 (47)	36	612 (107.3)	36	117 (16.7)	37	2,697 (407.4)	33	26.4 (2.4)	41
EDM dil	3,212 (601.9)	39	76 (40.7)	38	242 (55.9)	36	127 (33.5)	35	502 (120.8)	35	98 (12.8)	36	1,892 (392.6)	33	26 (4.6)	38
Mixed Na	164,776 (15,608)	36	3,493 (1369.2)	32	663 (265.6)	33	266 (118.1)	33	55,941 (6,538)	33	13,487 (12,878)	33	96816 (11,159)	32	8.9 (1.6)	35
Mixed Cl	175,122 (23,222)	36	32 (44.2)	36	20,721 (3,138)	33	9350 (1041)	32	17,130 (2,545)	32	92,100 (12,225)	31	968 (397.7)	32	5.1 (3.5)	35
NaCl	32,007 (2,763)	34	67 (72.4)	34	64 (36.7)	30	34 (18.7)	30	12,249 (1,561)	30	18,317 (2,576)	31	1,000 (739.3)	31	6.5 (2.5)	33
E-rinse	20,245 (2,591)	35	420 (418.3)	32	29 (15.9)	32	5 (4.5)	27	6,526 (7,964)	30	197 (96.7)	32	11,989 (1,899)	31	3.2 (2.9)	35

*TDS: Samples were not filtered before placement in oven.

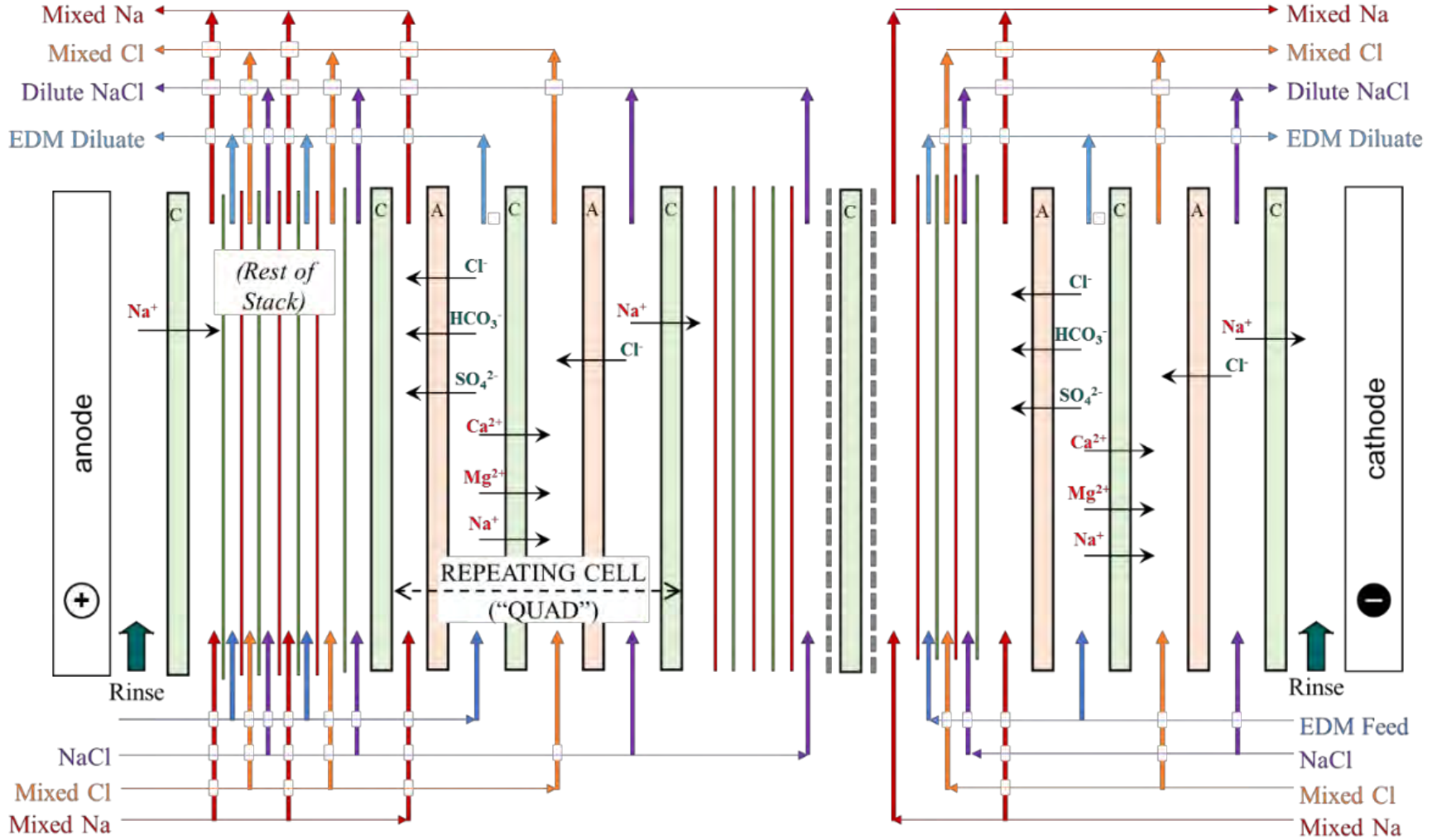


Figure 31.—EDM stack construction at center and electrodes.

Demonstration of Zero Discharge Desalination (ZDD)

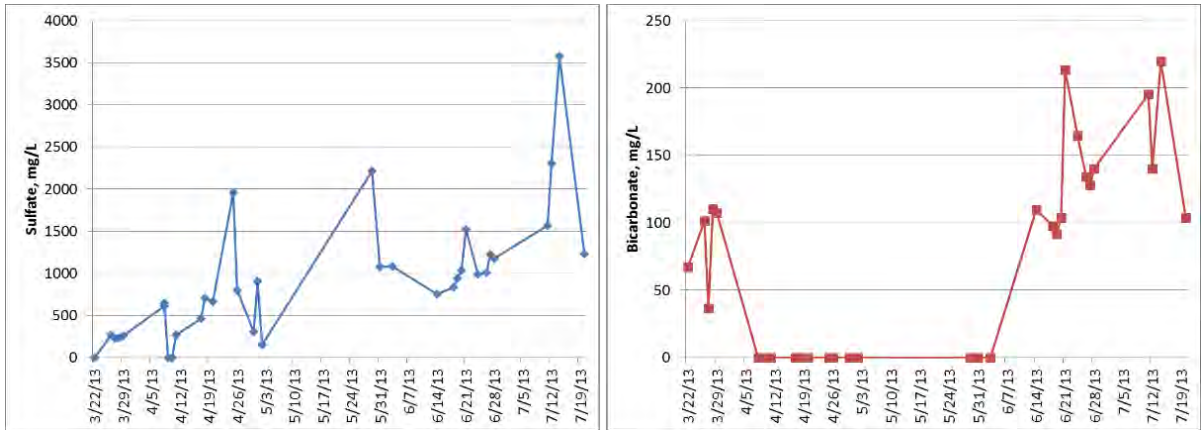


Figure 32.—Bicarbonate and sulfate contamination in NaCl.

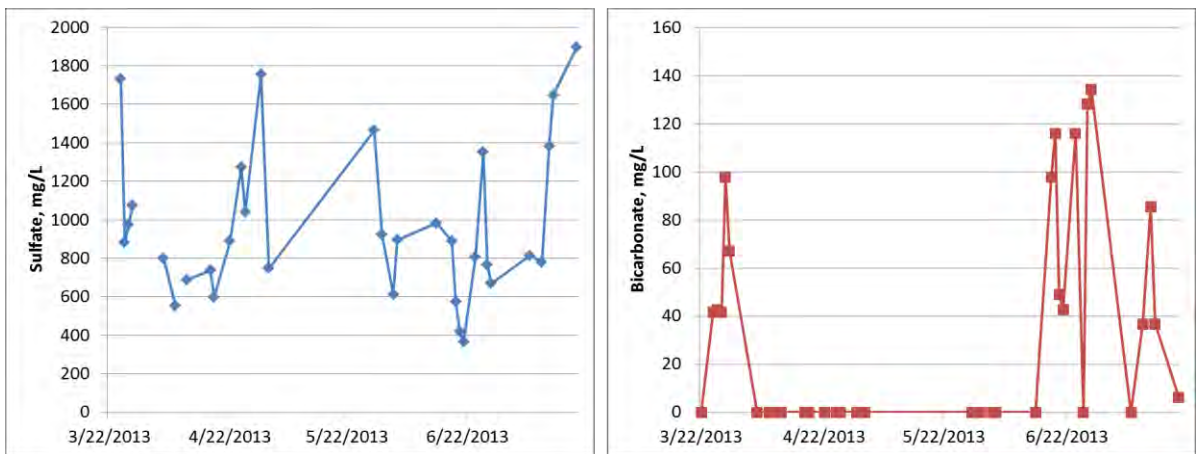


Figure 33.—Bicarbonate and sulfate increase over time - Mixed Cl.

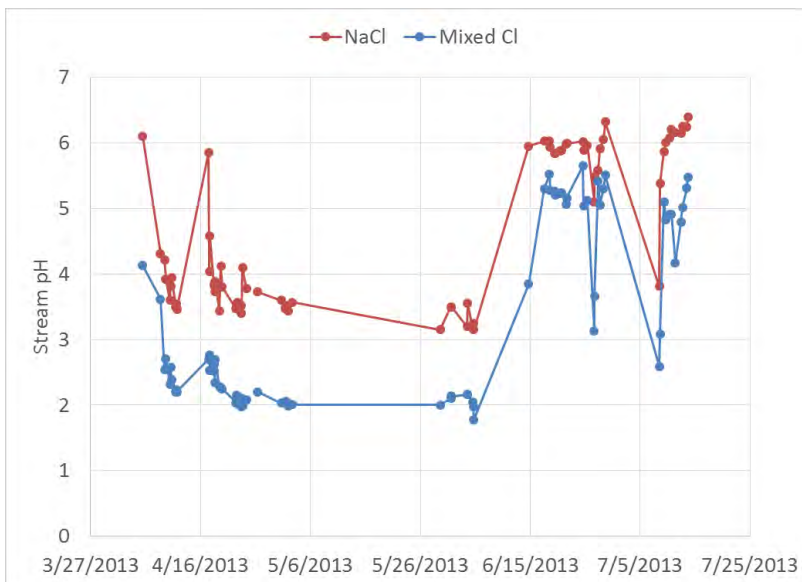


Figure 34.—pH increase over time – Mixed Cl and NaCl.

Demonstration of Zero Discharge Desalination (ZDD)

The anion membranes between the NaCl and Mixed Cl compartments were found to be much thicker than the other anion membranes in the stacks (see Figure 35). This swelling is likely the cause of the decreased flow in the Mixed Cl stream during the final experiment. A portion of the swollen membrane was soaked in a 500 molar NaCl solution and the solution was collected and analyzed by ion chromatography (IC) to see if calcium sulfate was released by the NaCl. NaCl increases the solubility of calcium sulfate, the expected cause of the membrane swelling. The IC analyses of the NaCl solution used to extract precipitate from the membrane are summarized in Table 9.

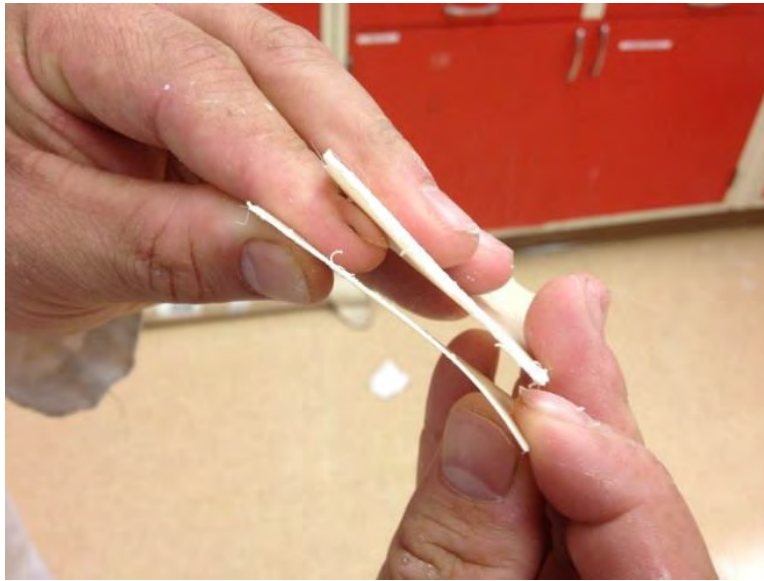


Figure 35.—Comparison of normal and swollen AEMs.

Table 9.—Anion Membrane Soak – NaCl Solution IC Results

	mg/L	meq/L
Ca	1,829	91
Na	65,621	2,853
Cl	93,154	2,662
SO ₄	5,173	108

Two quads of EDM membranes were shipped to MEGA for analysis of scale and fouling of the membranes at the conclusion of the Brighton study. The results are included here for continuity of the discussion regarding NaCl contamination. The project team was interested in determining whether the swollen AEMs were a primary cause of increased EDM stack resistance. MEGA's report is included as Appendix D2. The primary findings were:

- The AEMs in between the Na Cl and Mixed Cl (A1 and A1' in MEGA report) had high resistance that decreased after cleaning with acid.

Demonstration of Zero Discharge Desalination (ZDD)

- Cleaning also improved the AEM between the EDM feed and Mixed Na compartment (A2, A2' in MEGA report)

Analysis of the A1 and A1' membranes by burning the membrane and analyzing the ash for inorganic content. This procedure showed high levels of calcium and sulfate, as well as higher than expected phosphate and silica. MEGA suggests that $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$ (and calcium phosphate $[\text{Ca}_3(\text{PO}_4)_2]$) precipitated inside the membrane:

- MEGA recommended replacing the swollen AEMs.

Analyses by both UTEP and MEGA suggest that calcium sulfate, carbonate, and possibly calcium phosphate are present in or on the anion membranes. A likely mechanism for the swelling of this particular membrane in the quad is that:

1. Sulfate ions that contaminate the NaCl solution migrate through the AEM and encounter calcium ions that have accumulated in the concentrating boundary layers of the Mixed Cl compartments.
2. Because of incomplete Donnan exclusion due to high solution concentration in the Mixed Cl, calcium ions enter the AEM and react with the sulfate ions that are present in high concentration.
3. The precipitate forms a path for more calcium ions to invade the AEM and form more precipitate.
4. Accumulation of precipitate in the membrane causes it to become thicker.

Conductivity is known to have a non-linear relationship with concentration, and the ratio is specific to the composition of the fluid being measured. For this reason, specific gravity measurements were taken regularly to ensure that solubility limits for salts that have a steep decline with decreasing temperature (i.e. sodium sulfate, sodium carbonate, and, to a lesser degree, sodium bicarbonate – see Figure 8). Once a relationship is known (and the proper conductivity probes are installed in the proper location), a conductivity setpoint can be employed for maintaining the desired concentration in the Mixed Na and Mixed Cl streams. Figure 36 shows the relationship and reasonable linear fits for the regime in which the EDM stack was operated in Phase 2.

Demonstration of Zero Discharge Desalination (ZDD)

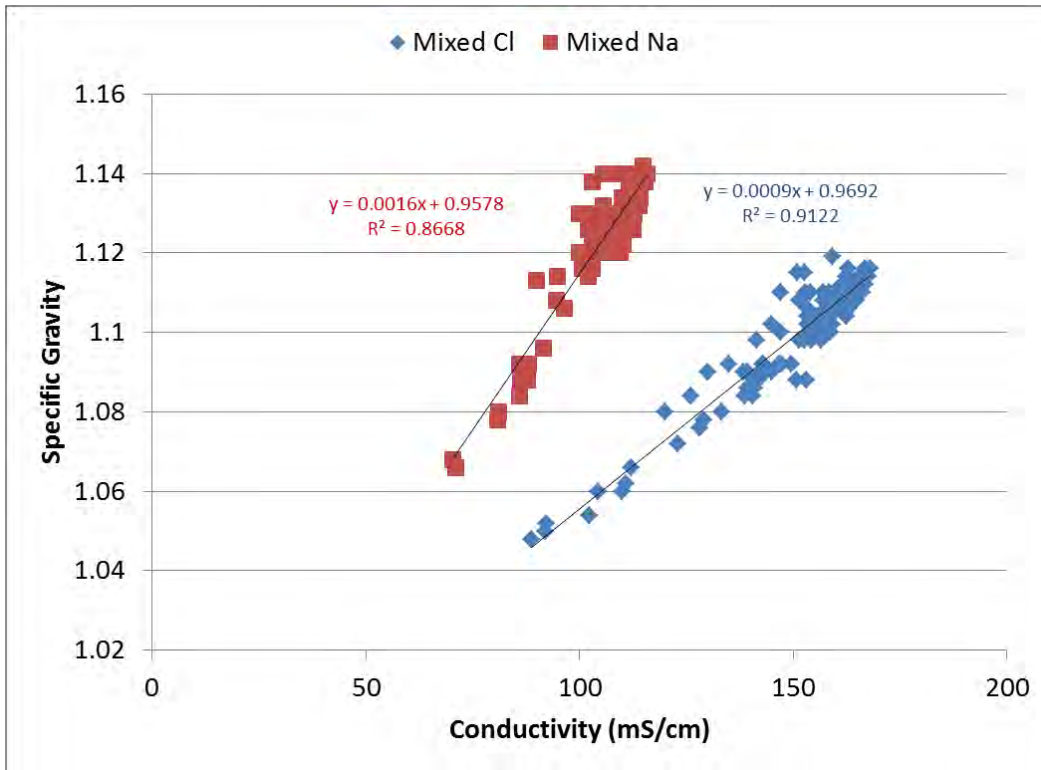


Figure 36.—Relationship between conductivity and specific gravity (Alamogordo – Phase 2) .

Demonstration of Zero Discharge Desalination (ZDD)

Table 10.—Summary of Alamogordo Phase 2 Operational Data

Experiment		0	1a	1b	2	3	4	6	
Dates		Bef. 4/22	4/22-4/24	4/25-4/26	4/30-5/2	5/29-6/5	6/17-6/21	6/22-6/28	
# EDM stacks		2	2	1	1	1	2	2	
		Average (Range)	Average (Range)	Average (Range)	Average (Range)	Average (Range)	Average (Range)	Average (Range)	
NF Feed Pressure	gpm	103.7	106.7	63.3	68.5	60.5	87.6		
		(98-109)	(103-112)	(62-64)	(65-70)	(59-62)	(82-99)	(98-109)	
NF Permeate flow	psi	40.2	40.3	21.2	21.6	22.3	36.8		
		(38.7-43.2)	(39.1-42.2)	(20.4-21.9)	(20.1-22.3)	(21-25.6)	(34.5-40.6)	(38.7-43.2)	
NF Pressure Drop	psi	37.5	37.0	18.7	33.8	18.5	33.1		
		(36-41)	(36-39)	(18-19)	(19-62)	(18-20)	(30-42)	(36-41)	
EDM Feed flow	gpm	56.6	55.1	26.7	26.4	31.8	56.4		
		(51.2-60.4)	(51.1-59.9)	(25.3-27.6)	(24.4-27.8)	(30.2-33.5)	(54.4-59.6)	(51.2-60.4)	
EDM Feed pressure	psi	17.8	15.6	11.7	13.1	24.6	18.6		
		(13.9-21.5)	(15.1-18)	(11.4-12)	(12.3-13.7)	(22.4-27)	(18.1-19.1)	(13.9-21.5)	
Mixed Cl flow	gpm	66.4	63.2	32.0	31.9	37.5	69.9		
		(57.5-76.3)	(57.2-68.2)	(30.9-33)	(31-32.7)	(35.3-40.2)	(67.2-75.3)	(57.5-76.3)	
Mixed Cl pressure	psi	18.1	15.9	11.7	13.5	25.0	18.9		
		(13.9-21.8)	(15-18.5)	(11.4-12.1)	(12.6-14.2)	(22.6-27.9)	(18-19.4)	(13.9-21.8)	
Mixed Na flow	gpm	68.8	64.4	31.3	31.2	35.8	73.4		
		(43.4-76)	(61-68)	(30-33)	(30-34)	(28-42)	(69-77)	(43.4-76)	
Mixed Na pressure	psi	18.1	15.7	12.0	13.4	24.8	18.9		
		(13.9-21.9)	(15.1-18.3)	(11.9-12.1)	(12.3-14.3)	(22.8-26.8)	(17.9-19.7)	(13.9-21.9)	
Mixed NaCl flow	gpm	76.4	71.9	31.0	33.7	42.0	74.1		
		(66-84)	(65-76)	(30-32)	(33-35)	(40-45)	(64-77)	(66-84)	
Mixed NaCl pressure	psi	18.1	15.6	11.8	13.5	24.9	18.7		
		(14-21.8)	(15-18.4)	(11.6-12.1)	(12.6-13.9)	(22.7-27.2)	(18.1-19.5)	(14-21.8)	
E-Rinse Flow	gpm	7.5	<i>Flow meter not functional</i>					11.6	
		(0-12.8)						(11-12.1)	(0-12.8)
E-Rinse Pressure	psi	11.8	11.1	12.2	13.6	20.2	12.8		
		(10-13.4)	(10.7-11.8)	(12-12.4)	(12.8-14.3)	(19.3-20.8)	(12-13.7)	(10-13.4)	
EDM-1 voltage	volts	160.0	163.0	156.0	154.0	142.0	124		
		(150-175)	(155-170)	(156-156)	(150-160)	(124-150)	(111-137)	(150-175)	
EDM-1 current	amps	61.0	62.0	52.0	54.0	50.0	48		
		(36-68)	(45-67)	(39-57)	(47-59)	(34-58)	(43-53)	(36-68)	
EDM-2 voltage	volts	151.0	160.0				122		
		(140-165)	(155-166)				(111-130)	(140-165)	
EDM-2 current	amps	49.0	48.0				49		
		(31-61)	(37-51)				(42-53)	(31-61)	
NaCl Consumption (est)	lb/day	Unreliable	1072	491	521	422	868		
NF Recovery	%	58.6	58.0	59.2	61.5	58.4	56.6		
		(57.3-60.1)	(57.2-58.6)	(58.6-59.6)	(59.7-64.1)	(57-61.7)	(55.2-58.7)	(57.3-60.1)	

Demonstration of Zero Discharge Desalination (ZDD)

Experiment		0	1a	1b	2	3	4	6
Dates		Bef. 4/22	4/22-4/24	4/25-4/26	4/30-5/2	5/29-6/5	6/17-6/21	6/28-7/2
# EDM stacks		2	2	1	1	1	2	2
		Average (Range)	Average (Range)	Average (Range)	Average (Range)	Average (Range)	Average (Range)	Average (Range)
NF rejection	%	60.5	60.4	60.4	57.4	63.9	64.9	64.9
		(57.6-61.8)	(58.3-62.1)	(58.3-62.1)	(56.1-58.5)	(63.2-64.7)	(63.3-66.5)	(63.3-66.5)
EDM removal- per pass	%	28.1	26.9	26.9	28.5	24.5	24.2	24.2
		(25.5-32.3)	(24.1-31.4)	(24.1-31.4)	(25.4-35.5)	(19.8-29.5)	(19.7-26.9)	(19.7-26.9)
EDM removal-overall	%	55.6	52.4	52.4	54.5	48.4	50.9	50.9
		(51.8-61.1)	(48.4-55.1)	(48.4-55.1)	(50.9-57.9)	(41.6-56.3)	(43.7-54.1)	(43.7-54.1)
ZDD removal-overall	%	61.0	59.8	59.8	57.6	61.8	64	64
		(59.2-62.7)	(58.5-60.7)	(58.5-60.7)	(56.2-60.1)	(60.4-63.9)	(63-64.5)	(63-64.5)
ZDD Recovery	%	96.6	96.3	97.2	97.1	97.5	97.7	97.7
		(95.8-97.7)	(96.1-96.5)	(97-97.5)	(96.7-97.5)	(97.5-97.9)	(97.2-98.2)	(97.2-98.2)
Longest continuous Run	Date	4/19-4/20	4/22-4/24	4/25-4/26	4/30-5/2	6/4-5	6/18-21	6/28-7/2
	Hours	24.4	49.3	23.4	58.1	20.8	79.1	79.1

*The combined total run for 7/8-14 was 137 hours (excludes 8 hours of shutdowns). There was a 7-hour shut down on 7/11 to CIP the EDM feed stream and a 1-hour shutdown on 7/12 to install new cartridge filters for the NF feed and fix a leak on the E-Rinse.

6.3 Evaluation of Current Density, Current Efficiency, Power Consumption – Alamogordo

6.3.1 Current Density

Operating beyond the limiting current density is inefficient and can lead to damage of membranes and equipment. A set of experiments was run in early April in an attempt to identify the limiting current density. Voltage and current data were gathered at three cell velocities in a single EDM stack. Unfortunately, conductivity data was not gathered at this time.

Figure 37 shows the current density versus voltage plots. The operational limiting current density is the intersection between the ohmic (blue line/dots) and plateau (red line/dots) lines and was calculated algebraically from the data gathered. A relationship between velocity and limiting current density was obtained so that the limiting current density could be calculated for the duration of the Phase 2 operations.

At the beginning of the Phase 2 operations, the operators thought that they had two 120-quad EDM stacks. This led to operating above the operational limiting current density (shown with blue dashed line in Figure 38). When operating beyond the limiting current density, ions are depleted from the boundary layer near the ion exchange membranes, and the only way to support increasing current is for water to split into hydrogen and hydroxyl ions. This causes a decreased pH in the Mixed Cl stream from H^+ ions and an increased pH in the Mixed Na stream from OH^- ions. The difference between the pH measurements in these two streams can be an indicator of water splitting. In Figure 38, this effect is visible in the earlier runs. The later runs were operated closer to the limiting current density, but in many cases still above the setpoint. While the pH difference does not show the effect of over-current, it is likely explained by the NaCl and Mixed Cl contamination described earlier (presence of bicarbonate in Mixed Cl stream buffered the pH so the difference was lower than expected).

Demonstration of Zero Discharge Desalination (ZDD)

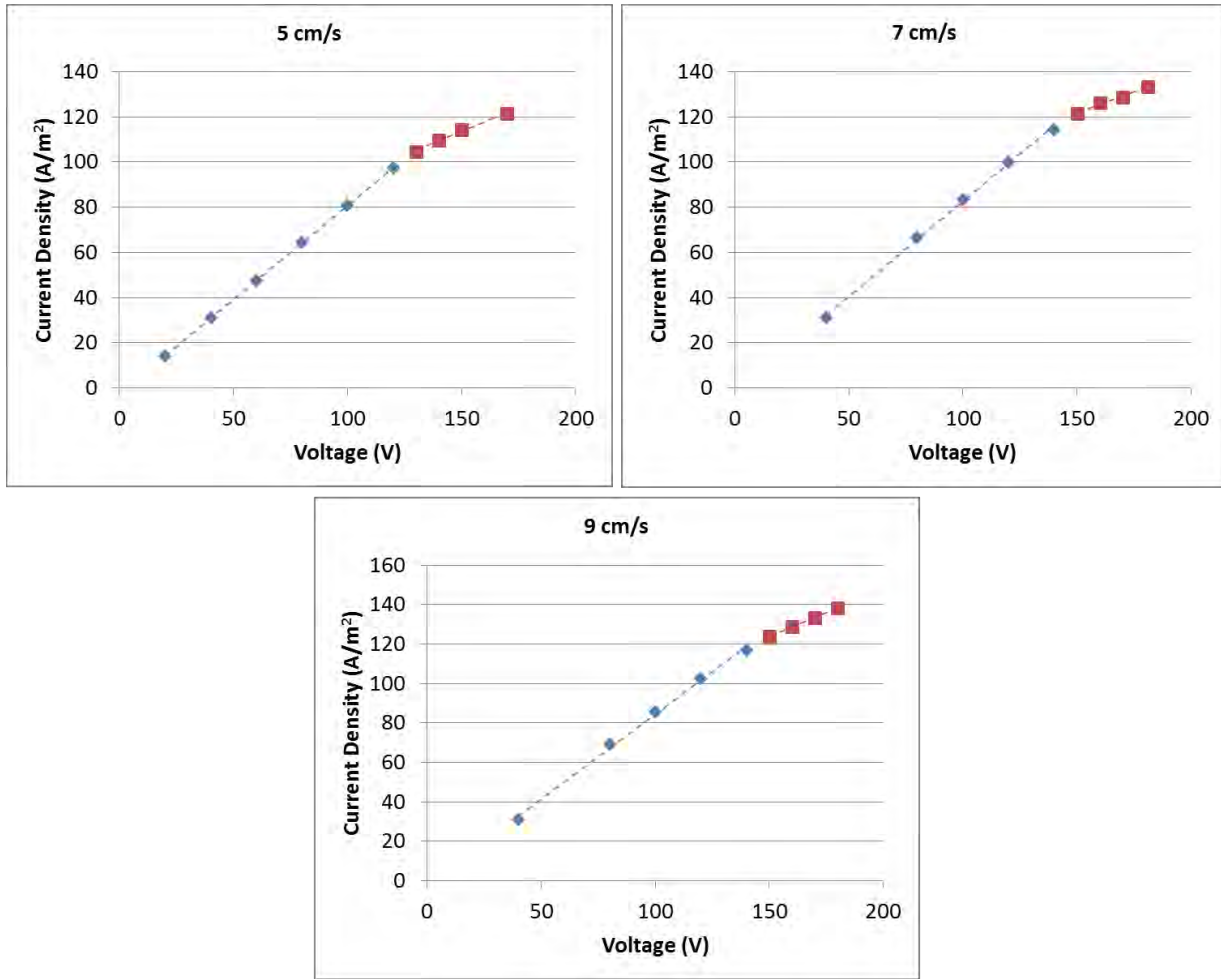


Figure 37.—Limiting Current Density – Alamogordo Phase 2.

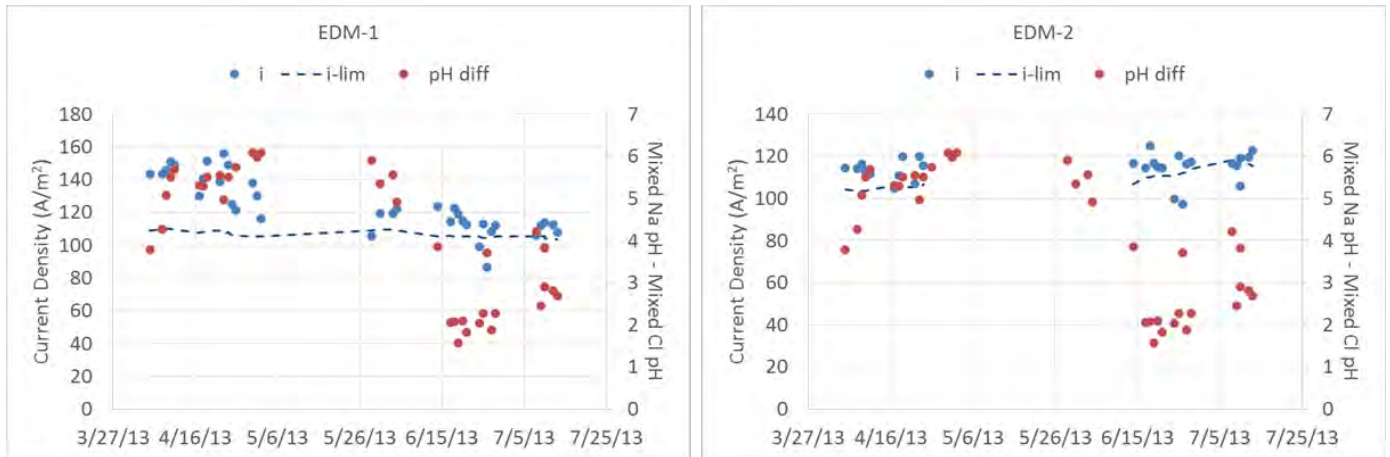


Figure 38.—Limiting current density and effects on EDM concentrate stream pH (Alamogordo Phase 2).

6.3.2 EDM Stack Performance Monitoring

One way to monitor EDM performance in the field is to track the EDM stack resistance (applied voltage divided by current). When stack resistance increases, it can be an indicator of membrane fouling or other problems within the stack. Increasing stack resistance can also be used to understand changes in diluate quality and temperature. The EDM feed-diluate compartment has the lowest concentration, so will be the determining factor for stack resistance: lower concentrations will increase the resistance. Also, there is increased resistance with reduced temperatures (see Figure 7).

Stack resistance did increase throughout some of the runs during Phase 2. However, the cause did not seem to be low diluate conductivity or low temperature. As described in Experiments 2, 5, and 6, iron particles were present in the NF concentrate and EDM feed streams. The source of iron was initially unknown and is thought to be the cause of the increased EDM stack resistance. When the particles were visible, the resistance seemed to increase. Several CIPs were performed and new 1-micron cartridge filters were eventually installed on the NF feed. This lessened the resistance increase (see data inside box in Figure 39).

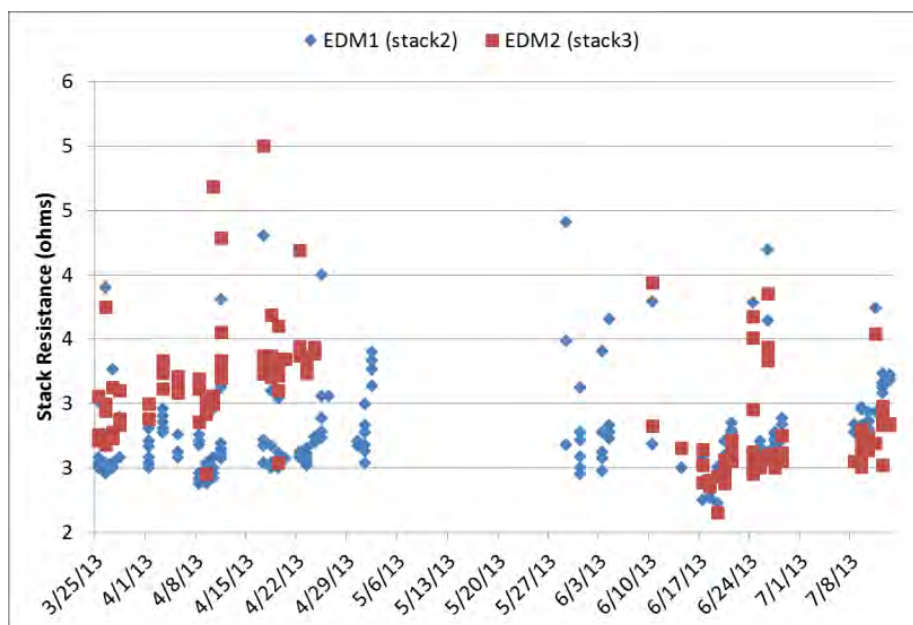


Figure 39.—EDM stack resistance – Phase 2.

The final reasons for reduced performance, as measured by ion removal in the EDM, could be explained by internal leakage, back diffusion of ions from the EDM concentrate streams to the EDM feed, and inadequate membrane area. Internal leaks were present throughout the study and are thought to be an important cause of scale formation at the center of the stack and reduced performance. Since nearly 10 quads were removed in each stack by the end of the

Demonstration of Zero Discharge Desalination (ZDD)

study, reduced performance was also likely caused by a combination of inadequate area and operator misunderstanding.

The predicted membrane area and quads needed for desired EDM performance can be calculated using the following equation,

$$EDM \text{ Area } (m^2) = \frac{Q_{feed} \cdot (C_{feed} - C_{diluate}) \cdot F}{i \cdot \eta}$$

Where:

Q_{feed} = EDM feed flow rate, L/s

C_{feed} , $C_{diluate}$ = Concentration of EDM feed and diluate, equivalent per liter (eq/L)

F = Faraday's Constant, 96,485 A-s/eq

i = Current density, amperes per square meter (A/m^2)

η = Current efficiency, %

Current efficiency was assumed to be 80 percent in the design by Veolia; however, it is good to calculate the actual value for a full scale design. The current efficiency can be calculated in several ways, but is essentially is a ratio of the charge (calculated from concentration) that leaves or enter a stream in the EDM. Each method is summarized. The first method is a ratio of the ions that are removed from the EDM feed to the current supplied:

$$\eta_{feed} = \frac{F \cdot Q_{feed} \cdot (C_{feed} - C_{diluate})}{A \cdot N_{quads}}$$

Where:

A = Measured EDM Stack current, amps

N_{quads} = Number of quads in EDM stack

The next method for estimating current efficiency is a ratio of the ions supplied by the NaCl stream to the current supplied:

$$\eta_{NaCl} = \frac{F \cdot M_{NaCl}}{A \cdot N_{quads}}$$

Where M_{NaCl} is the flow of NaCl into the EDM stack, eq/sec. This is calculated from the daily NaCl consumption.

Demonstration of Zero Discharge Desalination (ZDD)

The final method for estimating the current efficiency involves a ratio of the ions in the EDM concentrate overflow streams to the current supplied:

$$n_{M.Na \text{ or } M.Cl} = \frac{F \cdot Q_{OF} \cdot C_{M.Na \text{ or } M.Cl}}{A \cdot N_{quads}}$$

Where:

Q_{OF} is the waste flow rate of the Mixed Na or Mixed Cl stream, L/sec

$C_{M,Na \text{ or } M.Cl}$ is the concentrate of the Mixed Na or Mixed Cl stream, eq/L

Each of these methods is limited by the quality of the data gathered in the field. The NaCl consumption and EDM concentrate overflow rates were highly variable and open to operator errors during measurement. However, enough data was gathered to compare the various methods. The average current efficiency values for the Phase 2 operations in Alamogordo are summarized in Table 11 and Figure 40. The average current efficiency using these four methods is 80 percent and is in line with Veolia's design.

Table 11.—Average Current Efficiency (Alamogordo-Phase 2)

Method	Average Current Efficiency
EDM Feed	72%
NaCl	84%
Mixed Cl	76%
Mixed Na	90%
Average	80%

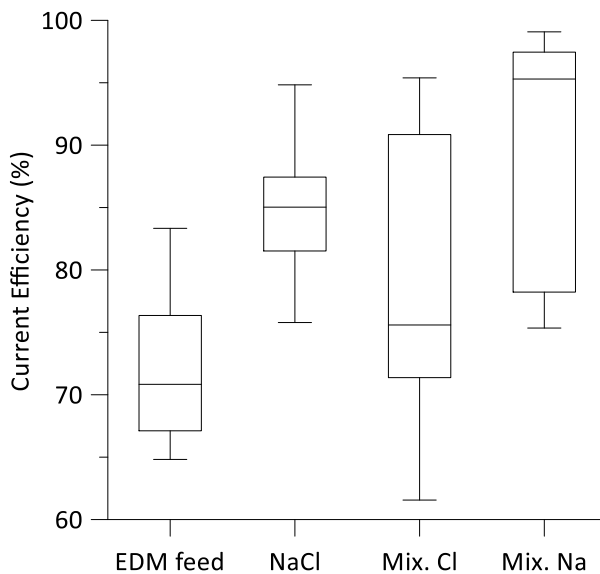


Figure 40.—Alamogordo current efficiency (Phase 2).

6.3.3 Energy Consumption

The goal of Phase 2 was to operate with fixed operating setpoints for long durations. This would allow for comparison to Phase 1 and provide information for sizing the full scale system for Alamogordo’s desalination plant. The EDM specific energy was estimated for all logsheet data points for Phase 2 using the same assumptions as made in Phase 1. Figure 41 compares the EDM stack resistance for both stacks with the estimated specific energy. As mentioned in earlier sections, operators applied more voltage than necessary; this is reflected in the higher specific energy (data in April and May). The recycle ratio was slightly lower in the earlier runs as well (0.91 vs 1.02). The combination of applying lower voltage and increasing the recycle ratio led to lower specific energy consumption by the EDM. The average specific energy for the early and late runs is comparable to what was observed during 1 activities (see Figure 42).

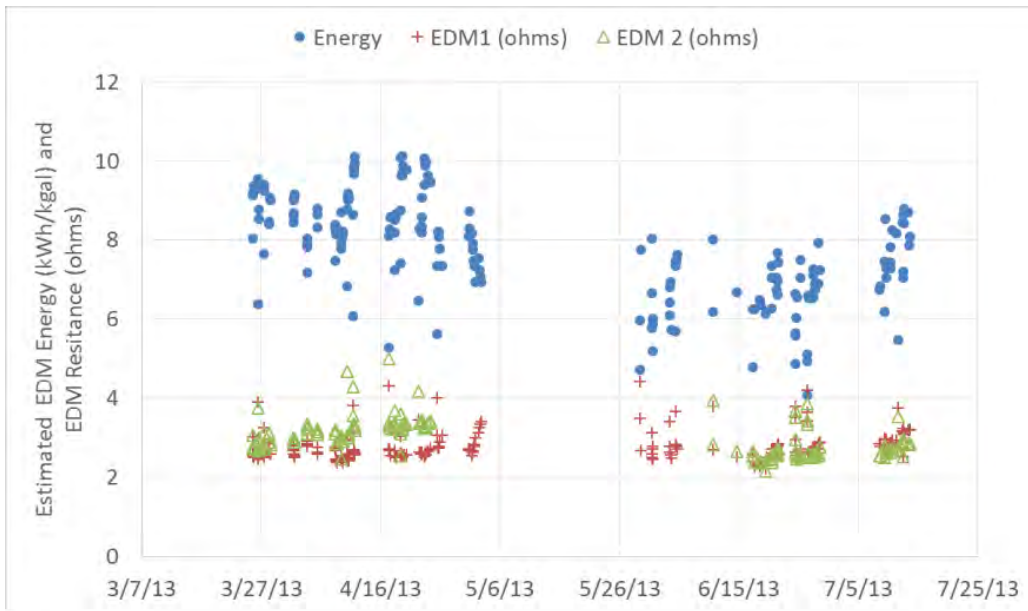


Figure 41.—EDM stack resistance and estimated specific energy (Phase 2).

Demonstration of Zero Discharge Desalination (ZDD)

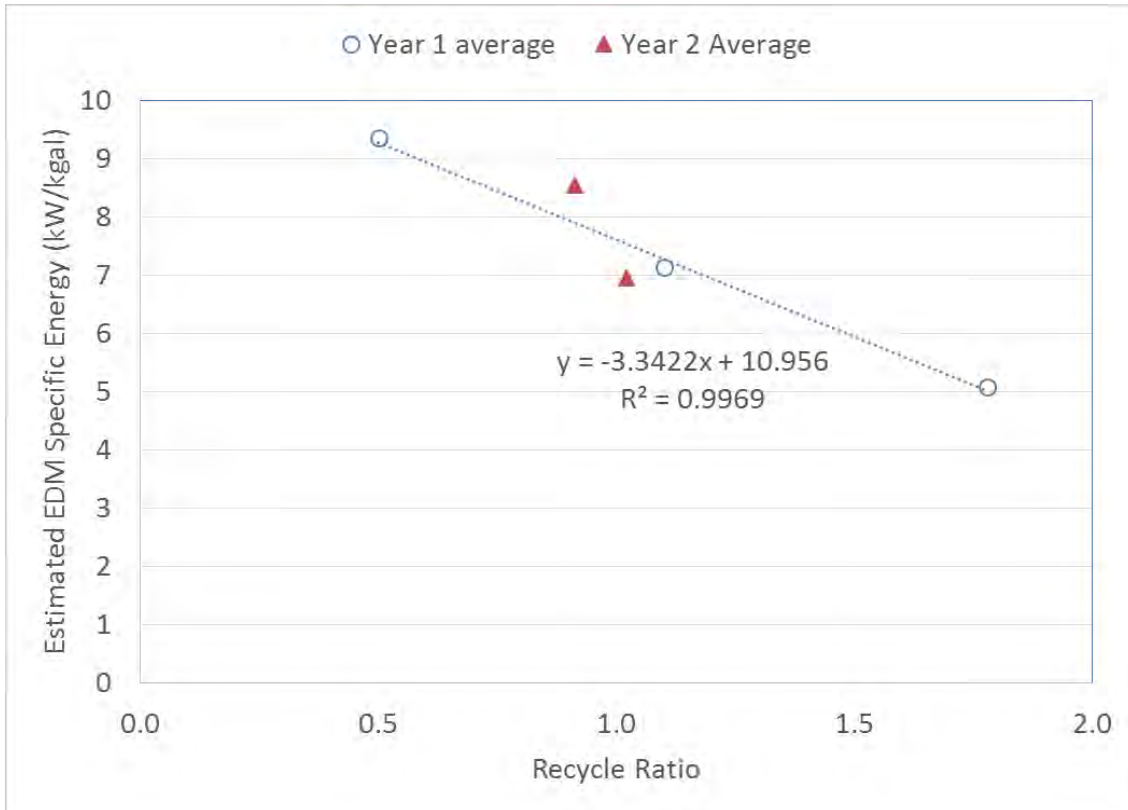


Figure 42.—Comparison of Phase 1 and Phase 2 specific energy.

Data from the Phase 2 demonstration activities were used to design a full-scale ZDD system for Alamogordo. Slight changes from what was used and observed during the pilot were made to in some cases and are summarized in Table 12 (middle column) and in Section 4.2.

Table 12.—Comparison of Full Scale and Pilot Demonstration Operating Setpoints

	Phase 2	Full Scale (4 MGD) Design
NF/RO Staging	4x4 (NF270-400)	Single Stage with 16 PVs (1 st three elements: XLE-440, last five: NF270-400)
NF/RO Recovery	55%	60%
NF Feed Pressure (psi)	90	91
EDM feed velocity (cm/s)	5-9	6-7
EDM Recycle Ratio	0.8-1.1	1.2
#Quads/EDM stack	100	120
EDM voltage/current	135-150 volts 40-50 amps	135-140 volts 44-46 amps

Demonstration of Zero Discharge Desalination (ZDD)

	Phase 2	Full Scale (4 MGD) Design
EDM current density (A/m ²)	10.5-11.9	10.9
EDM TDS reduction	45%	50%
Permeate TDS (mg/L)	<800	<400

The data in Table 12, along with flow rates and other design information, were used to develop a cost model. A key piece of this model includes the energy consumption for each of the major equipment systems. Table 13 summarizes the output from the power analysis. The calculations are made assuming the plant is operational 365 days per year and produces 4 MGD of product water (includes 570 gpm bypass blend) with TDS less than 800 mg/L at 98 percent recovery. The EDM specific energy is based on operating at the limiting current density for 6.5 cm/s velocity, 46 amps. An equation was fit to the same data obtained in the limiting current density experiment and was used to determine the voltage, 140 volts.

Table 13.—Alamogordo Estimated Specific Energy (4 MGD Design)

Process	kWh/kgal	% of total
NF energy	1.93	22.3%
EDM energy	6.61	76.4%
Other energy (e.g. process water, chemical pumps)	0.11	1.3%
Total energy	8.66	

7. La Junta Results and Discussion

As part of an effort to evaluate ZLD options available for Colorado utilities with existing or planned desalination facilities, HDR, Inc. led a pilot testing program that evaluated ZDD in the Colorado cities of La Junta and Brighton. A thorough report was written and co-published by WERF and IWA (Brandhuber et al., 2014), so only a brief description of the pilot and differences in (or additional) interpretation are described in this report. The WERF/IWA report is labeled WERF 5T10 in this report. Figure 43 shows a) NF container installed outside the La Junta Plant and b) EDM Container in background and NaCl Recovery Equipment in foreground, both installed inside the La Junta Plant

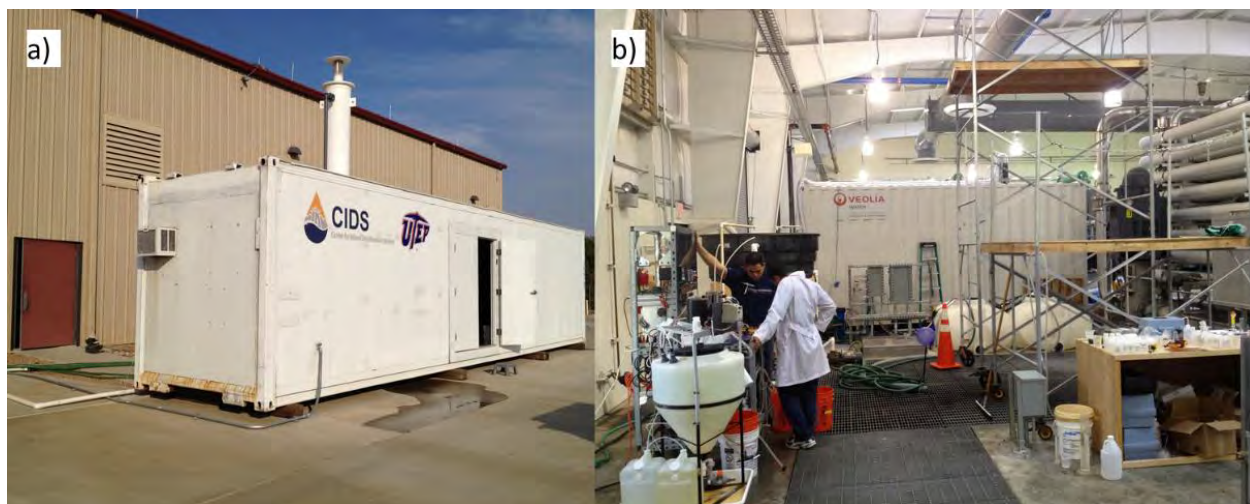


Figure 43.—ZDD and NaCl recovery systems installed in La Junta (20-gpm).

7.1 Equipment Description

The 20-gpm ZDD pilot system used in Alamogordo was shipped from Alamogordo to La Junta by Veolia and site work was performed by a combination of contractors, Veolia operators, and others. A new Veolia operator was trained and the system was modified during the summer of 2012. Four sets of experiments were performed in the fall of 2012 using the configuration in Figure 44 and a bench scale NaCl recovery system was tested in September 2012.

The NF staging and membranes were changed for the La Junta pilot. A 3x2x1(4M) array was chosen with NF90-4040 membranes in the first two stages (24 membrane elements) and the final stage had NF270-4040 membranes. The NF-90 membranes were chosen for their good salt rejection and low pressure (as compared to typical brackish water RO membranes). The NF270 membranes were chosen for their low silica rejection. Similar to Alamogordo, there was enough silica present in the brackish feed water that a strategy was required for mitigating scale formation in the ZDD system. While the La Junta brackish water feed has a lower salinity than what was treated in Alamogordo, the water quality

Demonstration of Zero Discharge Desalination (ZDD)

target was more stringent; the project set a target of 100 mg/L TDS (Brandhuber et al., 2014), which is well below the secondary maximum contaminant level (SMCL) of 500 mg/L TDS. NF270 membranes could produce good quality drinking water, likely below the SMCL (see Table 14). A design with only NF270 membranes would have the lowest feed pressure requirements and would also allow for the most protection against silica scale formation. Hydrex 4101 antiscalant was dosed at 2-3 mg/L to the NF system. Acid was added to the Mixed Na stream to prevent precipitation by sodium bicarbonate and carbonate species.

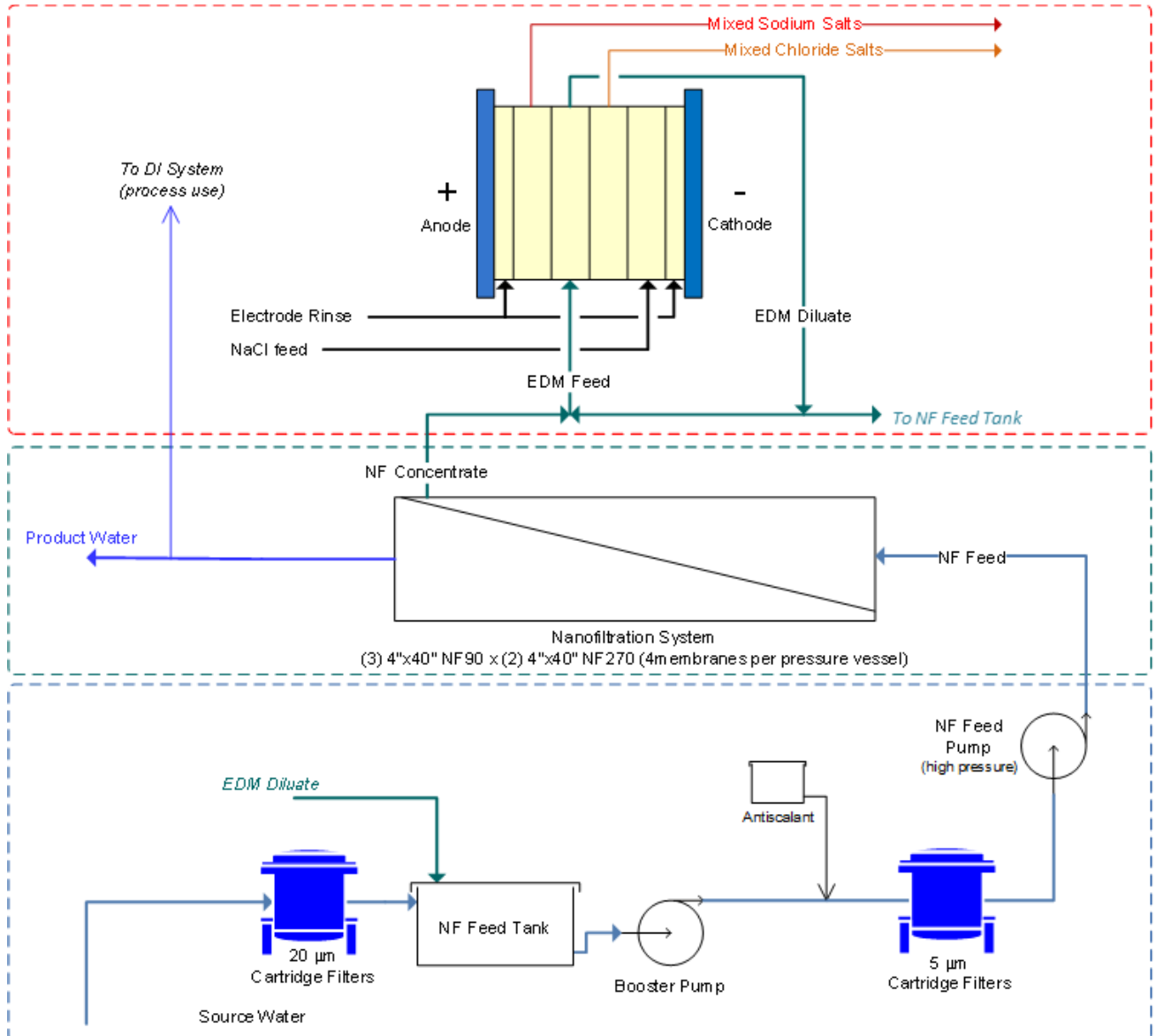


Figure 44.—Equipment flow diagram (La Junta).

Demonstration of Zero Discharge Desalination (ZDD)

Table 14.—Comparison of RO and NF Membranes for La Junta Desalination

Membrane	BW30	NF90/ NF270	NF270
Configuration	3x2x1 (4M)	3x2x1 (4M)	3x2x1 (4M)
Average Permeate Flux (gfd)	15.38	14.63	14.63
Feed Pressure (psi)	173.5	91.1	83.2
Permeate Back Pressure (psi)	0	0	15
Recovery	60%	60%	60%
RO Energy (kWh/kgal)	2.62	1.38	1.26
Permeate TDS (mg/L)	7.7	99.3	356.3

7.2 Summary of Pilot Testing

The ZDD system recovery calculation in the WERF 5T10 report used a different method than what is used by UTEP. The WERF 5T10 equation used was:

$$Recovery(WERF\ 5T10) = 100\% \cdot \left(1 - \frac{Waste\ Flow}{Feed\ Flow}\right)$$

While this is a correct formula, there are several challenges specific to the ZDD pilot system tested at La Junta that make this formula less accurate. First, the feed flow was calculated using a totalizer reading from a flow meter that showed 1 to 2 gpm flow when there was zero flow. Also, the method used in the WERF 5T10 report estimated the feed flow rate assuming the totalizer value represented a full 24-hour day's flow. The combination of these factors leads to the estimated feed flow being less than the NF permeate flow (see Figure 45).

The next problem with the WERF 5T10 recovery calculation was that all DI water consumption was considered an additional waste flow; this double-counts the DI water usage. While understanding the DI water consumption was important for sizing of the DI water system, virtually all DI water used in the ZDD system was accounted for in the EDM waste flow streams. Water was transported along with ions from the NaCl and EDM feed streams into the EDM's Mixed Na and Mixed Cl streams called water of hydration. Therefore, any DI water used to prepare the NaCl solution would be in the recirculating stream or in the EDM waste streams (evaporation is considered to be negligible). Any DI water added to dilute the Mixed Na and Mixed Cl streams would also be present in the streams' overflow (waste) streams. These waste streams would have periods of relatively high flow when DI water was being added and periods of relatively low flow when DI wasn't being added. To mitigate double-counting, the operators in La Junta measured the overflow rates during relatively low flow periods when DI wasn't being added; however, a portion of this overflow was still from DI water added to the system and so a portion of the DI water added is still double-counted. The

Demonstration of Zero Discharge Desalination (ZDD)

overflow measurements were taken once per day over a one minute period, which may not have been adequate to fully estimate the actual waste flow variation.

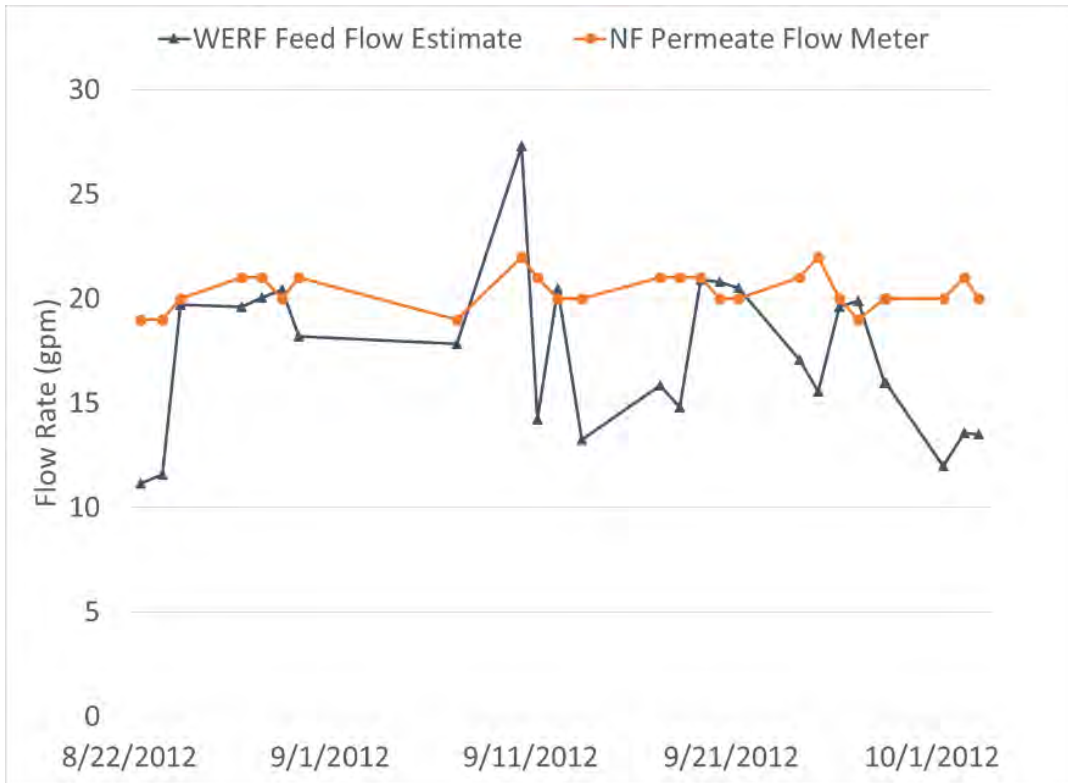


Figure 45.—Comparison of feed flow estimates and NF permeate flow.

The preferred ZDD recovery calculation method used by UTEP for reporting recovery is based on the NF Permeate, Mixed Na, and Mixed Cl flows. The feed flow can be estimated by summing these flows. However, to use this method a longer collection time (at least one hour) is needed for capturing the Mixed Na and Mixed Cl flows so that each stream's waste flow is properly calculated. A longer collection period will allow for variations in dilution volume and changes in the ion removal (and water transfer) rate. The ZDD flow diagram and available flow meters or flow measuring points are shown in Figure 46.

Demonstration of Zero Discharge Desalination (ZDD)

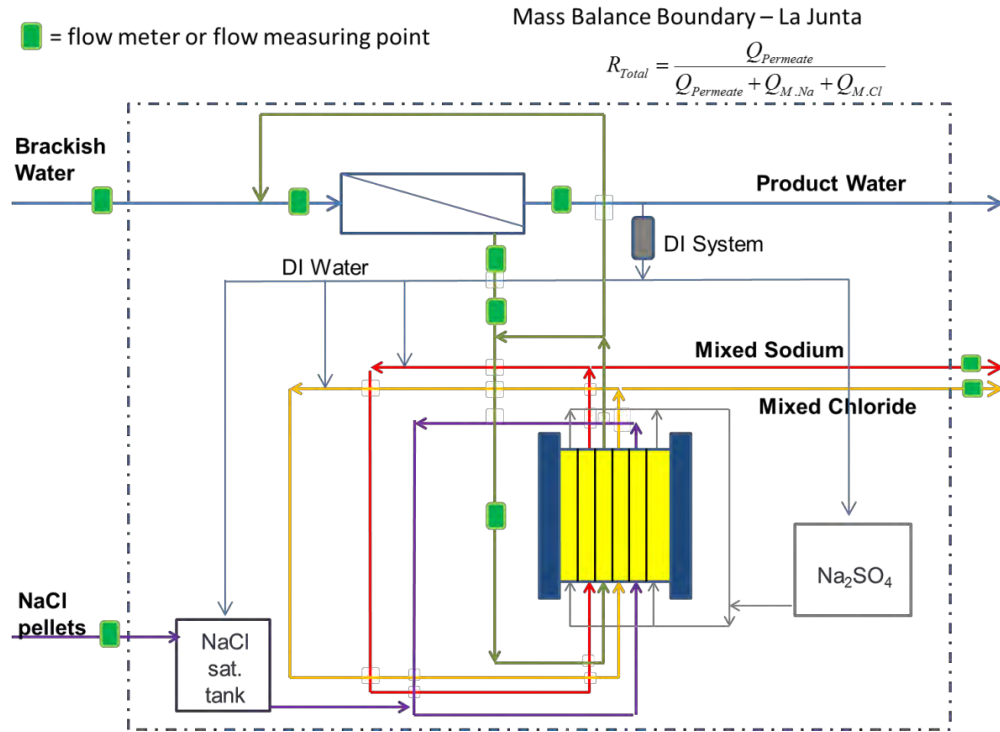


Figure 46.—ZDD flow diagram showing mass balance measuring points.

Since the measured overflow rates are likely underestimated, alternative methods for calculating ZDD recovery are summarized in the following equations. The first equation estimates the feed flow as the sum of the NF permeate flow (flow meter reading), average DI water use rate, and calculated Mixed Na and Mixed Cl overflow rates. The NF permeate, calculated DI, Mixed Na, and Mixed Cl flow values were taken directly from the logsheet data provided by HDR, Inc.

$$Rec1 = \frac{NF\ Permeate\ Flow}{NF\ Perm.\ Flow + DI + Mixed\ Cl,\ Na\ Waste\ Flows}$$

The second method eliminates the DI usage from the waste flows:

$$Rec2 = \frac{NF\ Permeate\ Flow}{NF\ Perm.\ Flow + Mixed\ Cl,\ Na\ Waste\ Flows}$$

The proportion of the EDM waste flows is expected to be predominantly from DI dilutions, so a third method ignores the overflow rates and solely uses the DI consumption for the waste flow:

$$Rec3 = \frac{NF\ Permeate\ Flow}{NF\ Perm.\ Flow + DI}$$

Since the first method (Rec1) underestimates the recovery and the second and third method (Rec2 and Rec3) overestimate the recovery, a better recovery is

Demonstration of Zero Discharge Desalination (ZDD)

estimated by taking the average of all methods. Each of the recovery method results and the average of Rec1 and Rec2 are compared against the data obtained in the WERF 5T10 study in **Error! Reference source not found.** (includes 26 ays' worth of data). Average values for the Mixed Cl and Mixed Na waste flows were used for days when overflow rates were not recorded. The large variation in the WERF 5T10 method (UTEP calculations using method described in WERF 5T10 report) reflects the wide variation in the estimated feed flow. Values from each of the calculation methods that incorporate the total DI flow have low points throughout the La Junta pilot operations. These low points correspond to days where the system was shutdown, started up, or had high DI consumption after process upsets such as the loss of NaCl makeup.

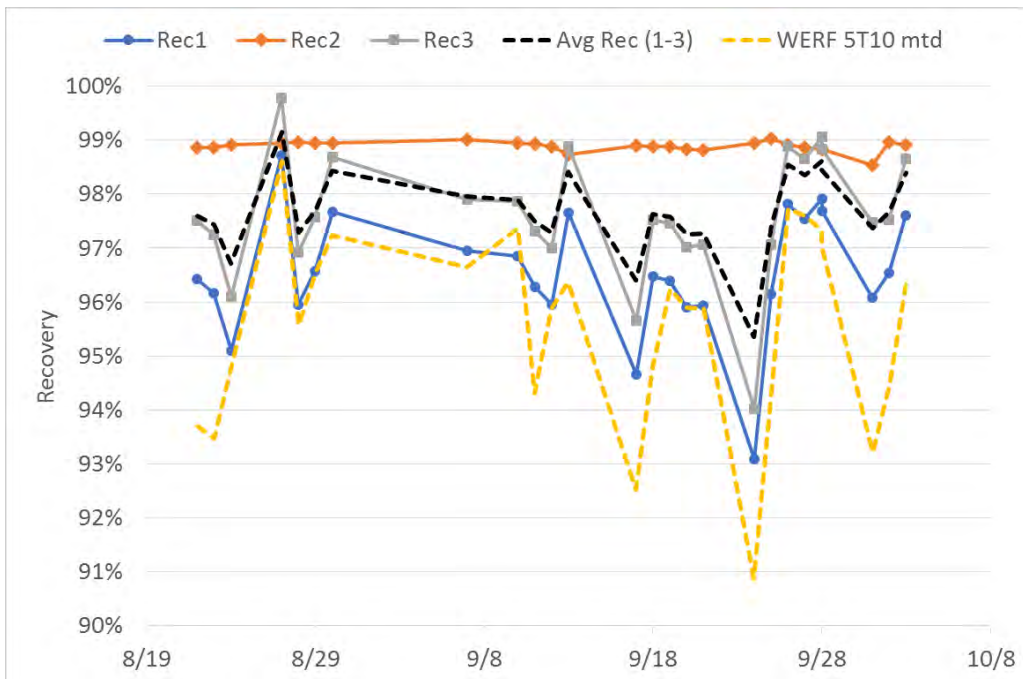


Figure 47.—Comparison of recovery calculation method results (La Junta).

Four time periods were summarized in table 3-3 of the WERF 5T10 report. Recovery using the method described in the WERF 5T10 report is compared against the recovery calculations summarized in Table 15 of this report. The average values (last column) are likely the most representative of the actual ZDD recovery, but are still considered to be somewhat conservative. DI water consumption was variable throughout the project and did not decrease with increasing conductivity setpoints for the Mixed Cl and Mixed Na streams (increased setpoints should lead to lower DI consumption for dilutions). DI water is used during startup and shutdown activities, for sample dilutions and washing, and was used for NaCl recovery experiments. Of these uses, the startup and shutdown activities should have consumed the most DI water and are not expected to be representative of a full scale system operating continuously. The pilot was shut down over weekends and other times when personnel were not

Demonstration of Zero Discharge Desalination (ZDD)

available; a full-scale plant would have a more continuous operation and would, therefore, use less DI water.

Table 15.—Comparison of Recovery Results by Experiment (La Junta).

Rec=Recovery method

Avg (1-3 is the average of Rec1, Rec2, and Rec3

Dates	# Quads	WERF 5T10 ¹	Rec1	Rec2	Rec3	Avg (1-3)
8/21-8/30	100	96%	97%	99%	98%	98%
9/6-9/13	80	96%	97%	99%	98%	98%
9/17-9/25	80	94%	96%	99%	97%	97%
9/26-10/3	80	96%	97%	99%	98%	98%

¹ Calculations made by UTEP match WERF 5T10 report except 9/26-10/3 (WERF 5T10 table 3-3 reported value is 97%).

Rec=Recovery method

Avg (1-3 is the average of Rec1, Rec2, and Rec3

Internal leaks are suspected in the EDM stack because sulfate and bicarbonate were present in the NaCl and Mixed Cl samples taken on September 21, 2013, and September 28, 2013. Another indicator is the higher than normal pH in both the NaCl and Mixed Cl streams for most of the La Junta pilot. Typically, the Mixed Cl stream pH is less than 4 and can be less than 3 when too much voltage is applied to the EDM stack. The NaCl stream pH is typically less than 5. Figure 48 shows the measured pH of both streams. The pH of both streams was around pH 6-7 for most of the samples. Small holes were detected in some membranes and the center dividers when the stack was taken apart in both La Junta and Brighton. Finally, the EDM stack bolts were not regularly tightened so low stack compression could be another source of internal leakage. ED and EDM stacks must be properly tightened and aligned to prevent internal leaks.

Demonstration of Zero Discharge Desalination (ZDD)

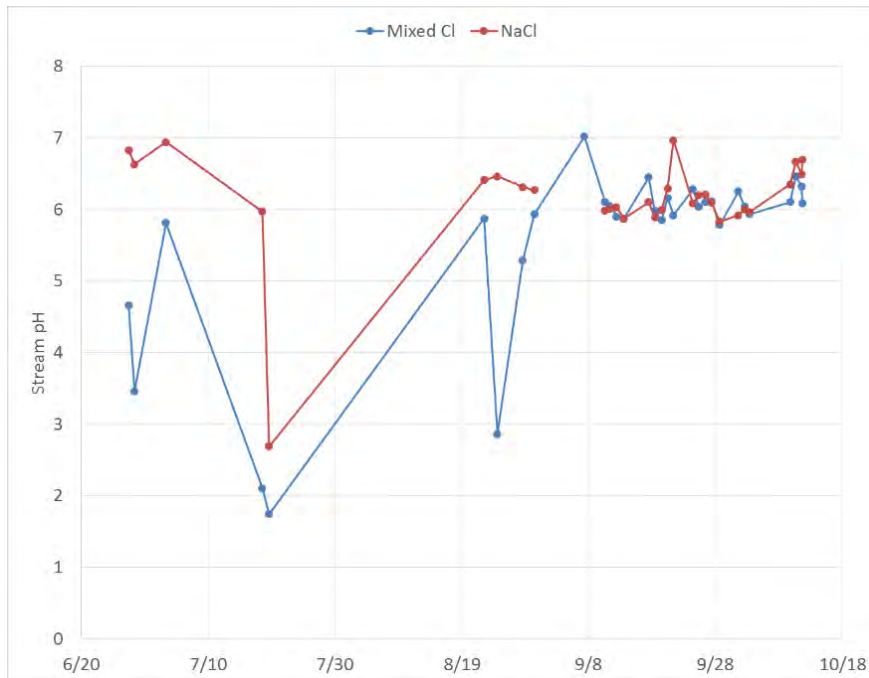


Figure 48.—NaCl and Mixed Chloride pH at La Junta.

One other type of operational problem occurred several times during the La Junta pilot. The pilot system's circulating NaCl stream conductivity dropped dramatically at least six times during the operations in August and September. This is because there was not enough NaCl in the saturator tank. Earlier piloting problems (scale formation) were attributed to NaCl contamination and to the fact that the overflow between the standpipe and saturator tank was routed to the drain instead of to the NaCl saturator tank. This caused excessive NaCl consumption. In addition, it is possible that the operators did not add enough NaCl pellets prior to departing for the night. After the loss of NaCl on August 29, the overflow was re-routed to the NaCl saturator tank. Unfortunately, this did not completely solve the problem as the NaCl conductivity dropped precipitously again on August 30, September 18, and September 19. All of these outages occurred during the night or early morning when operators were not onsite. The operators also did not have remote viewing or control access. When the conductivity dropped in the NaCl stream, there were no longer ions present to carry the current through the stack. This means the only current that could flow was by way of shunt currents. This caused the EDM feed and diluate conductivity to rise, because ions were not being removed from the NF system, but ions were still back diffusing from the EDM's Mixed Na and Mixed Cl streams. The impact of these NaCl losses led to a buildup of ions in the EDM and NF systems, as is evident from the EDM diluate conductivity increasing over time. Figure 49 and Figure 50 show the conductivity of the NaCl, NF Concentrate, and EDM feed streams along with the flow of the EDM feed pump.

Demonstration of Zero Discharge Desalination (ZDD)



Figure 49.—NaCl conductivity in La Junta (8/21 – 8/24).

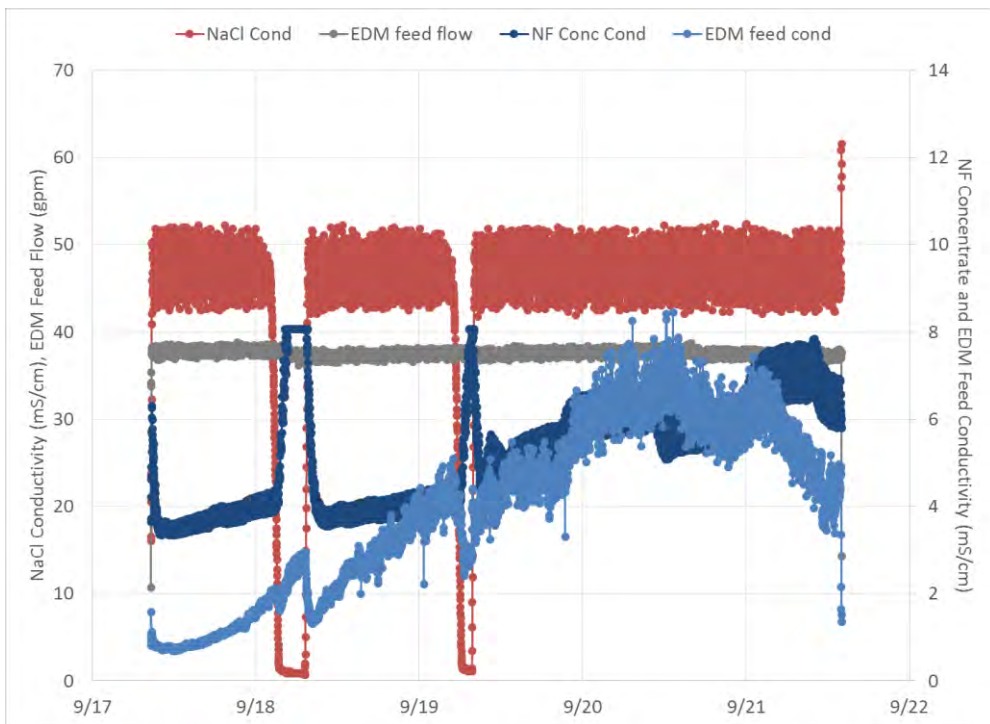


Figure 50.—Loss of NaCl conductivity in La Junta (9/17 - 9/21).

Late afternoon on August 30, the NF system shut down because the NF feed tank level was low. At about the same time, the EDM feed flow also dropped to zero.

Demonstration of Zero Discharge Desalination (ZDD)

All other pumps remained on and the stack power also remained on until operators returned on August 31. The lack of flow in the EDM feed compartment combined with the EDM's power supply remaining on led to a rapid and precipitous decline in conductivity in the EDM feed compartment. Also, temperatures in the EDM stack increased from 20-25°C to the maximum that the temperature probes could read (45-50°C). The MEGA membranes and spacers have a maximum temperature tolerance of 35-40°C. Five quads of membranes and spacers were melted together after this occurred, and the membranes were subsequently removed from the stack. This stack was sent to Alamogordo and was used during Phase 2. Figure 52 shows the conductivity of the streams and flows and Figure 52 the NF concentrate and EDM stream temperatures (bottom diagram).

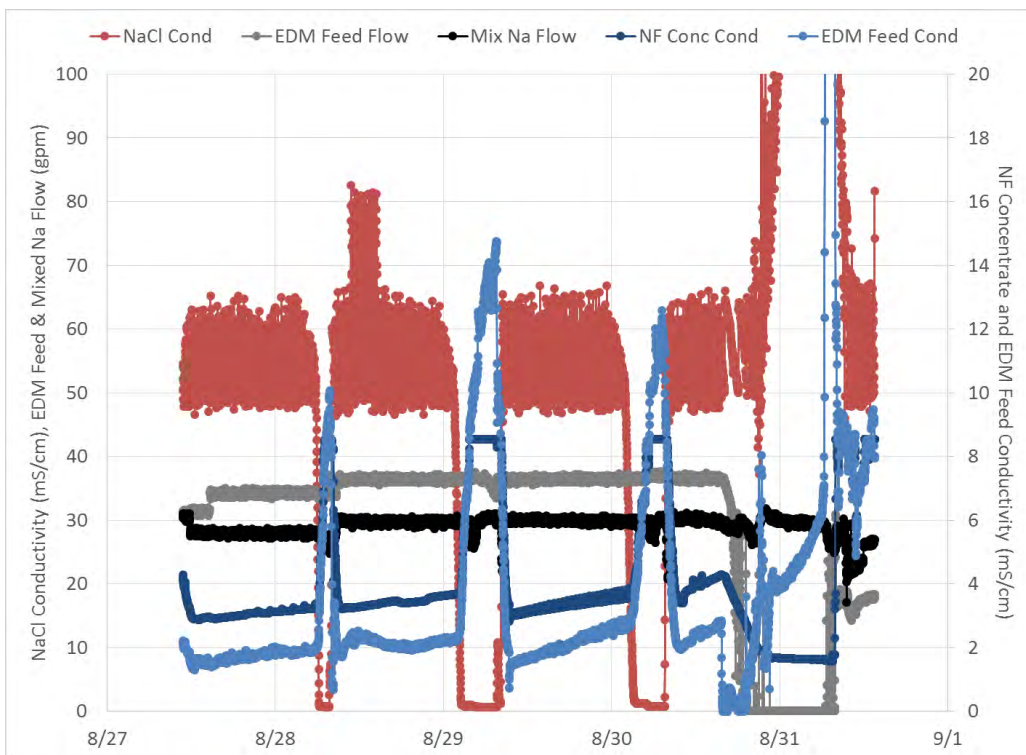


Figure 51.—NaCl conductivity in La Junta (8/27 -8/31).

Demonstration of Zero Discharge Desalination (ZDD)

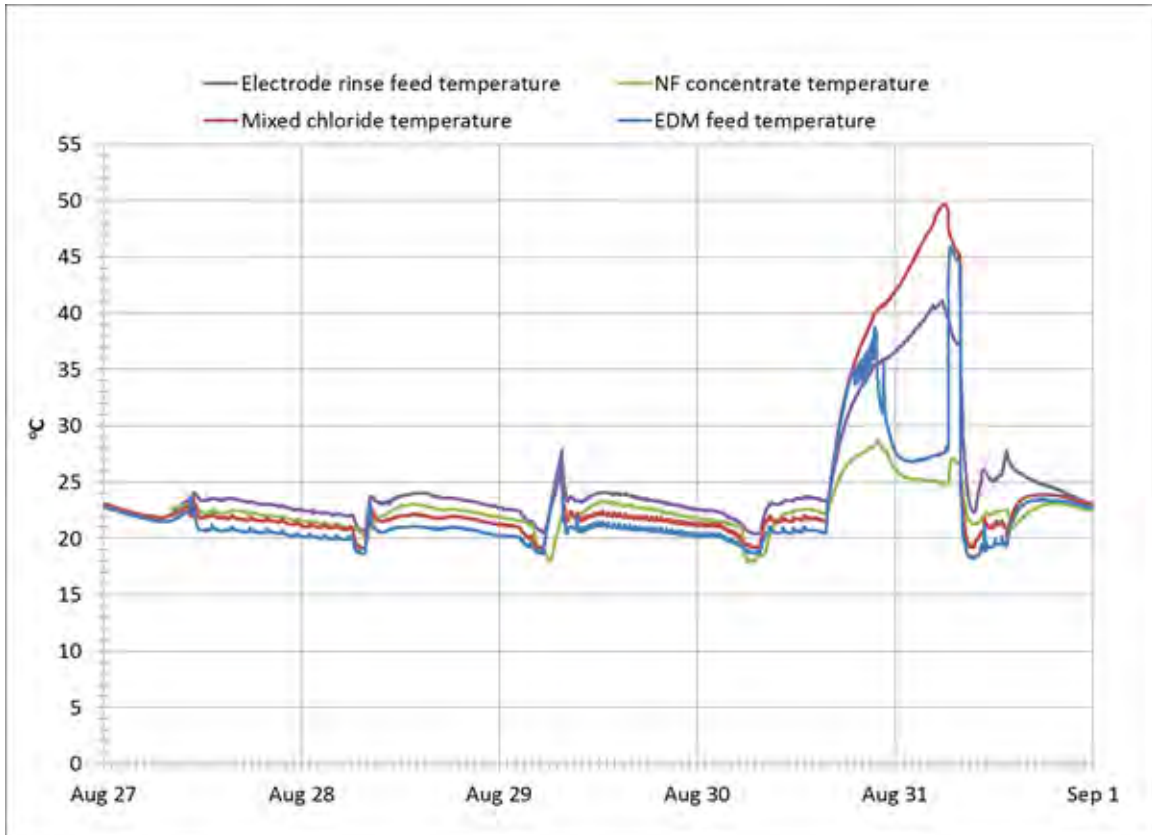


Figure 52.—EDM feed flow in La Junta (8/21 – 9/1).

7.3 Evaluation of Current Density, Current Efficiency, Power Consumption – La Junta

7.3.1 EDM Stack Performance Monitoring

The EDM stack resistance was relatively stable throughout the La Junta piloting in September. A similar miscommunication in the number of quads in the EDM stack as in Alamogordo (Phase 2) led to excessive voltage being applied to the 100-quad stack. The 80-quad stack was used for all of the September and October piloting in La Junta and the average resistance was 3.3 ohms. Figure 53 includes all of the EDM stack resistance points for the La Junta pilot.

Demonstration of Zero Discharge Desalination (ZDD)

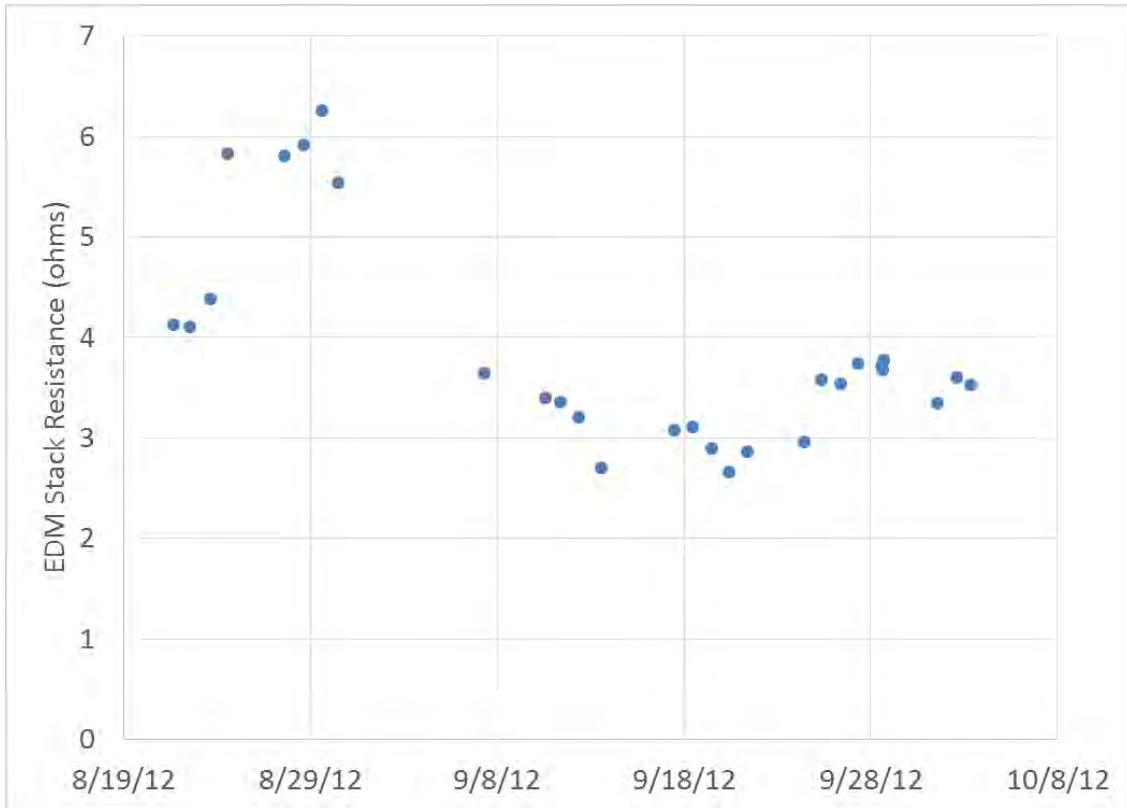


Figure 53.—EDM stack resistance (La Junta).

An attempt was made to calculate the mass balance and current efficiency for the two days with full water analyses. However, an accurate mass balance was not possible because of the problems with flow measurements and the lack of ability to accurately estimate the daily salt consumption. Also, the ion balance is off by 10 percent or more for several of the samples. There are other inconsistencies with the analyses (conductivity/TDS ratio suggests problems with one or both analyses for the EDM feed and diluate) that prevent estimations of the current efficiency. Finally, the way NaCl consumption is calculated is to sum up the number of bags added and divide by operational hours. Using this method is inaccurate because the amount of NaCl added varies from zero to as much as 10 bags in a period. A better method for NaCl consumption estimation was incorporated at the Brighton pilot.

A sample of membranes from the La Junta pilot was sent to MEGA for analysis of precipitates. One membrane was found to be damaged (tear or otherwise) upon inspection. Additionally, carbonate scale appeared to be present inside another membrane. MEGA's report is included as Appendix D1.

7.3.2 Power Consumption

Power meters were installed by Reclamation personnel on the pilot equipment trailers in La Junta to record power consumption during four periods in August and September. The data were reported in the WERF 5T10 report and are summarized in Table 16. The total energy consumption reported was quite high, at 13 to 15 kWh/kgal. As described in a footnote in the WERF 5T10 report, pilot equipment is not designed or sized to be energy efficient. In fact, the NF system pumps were grossly oversized for the flow and brackish water feed concentration at La Junta.

Table 16.—WERF 5T10 Report Energy Consumption

	8/21-8/23	9/6-9/7	9/10-9/12	9/19-9/21
NF	5.8	5.5	5.6	5.8
EDM	6.6	7.8	8.0	7.6
Total	12.3	13.3	13.6	13.4

The operating expenses are described later in this report for both the La Junta and Alamogordo sites. Briefly, the energy for the NF and EDM systems was estimated based on the pilot data and includes flow rate and pressure for the brackish feed, NF feed, EDM recirculating pumps, and the maximum VDC and current required by the EDM to achieve the water quality objectives (this point may be higher or lower than what was piloted and depends on the design of the full scale system).

The pilot brackish water pump, which pumps water from the NF feed tank through cartridge filters has a 5 horse power (hp) motor, or 400 percent higher than what is needed to pump 35 gpm at 30 psi. The NF feed pump had a 15 hp motor, 650 percent higher than what was required to boost the pressure from 30 psi to 90 psi. As shown in Table 14, Dow’s ROSA program predicted the energy for desalination to be 1.38 kWh/kgal, which corresponds to a motor size of 1.65 kW (2.2 hp); this is the amount of energy required if no pressure is supplied to the feed of the high pressure pump. The reported NF unit energy consumption is likely overestimating by a factor of 2.5 to 3.

Similar to the NF, the EDM recirculating pumps were also oversized. Most of the energy required for the EDM container should come from the EDM power supplies, which was 3 to 4 kWh/kgal for most of the operations in La Junta. The values in Table 16 suggest that more power was used in the EDM system in September than in August, which is odd since more power was required by the EDM stack power supply for at least a portion of August. The estimated unit energy is shown in Figure 54. The energy, kWh_{AC}, is estimated using an 82.4 percent conversion factor for the various transformers employed in the EDM container to convert from the 480 VAC supplied to the EDM container to 100-150 VDC supplied to the EDM stack. The reported EDM unit energy consumption is overestimated by 30-70 percent.

Demonstration of Zero Discharge Desalination (ZDD)

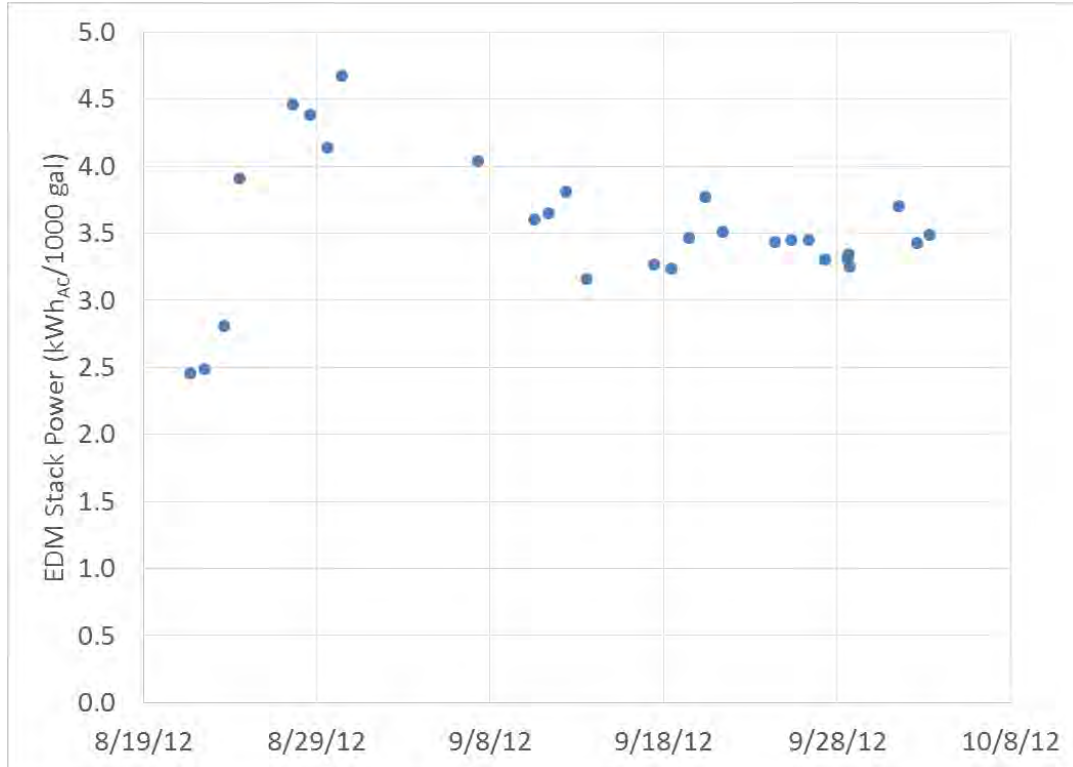


Figure 54.—Estimated power EDM consumption (La Junta).

An estimate was made for the unit energy consumption for the ZDD system components using the full scale flow rate of 6.6 MGD product water at the La Junta RO Plant. A full scale ZDD system would include:

- Brackish well water pumping: 4,670 gpm
- NF system (80 percent recovery with NF90/NF270 mixture piloted) capable of producing 4,130 gpm of permeate that would be blended with 603 gpm of brackish water
- Brackish feed pump: 467 gpm at 30 psi
- Three NF feed pumps: 1,690 gpm at 82 psi (5,097 gpm total NF feed flow at full flow)
- EDM system sized to reduce the concentration by 50 percent (from the NF reject to EDM diluate)
- 3,200 gpm recirculating rate for EDM feed, NaCl, Mixed Na, Mixed Cl pumps

Demonstration of Zero Discharge Desalination (ZDD)

- 800 gpm recirculating rate for E-rinse pump
- DC power supply needs: 120-140 VDC/40-45 amps
- Basis for operations is full production for 365 days per year

The maximum current from the La Junta pilot operations was 37 amps, but was likely low because of various operating problems and colder temperatures. The full scale system is expected to be able to operate at 40 amps or higher. This is an important design factor, as the number of EDM stacks required depends on the current density (current per unit of membrane area). The values for power consumption for the two ZDD designs are summarized in Table 17.

Table 17.—Estimated Full Scale Unit Energy Consumption (La Junta)

	100-Stack EDM Design (40 amps)		90-Stack EDM Design (45 amps)	
	kWh/kgal	% of total	kWh/kgal	% of total
NF energy	2.06	36.5%	2.06	37.8%
EDM energy	3.53	62.4%	3.35	61.2%
Other energy	0.06	1.0%	0.06	1.0%
Total energy	5.65		5.47	

8. Brighton Results and Discussion

The second pilot site for the WERF 5T10 project was in Brighton, where the “bolt-on” ZDD system was implemented (Figure 55). As with the La Junta pilot, data and analysis are provided in the WERF 5T10 report, so only additional or differences in analysis are described in this report.



Figure 55.—ZDD system installed in Brighton (20-gpm).

8.1 Equipment Description

The 20 gpm pilot system employed during Phase 1 in Alamogordo and La Junta was shipped and installed at the Brighton Water Treatment Plant. This plant includes a greensand treatment system for treating water with high iron and manganese and a RO plant for desalinating brackish groundwater.

The design for the bolt-on ZDD system involved blending the Brighton RO concentrate with NF concentrate from the pilot system. Sulfuric acid was added to the Brighton RO concentrate to reduce the alkalinity to less than 200 mg/L as CaCO₃ in the blended RO/NF feed to the EDM. The original design, prepared in May 2014, was designed to treat about 10 gpm of RO concentrate. (The WERF 5T10 report table 4-1 incorrectly notes this flowrate as the “ZDD Test Plan” flowrate, when in actuality it was an approximation based upon a model that does not consider the pilot pipe sizes.) However upon startup, it was evident that the feed piping to and from the EDM system could not accommodate 10 gpm of NF/RO reject. The standpipe inside the container would overflow, because the EDM diluate line was undersized to allow the gravity flow between the EDM and NF containers. The ZDD equipment was never designed with this configuration in mind. For this reason, the RO concentrate flow was reduced to 5 to 5.2 gpm,

Demonstration of Zero Discharge Desalination (ZDD)

which corresponds with an EDM feed flow of 9-9.2 gpm. The EDM reduced the concentration by at least 50 percent, and its diluate was fed to an NF system. The NF contained a 2x1(4M) array with NF90-4040 membranes in the first stage and NF270-4040 membranes in the second stage. The NF concentrate, as mentioned before, was blended with the pH-adjusted Brighton RO concentrate. A small purge stream from the NF concentrate was required to maintain silica levels below 150 mg/L. Figure 56 shows the flow diagrams for the ZDD system tested at Brighton. Flow rates for the RO concentrate and silica purge were measured manually and verified using rotameters installed outdoors adjacent to the waste tanks. Sulfuric acid dosage was adjusted manually and verified using field analysis of pH and alkalinity reduction. Hydrex 4101 antiscalant was dosed at 2 to 3 mg/L to the NF system.

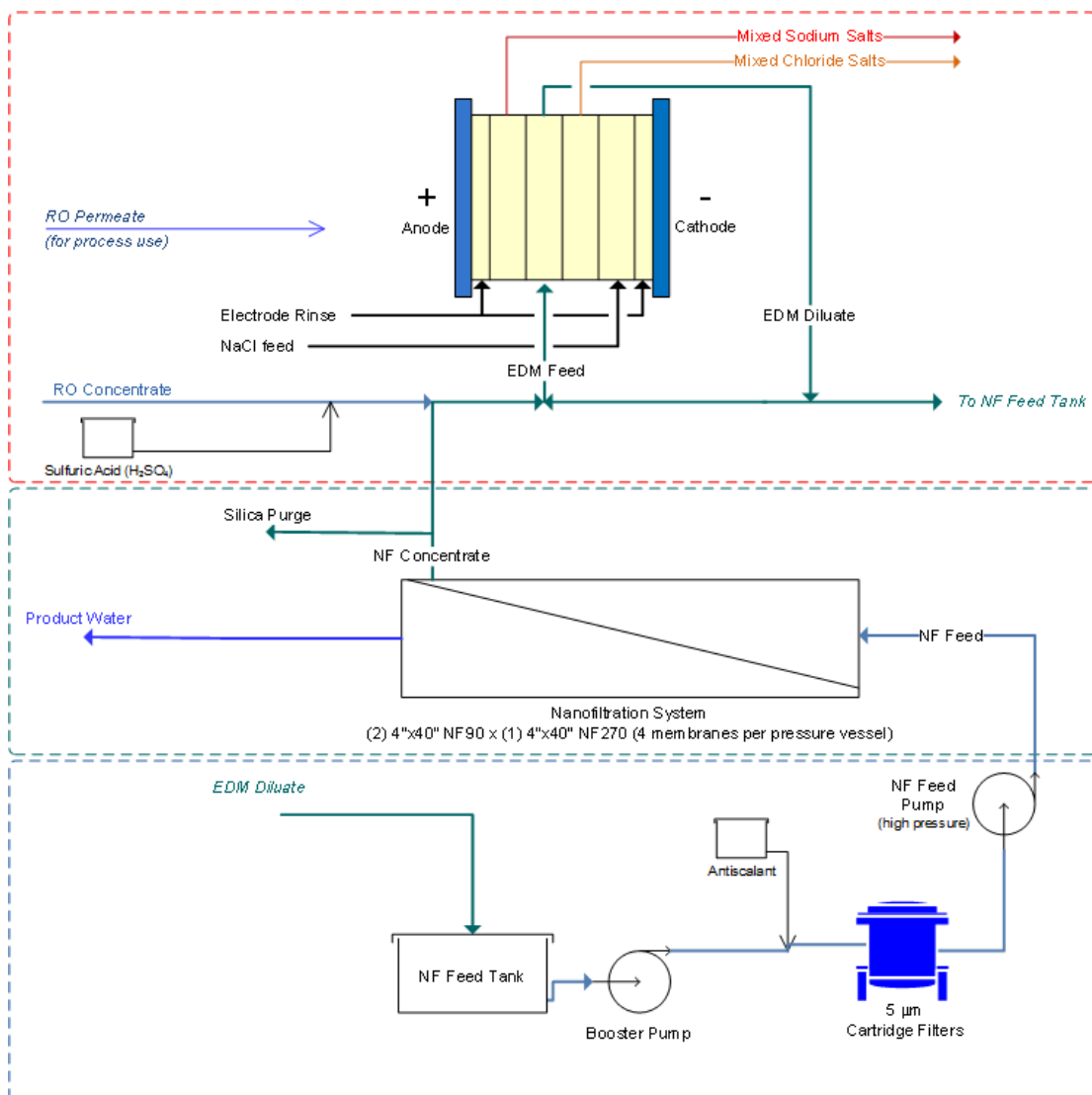


Figure 56.—Equipment flow diagram (Brighton).

The NF high pressure pump motor seized during commissioning and did not operate after September. Sufficient pressure and flow were available from the NF booster pump, so operations continued with the NF operated in manual mode. The NF system was designed to produce 20 gpm of permeate. By operating at 5 gpm of permeate, the online flow meters were out of range and did not provide adequate precision for flow adjustment (5 gpm displayed could easily be 4.5 gpm or 5.4 gpm). The operator estimated actual flows using a bucket and stop water and by comparison with other rotameters and online flow meters throughout the system. The combination of manual adjustment of RO concentrate and NF silica purge stream, low overnight temperatures, and a lack of additional operational support led to operating the system with an imbalance overnight. This means more RO concentrate was being added, which led to an intentional, but slight, overflow of the NF feed tank. This nighttime overflow was estimated to be 0.1-0.18 gpm.

8.2 Pilot System Upgrades and Commissioning

Prior to shipping the pilot equipment, multiple upgrades were performed to the pilot equipment:

- Conductivity probes on mixed salt streams were replaced with inductive conductivity probes (HACH 3726E2T).
- Improved interlocks were added between the EDM and NF systems to ensure automatic shutdown by the PLC for alarm conditions including high temperature, high pressure, and low flow; also a shutdown-enabled alarm was added for low EDM feed flow.
- Controls were upgraded so that additional data were automatically recorded and trended including EDM voltage and current as well as NF permeate and concentrate flow.
- The EDM stack flush system was upgraded to include a bladder tank to provide a more thorough flush of EDM stack upon shutdown.
- The E-Rinse pressure gauge and flow control valve were relocated to properly match the E-Rinse pressure to the other EDM stream pressures.

During commissioning, the Mixed Na and Mixed Cl conductivity probes were still unreliable. The operator moved their location from the standpipes to the overflow pipes. This allowed for isolation from suspected shunt current in the highly concentrated solutions (this lesson learned was also implemented in Alamogordo during Phase 2).

Several problems occurred during setup and commissioning, most of which were explained by internal EDM stack leaks, improper acid dosing location and rate,

Demonstration of Zero Discharge Desalination (ZDD)

unexpected solubility problems in the Mixed Na stream, and equipment damage. Each of these is described later in this section.

Budgetary concerns caused Veolia to withdraw on-site personnel after August 2013, therefore UTEP operated Brighton after Alamogordo Phase 2 operations were completed. Veolia supported the project by shipping a replacement stack to Brighton, purchasing supplies for and providing an operator to install new center dividers in the EDM stack, purchasing NaCl and sulfuric acid, and providing technical guidance throughout the pilot operations.

8.3 Experiment 1 (9/30/13-10/13/13)

The first three days of operation were used to identify a safe conductivity setpoint for the Mixed Na stream, which had caused several shutdowns during commissioning. Additionally, the time was used to determine if the new center divider was working properly and preventing internal leaks. A low conductivity setpoint (45 millisiemens per centimeter [mS/cm]) was the starting setpoint. After safe operation for around 24 hours, the setpoint was increased to 50 mS/cm for another 24 hours. After this, the setpoint was increased to 60 mS/cm, where it remained for the duration of the experiment (except for one overnight run where outdoor temperatures dropped below freezing). No dilutions were used for the Mixed Cl stream initially; however, dilutions were implemented overnight on October 12 (conductivity setpoint of 100 mS/cm), because the outdoor temperature had dropped below freezing. Additionally, it was thought that having the two concentrate streams be more balanced in terms of conductivity and density would improve operations.

The ZDD system operated nearly continuously during this experiment. The Brighton RO Plant shutdown briefly (5.3 hours) on October 3 for maintenance and there were a few times during the first week where RO concentrate flow was lost temporarily when the plant switched RO trains. The ZDD system operated for a total of 298.6 hours during this experiment (excludes the 5.3 hour shutdown); 233.6 hours were continuous and without setpoint changes.

All EDM flow and pressure data show no indication of scale formation. EDM performance was more stable at the beginning of the experiment, but the diluate conductivity and EDM stack resistance rose as the system operated. The system feed flows and the NF system were controlled manually. This, combined with low overnight temperatures and increasing RO concentrate conductivity, are believed to be the cause of the reduced EDM performance. The ZDD system was shut down on October 13 because the Brighton water tanks were full and the city's demand had decreased.

8.4 Experiment 2 (10/15/13-10/19/13)

The ZDD system was restarted on October 15 and operated continuously without shutdown until October 19 because the Brighton RO Plant needed to shut down to allow its product water tank levels to drop. During the 93.4 hours, all EDM streams (except the E-Rinse) were stable and showed no signs of scale formation. The E-Rinse would lose pressure overnight. There were several leaks in the E-Rinse system, and DI water was added regularly to make up for the losses. Addition of DI water also warmed the stream. The combination of these effects is the reason for the variation in pressure (see Appendix B3).

The Mixed Na and Mixed Cl conductivity setpoints remained constant at 120 mS/cm and 65 mS/cm, respectively. All high concentration EDM streams had stable conductivity except for the E-Rinse. DI water was added regularly to make up for the lost volume from leaks. The setpoint was reduced on the final day because the operator mistakenly thought this would improve the crystallization problem. The EDM performance was more stable towards the end of the run, which is related to the RO concentrate conductivity decreasing and also better flow balancing in the NF and EDM systems.

8.5 Experiment 3 (10/22/13-10/26/13)

The ZDD system was restarted on October 22 and operated continuously without shutdown until October 26, because the Brighton RO Plant needed to shut down to allow its product tank levels to drop. During the 94.4 hours, all EDM streams (except the E-Rinse) were stable and showed no signs of scale formation. The E-Rinse pressure increased gradually as the run progressed. About half of the E-Rinse piping is located outdoors and is uninsulated. This exposure likely caused some of the sodium sulfate to crystallize and settle in the electrode compartments, which is thought to be the cause of the increased pressure (see Appendix B3).

The Mixed Na and Mixed Cl conductivity setpoints remained constant at the setpoints from Experiment 2 for the first three days of operation. In an attempt to reduce back diffusion from the concentrate streams as well as to balance the stream densities, the setpoints were reduced stepwise until the Mixed Cl was at 100 mS/cm and the Mixed Na was at 55 mS/cm. The EDM diluate quality remained fairly constant during the day but increased overnight. This decreased performance was thought to be caused by a combination of factors including electrode scaling (later inspection of the electrode showed substantial scale, presumably from sodium sulfate), low temperatures, and flow imbalances between the NF and EDM.

8.6 Summary of Brighton Pilot Operations

8.6.1 Early Problem Discussion

During setup and commissioning activities, the EDM stack was not able to operate continuously without scale forming in the Mixed Na compartment. The operator initially located the sulfuric acid addition point inside the EDM container to add acid directly to the Mixed Na stream. The thought was this would target the single stream that needed bicarbonate reduction directly and could reduce the volume of acid used. However, pH measurement in this stream became problematic, and the acid addition point was relocated to the original design point (addition to RO concentrate).

Before the acid injection point was changed, carbonate and/or bicarbonate scale formation was regular in the Mixed Na stream. CIPs were performed, but were ineffective at increasing run time. The EDM stack was removed and inspected in late May. Holes were found in the center divider. The divider was patched, but not replaced. During the next week, the Mixed Cl stream did not overflow, but the EDM stack leaked warm water. The stack was removed and inspected again and damaged membranes were removed; the damaged center dividers were still not replaced.

The acid injection point was relocated to the RO concentrate in mid-June. The system operated more stably, but the EDM stack resistance continued to climb. An attempt was made to dose acid to the NF feed instead. However, this did not solve the problem.

In mid-July, both the Mixed Na and Mixed Cl streams had scale formation problems. The cause of the scale formation (carbonate based on acid addition test) in the Mixed Cl stream was suspected to be from internal leaks in the EDM stack. A leak test confirmed substantial leaks in late July, and the stack was removed so that new center dividers could be installed. The new dividers allowed for better operation; however, the Mixed Na stream was still scaling. Prior to mid-July, field alkalinity analyses were not performed, and the handheld and online pH probes were not calibrated. The field alkalinity tests revealed that there was more bicarbonate in the EDM feed than the Veolia design had anticipated. Even though changes were made to reduce bicarbonate in the NF/RO reject fed to the EDM the Mixed Na stream still had scale formation issues. This scale was suspected to be from reduced sodium sulfate and sodium bicarbonate solubility related to the concentration of NaCl present in the Mixed Na stream and from internal leakage in the EDM stack. Marion and Farren studied the effect of NaCl concentration on sodium sulfate and sodium bicarbonate solubilities in different studies (Marion and Farren, 1999 and Marion, 2001). Figure 57 shows a) how increasing Na_2SO_4 concentration and decreasing temperature reduce the solubility of NaHCO_3 , and b) how increasing NaCl concentration and decreasing temperature reduce the solubility of Na_2SO_4 .

Demonstration of Zero Discharge Desalination (ZDD)

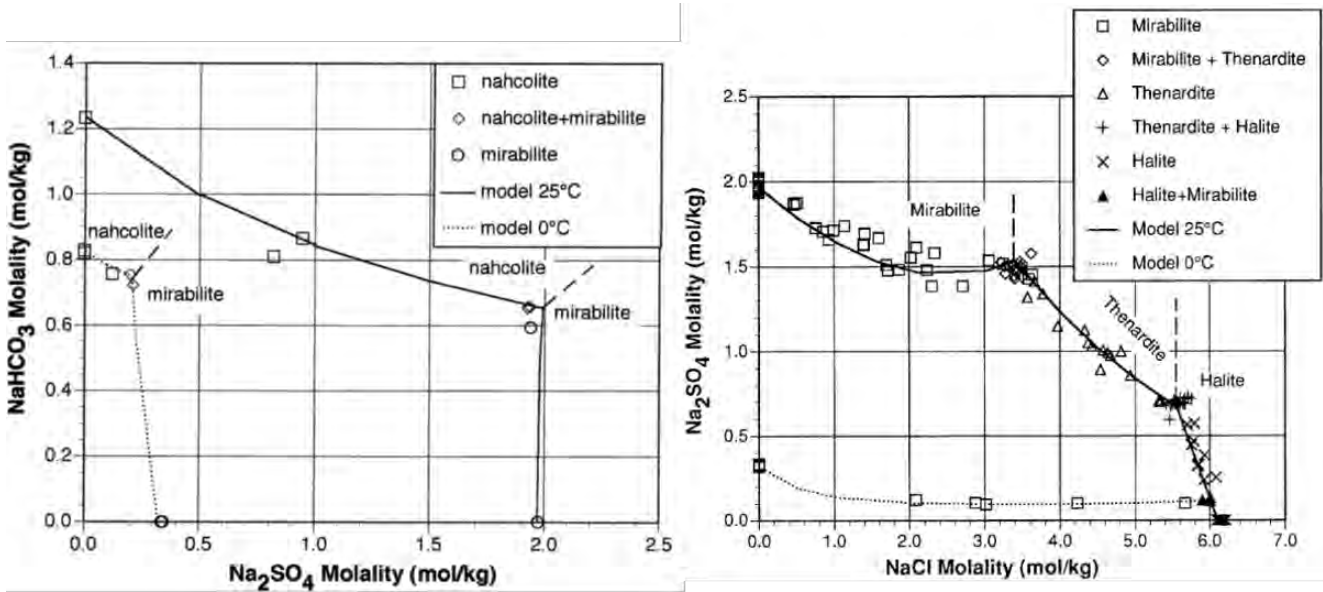


Figure 57.—Solubility and temperature relationships between a) Na₂SO₄-NaHCO₃ (Marion and Farren, 1999) and b) NaCl (Marion, 2001).

Operation of the ZDD system was relatively stable in August; however, the EDM stack resistance continued to increase. The E-Rinse stream showed signs of severe corrosion and did not allow for a successful handoff to the WERF 5T10 operators. The stack was inspected in early September and severe corrosion was found on the stainless steel cathodes. Inspection of the electrical connection identified the source of the corrosion and high stack resistance. Each electrode plate had an upper electrode and lower electrode with individual connections. The original electrical lug on the bottom cathode wire had been replaced in Alamogordo before the system was shipped to La Junta (and no changes were made before shipping to Brighton). This lug was made of aluminum and had a bolt to fasten the lug to the copper wire. This connection loosened over time, and less current flowed through the lower electrode. There was also evidence of arcing (see Figure 58). This bad connection is believed to have made the cathode act more like a sacrificial anode, and some of the stainless steel corroded.

Figure 59 shows the cathode end plate (top picture) during an inspection in late August; the bottom picture was taken during the inspection in early September. The lower cathode was very shiny and had no texture remaining. There were small holes showing pitting corrosion on several places on the lower anode, especially at the electrical connections.

Demonstration of Zero Discharge Desalination (ZDD)



Figure 58.—Damaged EDM cathode electrical connection. Photo taken on 8/29/13,

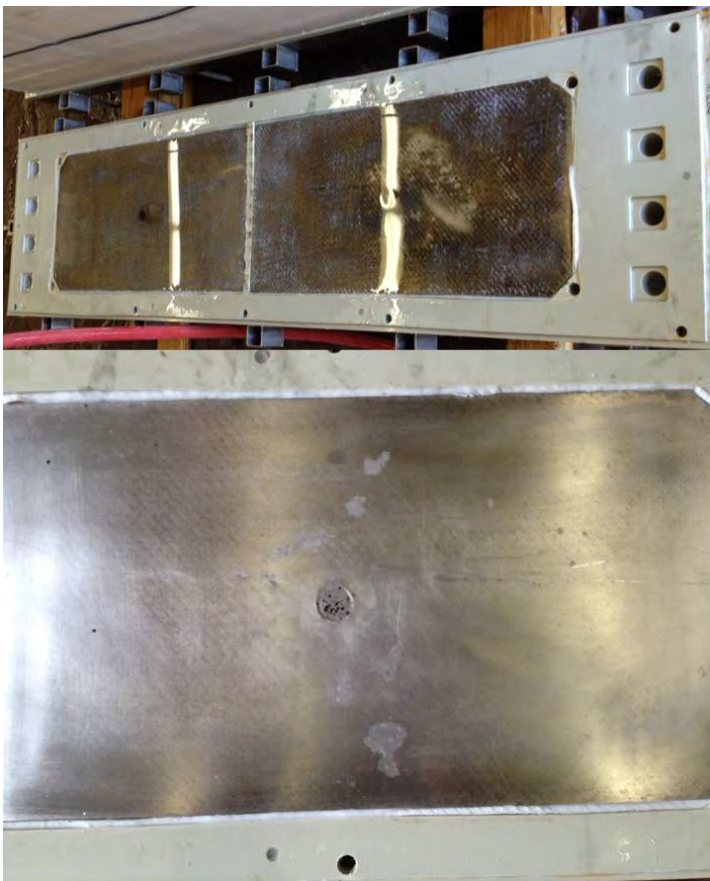


Figure 59.—EDM cathode inspection. Photo taken on 9/9/13)

8.6.2 Summary of Brighton Operations by UTEP (9/30/13-10/26/13)

Reliable operating setpoints were verified within the first days of a nearly two-week run. Except for overnight operations, the setpoints were followed throughout the three experiments.

As previously mentioned, the NF online flow meters were not precise (output whole numbers without decimals). Additionally, since the system was being operated in manual mode, no data was recorded. The actual permeate produced was estimated by subtracting the calculated Mixed Na and Mixed Cl flows from the EDM feed online flow meter. The NF concentrate flow was calculated by subtracting the silica purge flow from the NF concentrate flow (both measurements were taken from rotameters). The RO concentrate was piped to a storage tank, which was allowed to overflow to ensure adequate supply at all times to the pilot system. A submersible pump's outflow was modulated using a gate valve (red handle in Figure 60). Data from the three rotameters (Brighton RO concentrate, NF concentrate, and silica purge) located outside generally correlated with the EDM feed (called NF/RO reject) flow meter inside the EDM container. At night there was sometimes a discrepancy. The Mixed Na and Mixed Cl overflow streams were captured in 30-gallon tanks with markings at each gallon. The tanks were allowed to fill for at least an hour and the volume was recorded.



Figure 60.—RO concentrate and NF concentrate flow metering (Brighton).

Demonstration of Zero Discharge Desalination (ZDD)

The EDM system's datalogger recorded flow, pressure, conductivity for each of the streams (including the NF/RO reject), temperature of the EDM feed and E-Rinse, EDM applied voltage, and EDM current. The NF/RO reject pH was also recorded, but the data is not considered valid, since the probe would not calibrate properly. Table 18 summarizes the average value for various operating data points gathered during pilot operations at Brighton.

Table 18.—Summary of Brighton Operational Data

Data Point	Unit	Average (Range)
NF Permeate Flow	gpm	5 (N/A)
1st Stage NF Permeate (rotameter)	gpm	3.8 (3.8-3.9)
NF Reject (rotameter)	gpm	4 (3.9-4.2)
RO Reject (rotameter)	gpm	5.1 (4.8-5.2)
NF/RO reject	gpm	9.2 (9-9.4)
EDM Feed	gpm	31.4 (30.7-32.2)
M. Na	gpm	26.7 (25.7-27.6)
M. Cl	gpm	26.1 (25.4-26.6)
NaCl	gpm	27.1 (24.5-27.8)
E-Rinse	gpm	4.9 (4.6-5.2)
Silica Purge	gph	3.5 (1.3-4)
<i>Estimated Flows</i>		
<i>NF Feed Tank overflow</i>	<i>gpm</i>	<i>0.1 (0-0.2)</i>
<i>NF Permeate</i>	<i>gpm</i>	<i>4.6 (4.4-4.9)</i>
<i>NF Reject</i>	<i>gpm</i>	<i>4.1 (3.9-4.2)</i>
NF Feed	psi	40.1 (32.8-43.7)
NF Permeate	psi	2.1 (1.8-2.2)
NF Concentrate	psi	25.7 (21-28.9)
EDM Feed	psi	19 (17-147)
Mixed Na	psi	17 (16.2-17.7)
Mixed Cl	psi	17.1 (16.4-18.8)
NaCl	psi	17.1 (16.5-17.7)
E-Rinse	psi	17.3 (16.7-17.9)
NF Feed	°C	21.9 (20.2-23.6)
NF Permeate	°C	21.7 (20.5-22.8)
NF Concentrate	°C	21.7 (20.7-22.9)
NF/RO Concentrate	°C	20.1 (18.7-21.4)
EDM Feed	°C	20.7 (19-22.3)
EDM Diluate	°C	21.2 (19.2-22.9)
Mixed Na	°C	21.3 (19.5-23)
Mixed Cl	°C	21.5 (19.5-23.4)

Demonstration of Zero Discharge Desalination (ZDD)

Data Point	Unit	Average (Range)
E-Rinse	°C	21.8 (17.8-24.7)
EDM Voltage	volts	100 (79.2-120.4)
EDM Current	amps	26.3 (23.7-28.3)
NF Recovery	%	53.2 (51.2-55.1)
ZDD Recovery w/o NF purge	%	89.8 (87.1-91.4)
ZDD Recovery w/NF purge	%	88.1 (85.4-89.7)
ZDD Recovery (NF perm/RO Conc)	%	91.6 (85.4-95.4)
Longest Continuous Run		298 hours (9/30/13-10/13/13)

gph = gallon per hour

The ZDD recovery was calculated using three methods.

The first method is similar to Rec1 used in La Junta (called ZDD Recovery w/o NF Purge in Table 18) divides the calculated NF permeate by the sum of the calculated NF permeate and EDM concentrate streams.

The second method (called ZDD Recovery w/NF Purge in Table 18) assumes that the NF feed tank was overflowing continuously and adds the estimated flow to the denominator of the calculation:

$$\text{ZDD Recovery w/NF purge} = \frac{\text{NF permeate (calculated)}}{\text{NF perm (calc)} + \sum \text{overflows} + \text{NF purge}}$$

The third method (called ZDD Recovery (NF perm/RO Conc) in Table 18) simply divides the RO concentrate flow by the calculated NF permeate flow. Recoveries calculated by each of these methods varied by about 4 percent and the variation is caused by differences in the calculated NF permeate flow rate.

Again, the NF system was controlled manually, and the operator was not able to regularly balance all of the flows between systems and account for differences in concentration and temperature. Of course, a full scale system would be operated at constant flow using automated controls. The results of the Brighton pilot study indicate that recovery of 90 percent is achievable with this bolt-on approach. Samples were taken at least once per day and analyzed onsite to monitor performance of the NF and EDM systems. The results are summarized in Figure 61. The variation in the Mixed Na and Mixed Cl stream conductivity and specific gravity and Mixed Na alkalinity relate to the different conductivity setpoints employed throughout the study. The NF permeate was below 800 µS/cm before October 10 and around 900 µS/cm for the remainder of the pilot.

Demonstration of Zero Discharge Desalination (ZDD)

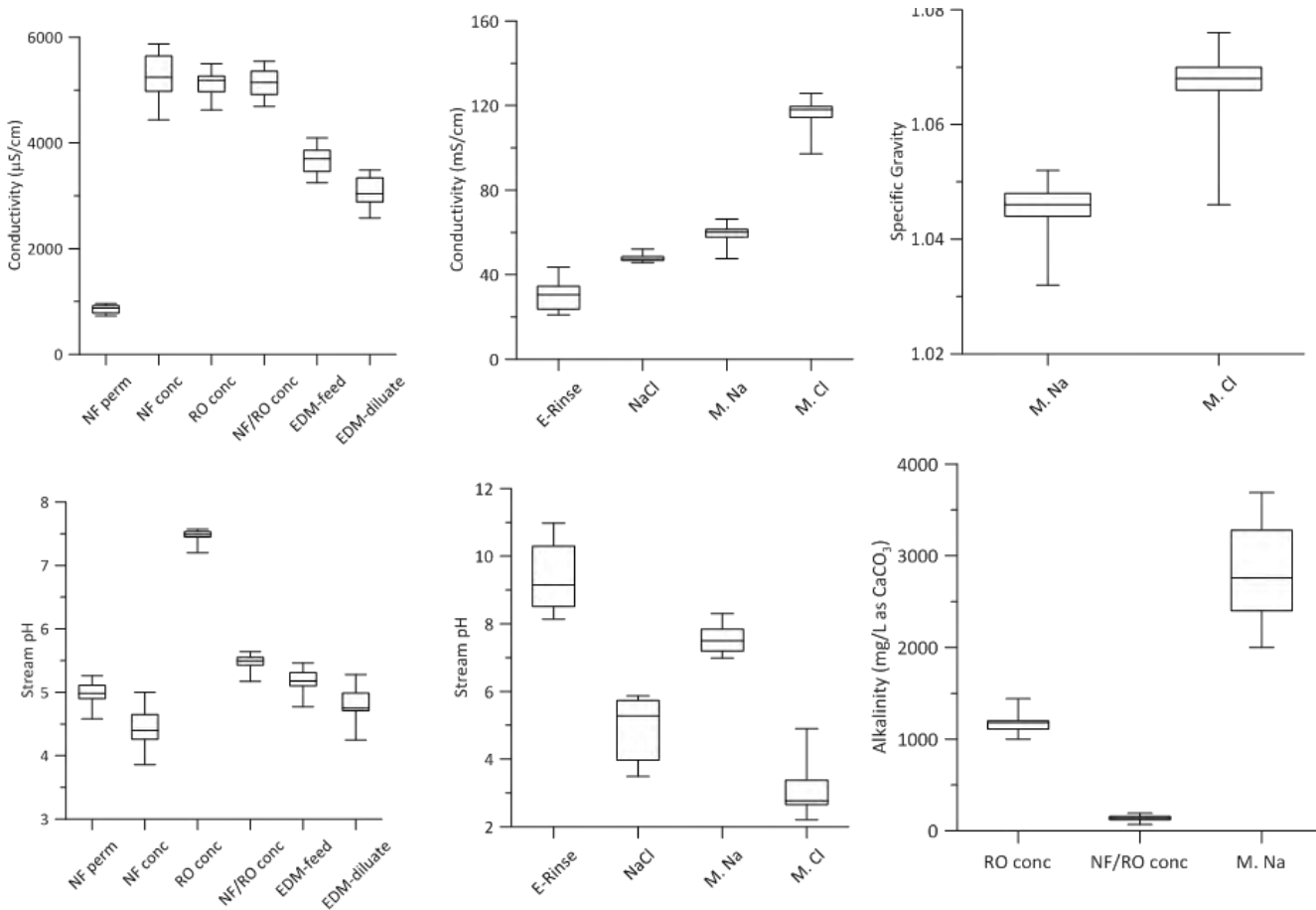


Figure 61.—Brighton grab sample data.

Samples were taken daily for most of the streams (RO concentrate, EDM feed and diluate, Mixed Cl, Mixed Na, and NF permeate) and at least weekly for the remaining streams. These samples were shipped to UTEP for analysis. The average data was summarized in Stiff Diagrams in Figure 62. From these, one can observe that the primary ions in the RO concentrate are calcium, sodium, chloride, sulfate, and bicarbonate. The EDM performed as expected transferring the cations into a stream with chloride salts (Mixed Cl) and the anions into the Mixed Na stream (see bottom figures in Figure 62).

Demonstration of Zero Discharge Desalination (ZDD)

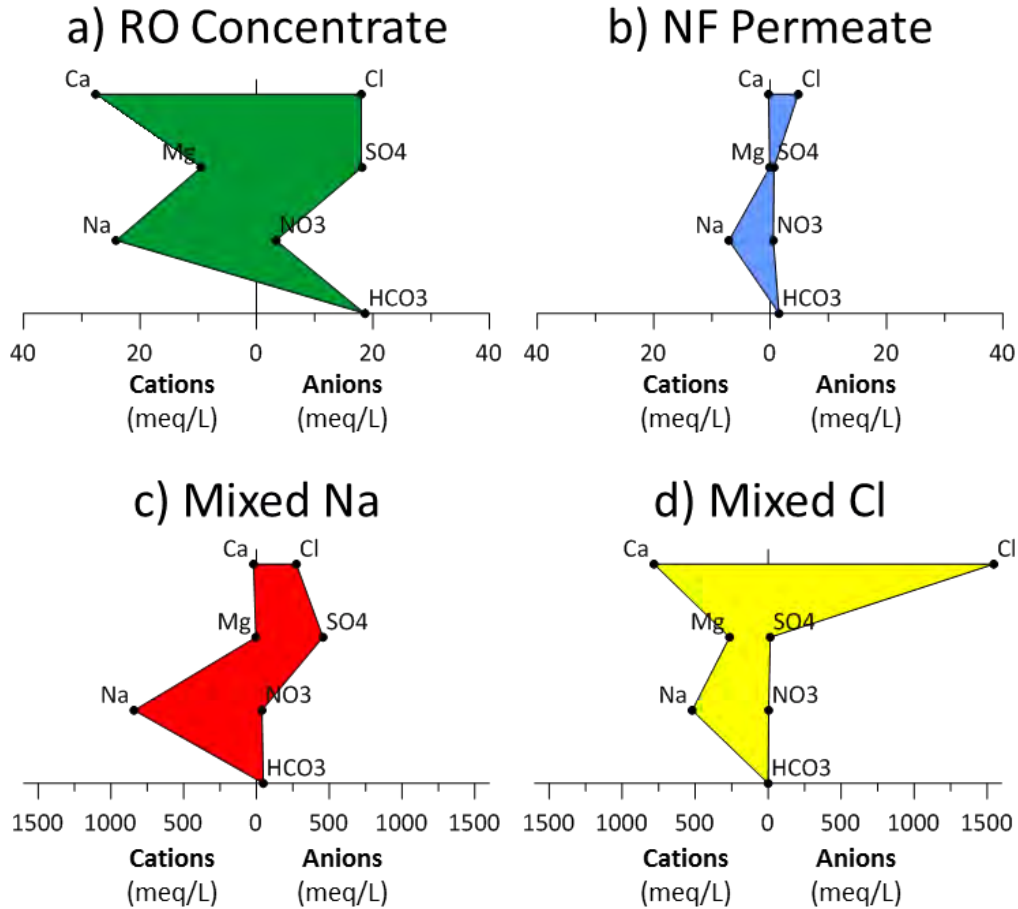


Figure 62.—Average Brighton water quality for ZDD Inlet (a) and outlet streams (b-d).

The variation for each of the main streams studied in Brighton is shown in Figure 63. As expected, the variation in the major ions in the Mixed Cl and Mixed Na streams is explained by the different conductivity setpoints. There is a larger variation and there also seem to be some outliers in the calcium data for the RO concentrate. This may be analytical error, since there isn't a similar degree of variability in the anions. The variability in the monovalent ions is similar to what was seen with conductivity variation for the NF permeate samples. In Figure 64, an upward trend in the RO concentrate is visible during the first run, along with the increase in conductivity of the NF permeate. During the second run, there is a consistent downward trend and a relatively flat trend with higher conductivity during the third run. Water quality for the remaining sample points is provided in Table 19 and Table 20.

Demonstration of Zero Discharge Desalination (ZDD)

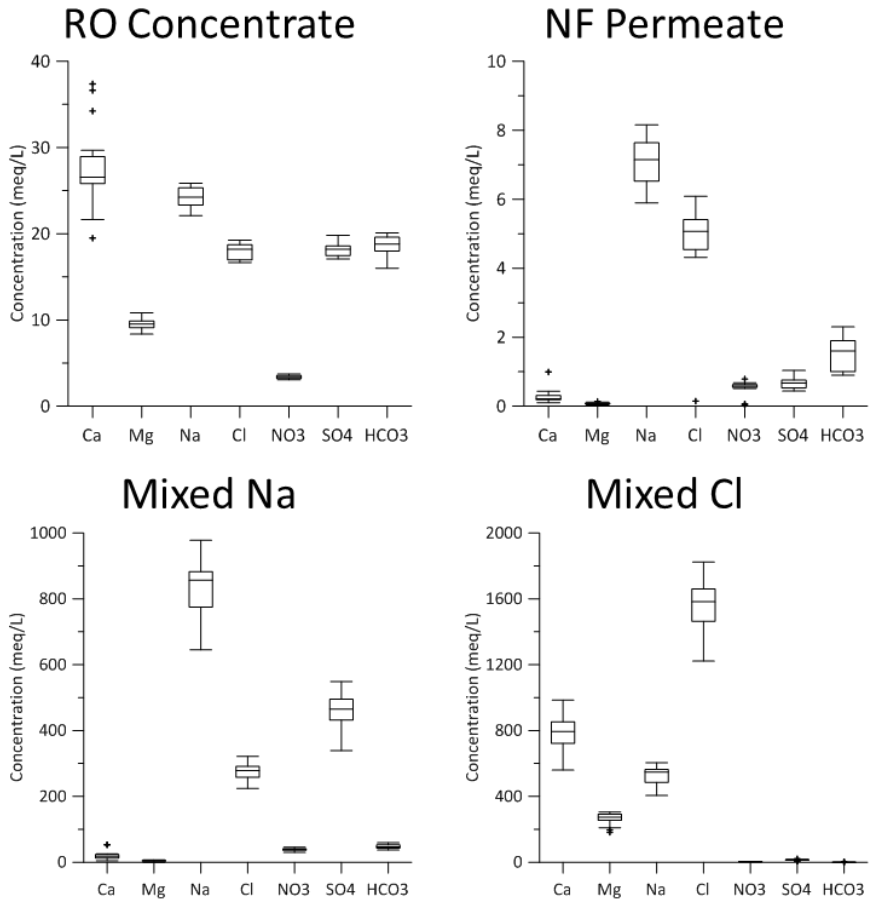


Figure 63.—Variability in Brighton water quality for ZDD inlet and outlet streams.

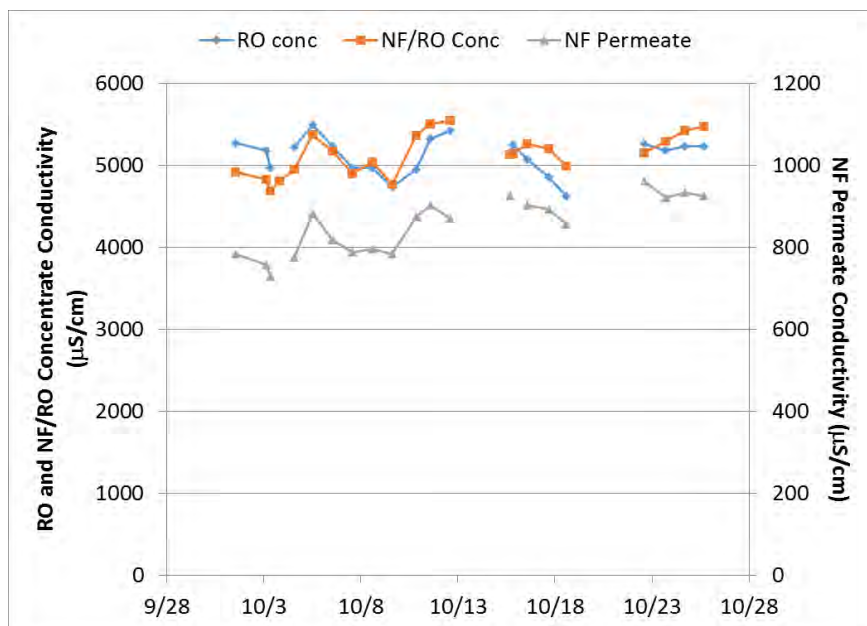


Figure 64.—ZDD inlet and outlet conductivity monitoring (Brighton).

Demonstration of Zero Discharge Desalination (ZDD)

Table 19.—Brighton IC Analyses (all in mg/L)

	Ca		K		Mg		Na		Cl		F		NO3		SO4	
	Avg (Std Dev)	N	Avg (Std Dev)	N	Avg (Std Dev)	N	Avg (Std Dev)	N	Avg (Std Dev)	N	Avg (Std Dev)	N	Avg (Std Dev)	N	Avg (Std Dev)	N
Raw	108.3 (3.66)	3	3.6 (0.35)	3	22.3 (3.52)	3	118.8 (4.15)	3	133.4 (3.32)	3	1.9 (1.36)	3	44.8 (3.25)	3	176.9 (8.48)	3
RO Conc	552.6 (82.7)	20	14.9 (2.04)	20	116.1 (7.27)	20	555 (27.78)	20	638.3 (30.77)	20	7.9 (2.14)	20	209.7 (13.09)	20	870.3 (35.79)	20
NF/RO Conc	449.5 (29.23)	7	14.6 (2.73)	7	99.8 (7.47)	7	724.1 (75.32)	7	374.6 (18.38)	7	6.3 (2.34)	7	119 (6.97)	7	2,164 (214.5)	7
EDM Feed	249.2 (27.21)	20	9.1 (1.58)	20	60.2 (6.85)	20	594.2 (64.74)	20	194.8 (12.86)	20	3.3 (0.84)	20	51 (3.61)	20	1,599 (179.2)	20
EDM Dil	152.2 (22.49)	20	7.3 (1.3)	20	40.4 (6.48)	20	530.9 (67.54)	20	117.2 (11.21)	20	3.1 (0.9)	20	21.9 (2.64)	20	1351 (184.3)	20
Mix Cl	15,636 (2578)	20	273.2 (35.27)	20	3,216 (430.32)	20	11,974 (1403)	20	54,795 (6,159)	20	86.3 (38.36)	20	212 (40.44)	20	678.7 (146.2)	20
Mix Na	397.4 (243.51)	20	47.6 (8.14)	20	52.3 (24.46)	20	19,346 (1971)	20	9,733 (950.2)	20	96.7 (45.72)	20	2,377 (245.3)	20	21,974 (2,558)	20
E-Rinse	119.8 (168.98)	4	11.7 (6.31)	4	3.8 (4.75)	4	8,148 (3444)	4	271.8 (118.6)	4	17.8 (19.8)	4	445.7 (454.3)	4	14,729 (6,443)	4
NaCl	209.6 (394.4)	4	20 (5.08)	4	14 (3.15)	4	12,050 (397.16)	4	18,257 (666.5)	4	21.5 (22.54)	4	47 (10.6)	4	361.9 (205.2)	4
NF Conc	330.1 (38.97)	4	14.4 (3.72)	4	83.8 (8.04)	4	984.5 (133.14)	4	45.3 (10.94)	4	3.2 (2.39)	4	5.6 (0.66)	4	2,959 (355.6)	4
NF Perm	5.36 (3.60)	20	2.69 (0.33)	20	0.96 (0.29)	20	163.16 (15.26)	20	171.27 (41.55)	19	1.55 (0.46)	20	34.18 (10.74)	20	32.12 (7.62)	18
RO Perm	8.53 (17.33)	4	0.64 (0.67)	4	1.91 (3.87)	4	30.54 (46.24)	4	5.92 (1.65)	4	0.5 (0.27)	4	5.56 (3.26)	4	68.92 (146.77)	4

Demonstration of Zero Discharge Desalination (ZDD)

Table 20.—Brighton General Water Quality Analyses

	pH		TDS* (mg/L)		Conductivity (μ S/cm)		SiO ₂ (mg/L)		Alkalinity (mg/L as HCO ₃)	
	Avg (Std Dev)	N	Avg (Std Dev)	N	Avg (Std Dev)	N	Avg (Std Dev)	N	Avg (Std Dev)	N
Raw	7.4 (0.11)	4	817.5 (55)	4	1,410 (66.19)	4	20.8 (0.5)	4	294.3 (17.52)	4
RO Conc	7.6 (0.05)	22	4,065 (211.6)	22	5,666 (304.5)	22	99.1 (5.23)	22	1,138 (64.34)	21
NF/RO Conc	5.7 (0.12)	9	4,686 (428)	9	5,643 (349.4)	9	109.8 (6.67)	9	167.4 (31.27)	9
EDM Feed	5.5 (0.31)	22	3,339 (332.5)	22	4,142 (313.5)	22	109.6 (6.37)	22	109.5 (28.25)	21
EDM Dil	5 (0.42)	22	2,725 (341.7)	22	3,442 (344.2)	22	110.4 (6.13)	22	79.6 (35.3)	20
Mix Cl	3.1 (1.02)	22	106,082 (10844)	22	128,336 (11,995)	22	10.1 (1.17)	22	7.8 (20.9)	22
Mix Na	7.7 (0.29)	22	59,416 (5770)	22	65,164 (5,443)	22	11.2 (2.11)	22	2,876 (371.4)	22
E-Rinse	9.3 (0.94)	6	30,308 (10763)	6	30,810 (10,970)	6	2.3 (1.03)	6	244 (29.6)	6
NaCl	4.6 (1.36)	6	31,738 (1235)	6	53,833 (2,917)	6	5.2 (1.47)	6	71.2 (78.9)	6
NF Conc	4.4 (0.53)	6	5,328 (705.6)	6	6,053 (645)	6	127.3 (8.91)	6	37.6 (38.66)	5
NF Perm	5.44 (0.52)	22	547.1 (131.5)	21	923.7 (75.86)	22	99.64 (5.778)	22	92.11 (26.91)	20
RO Perm	5.96 (0.13)	5	265 (197.96)	5	62.92 (13.04)	5	1.4 (0.55)	5	82.96 (18.1)	5

*TDS samples are not filtered before being placed in oven.

8.7 Evaluation of Current Density, Current Efficiency, Power Consumption – Brighton

Intentional experiments were not performed to identify the limiting current density at Brighton due to time limitations, but useful data can be obtained from the startup data. The EDM voltage was slowly increased during startup and the datalogger recorded all of the data that would be needed to determine the limiting current density for the flow rate used throughout the study. Data from five startups were graphed (Figure 65 is from 10/15/13). The intersection between the ohmic and limiting sections was calculated for each set of data and averaged. The limiting current density using the shoulder method was determined to be 59.6 A/m².

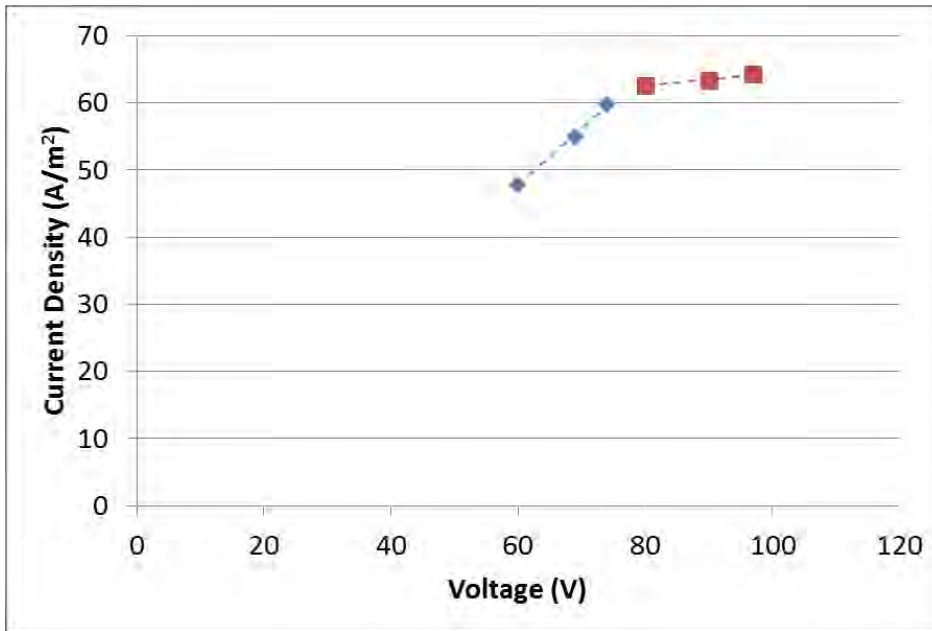


Figure 65.—Typical limiting current density – Brighton (7 cm/s velocity).

8.8 EDM Stack Performance Monitoring

During each of the runs EDM performance declined, as measured by stack resistance and EDM diluate conductivity. The cause seemed to be related to temperature, as ED systems perform better at warmer temperatures, but a direct correlation can't be shown with the data obtained. In some cases, decreasing temperature led to lower current efficiency, but in others, it did not. Instead of having a direct effect, the lower temperatures may have contributed indirectly. RO and NF membranes produce better quality permeate (and higher conductivity

concentrate) at lower temperatures. The combination of lower temperature and overnight flow imbalances tended to lead to imbalances in ions entering and leaving the system. A mass balance using the measured or calculated flow rates and TDS analyses showed a gradual upward trend—suggesting that more ions were entering than leaving during the first run and that the performance was more stable during the final two runs. The current efficiency (calculated based on ions removed from the EDM feed and diluate), daily outdoor low temperature (at nearby Denver airport), mass balance, and EDM stack resistance are graphed in Figure 66. Since the second and third runs were shorter than the first it is unclear whether the stack resistance would have stabilized.

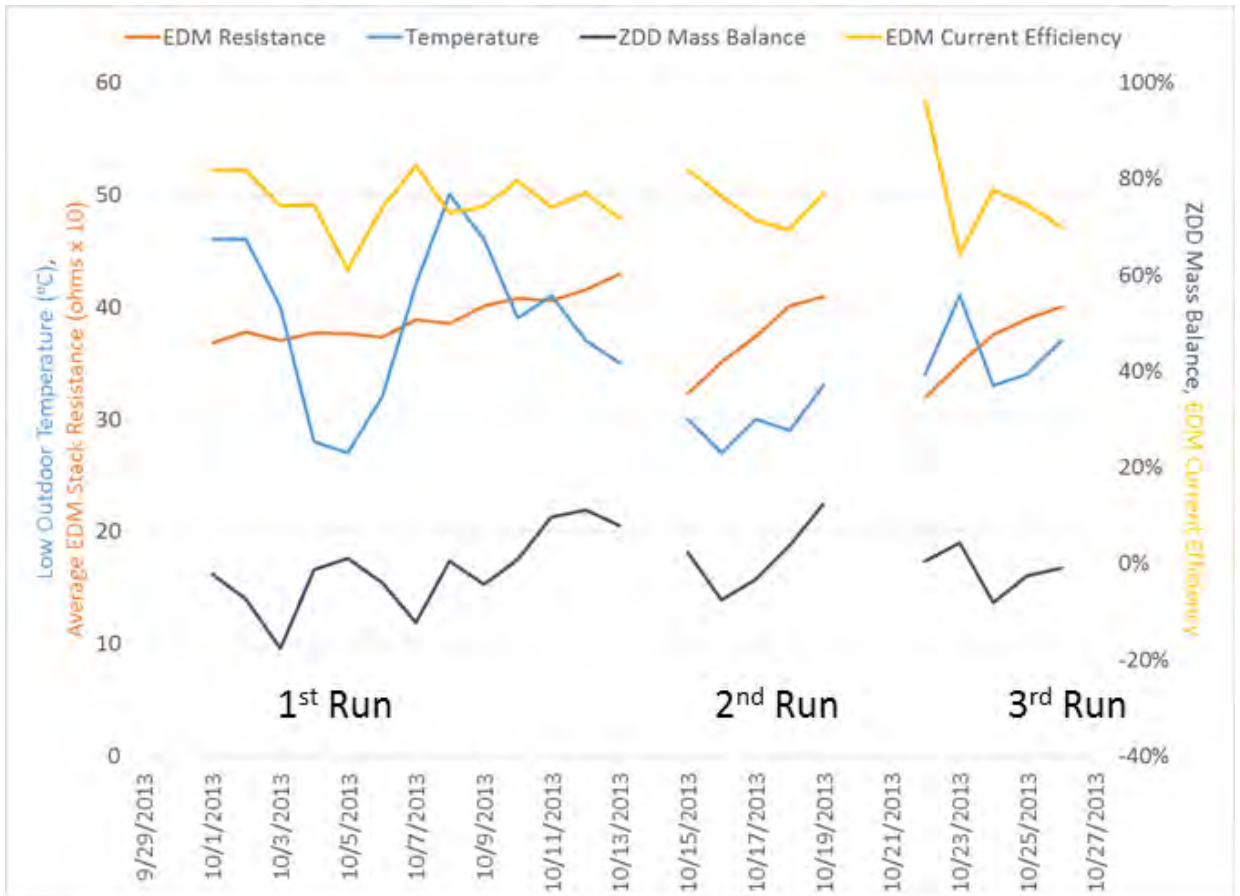


Figure 66.—Brighton EDM performance and outdoor temperature data near Brighton (Denver International Airport (DIA), Denver, Colorado).

The stack was operated beyond the limiting current density identified using the shoulder method. This led to an increased difference between the Mixed Cl and Mixed Na pH values, which is expected to be caused by water splitting (see Figure 67).

Demonstration of Zero Discharge Desalination (ZDD)

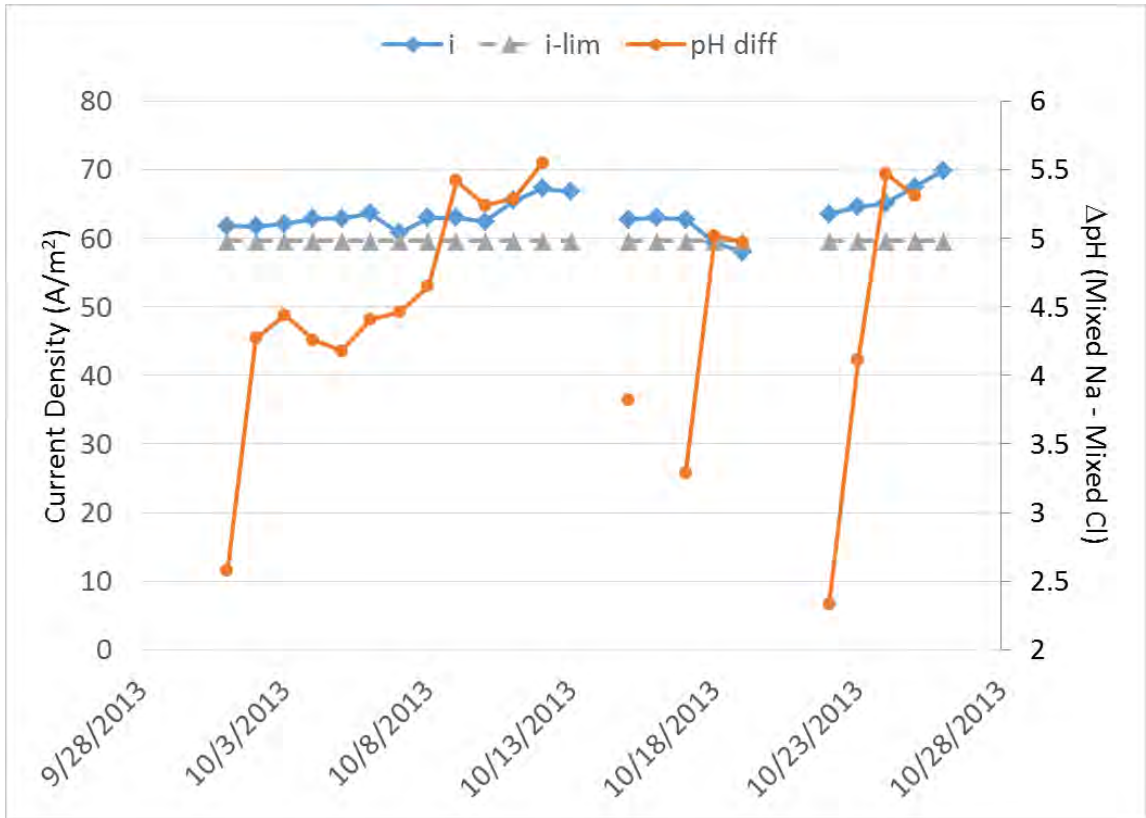


Figure 67.—Current density and pH difference (Brighton).

Using methodology similar to what was described for Phase 2 operations in Alamogordo, the current efficiency for the EDM stack was calculated using four methods for each day with reliable water quality, NaCl consumption, and flow rate data (22 sets of data for each method). Average data for the 22 sets are summarized in Table 21 and the range of the calculated results is shown in Figure 68. The individual method results and average of the method yields lower average current efficiency than what was calculated for Alamogordo. The lower current efficiencies were attributed to operation above the limiting current density, cold temperature effects, and possibly back diffusion.

Table 21.—Average Current Efficiency (Brighton).

Method	Average Current Efficiency
EDM Feed	76%
NaCl	73%
Mixed Cl	62%
Mixed Na	71%
Average	70% (74%-EDM feed & NaCl only)

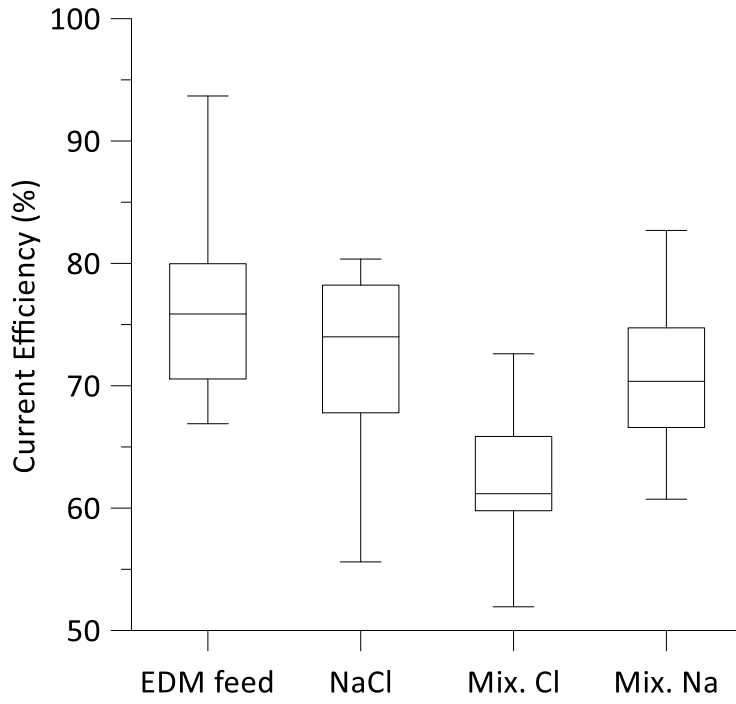


Figure 68.—Brighton current efficiency.

9. Beverly Hills Pilot

The EDM container used in Phase 2 operations in Alamogordo was shipped to the Beverly Hills Water Treatment Plant for a pilot project led by Black & Veatch and funded by the California Energy Commission (Bond et al., 2015). The experimental objectives differed somewhat from the ZDD demonstrations and pilots in Alamogordo and Colorado, but some observations can be made from the work performed by Black & Veatch. Some raw operating data and water analyses were shared with UTEP for inclusion in this report.

- The EDM stack used for the pilot studies at Beverly Hills still had the swollen AEMs between the NaCl and Mixed Cl compartments for the experiments performed in fall 2013. Two quads of membranes were shipped from Brighton to the manufacturer, MEGA, for analysis. Calcium sulfate and possibly calcium phosphate (from antiscalant) was present inside the anion exchange modules and MEGA recommended replacement. The swollen membranes were replaced in February 2014 in the one stack operated at Beverly Hills.
- The new center divider, installed after Alamogordo Phase 2 operations, seemed to correct problems with major internal leakages. The stack did not show signs of extensive scale formation during the project. Some internal leaks seemed to be present based on sulfate and bicarbonate being present in the NaCl and Mixed Cl streams. At the beginning of the Beverly Hills pilot, the amount of sulfate in both the Mixed Cl and NaCl streams was high. The source of the contamination was not identified or reported by the project team; however, it is possible that small leaks were still occurring at the center divider or elsewhere in the stack or that the RO system used to prepare process water was not performing adequately. Figure 69 shows the concentrations of sulfate and bicarbonate in mg/L (left) and the Na/Ca and Cl/SO₄ ratios (right; based on concentrations in meq/L).
- High recovery was demonstrated with the EDM system. The EDM recovery reported in the California Energy Commission report is calculated differently from the previous studies. EDM recovery is defined as the EDM diluate volume divided by the EDM feed (Beverly Hills RO concentrate). Total system recovery is defined as the EDM diluate flow rate divided by the sum of the EDM feed rate (RO concentrate) and DI water used for NaCl makeup and dilutions. The total system recovery method is expected to provide similar recovery as the method used in this report. Black & Veatch reported total system recovery of 91 to 92 percent, which would boost the Beverly Hills RO plant recovery to 98 percent or

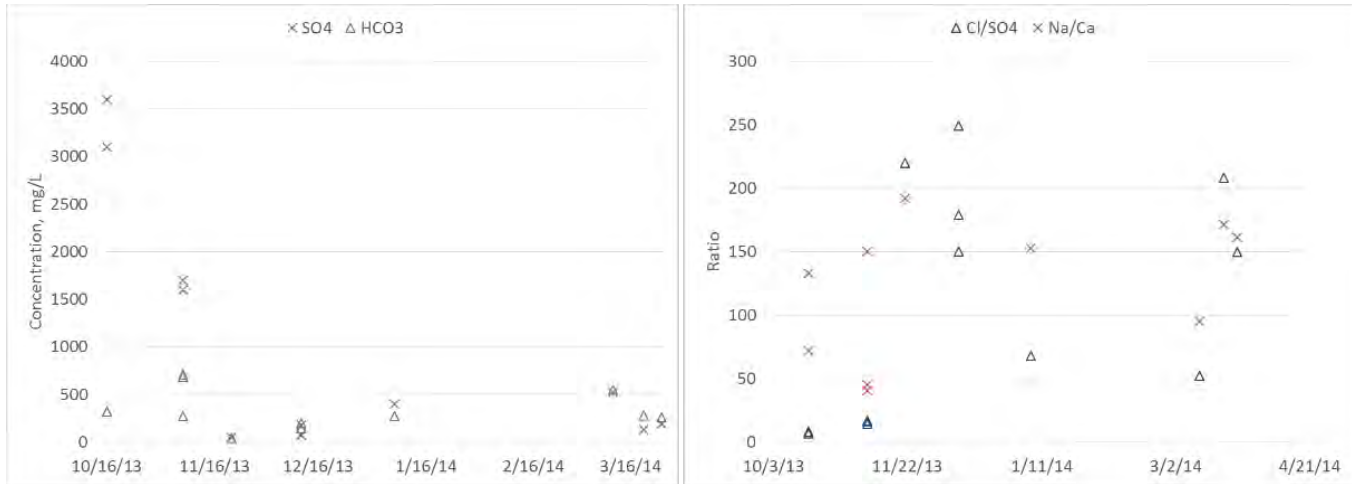


Figure 69.—NaCl contamination – Beverly Hills pilot.

higher. A substantial amount of DI water was used to dilute the Mixed Na Stream to prevent carbonate- and bicarbonate-based scale formation. If sulfuric acid were to be added to the RO concentrate stream to reduce bicarbonate levels, higher recovery is possible.

- Black & Veatch acknowledged that the complexity of EDM is higher than that of conventional water treatment and suggested that, “Concentrate treatment is inherently more complex than conventional water treatment, because concentrations of contaminants that must be removed are four to five times greater” (Bond et.al, 2015). Additionally, the report stated that EDM has two advantages. First, EDM is capable of achieving recovery comparable to thermal approaches while requiring substantially less energy. Second, EDM allows for beneficial recovery of saleable salts, which could reduce concentrate treatment costs.
- Black & Veatch developed an EDM cost model and performed calculations for three sites in California, including Beverly Hills, Santa Rosa, and Arcadia. Selected details from the model inputs and outputs are summarized in Table 22.
- EDM was described as a “promising solution for concentrate management” and Black & Veatch recommended that utilities test it on their water to best understand if EDM (or other technologies) will meet the needs of the utility with respect to cost, complexity, and sustainability (Bond et.al., 2015).

Table 22.—Black & Veatch EDM Study Summary (Bond et al., 2015)

	Units	Beverly Hills	Santa Rosa	Arcadia
# of EDM Stages		1	3	1

Demonstration of Zero Discharge Desalination (ZDD)

	Units	Beverly Hills	Santa Rosa	Arcadia
RO Concentrate flow	gpm	300	550	270
TDS	mg/L	3,190	4,791	4,333
Conductivity	mS/cm	4.126	5.99	5.53
% Ca, Mg to cations*	%	42.6%	38.9%	72.7%
% SO ₄ to anions*	%	26.2%	25.9%	37.0%
Projected EDM Recovery	%	95%	89%	93%
New Plant Recovery	%	99%	97%	99%
RO Concentrate Treatment Cost	\$/kgal	\$2.28	\$4.48	\$2.61

\$/kgal = dollars per thousand gallons

10. Cost Analysis

As part of the Reclamation's Alamogordo Phase 1 Study, a cost estimate was developed for a 3.6 MGD production ZDD system (Biagini et.al, 2012). These cost estimates were considered preliminary at that point due to the lack of long-term ZDD operating data. Cost information was updated as part of the Phase 2 Study using additional ZDD pilot operating data at BGNDRF, La Junta, Brighton, and Beverly Hills, California; project-specific equipment quotes for ZDD controls and ancillaries; and a new EDM skid design which minimizes onsite installation time.

Alamogordo is planning to build desalination capacity in stages, starting with a 1 MGD plant to replace the loss of its Bonito Lake surface water supply. This will be installed at a location that will allow Alamogordo to use the existing Bonito Lake pipeline with minimal cost to connect. Later, the city is planning a 4 to 5 MGD desalination plant to be located across the street from the BGNDRF. Cost estimates were updated for 1 MGD production based on the average Alamogordo Snake Tank water quality (see Table 5). Since the EDM and NF systems are modular, these costs were scaled-up to 4 MGD. Estimates for both are presented in this chapter, along with the conceptual drawings of the treatment process for Alamogordo.

10.1 Cost Development Approach

This report's objectives were to provide study or feasibility-level costs. The American Association of Cost Engineers (AACE) defines five categories of estimates that describe the accuracy range for cost estimates (see Table 23). It is expected that an estimate of this type would be a Class 4 estimate (accurate within -15 percent to +30 percent). Project-specific process and instrumentation diagrams were developed, and ancillaries were specified and selected based upon years of operating data at the pilot-scale. However, quotes for the EDM skid were based upon designs of similarly-sized ED systems from MEGA instead of mechanical design drawings specific to the EDM skids. Additional details on the selection of the estimated class are provided in the AACE International Recommended Practice document (AACE, 2000). Capital and operating costs could change, depending on the actual brackish water feed or on further optimization of the EDM to reduce NaCl and/or energy consumption.

Demonstration of Zero Discharge Desalination (ZDD)

Table 23.—AACE Cost Estimation Classification (AACE, 2000)

Estimate Class	Level of Project Definition ¹	End Usage – Typical Purpose Of Estimate	Typical Budget Estimate Accuracy
Class 5	0 to 2%	Concept screening	-30% to +50%
Class 4	1 to 15%	Study or feasibility	-15% to +30%
Class 3	10 to 40%	Budget, amortization or control	-15% to +30%
Class 2	30 to 70%	Control or bid	-5% to +15%
Class 1	50 to 100%	Check estimates on bid	-5% to +15%

¹ Expressed as % of complete definition.

Cost information was requested from vendors for the 1 MGD design:

- MEGA EDM stacks
- Veolia-Dayton for NF skids
- Bardac DC drive
- Bryneer salt make-up system
- Verderflex NaCl slurry pumps
- Prominent chemical dosing pumps
- Grundfos pumps (NF and EDM)

The economic analysis assumes the following multipliers:

- Install cost: 1.4 times equipment price
- 20-year present worth of operating costs: 11.47 times annual operating cost (or, 20 years at 6 percent)

Typical flow diagrams for a ZDD and BWRO system are shown in Figure 70 and Figure 71. Letters are placed at key positions where key design information is provided for economic analysis. Recovery is defined as product flow (G) and well water (A) in Figure 70 and Figure 71 for both designs. Key labels are included and are incorporated into the operational cost summary in **Error! Reference source not found.**Table 24.

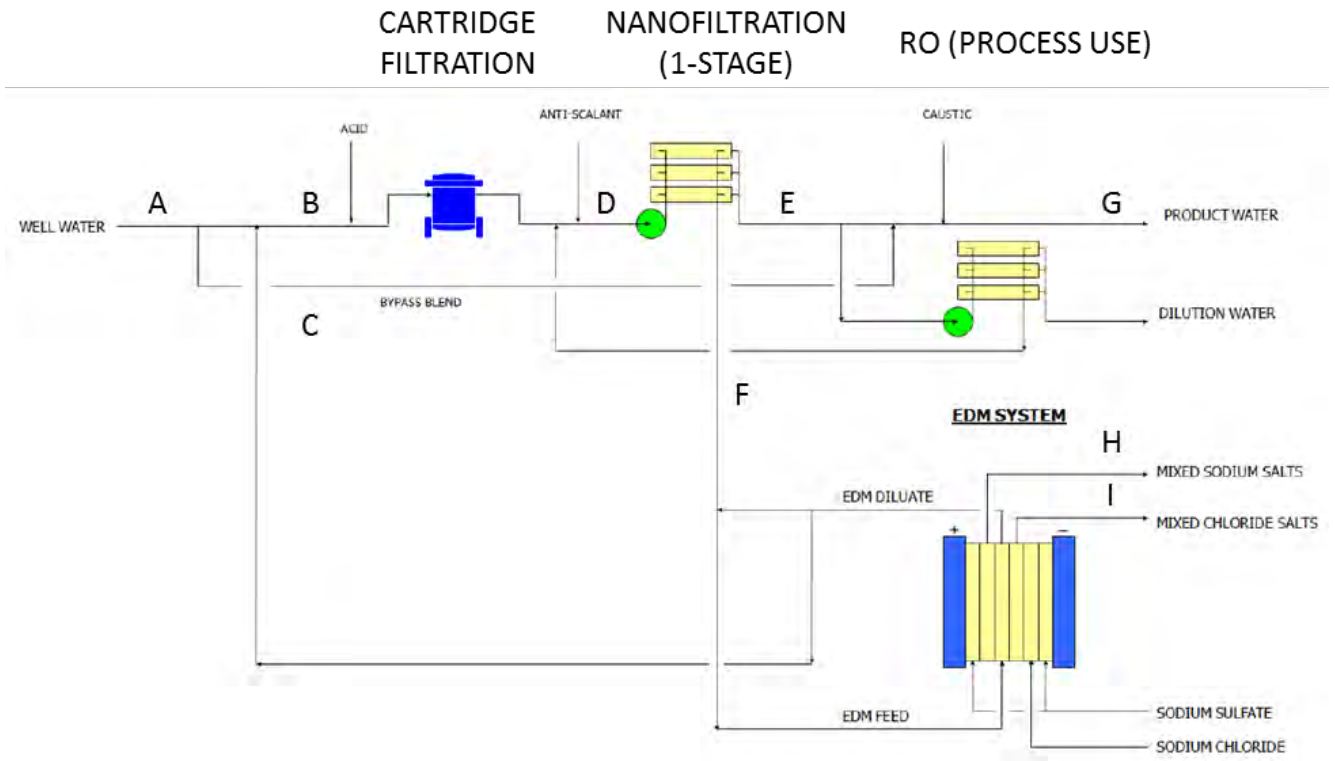


Figure 70.—Process flow diagram of a ZDD system.

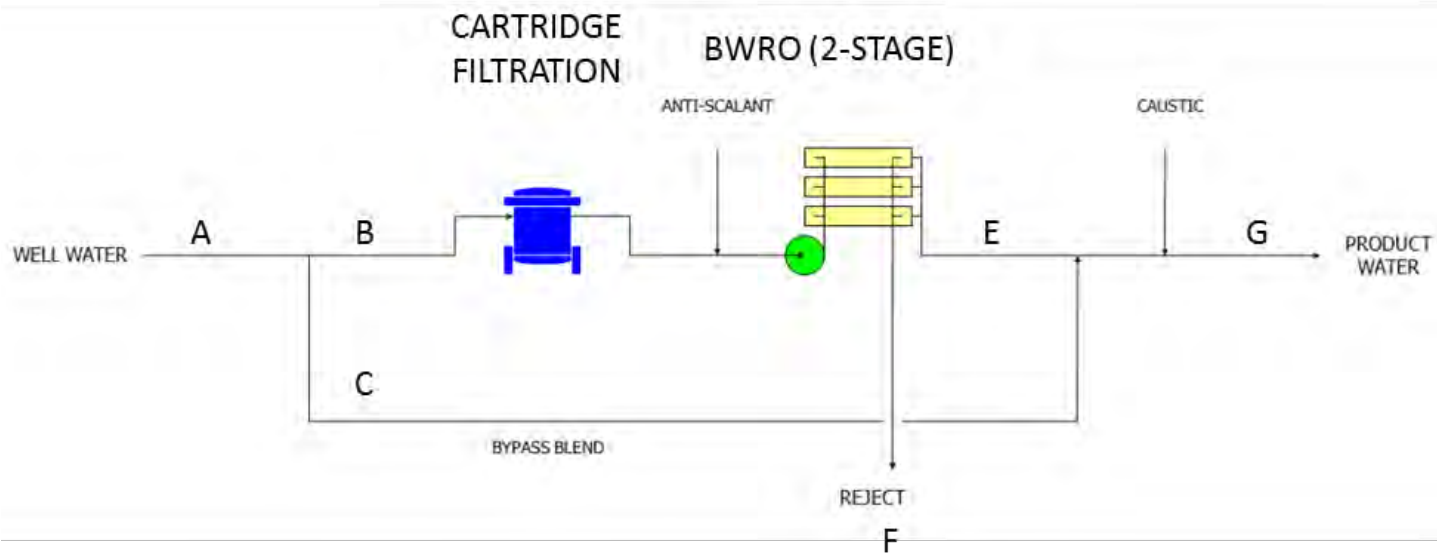


Figure 71.—Process flow diagram of a BWRO system.

Demonstration of Zero Discharge Desalination (ZDD)

Table 24.—Major Operational Cost Comparison by Product Flow Rate and System (Alamogordo)

Location	1 MGD		4 MGD	
	ZDD	BWRO	ZDD	BWRO
A (BW feed)	709 gpm / 0 psi	868 gpm / 0 psi	2,835 gpm / 0 psi	3,473 gpm / 0 psi
B (BW feed to ZDD/BWRO)	561 gpm / 30 psi	720 gpm / 30 psi	2,263 gpm / 30 psi	2,875 gpm / 30 psi
C (Bypass)	148 gpm / 30 psi	150 gpm / 30 psi	572 gpm / 30 psi	600 gpm / 30 psi
D (NF/RO feed)	950 gpm / 90 ¹ psi	720 gpm / 174 psi ¹	3,800 gpm / 90 psi ¹	2,875 gpm / 174 psi ¹
E (NF/RO permeate)	570 gpm / 0 psi	544 gpm / 0 psi	2,280 gpm / 0 psi	2,178 gpm / 0 psi
F (NF/RO concentrate)	380 gpm	176 gpm	1520 gpm	697 gpm
G (Product)	694 gpm / 30 psi	694 gpm / 30 psi	2,778 gpm / 30 psi	2,778 gpm / 30 psi
H + I (ZDD Waste)	17.7 gpm / 25 psi		55.6 gpm / 25 psi	
PROCESS RO design	25 gpm / 90 psi ²		75 gpm / 90 psi ²	
EDM pumps (total)	835 gpm / 25 psi		3,339 gpm / 25 psi	
EDM E-rinse pumps (total)	209 gpm / 25 psi		209 gpm / 25 psi	
EDM Power Supply	24 @ 140 V _{DC} /46 A		93 @ 140 V _{DC} / 46 A	
EDM NaCl use	12,205 lb/day		48,350 lb/day	

¹ The pressure listed is the pressure estimated by Dow's ROSA program; however, the pressure in the cost model accounts for the pressure in the BW feed (NF feed pressure = ROSA value – BW feed)

² The cost model is based on the design flow, but actual flow (and energy) is expected to be lower

10.2 Technology Cost Estimates – Alamogordo

10.2.1 Treatment Process Descriptions – Alamogordo

ZDD with 98 percent recovery was compared against a conventional BWRO with 80 percent recovery. Both designs include a bypass blend intended to improve the stability of the RO/NF permeate. The ZDD system consists of an NF membrane system, an EDM system, and all ancillary feed tanks and pumps. The BWRO

system consists of a RO membrane system and all ancillary feed tanks and pumps. Snake Tank brackish water is fed to each system. Sulfuric acid is added to reduce the pH to 7 on the ZDD system and both systems include antiscalant added to the NF or RO feed (2 mg/L). The product water flow is the same for each BWRO and ZDD system, but the amount of brackish groundwater differs because of the difference in recovery.

Both ZDD systems include a single stage NF with a blend of Dow Filmtec XLE-440 (first three) and NF270-400 (last five) membranes. The 1 MGD design includes a single skid with 16 pressure vessels and the 4 MGD design includes four skids, each with 16 pressure vessels. The mixture of the XLE and NF270 membranes allows for the flux and product water quality necessary for blending with the bypassed brackish water. A portion of the NF permeate is fed to a BWRO system for process water production. This water is used for dilutions and salt preparation for the EDM. The system is sized to equal the full waste flow (2 percent of the brackish water feed). Chemical costs include the initial sodium sulfate charge for the E-Rinse as well as daily consumption for sulfuric acid, antiscalant, and caustic (post treatment) for the NF and NaCl for the EDM. The NF concentrate is fed to the EDM system and the EDM diluate (and concentrate from the Process RO) is blended with acidified brackish water and fed to the NF. The NF is designed for 60 percent recovery, which reduces the risk of scale formation. The EDM is design to reduce the concentration of the NF concentrate by 50 percent and nearly all of the NF reject is returned to the NF system (>99.5 percent recovery).

The BWRO system includes a two-stage BWRO with Dow Filmtec BW30-400/34i membranes in a 12x6 array with 6 membranes in each pressure vessel. The 1 MGD design includes a single skid and the 4 MGD design includes four skids, each with 16 pressure vessels. There is an extra skid included with the 4 MGD design to allow for maintenance on the other skids. The BW30-400/34i membranes were chosen because of their relatively high flux as well as having a larger spacer. Chemical costs include antiscalant and caustic (post treatment). The BWRO is operated at 76 percent recovery, which is technically feasible for commercially available antiscalants; however, it is close to the limit on calcium sulfate solubility (concentrate is estimated to be at 320 percent of saturation).

10.2.2 Capital Cost Development – Alamogordo

As mentioned previously, BWRO and ZDD designs were developed for a 1 MGD ZDD system, and were modularly scaled-up to 4 MGD. The BWRO capital cost is simply based on a design cost of \$2.23/GPD obtained from the Alamogordo Environmental Impact Study (SWCA, 2012). The ZDD includes an NF and EDM system. The 1 MGD NF system capital cost (CapEx) includes a Veolia NF package system containing:

- Single-stage NF unit, including 16 - 8 M vessels, each loaded with three XLE-440 membranes followed by five NF 270-400 membranes. Also

Demonstration of Zero Discharge Desalination (ZDD)

included on this skid are a horizontal centrifugal high pressure pump and two 100 percent cartridge filters.

- NF feed pump skid, including two horizontal centrifugal pumps.
- NF permeate forwarding pump, including two 100 percent permeate forwarding pumps.
- NF CIP skid with tank, pump, heater, and cartridge filter.
- 48 Dow XLE-400 and 80 Dow NF270-400 membranes
- Sulfuric acid and antiscalant dosing pumps

The 1 MGD EDM system consists of 24 EDM stacks configured into skids of six quads per skid. Each EDM stack has 120 quads, stainless steel frame, membranes, spacers, and electrodes, all manufactured by MEGA. Each EDM stack has its own power supply (DC drive manufactured by BARDAC). All of the stack skids share a common feed pump and feed tank for each of the five compartments in the stack:

- EDM feed/NF concentrate tank (3,500-gallon), and one duty plus one standby 835 gpm pump
- E-Rinse (Na_2SO_4) tank (600-gallon), and one duty plus one standby 209 gpm pump
- Sodium chloride storage tank (2,500-gallon), and one duty plus one standby 850 gpm pump
- Mixed sodium tank (2,500-gallon), and one duty plus one standby 850 gpm pump
- Mixed chloride tank (2,500-gallon), and one duty plus one standby 850 gpm pump

The 4 MGD ZDD system includes four 1 MGD NF skids with one backup 1 MGD skid. The EDM includes 16 EDM skids with 93 stacks total. Tanks, controls, and other ancillary systems were scaled up for the 4 MGD design.

Each of the EDM recirculating/transfer pumps delivers a pressure of 25 psi to the stack, and provides enough residual pressure to return the streams to the storage tank, creating one large recycle loop from the tank, through the stack, and back to the tank. Automatic valves on each skid ensure even distribution among the stacks, based upon pressure readings for each stream. ED stacks are operated with equal pressure on each side of the membranes to mitigate internal leakage and prevent membrane damage. As with the pilot and demonstration systems, the

NaCl, Mixed Na, Mixed Cl, and E-Rinse pressures will be set to match the EDM feed pressure.

There are make-up tanks for sodium chloride and sodium sulfate solutions, which will deliver to site as dry products. The NaCl brine make-up system for the 1 MGD Alamogordo design is a Bryneer system of 42-ton capacity (84,000 lbs); the 4 MGD system will include four 42-ton capacity Bryneer systems.

There is also a degasifier fan and tower to remove hydrogen and oxygen gas produced at the EDM electrodes.

A secondary RO system provides process water to dilute the Mixed Na and Mixed Cl streams. The system is designed to equal the waste flow from the EDM (2 percent of the brackish water feed).

Instrumentation and controls include rotameters, flow switches, and pressure switches for every EDM stream on every stack, ancillary instruments for tanks and pumps, master control panel/SCADA, and skid remote input/output (I/O) panels.

Flush pumps, tank heaters, and CIP pumps are provided for both the EDM and NF systems.

Equipment to transfer water from the brackish wells to the NF feed pump are not included in the Veolia pricing estimate, since these costs vary widely on every site; however power required for this transfer is included in the operation and maintenance cost development (OpEx) evaluation. Additionally, the air compressor required for automatic valve actuation is not included.

10.2.3. Operation and Maintenance Cost Development – Alamogordo

Operating cost assumptions used for the ZDD and/or BWRO systems include the following:

- Power cost: \$0.08/kWh flat rate
- EDM sodium chloride: \$0.075 per pound (lb)
- Annual CIP for NF lasting one hour, includes operation of NF flush pump, CIP pump, CIP heater (same for BWRO and ZDD)
- Monthly CIP for EDM systems, lasting one hour per skid, includes operation of NF flush pump, CIP pump, CIP heater
- Antiscalant: 2 mg/L dosage: \$2.95/lb (same for BWRO and ZDD)
- Sulfuric acid for reducing bicarbonate in feed: $\text{pH} \leq 7.0$, \$0.20/lb

Demonstration of Zero Discharge Desalination (ZDD)

- Sodium sulfate for the E-Rinse: \$0.08/lb
- Caustic post treatment: 5 mg/L dosage, \$0.45/lb (same for BWRO and ZDD)
- NF and EDM CIP chemicals: <\$0.01/1,000 gallons product (same for BWRO and ZDD)
- Source water cartridge filters replaced monthly for annual cost: \$38,720 for the 1 MGD system and \$69,750 for the 4 MGD (same for BWRO and ZDD)
- NF and RO membranes replaced every five years: \$600/element
- EDM membranes and spacers will be replaced every 10 years: \$354/quad
- EDM electrode replacement schedule includes replacing the anodes every two years and cathodes every five years.

The ZDD O&M costs are dominated by the EDM power cost (77 percent of annual power expenditure) and sodium chloride (91 percent of annual chemical cost; 42 percent of annual OpEx). The EDM design is conservative, and additional optimization is possible to improve the energy efficiency by increasing the current density (or decreasing stack resistance) or reducing the recirculating rate for the EDM pumps. Although not tested at pilot scale, cost savings may be realized if sodium chloride is recovered by any of the methods suggested in this report; this will be tested in future demonstration studies. While not included in the cost estimate, there is an expected avoided cost by way of power savings with ZDD since the brackish water pumps will pump less water to achieve the same amount of product water. This savings is estimated to be \$31,000/year for the 1 MGD design and \$124,000/year for the 4 MGD design. This only includes the power required to lift water from 1,000 feet. Additional savings may be realized if additional pressure is required by the well pumps or if booster pumps are required for transporting on the 32 mile pipeline from the Snake Tank Wells.

The major flows, pressures, EDM power supply, and NaCl consumption are summarized for the two design flow rates in Table 24. Process flow diagrams are included as Figure 70 and Figure 71. Key labels are included and are incorporated into the operational cost summary in **Error! Reference source not found.** Table 24. The recovery is calculated by dividing the product water by brackish feed water (G/A in Table 24).

10.2.4 ZDD & BWRO Residuals Cost Development – Alamogordo

All of the BWRO concentrate and only the EDM waste streams will require final disposal, likely using evaporation ponds or deep well injection. The residual streams from the ZDD consist of a total of 2 percent of the brackish feed, in the

form of two high-concentration streams called the Mixed Cl stream and Mixed Na streams. For the 1 MGD system, the combined ZDD waste flow is 17.7 gpm and the 4 MGD is 55.6 gpm (see H+I in Table 24 **Error! Reference source not found.**). The BWRO concentrate is 176 gpm for the 1 MGD system and 697 gpm for the 4 MGD system (see Faraday's constant (F) in Table 24). ZDD achieves a 92 percent reduction in waste volume compared to BWRO.

Four evaporation pond estimation methods were evaluated and averaged for the final disposal of BWRO and ZDD waste flows. Costs are in normalized to 2013 dollars and were updated using the Producer Price Index for capital equipment (Bureau of Labor Statistics, n.d.). The capital costs for the evaporation pond used for this model is the average of the values estimated using the Voutchkov method and the scaled versions from El Paso (design cost) and BGNDRF (actual cost). The method described by Mickley (Mickley and Associates, 2006) allows for multiple dike heights, so a four foot and 12 foot height model are included. The net evaporation rate for Alamogordo used in the model is 14.8 cm per month, based upon evaporation rates from nearby Holloman Air Force Base (Livingston Associates, P.C., 2003). The estimated cost for evaporation ponds for each system is summarized in Table 25.

Table 25.—Comparison of Evaporation Pond CapEx Estimation Costs (\$ Million)

System	1 MGD		4 MGD	
	BWRO	ZDD	BWRO	ZDD
Pond size (acres) ¹	69	8	252	25
Method:				
Voutchkov (ref)	\$5.59	\$0.39	\$13.1	\$1.21
Mickley (Mickley & Associates, 2006)				
8-ft dike	\$3.79	\$0.38	\$14.2	\$1.31
12-ft dike	\$4.45	\$0.54	\$15.9	\$1.65
BGNDRF	\$7.36	\$0.59	\$29.1	\$3.73
EPWU (Gorder, 2009)	\$5.14	\$0.41	\$20.3	\$1.29
Average	\$5.27	\$0.46	\$19.2	\$1.83

¹Acreage includes adjustment for 8-ft dike height.

DWI can be an affordable concentrate disposal method in areas with the geology to support it and regulations to allow it. The CapEx is dependent on the style of construction (affected by regulations and geology), and the OpEx is affected by the geology (tight formations may require pressurization of the concentrate). It is estimated that a single disposal well is required for the 1 MGD and 4 MGD BWRO and ZDD designs based upon Mickley and Associates (2006), which states, as a generality, one well of 6 inch nominal internal pipe diameter can receive 1.27 MGD of flow at the recommended velocity of 10 feet per second (ft/sec) (Mickley and Associates, 2006). A well size of 4 inches was assumed for

Demonstration of Zero Discharge Desalination (ZDD)

flows equal to or less than 0.25 MGD for the calculation using Mickley and Associates (2006), Voutchkov (2013, and M46 (AWWA, 2007) methods scale with concentrate flow. The well is assumed to be 2,500 feet deep and constructed similar to the wells in El Paso. Minimal pressure (50 psi) is estimated for the CapEx. Estimates for disposal using DWI were made using a similar approach as with the evaporation ponds and are summarized in Table 26. Four methods for DWI well costs were used and the results averaged. Costs are in 2013 dollars and were updated using the Producer Price Index for capital equipment (Bureau of Labor Statistics, n.d.).

Table 26.—Comparison of DWI CapEx Estimation Costs (\$ Million)

System	1 MGD		4 MGD	
	BWRO	ZDD	BWRO	ZDD
EIS (SWCA, 2012)	\$2.85	\$2.85	\$2.85	\$4.17
Voutchkov (Voutchkov, 2013)	\$1.46	\$1.13	\$2.50	\$1.17
AWWA M46 (American Water Works Association, 2007)	\$2.63	\$2.40	\$3.29	\$2.46
Mickley (Mickley & Associates, 2006)	\$3.79	\$3.79	\$4.22	\$3.79
EPWU (Gorder, 2009)	\$4.17	\$4.17	\$4.17	\$4.17
Average	\$2.98	\$2.87	\$3.41	\$2.89

10.3 Technology Cost Estimates – La Junta

10.3.1 Treatment Process Descriptions – La Junta

ZDD was piloted at La Junta as part of WERF Project 5T10. Operational data from this pilot were combined with lessons learned at other pilot studies, to create a full-scale design for a 6.6 MGD greenfield ZDD solution for La Junta. The ZDD system would consist of a NF membrane system, an EDM system, and all ancillary feed tanks and pumps.

Process flow diagrams are similar to those described earlier (Figure 70 and Figure 71. Key labels are included and are incorporated into the operational cost summary in Table 24). A portion (13 percent) of the brackish feed is bypassed for blending and sulfuric acid is added to the remainder. The NF feed is dosed with antiscalant (2 mg/L). The NF concentrate is fed to the EDM system, and the EDM diluate (and concentrate from the Process RO) is blended with acidified brackish well water and fed to the NF. The NF is designed for 80 percent overall recovery. The EDM is designed to reduce the concentration of the NF concentrate by 40 percent, and nearly all of the NF reject is returned to the NF system (98 percent overall recovery). The EDM produces two waste streams: a Mixed Sodium Salt and Mixed Chloride Salt stream.

The ZDD system includes a two-stage NF where Stage 1 consists of 30 pressure vessels with six Dow NF 90-400 modules and Stage 2 consists of 15 pressure vessels with a blend of Dow NF 90-400 (first three) and NF270-400 (last three) membranes. Four skids are provided, each capable of producing 2.2 MGD. An extra skid is included to allow for maintenance on the other skids. The mixture of the NF90 and NF270 membranes allows for the necessary flux and product water quality for blending with the bypassed brackish water. The NF270 modules were selected for their ability to pass silica into the product water, thereby bypassing the EDM loop and acting as a silica purge. A portion of the NF permeate is fed to a small BWRO system for process water production, used for dilutions and salt preparation for the EDM. Chemical costs include the initial sodium sulfate charge for the E-Rinse and NaCl for the EDM, as well as daily consumption for sulfuric acid, antiscalant, and caustic (post treatment) for the NF.

10.3.2 Capital Cost (CapEx) Development – La Junta

ZDD designs were developed for a 6.6 MGD system. The ZDD includes an NF and EDM system. The NF system CapEx includes four Veolia NF package systems containing:

- Two-stage NF unit including 45 – 6M vessels. Stage 1 consists of 30 vessels loaded with Dow NF90-400 modules. Stage 2 consists of 15 vessels, each loaded with three Dow NF90-400 modules in the lead position followed by three NF 270-400 membranes.
- Cartridge filter skids, three including one standby.
- NF feed pumps, four horizontal centrifugal pumps, including one standby.
- Two product water pumps.
- NF CIP skid with tank, pump, heater, cartridge filter.
- 225 Dow NF90-400 and 45 Dow NF270-400 membranes.
- Sulfuric acid and antiscalant dosing pumps.

The EDM system consists of ninety (90) EDM stacks configured into skids of six stacks per skid. Each EDM stack has 120 quads, stainless steel frame, membranes, spacers, and electrodes, all manufactured by MEGA. Each EDM stack has its own power supply (DC drive manufactured by BARDAC). All of the stack skids share a common feed pump and feed tank for each of the five process streams to the stack:

- EDM feed/NF concentrate tank (4,000-gallon) and one duty plus one standby 900 gpm pump

Demonstration of Zero Discharge Desalination (ZDD)

- E-Rinse (Na_2SO_4) tank (600-gallon) and one duty plus one standby 225 gpm pump
- Sodium chloride storage tank (2,500-gallon) and one duty plus one standby 900 gpm pump
- Mixed Sodium tank (2,500-gallon) and one duty plus one standby 900 gpm pump
- Mixed Chloride tank (2,500-gallon) and one duty plus one standby 900 gpm pump

Each of the EDM recirculating/transfer pumps deliver a pressure of 25 psi to the stack, and provide enough residual pressure to return the streams to the storage tank, creating one large recycle loop from the tank, through the stack, and back to the tank. Automatic valves on each skid ensure even distribution among the stacks, based upon pressure readings for each stream. ED stacks are operated with equal pressure on each side of the membranes to mitigate internal leakage and prevent membrane damage. As with the pilot and demonstration systems, the NaCl, Mixed Na, Mixed Cl, and E-Rinse pressures will be set to match the EDM feed pressure.

There are make-up tanks for sodium chloride and sodium sulfate solutions; these products will deliver to site as dry products. The NaCl brine make-up system design is for three 80 ton Bryneer system.

There is also a degasifier blower and tower to remove hydrogen and oxygen gas produced at the EDM electrodes.

A secondary RO system provides 75 gpm of process water to dilute the Mixed Na and Mixed Cl streams. The system is designed to match the waste flow from the EDM (2 percent of the brackish water feed).

Instrumentation and controls include: rotameters, flow switches, and pressure switches for every EDM stream on every stack; ancillary instruments for tanks and pumps; master control panel/SCADA; and EDM skid remote I/O panels.

Flush pumps, tank heaters, and CIP pumps are provided for both the EDM and NF.

Equipment to transfer water from the brackish wells to the NF feed pump are not included in the Veolia pricing estimate, since these costs vary widely on every site. Further, the air compressor required for automatic valve actuation is not included.

10.3.3 Operation and Maintenance Cost Development

Operating cost assumptions used for the ZDD and/or BWRO systems includes the following (same as for Alamogordo):

- Power cost: \$0.08/kWh flat rate
- EDM sodium chloride: \$0.075/lb
- Annual CIP for NF lasting one hour, includes operation of NF flush pump, CIP pump, CIP heater (same for BWRO and ZDD)
- Monthly CIP for EDM systems, lasting one hour per skid, includes operation of EDM flush pump, CIP pump, and CIP heater
- Antiscalant: 2 mg/L dosage: \$2.95/lb
- Sulfuric acid for reducing bicarbonate in feed: $\text{pH} \leq 7.0$, \$0.20/lb
- Sodium sulfate for the E-Rinse: \$0.08/lb
- Caustic post treatment: 5 mg/L dosage: \$0.45/lb
- NF & EDM CIP chemicals: $< \$0.01/1000$ gallons product
- Source water cartridge filters replaced every month for annual cost: \$115,000
- NF and RO membranes replaced every five years: \$600/element
- EDM membranes and spacers will be replaced every 10 years: \$354/quad
- EDM electrode replacement schedule includes replacing the anodes every two years and cathodes every five years

The ZDD O&M costs are more evenly distributed than the Alamogordo design because more of the work is being done by the NF system, which is designed for 80 percent recovery. The higher recovery (80 percent vs 60 percent) leads to a higher relative specific energy for the NF and lower relative specific energy by the EDM. Another important difference is the EDM power supplies are operating at a lower voltage and current than the Alamogordo design. The current density is assumed to be 10 percent higher than what was piloted and differs from what was provided to the WERF authors for inclusion in their report. As stated in that report, the EDM stack was not operated at 80 amps (Brandhuber et. al, 2014) and it is possible that this is infeasible for many reasons. This revised lower current density means that more EDM stacks are included in this design than what was provided to the WERF authors for inclusion in their report. The EDM power cost

Demonstration of Zero Discharge Desalination (ZDD)

is 31 percent of annual power expenditure, NF feed pumps power cost is 32 percent of annual power expenditure, and the NaCl cost is 46 percent of annual chemical expenditure. The EDM design is conservative, so it is possible that additional optimization could improve the energy efficiency by increasing the current density (or decreasing stack resistance) or reducing the recirculating rate for the EDM pumps. Cost savings may be realized if sodium chloride is recovered by any of the methods suggested in this report. NaCl recovery using lime treatment of the Mixed Cl and ED fitted with monovalent selective membranes was tested at La Junta.

The major flows, pressures, EDM power supply, and NaCl consumption are summarized for the design flow rates in Table 27. The letters refer to the locations in Figure 70 and Figure 71. The recovery is calculated by dividing the product water by brackish feed water (G/A in Table 27).

Table 27.—Major Operational Cost Comparison by Product Flow Rate and System (La Junta, 6.6 MGD)

Location	ZDD	BWRO	Dual RO ²
System Recovery	98%	80%	92%
A (BW feed)	4,670 gpm / 30 psi	5486 gpm	4,987 gpm
B (BW feed to ZDD/BWRO)	4,067 gpm / 30 psi	4,514 gpm / 225 psi	4,583 gpm
C (Bypass)	603 gpm / 30 psi	972 gpm	404 gpm
D (NF/RO feed)	5,097 gpm / 112 psi ¹	4,514 gpm / 225 psi	1 st RO: 4,583 gpm 2 nd RO: 916 gpm
E (NF/RO permeate)	4,077 gpm / 0 psi	3611 gpm	1 st RO: 3,667 gpm 2 nd RO: 513 gpm
F (NF/RO concentrate)	1,019 gpm	903 gpm	403 gpm
G (product)	4,583 gpm / 30 psi	4,583 / 30 psi	4583 gpm
H + I (ZDD waste)	72.8 gpm / 25 psi		
PROCESS RO design (feed)	97 gpm / 90 psi ²		
EDM pumps (total)	3,600 gpm / 25 psi		
EDM E-rinse pumps (total)	900 gpm / 25 psi		
EDM Power Supply	90 @ 110 V _{DC} / 37 amps		
EDM NaCl use	42,327 lb/day		

¹ The pressure listed is the pressure estimated by Dow's ROSA program; however, the pressure in the cost model accounts for the pressure in the brackish water feed (NF feed pressure = ROSA value – BW feed).

² WERF report figures were scaled from 5.2 MGD permeate flow to 6.6 MGD permeate flow.

10.3.4 ZDD and BWRO Residuals Cost Development – La Junta

The EDM and BWRO waste streams will require final disposal via evaporation ponds. The residual streams from the ZDD consist of 2 percent of the brackish feed, in the form of two high-concentration streams called the Mixed Cl stream and Mixed Na stream. The residual stream from the BWRO consists of 20 percent of the brackish feed.

Four evaporation pond cost estimation methods were evaluated for the brine disposal. Costs are in 2013 dollars and were updated using the Producer Price Index for capital equipment (Bureau of Labor Statistics, n.d.). The capital costs for the evaporation pond used for this model is the average of the values estimated using the Voutchkov method, Mickley method with a 4-foot dike height for comparison with the WERF 5T10 evaluation, and the scaled versions from El Paso (design cost) and BGNDRF (actual cost). The net evaporation rate for La Junta used in the model is 32.4 inches per year (6.9 centimeter per month), based upon the WERF 5T10 report (Brandhuber et al., 2014). The estimated cost for evaporation ponds for each system is summarized in Table 28.

Table 28.—Comparison of Evaporation Pond CapEx Estimation Methods (La Junta, 6.6 MGD full capacity)

System	ZDD	BWRO	Dual RO, 91.2% Recovery ³
Required Evaporative Area (acres)	56	682	240
Capital Cost Using			
Voutchkov (ref)	\$4.9 M	\$41.0 M	--
Mickley (ref) ²	\$5.9 M	\$68.1 M	\$15.8M ⁴
BGNDRF ²	\$9.2 M	\$103 M	--
EPWU (ref)	\$2.1 M	\$33.3 M	--
Average	\$5.5 M	\$61.3 M	--

¹ Both the BWRO and ZDD evaporation ponds assume an 8-foot dike height.

² Excluded from average for BWRO.

³ From WERF 5T10 Report, assumes a 4-foot dike height.

10.3.5 Comparison of BWRO and ZDD Cost Estimates – La Junta

The capital cost estimate for the BWRO and ZDD systems is summarized in Figure 72. At this flow rate and with evaporation ponds as the only available brine disposal method, ZDD is less expensive than conventional BWRO. The Dual BWRO costs for a 6.6 MGD system were estimated by scaling the costs included in the WERF report from 5.2 to 6.6 MGD. Also, the WERF report used the average flow for designing the evaporation ponds (Brandhuber et. al., 2014). While

Demonstration of Zero Discharge Desalination (ZDD)

this is a reasonable assumption, our report uses the design flow for all systems, so the WERF 5T10 reported installation cost (\$6,400,000 for 96.3 gpm of concentrate) was used to scale up the full Dual RO concentrate flow (403 gpm).

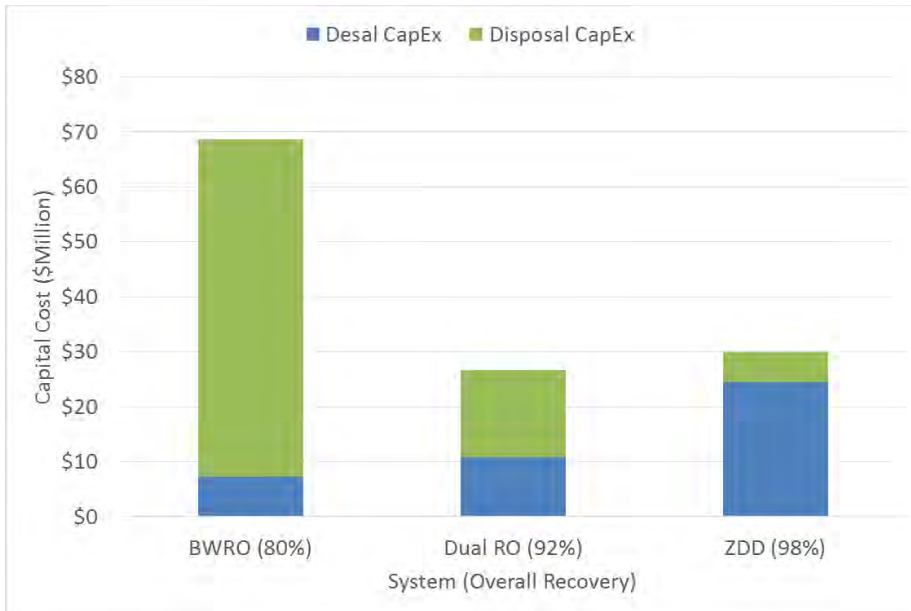


Figure 72.—Capital cost estimates for ZDD, BWRO, and Dual RO (6.6 MGD).

The total cost of water produced by the BWRO, ZDD, and Dual RO is summarized in Figure 73. The estimated annual capital cost for the ZDD system, which includes ZDD equipment and Evaporation Pond installation, accounts for 38 percent of the total cost of water production for ZDD and 73 percent of the total cost of water by BWRO. The capital cost of ZDD is very competitive with the Dual RO system, as the unit cost is within 33 percent of the Dual RO design and the main difference is membrane replacement cost. It is possible that both the RO and EDM membranes could last longer than the budgeted five year timeframe. It is important to note that the Dual RO would still require 278 acres of evaporation ponds, and ZDD would only require 94 acres. As mentioned earlier, the ZDD primary OpEx cost drivers are the NaCl and EDM power consumption. Future optimization and research will aim to reduce the cost of both of these.

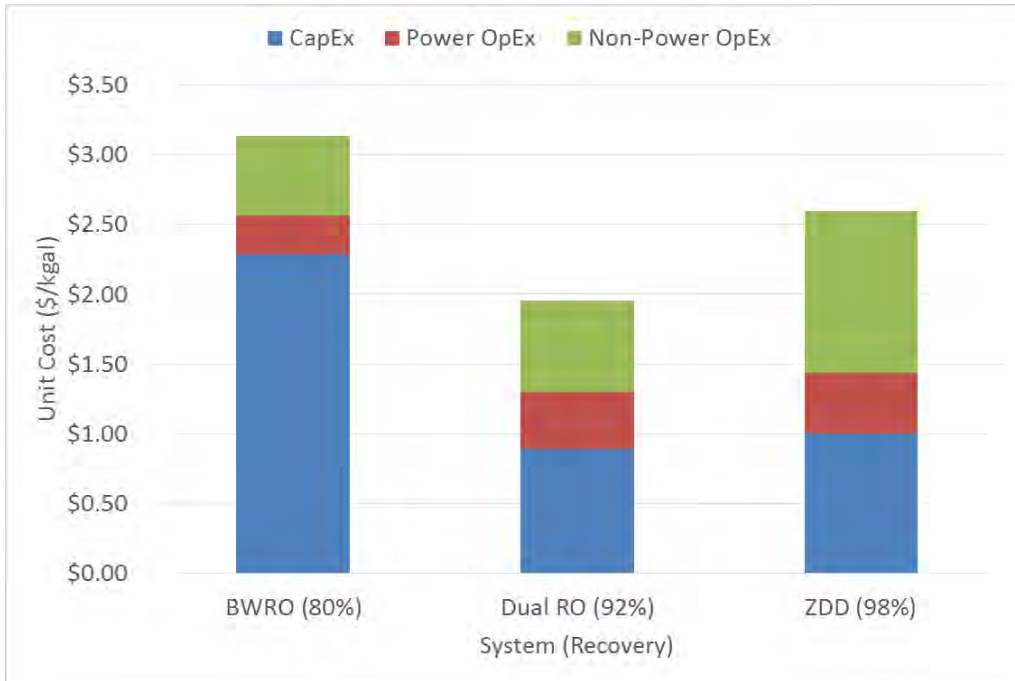


Figure 73.—Unit cost estimates for ZDD, BWRO, and Dual RO (6.6 MGD).

11. Salt Recovery

11.1 NaCl Purity Requirements

Experiments were performed to determine the level of purity required for the NaCl feed to the EDM stack. The results, described elsewhere in this report, indicated that the ion ratios of Na/Ca and Cl/SO₄ need to be at least 12 to prevent precipitation of CaSO₄ in the Mixed Cl and Mixed Na solution compartments of the EDM. Higher ratios will definitely reduce the chances of formation of CaSO₄ precipitation in the concentrate streams. The supernatant from the precipitation of CaSO₄, which typically has 1.5 M NaCl and Na/Ca and Cl/SO₄ ion ratios of 15, might be marginally acceptable for use as the NaCl supply, but it does not meet the criterion of a 3 M solution. If a 1.5 M NaCl solution were fed into the EDM stack, the rate of NaCl transport through the membranes would be about twice the rate of water transport, so the salt concentration in the NaCl solution compartments would be depleted. However, this shortcoming does not preclude the use of supernatant in the salt supply; it just means that the NaCl concentration of the solution would have to be boosted by the addition of NaCl crystals. If crystals of NaCl were added to the supernatant to boost its concentration to 3 M, the value of the Na/Ca and Cl/SO₄ ratios would double to about 30. Essentially half of the NaCl required for the EDM would be supplied by the supernatant, and half would be supplied by purchased salt.

The importance of salt purity is illustrated in Figure 74 where the cations in the salt stream include Na⁺ and Ca²⁺, and the anions include Cl⁻ and SO₄²⁻. If it is assumed that the membranes on either side of the NaCl compartment have equal permeability for monovalent and divalent ions, the undesirable Ca²⁺ ions will migrate into the Mixed Na stream along with the desirable Na⁺ ions. Ca²⁺ ions that enter the Mixed Na compartment will encounter a high concentration of SO₄²⁻ ions. If the rate of entry of Ca²⁺ ions is sufficient to exceed the solubility of CaSO₄, the condition will exist for precipitation of CaSO₄ in the Mixed Na solution compartments. Assuming that the NaCl stream is the only source of Ca²⁺ ions entering the Mixed Na compartment, one could determine experimentally what ratio of Na⁺ to Ca²⁺ ions would be sufficient to avoid a condition of supersaturation of CaSO₄ in the Mixed Na compartments. Indeed, after consideration of the utility of this ratio, it was decided to adopt the Na/Ca ratio as a criterion for assessing the quality of the NaCl supply. Similarly, Cl⁻ ions and SO₄²⁻ ions are transported through the AEM into the Mixed Cl compartment, and excessive transport of SO₄²⁻ ions leads to precipitation of CaSO₄ in that compartment. Therefore, the Cl/SO₄ ratio is also a useful criterion for measuring salt quality. Both ratios are based on measured or calculated concentrations of the ions in equivalents per unit volume.

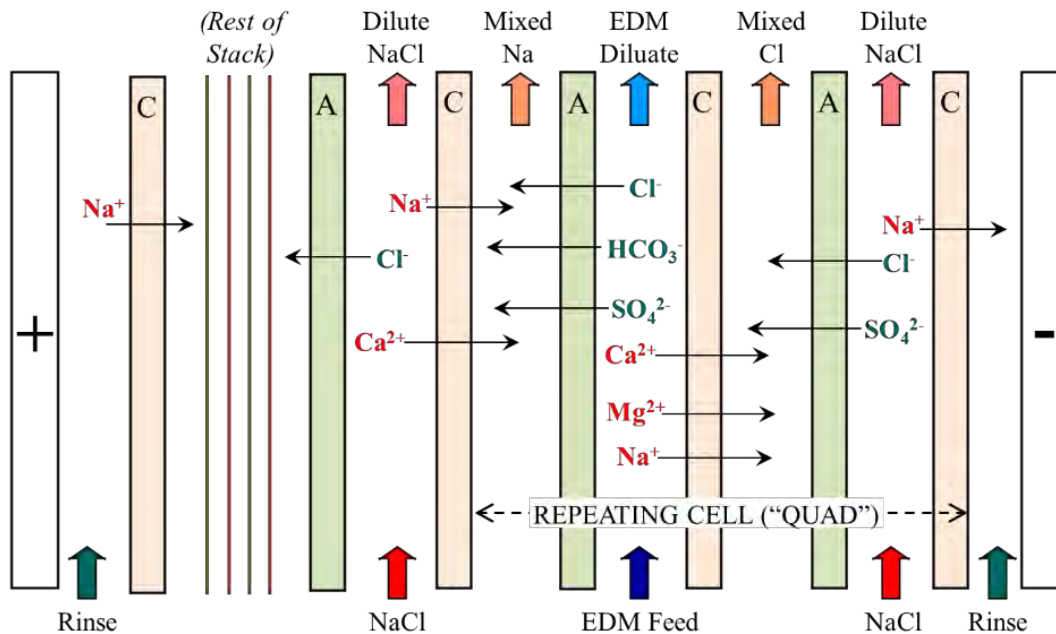


Figure 74.—Transport of ions through ion-exchange membranes in EDM stack.

It is extremely useful to have an understanding of how the solubility of CaSO_4 is affected by the presence of NaCl in solution. Figure 75 shows solubility data reported by Ostroff and Metler (1966). The data show that the presence of NaCl causes a substantial increase in the solubility of CaSO_4 . The solubility of CaSO_4 is 2 g/kg of H_2O in pure water, and it increases to a maximum of 7.7 g/kg of H_2O in 2.7 molar NaCl . Dashed lines in Figure 75 indicate the concentration of NaCl in the supernatant of the precipitation tank and the concentration of NaCl that is needed for the salt supply to the EDM stack. The 1.5 M supernatant concentration is based on typical experimental data obtained when Mixed Cl and Mixed Na streams were combined in stoichiometric proportions to precipitate CaSO_4 . The 3 M requirement is based on data obtained for the amount of water transported through the membranes along with the Na^+ and Cl^- ions.

As mentioned earlier, the Snake Tank wells that have been selected for the supply of groundwater to the planned desalination plant in Alamogordo have an excess of sulfate compared to calcium (see Table 3). Since the NF membranes reject divalent anions and cations almost equally well, this same excess of sulfate occurs in the NF reject stream and in the concentrate streams produced by the EDM. This means that some of the Mixed Na stream would need to be diverted to achieve stoichiometric proportions with the Mixed Cl stream in the precipitation of CaSO_4 .

Demonstration of Zero Discharge Desalination (ZDD)

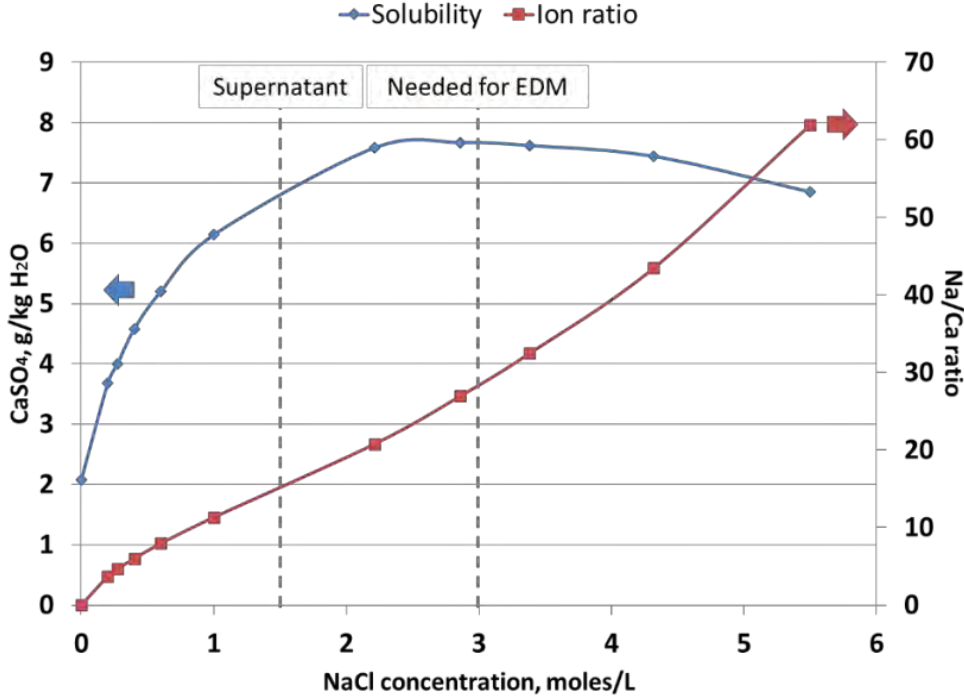
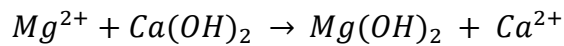


Figure 75.—CaSO₄ solubility in NaCl solution at 28°C and calculated ion ratios in saturated CaSO₄ solution.

However, the total equivalents of calcium and magnesium in the groundwater almost match the sulfate level. Therefore, if the Mg²⁺ ions were to be replaced with Ca²⁺ ions, then essentially all of the two concentrate streams from the EDM could be combined to make stoichiometric proportions for precipitation of CaSO₄.



This chemical reaction is aided by the fact that Ca(OH)₂ is slightly soluble (1.73 g/L) whereas Mg(OH)₂ has very low solubility (0.009 g/L). It is also aided by the fact that the Mixed Cl stream has negligible amounts of sulfate and carbonate, which would react with the calcium ions released from the lime.

Another motivation for removing the magnesium is that the presence of magnesium increases the solubility of CaSO₄. The graphs in Figure 76 show that both NaCl and MgCl₂ cause the solubility of CaSO₄ to increase, and the effects are additive.

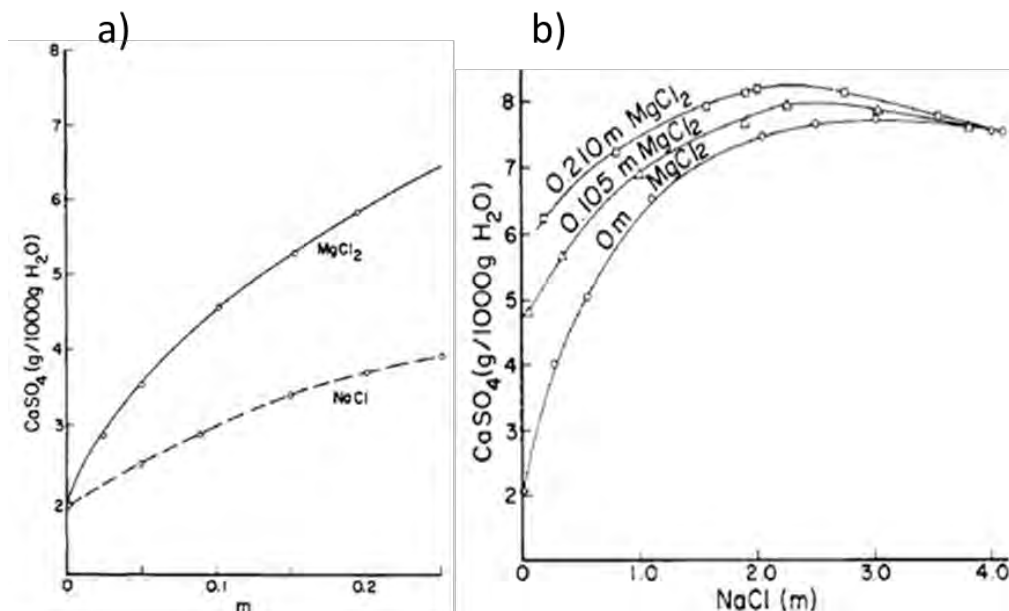


Figure 76.— a) Gypsum solubility in NaCl and MgCl₂ solutions and (b) Gypsum solubility in NaCl solutions with increasing concentrations of MgCl₂ (Ostroff & Metler, 1966)

11.2 Feasibility of Supernatant Use in EDM for NaCl Supply

To demonstrate that the use of supernatant is technically feasible, a lab-scale EDM stack containing the same MEGA membranes that were in the demonstration stacks was operated in the pilot area of BGNDRF for one week during Phase 2. The Mixed Cl and Mixed Na streams from the 40 gpm demonstration system were collected, magnesium ions were removed from the Mixed Cl stream by adding Ca(OH)₂ to precipitate Mg(OH)₂, and approximate stoichiometric proportions of the two concentrate streams were combined to precipitate CaSO₄ (previous sample analyses and ratios with conductivity were used to approximate the calcium and sulfate in the streams). Three batches of supernatant were prepared using this method, and samples were analyzed. Two sets of samples of the mixture of supernatant and NaCl pellets (same brand used in the demonstration system) were also analyzed. The average of each of these batches, along with the recirculating NaCl streams in the bench scale and demonstration scale systems are summarized in Table 29. The average supernatant from that precipitation was determined to have a NaCl concentration of 1.65 M. The NaCl concentration was boosted to 3 M by addition of the same NaCl crystals that were used in the 40 gpm demonstration plant.

Demonstration of Zero Discharge Desalination (ZDD)

Table 29.—Summary of Prepared NaCl Solutions and Circulating NaCl Streams
(all in meq/L)

	Ca	Mg	Na	Cl	SO ₄	HCO ₃	Na/Ca	Cl/SO ₄
Supernatant (average of 3)	121	7.4	1,565	1,602	136	8.7	14.7	12.6
Supernatant + NaCl (average of 2)	121	14.6	4,880	4,696	205	8.0	40.8	26.8
NaCl to EDM (bench - beginning)	3.4	1.0	238	230	15.0	2.3	70.8	15.4
NaCl to EDM (bench - end)	7.3	3.6	474	468	24.4	3.2	65.2	19.2
NaCl to EDM (demo - low Cl/SO ₄)	2.6	2.0	540	534	15.7	1.8	209.3	34.1
NaCl to EDM (demo - high Cl/SO ₄)	8.6	6.7	597	494	74.6	3.6	69.7	6.6

The system was intended to be operated at a current density similar to the demonstration scale EDM stacks, but the density ended up being lower than expected because of a miscommunication about the available surface area in the full scale EDM stacks. Even though the current density was lower at the bench scale (bench scale: 65-70 A/m², demonstration stacks: 110 A/m²), the EDM performance was similar in terms of concentrations. Since the bench-scale system is not automatically controlled, it was not operated overnight or for extended periods of time. At the end of each day's operation, a portion of the Mixed Na and Mixed Cl streams were captured before the concentrate compartments were rinsed with DI water. The captured Mixed Na and Mixed Cl would be used to start up the experiment on the following day. The NaCl concentration would go down over time and the infusion pump would be turned on when the conductivity was below 50 mS/cm. The EDM stream pressures were controlled manually by opening or closing a valve downstream of each of the pumps. The valves on the Mixed Na, Mixed Cl, NaCl, and E-Rinse streams were adjusted to match the pressure in the EDM feed (similar to how the demonstration equipment is controlled automatically). Pressure drop across the EDM stack increased slightly over time; however, this is related to iron particulates in the EDM feed stream. A 1 micron cartridge filter was installed on the EDM feed and as it clogged, the EDM feed pressure (and therefore the other stream pressures) increased. Figure 77 includes conductivity, pressure, voltage, and current data obtained during the six day study.

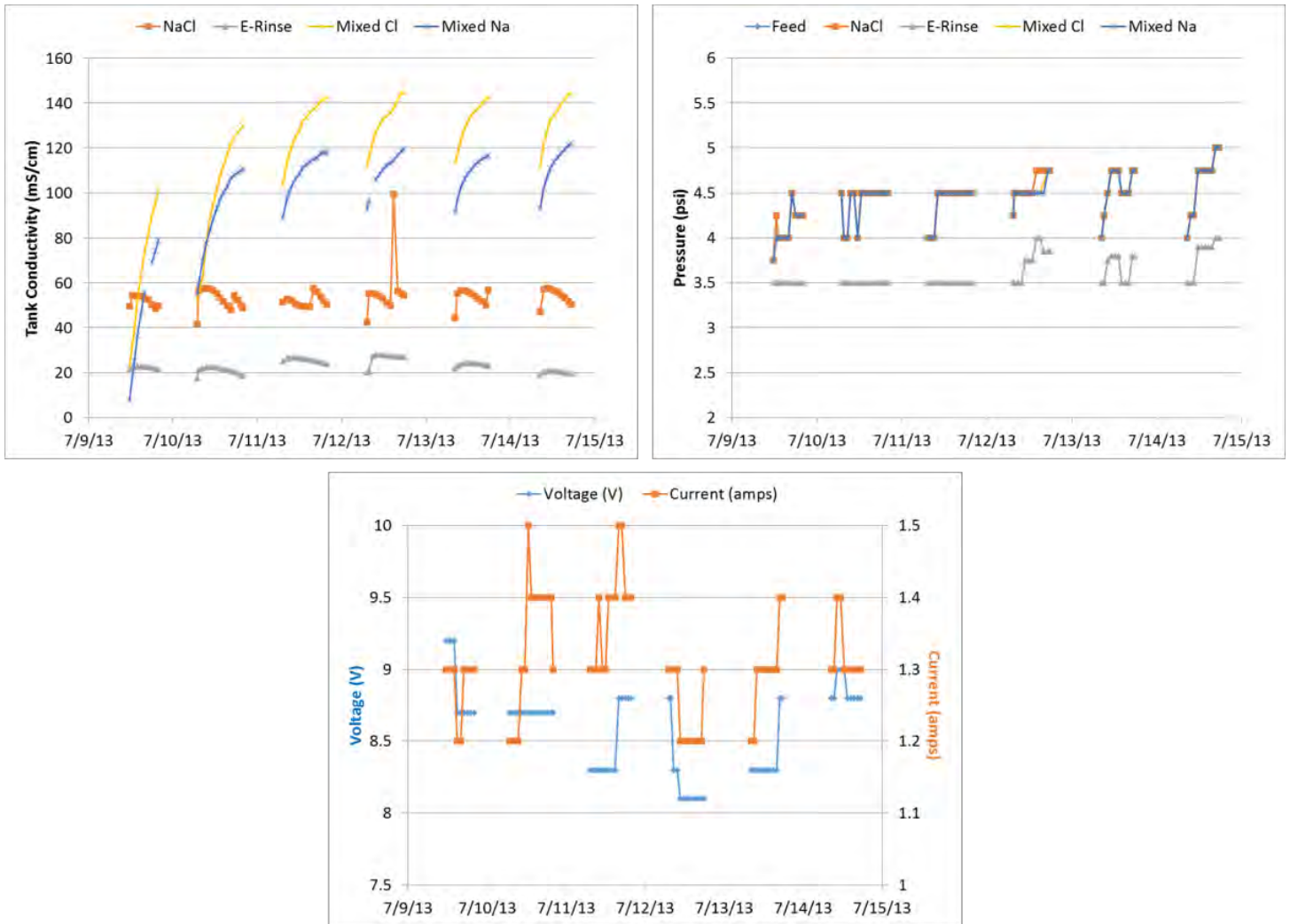


Figure 77.—Operations data from Supernatant Study at BGDRF.

There was no evidence of precipitation in the concentrate streams or of membrane swelling during the six days of operation. The results of this experiment offered conclusive evidence that the supernatant can be used to dissolve NaCl crystals for the salt supply to the EDM stack, and the benefit is that the consumption of purchased salt is reduced. Moreover, the use of the supernatant for this purpose consumes about half of the volume of supernatant produced and pushes the overall yield of fresh water from 98 percent to 99 percent.

11.3 Feasibility of Using Evaporation Pond to Recover NaCl

Full-scale implementation to use supernatant in the NaCl supply would require a process for the precipitation of CaSO_4 . That process was performed in the laboratory and in the ZDD pilot in La Junta by combining the two concentrate streams (after treating the Mixed Cl stream with $\text{Ca}(\text{OH})_2$) in a conical-bottom plastic tank equipped with an agitator. The Mixed Cl and Mixed Na streams were introduced to a zone that contained the previously formed CaSO_4 precipitate, which provided seed crystals for the precipitation of Ca^{2+} and SO_4^{2-} ions being introduced with the concentrate solutions. The agitation was confined to the bottom of the tank so that supernatant could disengage from the crystals. Because the crystals were allowed to grow, they became large enough to remain in the lower zone of the tank. Supernatant was decanted from the top of the tank. This operation is illustrated in Figure 78. For a commercial-scale ZDD process it is envisioned that a conventional clarifier similar to what is used for lime softening would be employed. Figure 79 shows the anticipated configuration of an appropriate clarifier.

The CaSO_4 crystals from the clarifier would be removed continuously and sent to a belt filter or centrifuge where the liquid would be drained from the crystals. The centrate or drained supernatant would be returned to the mixing zone of the clarifier, and that liquid would ultimately appear in the overflow of the clarifier. It would likely be necessary for the crystals to be washed with groundwater to remove excess NaCl if the CaSO_4 crystals are to be sold. Washing would not be necessary if the CaSO_4 is to be placed in a landfill. The rinse water, which might represent 1 percent of the groundwater pumped from the wells, could be combined with the NaCl feed to the EDM, as illustrated in Figure 80.

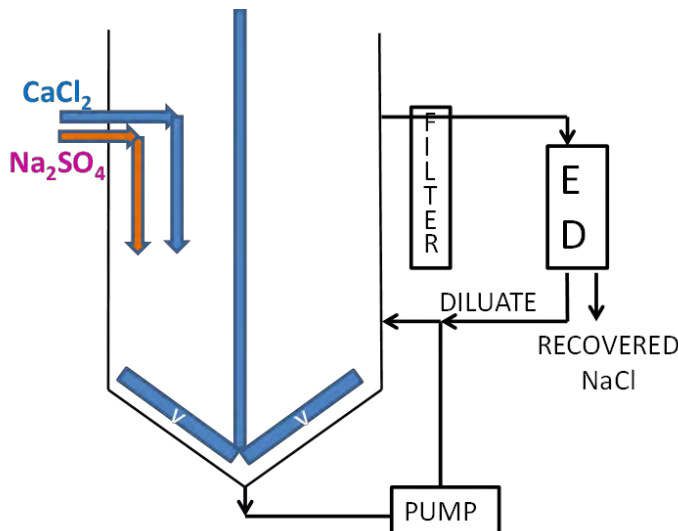


Figure 78.—Setup for lab ED of supernatant.

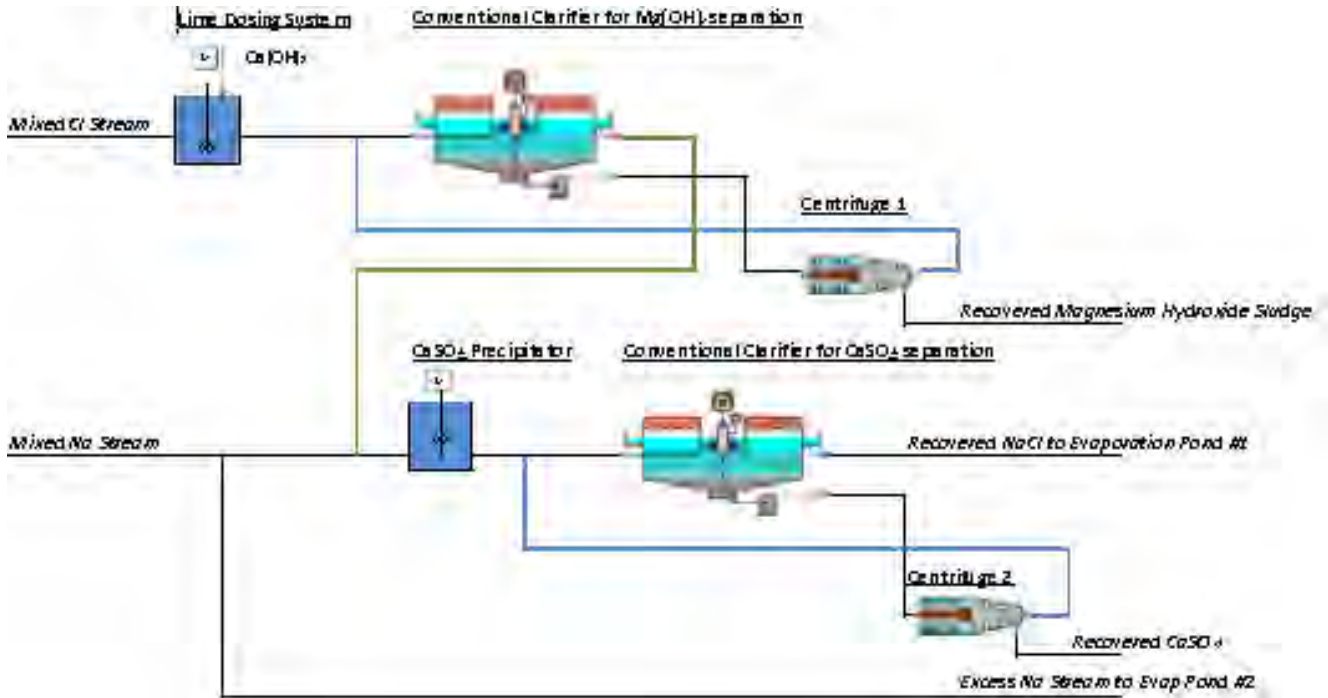


Figure 79.—Full scale salt (NaCl, CaSO₄, Mg(OH)₂) recovery diagram.

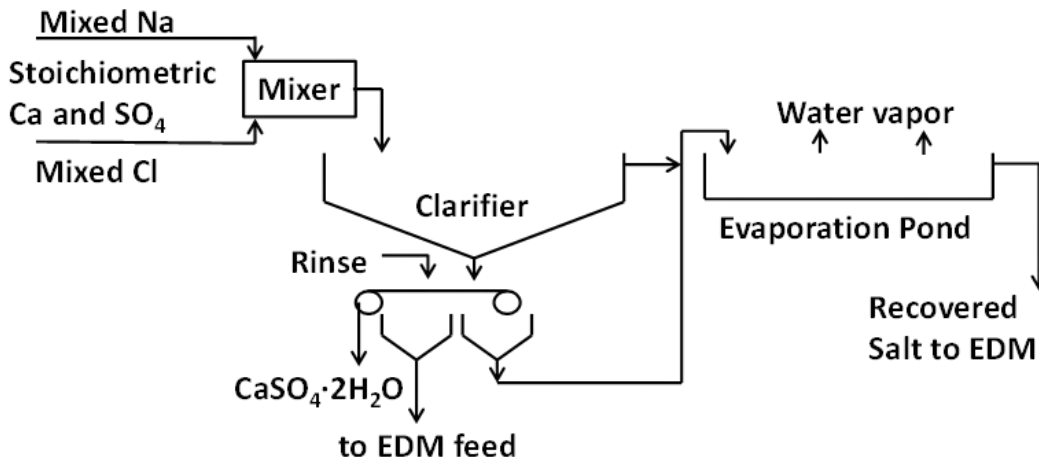


Figure 80.—Recovery of NaCl from EDM concentrate streams by precipitation and evaporation.

Returning to Figure 75, one can observe that the equilibrium values of the ion ratios (Na/Ca and Cl/SO₄) will continue to rise as the concentration of NaCl in the solution increases. That increase in NaCl concentration could be achieved by evaporation of water from the supernatant, and that evaporation would be most economically performed in an evaporation pond. If the NaCl concentration is raised to 3 M in the evaporation and the calcium and sulfate are fed to the pond in stoichiometric proportions, then the ion ratio would be about 28, which should be

Demonstration of Zero Discharge Desalination (ZDD)

adequate for direct use in the EDM stack. However, since some NaCl still needs to be purchased, the blended purity is expected to have a ratio greater than 50. Experience with operation of the pilot and demonstration plants confirmed that an ion ratio of 28 in the NaCl stream was quite sufficient to avoid precipitation problems in the EDM stack. Figure 81 shows the various ratios and pH measurements for Phase 2 operations at the BGNDRF. Managing the evaporation pond for NaCl recovery would be somewhat more complicated than managing a pond in which all of both concentrated streams are disposed. It would be necessary to monitor the Ca^{2+} and SO_4^{2-} ion concentrations of the water in the salt-recovery pond and divert an appropriate amount of one of the concentrate streams into another pond to maintain nearly equal concentrations of the Ca^{2+} and SO_4^{2-} ions. Such imbalances could happen due to changes in the proportions of water drawn from wells with different water chemistries.

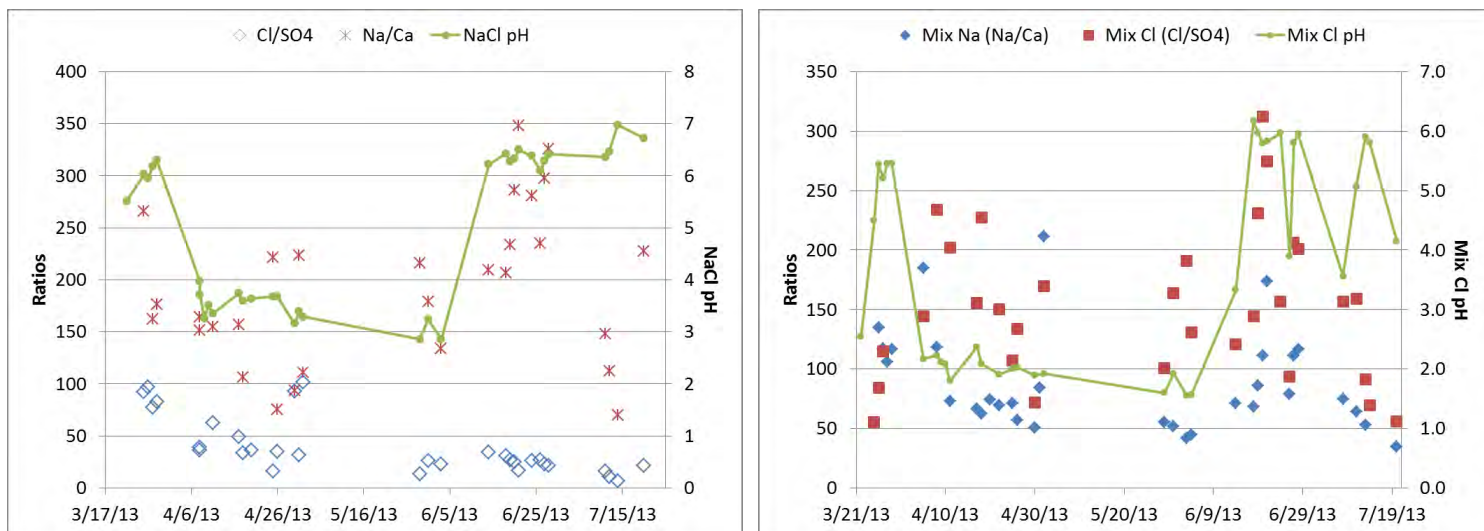


Figure 81.—Comparison of Na/Ca and Cl/SO ratios and stream pH during Phase 2 Operations at BGNDRF. Left: NaCl stream ratios & pH; Right: Mixed Na & Mixed Cl ratios, Mixed Cl pH

A laboratory experiment (see f) was performed to demonstrate that evaporation would indeed lead to the production of NaCl solution of adequate purity for use in the EDM stack. The Mixed Cl and Mixed Na solutions from the 40 gpm demonstration plant at BGNDRF were analyzed, and stoichiometric proportions of those solutions, with respect to calcium and sulfate, were combined in a 2 liter (L) beaker and stirred. The precipitate was recovered by filtration. The filtrate (supernatant) was boiled slowly to concentrate the NaCl. As water was removed by evaporation, the solution became supersaturated with CaSO_4 , and precipitation occurred. The evaporated solution was 4.6 M, nearly saturated in NaCl, and the ion ratios were $\text{Na}/\text{Ca}=185$ and $\text{Cl}/\text{SO}_4=16$. The square root of the product of the ion ratios is 54.4, a value very close to the curve for ion ratio in Figure 75 for the final salt concentration. The distribution of the calcium and sulfate originally in the Mixed Cl and Mixed Na solutions introduced to the experiment was as follows: 73 percent was removed in the precipitation tank, 25 percent was

removed in the evaporation tank, and 2 percent remained in the recovered NaCl solution.

It is obvious that the experiment was started with an excess of the Mixed Na solution; however, considering that only 2 percent of the calcium and sulfate remained in the evaporated liquid, the difference was likely due to imprecision in the original analyses. This underscored the need for periodic adjustment of the mixing proportions based on analysis of the solution in the evaporation pond.

The results of the experiment described in Figure 82 indicate that the solution of NaCl produced in an evaporation pond would be of sufficient quality to provide the supply of NaCl to the EDM stack. The next question relates to the quantity of NaCl that could be available from the evaporation pond. The composition of water in the wells of Alamogordo will be used as an example. Data in Table 3 indicate that there is more sulfate than calcium in the well water, so just as in the experiment described in Figure 82, a larger portion of Mixed Cl solution would be needed to achieve stoichiometric proportions of calcium and sulfate. However, if the magnesium is replaced with calcium, which can be achieved by adding lime to precipitate $Mg(OH)_2$, then the proportions of calcium and sulfate would be nearly the same, and almost all of the volume of both concentrate streams could be fed to the precipitation chamber, and all of the supernatant could be sent to the evaporation pond. Therefore, except for small amounts of solutions that would be discarded in a second evaporation pond and the solution that was not separated by the centrifuge, all of the NaCl that enters the EDM would enter the evaporation pond for recovery. To provide a margin for loss, it will be assumed that 90 to 95 percent of the available NaCl in the Snake Tank source water is recovered.

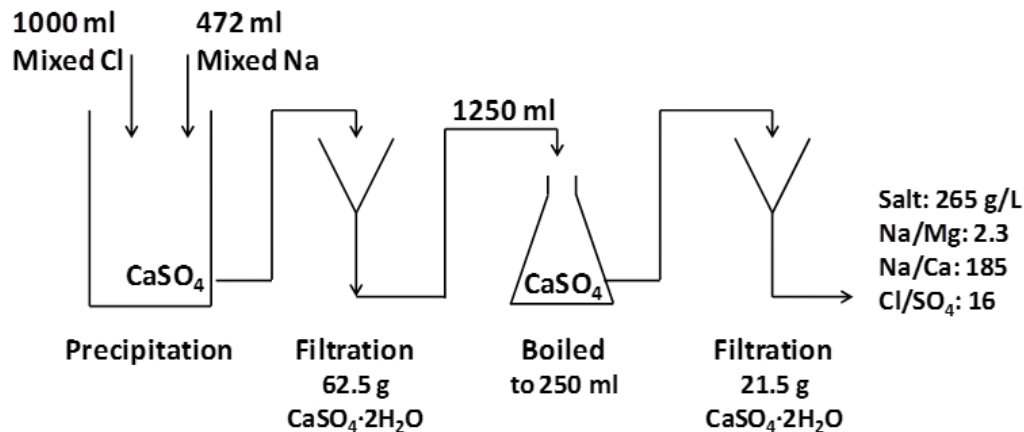


Figure 82.—Evaporation experiment using EDM concentrates from BGNDRF 40 gpm demonstration plant.

The quantities of $CaSO_4$ involved can be estimated for the case of the Snake Tank Wells that will supply the water for the desalination plant to be built in Alamogordo. Examination of the analyses in Table 3 reveals that sulfate represents 85 percent of the anions in the feed. Removal of that SO_4 as a solid byproduct

Demonstration of Zero Discharge Desalination (ZDD)

would result in the production of water of high quality. Assuming total conversion of sulfate to calcium sulfate dihydrate ($\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$) would result in the daily production of metric 32 metric tons per day (mT/d) of solids from a 4 MGD desalination plant. Data from the experiment described in Figure 82 indicate that about 73 percent of the solids would precipitate quickly when the two concentrate streams are combined, and 25 percent would precipitate gradually in an evaporation pond used for recovery of NaCl for return to the EDM. The net annual evaporation rate for the Holloman Air Force Base, which is close to the site for Alamogordo's desalination plant, is 70 in/yr (Livingston Associates, 2003). Using the monthly average net evaporation rate in at Alamogordo, 14.8 cm per month, an evaporation pond with 7.7 acres would be needed to concentrate the salt solution. A smaller pond (3 acres) would also be needed for excess Mixed Na solution and other waste streams. Assuming dense gypsum crystals at the bottom of the pond (1 metric tons per cubic meter [mT/m^3]), the accumulation from the 25 percent would be 0.093 meters per year (m/yr). If the life of the evaporation pond is assumed to be 20 years, the pond would need to be 1.85 meters deeper than an evaporation pond with no precipitation. After its 20-year life, the pond would be removed from service, allowed to evaporate to dryness, and covered with an impervious film and soil.

Construction of the evaporation ponds and clarifiers represents a capital cost. Assuming that the rate of flow of supernatant into the evaporation pond is 2 percent of the well water in a 4 MGD desalination plant and that half of the water would be evaporated to raise the salt content from 1.5 moles per liter (M) to 3 M, the evaporation rate would be 40,000 GPD. Capital equipment would include the evaporation pond, two clarifiers (one for lime addition, one for CaSO_4 precipitation and recovery), and pumps and piping for transferring fluid between processes. Operating costs would include lime. The cost for this method is estimated to be \$0.022/lb of NaCl recovered, which would reduce the NaCl purchases by 53 percent. Assumptions for the cost for implementation of this strategy include:

- 7.7 acre evaporation pond for NaCl recovery, 3 acre evaporation pond for excess Mixed Na waste. Capital Cost of \$80,000/acre is based on average of four methods described in the Chapter 8 (see Table 25).
- Clarifiers for Mixed Cl treatment with lime and CaSO_4 recovery. Capital cost of \$14,427/gpm treated is based on the capital cost for a 500 gpm system treating produced water (Hamilton Engineering Inc., 2009).
- Belt filter capital cost estimated at \$71,532 is based on 500 dry pounds per hour (lb/hr) (U.S. Environmental Protection Agency [EPA], 2000).
- Annual operating cost includes the purchase of lime is based on \$0.04/lb.
- Annual electrical cost for clarifiers is based on 15 kWh/kgal.

A less expensive alternative (in terms of capital cost) exists for recovery of NaCl in an evaporation pond. If there is no market for the $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$ that is precipitated and washed as shown in Figure 80, the clarifier and belt press equipment could be eliminated, and the precipitation could be done in a mixing channel at the entrance to the evaporation pond. After magnesium is removed from the Mixed Cl stream by lime addition, the Mixed Cl and Mixed Na streams are combined, agitated and discharged into a concrete-lined channel with a weir that overflows to the evaporation pond. This alternative would require an evaporation pond of the same size as the one used to evaporate water from the supernatant of the process shown in Figure 80. Most of the 75 percent of the CaSO_4 that would have been collected on the filtration belt (or centrifuge) would now accumulate in the channel. When the channel is filled with solids, the slurry would be diverted to another channel, and the CaSO_4 in the first channel would be dewatered, excavated and transported to a landfill. Assuming that a contractor would perform the excavation and hauling monthly, a channel 40 meters long and 20 meters wide would fill to a depth of 1.7 meters, which is enough material to fill twenty five 35 metric ton dump trucks. The cost for this method is estimated to be \$0.024/lb of NaCl recovered, which would reduce the NaCl purchases by 50 percent. Assumptions for the cost for implementation of this strategy include:

- Same cost for evaporation ponds as using CaSO_4 recovery method
- One clarifier for lime precipitation using the same price as the CaSO_4 method
- Annual CaSO_4 disposal cost based on a dumping fee of \$10/m/T and a contractor charge of \$5,000 per month
- Annual lime and electricity cost is the same as for the CaSO_4 + lime recovery method

11.4 High Purity NaCl Recovery Using Electrodialysis and Recovery of $\text{Mg}(\text{OH})_2$

Before the potential cost savings of evaporation became evident, the NaCl recovery efforts were focused on the use of ED to recover NaCl from the supernatant. ED is used on a large scale to recover NaCl from seawater in Japan where there are no natural salt deposits and where abundant rainfall makes evaporation ponds ineffective. To prevent precipitation of CaSO_4 in the concentrate streams of the ED stacks, companies in Japan developed ion-exchange membranes that were selectively permeable to monovalent ions and limited the transport of divalent ions into the recovered salt. Figure 83 illustrates the selective transport of monovalent ions in ED. Monovalent selective ion exchange membranes are able to allow passage of mostly Na^+ and Cl^- ions and block most of the troublesome ions such as Ca^{2+} and SO_4^{2-} .

Demonstration of Zero Discharge Desalination (ZDD)

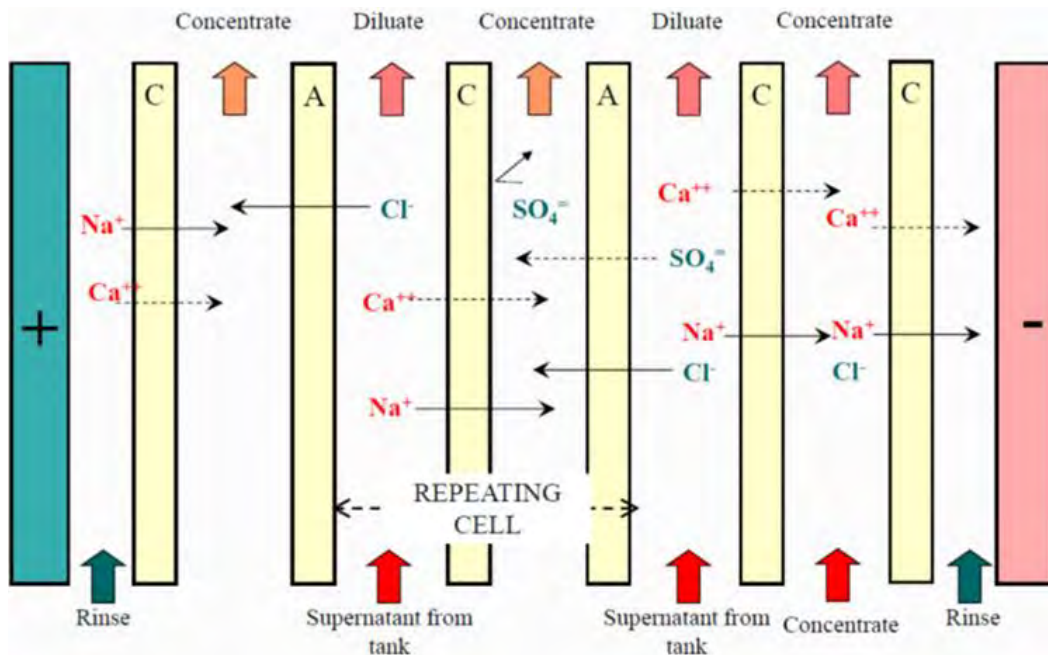


Figure 83.—Selective transport of monovalent ions in electrodialysis.

An extensive experimental program was carried out to study the use of ED to recover NaCl from the supernatant of the process where CaSO₄ precipitates when the Mixed Cl and Mixed Na concentrate streams from the EDM are combined. ED has the advantage that it can concentrate salts to very high levels. ED has been used in Japan to recover NaCl from seawater (3.5 percent NaCl) and concentrate it to high levels (20 percent) in the commercial production of edible salt (Strathmann, 2004). To avoid the precipitation of CaSO₄ in the concentrated salt solution, the major manufacturers of ED membranes in Japan developed monovalent-ion-selective membranes that block the transport of divalent ions. The monovalent-anion-selective membranes are extremely effective in allowing the passage of Cl ions and blocking SO₄²⁻ ions. This high selectivity for rejecting sulfate in ED is fortuitous, because there are really no other practical methods for removing sulfate from a concentrated NaCl solution. The monovalent-cation-selective membranes are adequately effective for rejecting Ca²⁺ and Mg²⁺ ions and passing Na⁺ ions. Fortunately there are other methods for removing divalent cations if further purification were required. Adding Na₂CO₃ or using chelating resins would provide means of removing divalent cations from the concentrated NaCl solution.

With the prospect available for employing an inexpensive base to recover a valuable byproduct and solve a stoichiometry problem, a study using salt values from (National Lime Association, 2000) was devised to determine the practicality of adding $\text{Ca}(\text{OH})_2$ (\$80/ton) to the Mixed Cl stream to precipitate $\text{Mg}(\text{OH})_2$ (\$600/ton). Preliminary experiments showed that feeding a slurry of lime to the Mixed Cl stream was effective, but calculations indicated that the water to make a 10 percent lime slurry would dilute the Mixed Cl stream by more than 20 percent. Therefore, experimental efforts were focused on the use of powdered hydrated lime, which is easily handled with commercial equipment (Figure 84).

In the precipitation experiments a solution with composition simulating the Mixed Cl from the Phase 1 Alamogordo pilot plant was placed in a beaker fitted with a magnetic stirrer and a pH probe, and measured amounts of powdered $\text{Ca}(\text{OH})_2$ were added. The pH was recorded after it stabilized after each incremental addition. The graph in Figure 85 shows the pH versus amount of lime added. The pH remained nearly constant, because $\text{Ca}(\text{OH})_2$ was dissolving, and the OH^- ions were reacting with Mg^{2+} ions. When the Mg^{2+} ions had been consumed, further dissolution caused an increase in pH until the solution was saturated with $\text{Ca}(\text{OH})_2$. After the second plateau was reached, all of the lime remained in suspension.

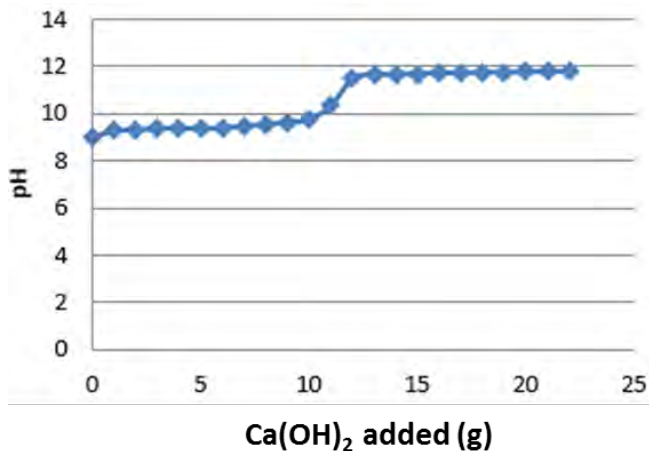


Figure 85.—Addition of $\text{Ca}(\text{OH})_2$ powder into simulated Mixed Cl solution.



Figure 84.—Hydrated lime injection system (Con-V-Air Solutions, n.d.).

Demonstration of Zero Discharge Desalination (ZDD)

In a second experiment, the addition of lime was stopped as soon as the pH increased (Figure 86) to avoid accumulation of undissolved lime. A sample of the supernatant analyzed by ion chromatography (IC) revealed that the magnesium level was below detection limits. For a determination of the quality of the $\text{Mg}(\text{OH})_2$, duplicate samples of the precipitate were washed with DI water and dissolved in 37.1 percent HCl acid, and the composition of the solution was analyzed by the IC. The analysis of the solution from the dissolved precipitate, shown in table 2, reveals the high quality of the precipitated $\text{Mg}(\text{OH})_2$. A small amount of CaSO_4 was the only noteworthy contaminant. Based on the cations, the purity is 97.31 and 98.17 for the two samples. If the precipitation shown in Figure 86 had been stopped before the last addition of lime, it is likely that the 99 percent target could have been achieved.

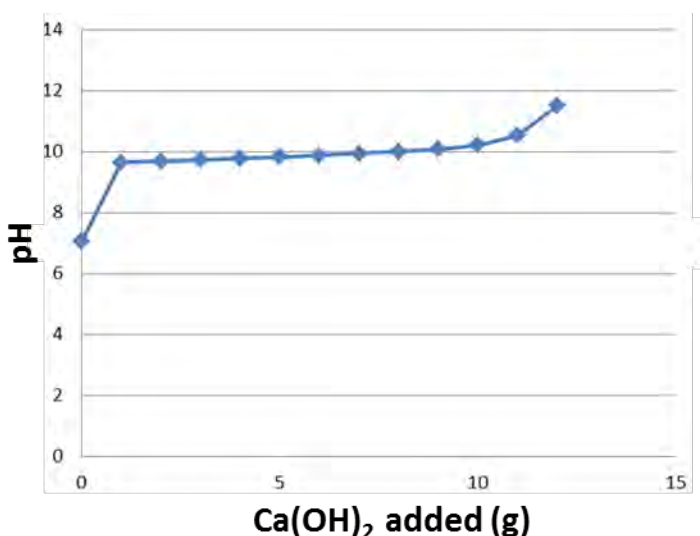


Figure 86.—Stopping the addition of $\text{Ca}(\text{OH})_2$ powder at the endpoint of the plateau eliminates undissolved lime in the $\text{Mg}(\text{OH})_2$ product.

Alibaba advertises 92 percent as high purity with a retail price range of \$400 to 600 per metric ton (Alibaba, n.d. a), and advertises 99 percent purity with a retail price range of \$800-850 per metric ton (Alibaba, n.d. b). The magnesium content of 88 mg/L in the Snake Tank wells would produce 3262 kg/day of $\text{Mg}(\text{OH})_2$ from a 4 MGD desalination plant. At \$600 per metric ton, monthly sales exceeding \$50,000 could induce a contractor to invest in the operation of a demonstration of the precipitation in Alamogordo's 1 MGD plant to see if a high-purity product and a premium price could be obtained. There are markets for $\text{Mg}(\text{OH})_2$ slurry as well as the dried product.

All of the laboratory ED experiments used as feed the supernatant from the precipitation of CaSO_4 and specifically used the apparatus illustrated in Figure 78. The ED stack was assembled with nine cell pairs comprising Neosepta ACS monovalent-anion-selective and CMX-S monovalent-cation-selective membranes. The diluate flowed once through the ED stack and back into the precipitation

tank. The concentrate stream was recirculated through the ED stack at a flow rate that resulted in balanced pressures between the diluate and concentrate. There was no solution fed to the concentrate streams; the water in the concentrate was entirely derived from molecules of water that were transported through the membranes along with the ions. In a typical experiment the concentrate tank would be filled initially with DI water, and the concentrate built up gradually over time. The ED would be operated during the day and shut down for evenings and weekends. Because the precipitation tank had a large capacity relative to the ED, the introduction of Mixed Cl and Mixed Na streams to the tank was intermittent and led to depletion of the salt in the feed tank. Figure 87 shows the concentrations of ions in the diluate and concentrate based on measurements of solution conductivity. With this mode of operation, the NaCl level of the concentrate was pushed as high as 4.36 M (22 percent NaCl by weight), which is far above the 3 M level needed to feed the EDM stack. At that level of NaCl concentration the current efficiency was 64 percent. The graduate student who was performing the ED experiments infused DI water to dilute the concentrate to 2.7 M (the last two data points on Figure 87), slightly below the target of 3 M, and he reported in his thesis that the current efficiency improved to 84 percent (Yestayew, 2013). Evidently, back diffusion of ions is enhanced when the ED is pushed to its limits, and reducing the driving force for back diffusion provides a substantial improvement in current efficiency. Since DI water would be used to dilute the recovered salt before it is returned to the EDM, it makes sense to perform that dilution in the ED stack and benefit from the improved current efficiency.

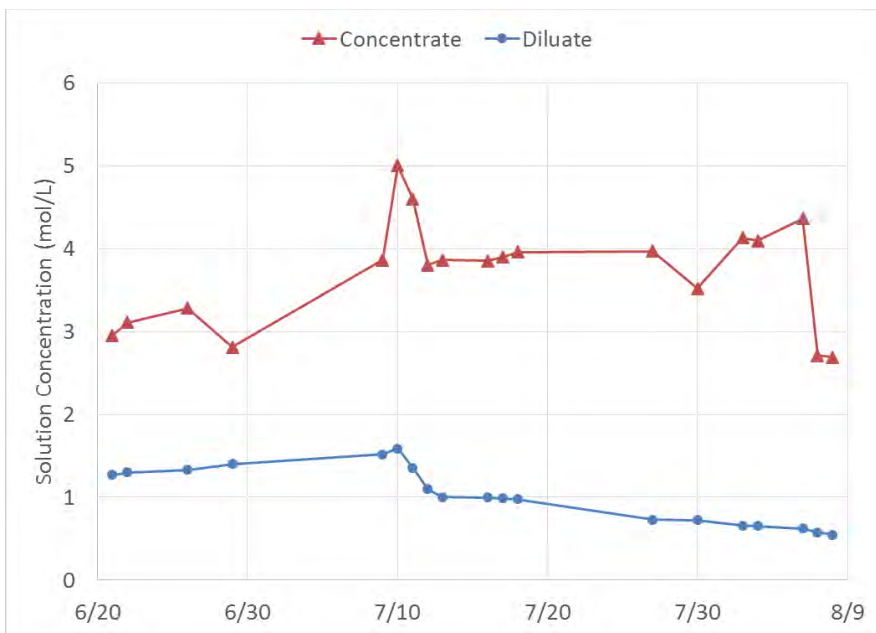


Figure 87.—Molar concentrations (based on conductivity measurements) of NaCl in diluate and concentrate of ED stack.

Demonstration of Zero Discharge Desalination (ZDD)

The experiments described above were performed in a feed-and-bleed mode of operation in which the CaCl_2 -rich concentrate and the Na_2SO_4 -rich concentrate from the EDM were added continuously to the bottom of the precipitation chamber so that they contacted the previously precipitated CaSO_4 , and the supernatant was drawn off continuously for treatment by ED. In the feed-and-bleed mode, the solution being treated by the ED has a constant concentration, and that concentration is low. An alternative is to operate the ED in batch mode wherein the initial concentration is 1.5 M and the final concentration is the low value. The data in Figure 88 show the implications of operation in the batch or feed-and-bleed modes. The longer of the two vertical dashed lines in Figure 88 shows the expected concentration of NaCl (1.5 eq/L) in the supernatant when the Mixed Cl and Mixed Na concentrates from the EDM are combined. (The 1.5 eq/L is given as an example based on pilot plant experience. Larger values of NaCl concentration in the supernatant are possible.) The shorter dashed line shows the concentration of NaCl (0.215 eq/L) after 90 percent of the NaCl is recovered by ED. In the batch mode of operation the Na/Ca ratio in the ED feed would begin at 15 and drop to 4 at the end of the batch. The solution being fed to the ED in the feed-and-bleed mode would have a constant Na/Ca ratio of 4, so the recovered NaCl solution would have a Na/Ca ratio less than half of that achievable with batch ED.

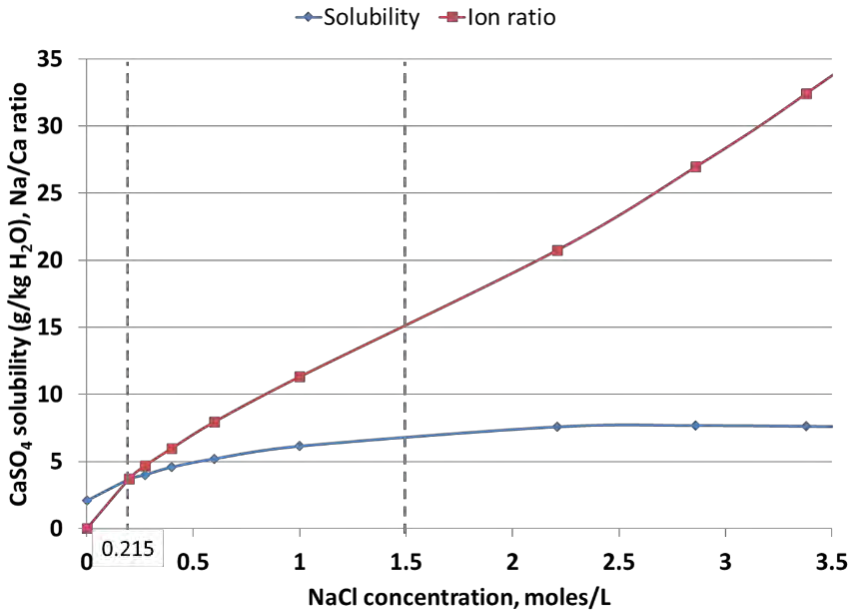


Figure 88.—on ratios in feed-and-bleed mode of ED operation.

Batch treatment requires less electrical energy than feed-and-bleed treatment, because the higher concentration of salt in the depleting compartments of the ED at the beginning of the batch provides lower electrical resistance, so the voltage applied to the ED stack can be lower at the beginning of the batch than at the end. With feed-and-bleed operation, the required voltage would be the same as the

voltage at the end of a batch operation, because the NaCl concentration fed to the ED would have a constant value of 0.215 eq/L.

In the procedure for batch ED, a batch of supernatant and previously precipitated CaSO₄ are agitated to promote contact between the calcium and sulfate ions in solution with the solid CaSO₄. The precipitate is allowed to settle somewhat in order to form a clear supernatant containing NaCl and saturated with CaSO₄. The supernatant passes through a filter to remove particles of CaSO₄ that might remain suspended in the solution. Then the supernatant flows through the ED stack and back into the agitated precipitate in the bottom of the tank.

Because the supernatant solution enters the ED saturated with CaSO₄, removal of NaCl causes the diluate to be somewhat supersaturated with CaSO₄. However, the slow kinetics of precipitation of CaSO₄ and the short residence time of the solution in the ED stack favor a delay in precipitation until the diluate enters the precipitation chamber where it comes into contact with the previously formed precipitate that provides sites for crystallization of CaSO₄·2H₂O. The point of entry of the diluate is preferably in the bottom region of the precipitation chamber where it can have maximum contact with the previously formed precipitate. ED treatment of the batch of supernatant continues until a target concentration of NaCl is reached in the supernatant or until the Na/Ca ratio or the Cl/SO₄ ratio in the recovered NaCl solution drops to a target level.

Controlling the composition of the batch solution being treated by ED is important to quality control for the recovered NaCl. Since the monovalent-anion-selective membranes exclude sulfate ions more effectively than monovalent-cation-selective membranes exclude calcium ions, the removal rate of calcium ions from the feed solution exceeds the removal rate of sulfate ions when the calcium and sulfate are present in equal ionic concentrations. If it is assumed that an optimum condition is to have equal amounts of calcium and sulfate in the feed solution to the ED, it would be necessary to provide supplemental calcium ions to the solution in order to avoid an imbalance in the proportions of calcium and sulfate ions. Control of the solution composition can be accomplished by measuring the concentration of calcium ions or the concentration of sulfate ions or both. With information on the solubility of CaSO₄ as a function of NaCl concentration and analytical information about the calcium or sulfate concentration or both, a control algorithm provides a signal to control an infusion pump that infuses the Mixed Cl concentrate stream produced by the EDM and thus achieve the desired addition rate of calcium to match the rate of calcium removal through the cation-exchange membranes of the ED. The Orion™ 2120XP calcium hardness analyzer (Thermo Scientific, 2013) could be used to measure the calcium level of the solution in real time. It is not necessary to maintain calcium and sulfate concentrations at equal levels throughout the batch. If specifications require a reduction in the sulfate level of the recovered NaCl, the control algorithm can be set to maintain the calcium level higher than the sulfate level.

Demonstration of Zero Discharge Desalination (ZDD)

Experiments were performed in December 2012 and January 2013 to determine the quality of NaCl that could be achieved with batch ED treatment of the supernatant. The results are shown in Table 30 and Figure 89. The total duration of this series of experiments was 25 hours.

Table 30.—Ion Ratios and Relative Transport during Batch ED of Supernatant Prepared with Simulated Alamogordo EDM Concentrate Streams

Sample	Day	Ion Ratio		Relative Transport	
		Na/Ca	Cl/SO ₄	Na/Ca	Cl/SO ₄
Feed-12-17-2013-end	1	28.2	6.2	-	-
Feed-12-18-2013-end	2	26.3	5.8	-	-
Feed-12-19-2013-end	3	26.7	5.4	-	-
Feed-12-20-2013-end	4	28.0	5.2	-	-
Feed-12-21-2013-end	5	26.6	4.6	-	-
Feed-01-16-2013-end	6	27.8	4.3	-	-
Feed-01-17-2013-end	7	27.4	4.0	-	-
Feed-01-24-2013-end	8	28.4	3.8	-	-
Concentrate-12-17-2013-end	1	59.9	132.4	2.1	21.3
Concentrate-12-18-2013-end	2	68.0	165.5	2.6	28.8
Concentrate-12-19-2013-end	3	70.9	156.4	2.7	29.0
Concentrate-12-20-2013-end	4	69.1	122.8	2.5	23.4
Concentrate-12-21-2013-end	5	70.2	125.5	2.6	27.3
Concentrate-01-16-2013-end	6	64.5	67.3	2.3	15.6
Concentrate-01-17-2013-end	7	67.8	91.0	2.5	22.7
Concentrate-01-24-2013-end	8	58.2	34.6	2.1	9.1

The data in Figure 89 indicate that the concentration of Na⁺, Cl⁻, and Ca²⁺ in the feed tank decreased while Mg²⁺ and SO₄²⁻ ions remained unchanged. No explanation was found for the jump in concentrations on Day 6. The quality of the recovered NaCl was excellent up to the last day. Ion ratios for Na/Ca were consistently in the range of 60 - 70. Ion ratios for Cl/SO₄ began very high and decreased somewhat. It was fortuitous that there was an excess of SO₄²⁻ ions in the feed tank, because the monovalent AEM provides excellent rejection of sulfate.

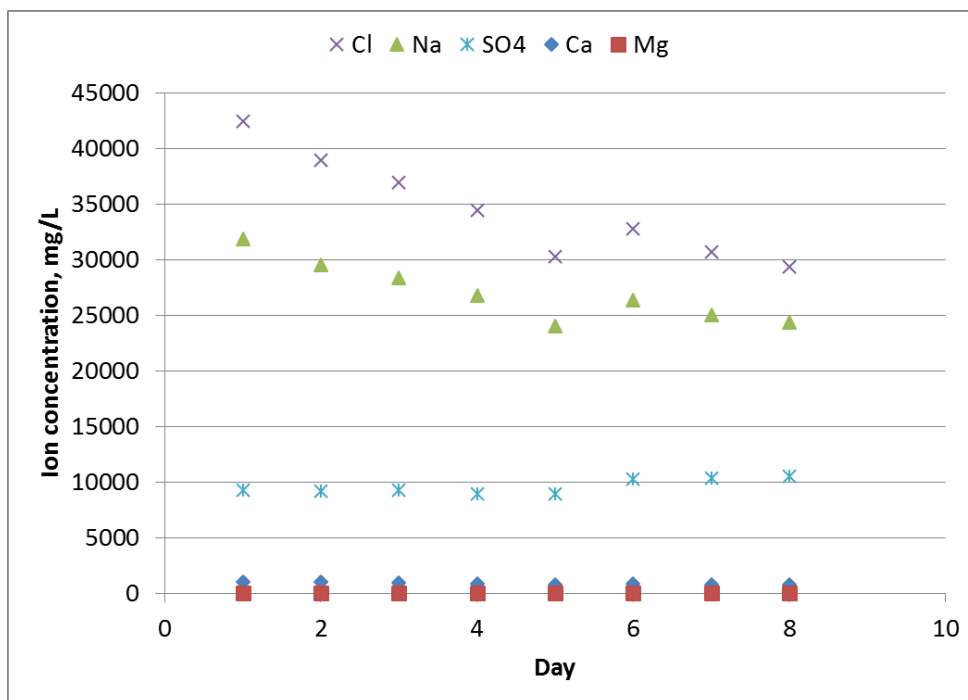


Figure 89.—Composition of solution in feed tank during batch ED experiments.

The selectivity of these membranes is usually described by a term called the relative transport number (RTN) with is defined as:

$$RTN_{Cl/so_4} = \frac{[Cl]_c}{[SO_4]_c} / \frac{[Cl]_d}{[SO_4]_d}$$

So the RTN can be calculated simply by dividing the ion ratio in the ED concentrate by the ion ratio of the ED feed. Calculated values for RTN shown in Table 30 indicate that RTN_{Cl/so_4} for the ACS membrane is an order of magnitude higher than $RTN_{Na/Ca}$ for the CMX-S membrane

On January 24, the flow rate of diluate was decreasing, so the experiment was terminated, and the ED stack was disassembled. White precipitate, presumed to be $CaSO_4 \cdot 2H_2O$, was observed in the depleting compartments.

The crystallization of $CaSO_4$ could be prevented by reversing the polarity of the electrodes in the ED stack, in the EDR process. Reversal of the polarity causes the electrical current to flow in the opposite direction, and the compartments that had been depleting compartments become concentrating compartments. When nascent crystals of $CaSO_4$ in the concentrating compartments are exposed to increasing NaCl concentrations, the increased solubility will cause $CaSO_4$ crystals to dissolve. The dissolution will add some calcium and sulfate ions to the recovered salt, but those amounts will be small. The main purpose of the current reversal is to eliminate nascent crystals so that they do not become sites for further crystal

Demonstration of Zero Discharge Desalination (ZDD)

growth, which could happen if the ED stack is used to treat multiple batches of supernatant without current reversal.

The values of $RTN_{Na/Ca}$ shown in Table 30 were lower than expected. The monovalent-cation-selective membrane used in these experiments was Neosepta CMX-S. Another monovalent-cation-selective membrane Neosepta CMS had been used in the EDM stack in the early pilot-scale work, and CMS was found to be the most fragile membrane in the EDM stack. Switching to CMX-S membranes definitely resulted in improved membrane life. A decision was made to do some comparative experiments with an ED stack containing CMS and ACS membranes.

Because the CMS membranes were in short supply, the ED stack was assembled with only 6 cell pairs and hence a potential of only 6 V was applied. This substantially lower voltage resulted into a lower current flow in the system (17 mA/cm^2 vs 50 mA/cm^2 in the December experiments). The analytical data and calculated values of ion ratios and RTN are shown in Table 31. The RTN values for Cl/SO_4 in the ACS membrane were nearly the same as in the previous experiments, but the RTN values for Na/Ca in CMS were doubled the values for CMX-S. The CMS membranes block more calcium ions than the CMX-S membranes did and produce NaCl solution with higher purity for the EDM process.

Table 31.—Ion Ratios during Batch ED of Supernatant with Neosepta CMS and ACS Membranes at Low Current Density

Sample	Ion Ratio		Relative Transport		Current Density
	Na/Ca	Cl/SO ₄	Na/Ca	Cl/SO ₄	mA/cm ²
Feed 02-11-2013-end	29.95	3.61			
Feed 02-12-2013-end	29.53	3.52			
Feed 02-13-2013-end	27.53	3.36			
Concentrate 02-11-2013-end	156	109	5.22	30.11	16.65
Concentrate 02-12-2013-end	174	145	5.88	41.03	16.95
Concentrate 02-13-2013-end	188	167	6.83	49.79	17.2

The same ED stack with ACS and CMS membranes was operated with sufficient applied voltage to achieve the current densities that were attained in the December experiment. Raising the current density caused an improvement in the quality and the concentration of the recovered NaCl solution. The conductivity of the ED concentrate indicated a 5.9 M concentration, and the ion ratios were about 300 for Na/Ca and 200 for Cl/SO_4 (see Table 32). This is the highest quality of NaCl recovered by ED and is of similar quality of salt that has been purchased for use in the EDM. These data indicate that the RTN values for both membranes improved at the high current density.

Table 32.—Ion Ratios During Batch ED of Supernatant with Neosepta CMS and ACS Membranes at High Current Density

Sample	Ion Ratio		Relative Transport		Current Density
	Na/Ca	Cl/SO ₄	Na/Ca	Cl/SO ₄	mA/cm ²
Feed 02-21-2013-end	28	3			
Feed 03-05-2013-end	27	3			
Concentrate 02-21-2013-end	369	208	13	68	57.8
Concentrate 03-05-2013-end	302	190	11	66	51.2

mA/cm² = milliampere per square centimeter

Implementation of NaCl recovery using ED with the mixing channel approach (i.e., single clarifier) could involve the purchase of an ED with 680 m² of exposed membrane. The installed cost for a very large ED system fitted with monovalent selective membranes is \$1,800/m² (private communication from Daniel Bar of Ameridia). Since the ED for NaCl recovery would be a smaller scale system, a multiplier of 1.5 is used to scale up the capital cost. Experimental data suggest that the power cost for recovering NaCl by ED would be 2.4 cents per pound. The cost for this strategy is estimated to cost between \$0.036-0.038 per pound of NaCl recovered, which would reduce the NaCl purchases by 33 to 35 percent. If Mg(OH)₂ is sold at half the market rate of \$800 per metric tons, annual operating costs could be reduced by nearly \$375,000. If a utility doesn't want to invest in the capital equipment (clarifiers, belt presses, etc), it is possible that a private company would make the investment to sell Mg(OH)₂ and the city could reduce its operating cost.

11.5 Voltage Drop in ED Stacks

In lab-scale ED, a large portion of the voltage applied to the electrodes is consumed in the electrode reactions and in the E-Rinse solution compartments. To obtain a true value of the voltage drop attributable to the cell pairs, it is necessary to subtract the voltage associated with the electrodes. Experiments were performed with the ED electrodes separated by a single cation-exchange membrane to determine the voltage that should be subtracted. Data for those experiments are plotted in Figure 90. The individual curves are associated with different concentrations of Na₂SO₄ in the E-Rinse. The data were well described by the equation:

$$V = 2.943 + 0.1205 \times I + 8.434 \times I/K$$

Where:

V is the applied potential (volts [V])

I is the current (amps [A])

K is the conductivity (mS/cm) of the rinse solution

Demonstration of Zero Discharge Desalination (ZDD)

The conductivity of the E-Rinse solutions was routinely measured during the ED experiments, so there were sufficient data to determine the portion of the applied voltage that should be attributed to the electrodes.

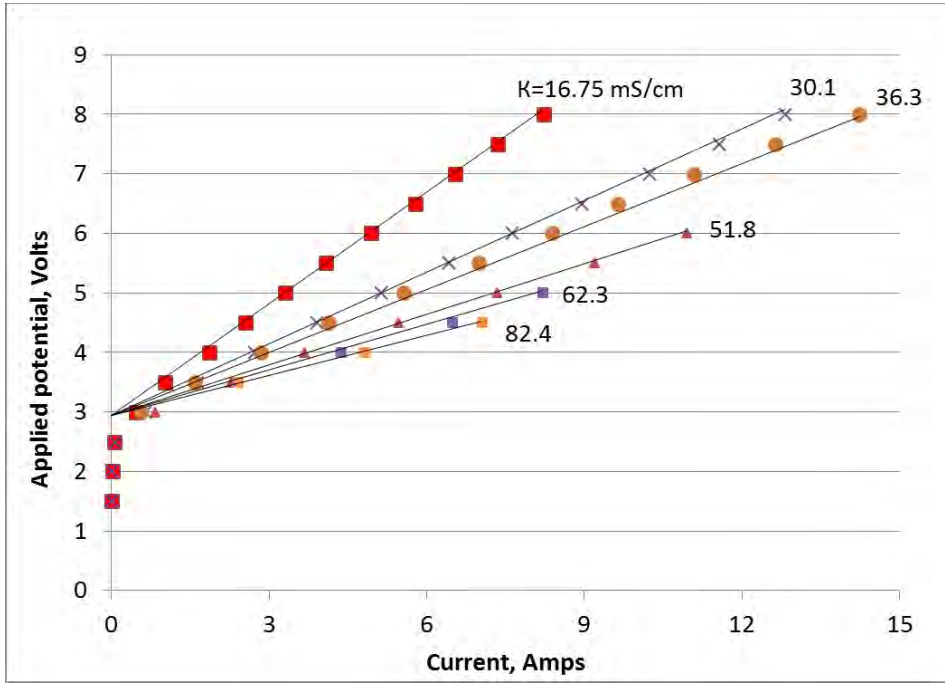


Figure 90.—Voltage drop in TS-2 stack with electrodes and a CMB membrane.

In a typical ED experiment, the measurements of the current and applied voltage are routinely performed at the end of an experiment simply by reducing the applied voltage incrementally and recording the voltage and current. Figure 91 shows the voltage measured when the ED was being shut down on December 21, 2012. In this experiment, the potential applied to the ED stack was 13 V, and 4.32 V is attributed to the electrodes. The power consumed by the ED was $(13-4.32) \text{ V} \times 10.68 \text{ A} = 89 \text{ W}$. Calculation of power consumption by the ED is shown in Table 33 for three laboratory experiments with an assumed current efficiency of 0.8 and power cost of 10 cents per kWh.

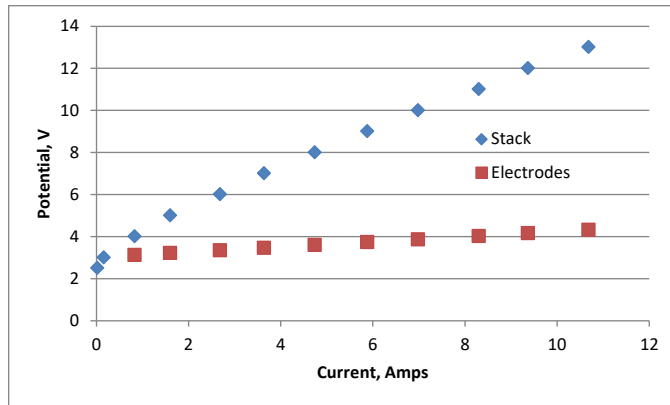


Figure 91.—Measured potential and current of the ED stack as it was being shut down on 12/21/2012.

Table 33.—Energy Consumption for NaCl Recovery by ED

Date	Applied to Stack		E-Rinse Conc.	Electrodes	Power	Cell Pairs	Power Cost
	Volts	Amps					
12/21/12	13	10.19	98	4.26	89	10	2.28
2/13/13	6	3.44	53	3.52	9	6	1.08
3/5/13	10	10.24	76	4.29	58	6	2.48

Note: Current efficiency = 80%, power cost = 10 cents per kWh

12. Conclusions and Recommendations

12.1 Technical Performance and Cost Evaluation

Outside of this project, Black and Veatch has evaluated EDM in Florida (Bond et.al., 2011) and most recently in Beverly Hills (Bond et.al., 2015). UTEP and Veolia participated by providing equipment, training, and support for the research sponsored by the California Energy Commission. High recovery with acceptable water quality for blending was achieved. The reported total EDM recovery of water from the RO concentrate was 91 to 92 percent, which would improve the overall recovery of the Beverly Hills RO plant from 80 percent to 97-98 percent. Details can be found in the WERF 5T10 report and in Section 9 of this report.

The capital costs for the BWRO and ZDD systems with deep well injection are summarized in Figure 92. The capital cost for the ZDD system is similar to the BWRO system capital costs, but the relative proportion for the concentrate disposal is quite different between ZDD and BWRO. The capital cost of BWRO with evaporation ponds is greater than ZDD due to the large evaporation ponds required for BWRO. If a utility does not have land available for large evaporation ponds, and, therefore, considers deep well injection (DWI), the capital cost for ZDD with a small evaporation pond is 13 percent higher than BWRO. Since the EDM concentrate streams are above 100,000 mg/L TDS, it is unlikely that they would be disposed of using DWI since the concentration may not be allowed to be disposed of using this method.

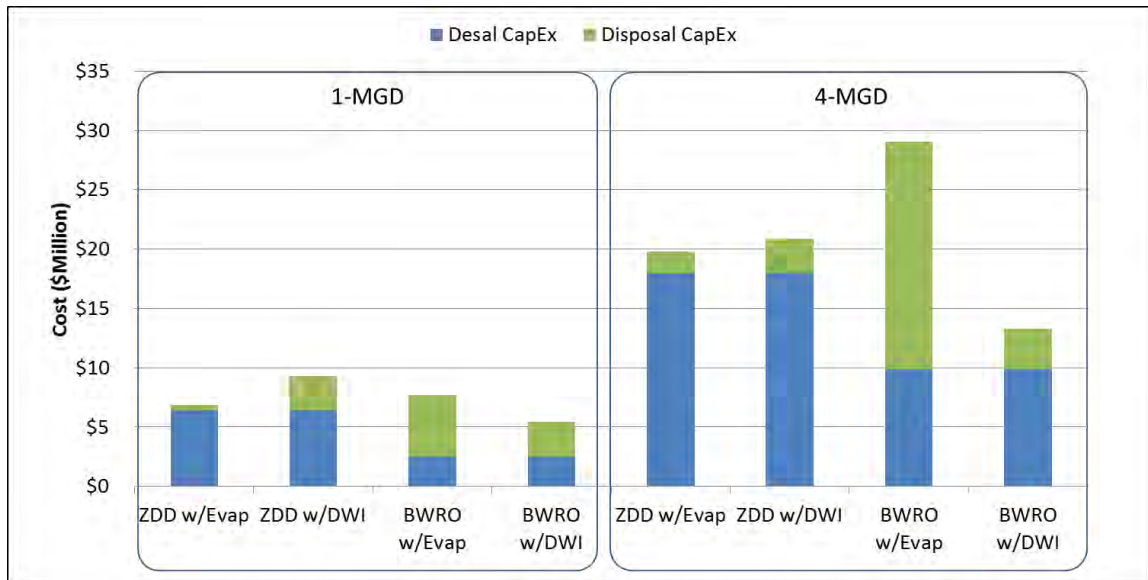


Figure 92.—Capital cost estimates for ZLD, ZDD and BWRO for 1 MGD and 4 MGD designs.

This BWRO unit cost range is likely low because our analysis excludes the cost of labor. An example of an existing full-scale BWRO facility is El Paso Water Utilities' Desalination Plant (concentrate disposal with deep well injection) which reports to be around \$1.53 per kgal for its 27.5 MGD facility. While site-specific details can play a larger role, in general, the unit cost to produce 1-4 MGD should be higher than the unit cost to produce 27.5 MGD because of economies of scale. ZDD, at 98 percent recovery, maximizes the amount of water available for consumption while minimizing the environmental and economic cost of concentrate disposal.

The total cost of water produced is summarized in Figure 93. As expected, there is a decrease in the unit cost of water as the production increases due to there being some effect of economies of scale. The estimated capital cost for the ZDD system accounts for 35 to 37 percent of the total cost of water production for ZDD and nearly 63 to 77 percent of the total cost of water by BWRO. The ZDD primary operations and maintenance (OpEx) cost drivers are the NaCl and EDM power consumption. Future optimization and research will aim to reduce the cost of both of these. Using methods proven to be feasible at a bench scale (see Section 11), it is reasonable to expect that up to 90 to 95 percent of the NaCl already in the EDM waste streams can be recovered and used in the EDM, which would reduce the cost of NaCl purchased by up to 50 percent, reduce the amount of deliveries by more than 70 percent, and reduce or eliminate the capital cost of the larger evaporation pond (the combined size for the NaCl recovery evaporation ponds is 10.7 acres vs 18.2 acres for ZDD without NaCl recovery). Improved energy efficiency of the EDM system could also lead to reduced capital cost, because fewer EDM stacks would need to be purchased. Further optimization will minimize recirculating flow to each stack, improve current density and efficiency, and minimize back diffusion from the concentrate streams. Finally, while not included in Figure 93, ZDD will reduce the amount of power used by the Snake Tank wellfield because less volume will be required to produce the same amount of energy. ZDD would reduce the wellfield pumping cost by at least 30 percent (assuming 1000-ft depth for wells). If additional pressure is needed to transport the water from the wellfield to the desalination plant, additional energy savings will be realized if a ZDD system is installed.

Demonstration of Zero Discharge Desalination (ZDD)

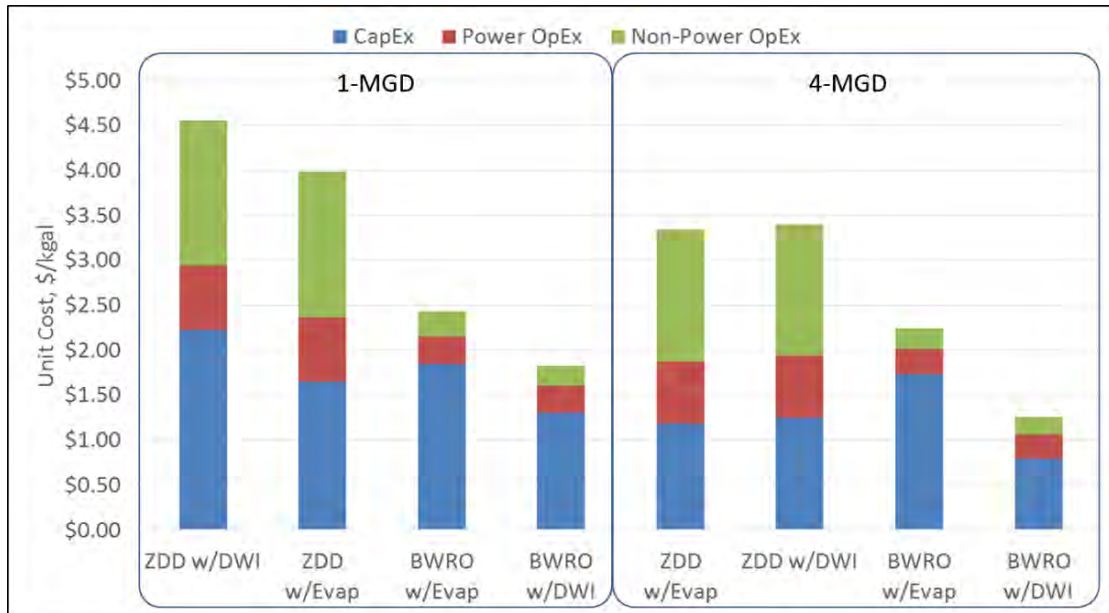


Figure 93.—Production cost estimates for ZLD, ZDD and BWRO for 1 MGD and 4 MGD designs.

Conceptual costs for NaCl recovery were estimated for four methods of production, ranging from low capital cost to high capital cost. There is increasing purity expected with increasing cost. All three methods use lime to remove magnesium from the Mixed Cl stream using a clarifier, but the potential revenue for $Mg(OH)_2$ (\$375,000 per year) is not included. In Method A (see Table 34) the Mixed Cl stream is combined with the Mixed Na stream stoichiometrically to precipitate calcium sulfate in a clarifier that overflows to an evaporation pond. The remaining fluid, called the supernatant, would be allowed to concentrate in the evaporation pond and then be used directly as the NaCl supply for the EDM. The second method (B in Table 34) would use a mixing channel in place of the $CaSO_4$ clarifier, but all other aspects are the same (the anticipated recovery is expected to be lower than the clarifier method). The final methods (C and in Table 34) would employ an ED system fitted with monovalent-selective ion exchange membranes to recover a high purity NaCl solution from the mixing channel (C) or clarifier effluent (D). Both C and D assume that the process is expected to recover 10 percent less than the clarifier methods. The data in the first column of Table 34 are for the case in which all of the NaCl is purchased. All of the methods have been evaluated at a laboratory scale and are deemed feasible. If a NaCl recovery method is implemented, the annual NaCl purchase cost is expected to be reduced by as much as 28 percent based on a NaCl purchase price of \$0.075/lb. If a buyer could be identified for the $Mg(OH)_2$, the revenue would cover the annual cost of NaCl recovery. Detailed calculations are provided in Appendix E.

Table 34.—Conceptual NaCl Recovery Costs (4 MGD, Alamogordo)

		4 MGD	A	B	C	D
NaCl needed	lb/day	48,350	--	--	--	
NaCl produced	lb/day	--	35,910	35,910	32,319	32,319
% NaCl recovered	--	--	90%	90%	81%	81%
CapEx of equipment	\$/yr	--	\$159,377	\$88,942	\$248,686	\$319,122
OpEx	\$/yr		\$159,126	\$267,010	\$267,087	\$159,203
NaCl purchased	lb/day	--	12,440	12,440	16,031	16,031
Blended NaCl purity	Cl/SO ₄	--	75	75	194	194
Solar salt (99.1%)	Cl/SO ₄	232	--	--	--	--
Morton (99.5%)	Cl/SO ₄	128	--	--	--	--
Unit cost	\$/lb	\$0.075	\$0.022	\$0.024	\$0.038	\$0.036
Annual NaCl cost (\$1000)		\$1,324	\$633	\$661	\$885	\$859

Incorporation of NaCl recovery would lead to a considerable decrease in the overall cost of desalination by ZDD. Operating costs would decrease using any of the methods described, but the lower cost methods show potential to reduce the overall unit cost of treatment by nearly 30 percent. Figure 94 compares the unit cost of a 4 MGD system with evaporation ponds and the same system with the four NaCl recovery methods summarized in Table 34.

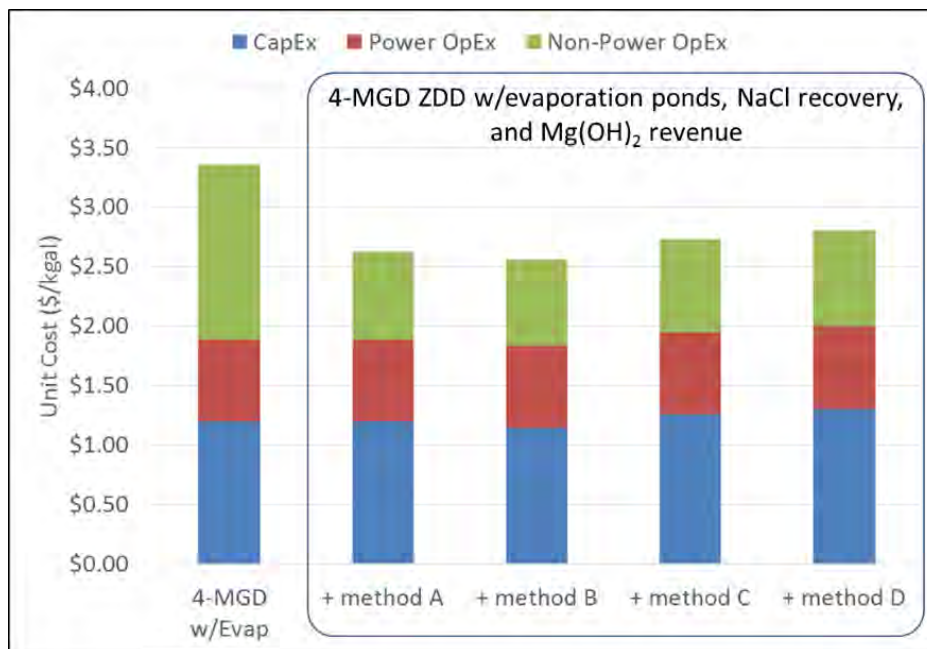


Figure 94.—Comparison of 4 MGD ZDD System with and without NaCl Recovery (Alamogordo).

12.2 Problems Encountered & Solutions

Developing and demonstrating a new technology is challenging, and problems can and do arise, offering opportunities for improvement. Several challenges were encountered during the piloting and demonstration activities. The most important challenges and the solutions identified are summarized here.

Many of the problems encountered in Colorado and some in Alamogordo can be attributed to inadequate instrumentation and controls and/or lack of remote viewing and control access to the equipment. Specific issues encountered include:

- Improper (or lack of) automated shutdown of equipment. Alarm conditions such as low flow, high pressure, and high temperature were intended to shut down the 20-gpm equipment. Additionally, the interlock, which would shut down the NF if the EDM shut down (and vice versa), between the NF and EDM containers was not working as intended. After the La Junta pilot, the project team found problems in the code that were modified before the system was restarted in Brighton.
- An additional conductivity alarm setpoint will be added for automated shutdown based on low conductivity measured in any of the EDM streams. This will protect the EDM stack from loss of chemical feed or DI valves getting stuck in the open position.
- The Mixed Na and Mixed Cl conductivity probes did not read or calibrate properly, likely because of stray current in the streams. The probes were relocated in both of these concentrate streams (in both the Brighton and Alamogordo locations), and the relocated probes provided accurate signals. This design change will be implemented at the full scale.
- Remote access and control is crucial for long-term operation; however, piloting budgets don't always allow for all of the necessary instrumentation and interfaces for remote access. The 20 gpm pilot does have remote viewing capability, so the operators must make modifications in person if problems occur. The 40 gpm pilot has remote viewing and control capability.

Training new operators was another challenge for the project team. This is part of commercialization, and many lessons were learned at the project sites. A training manual, such as EDM stack tightening, daily sampling, and operational checklists and alarm/shutdown setpoints, was written and used for Brighton, and that manual was the basis for training operators in Beverly Hills. Adherence to the procedures in the manual seemed to reduce the number and frequency of equipment problems. This manual will be the basis for full scale training operators at full scale and other pilot demonstration sites. It will be updated as new lessons are learned and better practices identified.

The EDM stacks are constructed with center dividers intended to reduce shunt currents by blocking direct current flow between the electrodes through the solution manifolds. These dividers were the source of internal leakage that caused scale formation and reduced electrical efficiency. The solution compartments on each side of the divide are NaCl and Mixed Na. When these streams mixed, sulfate that entered the NaCl solution compartment was subsequently transported from the NaCl compartment into the Mixed Cl compartment. This is believed to be the cause of membrane swelling and led to loss of flow and increased pressure in the EDM stacks in Alamogordo and in early Beverly Hills activities. New dividers made of thicker HDPE material were constructed by project personnel after the Alamogordo Phase 2 activities were completed. The swollen membranes were replaced in January 2014 in Beverly Hills. The new dividers appeared to eliminate major internal leaks, as scale formation was no longer evident in Brighton and Beverly Hills. To further mitigate this problem, a minor stack modification is being evaluated that could eliminate the scale formation problem even if slight leaks occur. Additionally, MEGA is evaluating different center divider designs that could reduce or eliminate leakage at the center of the stack. Future work will implement solutions identified.

12.3 Recommendations and Future Research

ZDD has shown potential to greatly improve desalination recovery and provide good quality drinking water or blend water at four sites with different water quality. Multiple configurations have been evaluated and most of the operating parameters necessary for full scale design have been identified, tested, and validated. Future research is needed to better optimize the EDM system for improved electrical efficiency and to better understand the interaction between salts and solubility in the Mixed Na stream. Finally, the NaCl recovery options tested at the bench scale should be tested at the pilot scale for verification and eventual implementation at the full scale. Each of these opportunities is described here briefly.

Future research will include better characterization of the upper operating limits for solubility in the Mixed Na stream to provide utilities with better information for conductivity setpoints at various temperatures. All anions in the brackish water feed will end up in the Mixed Na stream at very high concentrations. This is not a problem for chloride and sulfate, as sodium chloride and sodium sulfate are both highly soluble at normal operating temperatures (i.e. 25 °C or higher). However, bicarbonate (and carbonate) can be a problem. Sodium carbonate is similar to sodium sulfate, both in terms of solubility and in terms of its solubility-temperature relationship. Both salts have steep declines in solubility at low temperatures. This is not expected to be a problem at the full scale, where the equipment will be inside a building (pilot systems were in uninsulated containers with exposed piping subject to cold outdoor temperatures in fall and spring). Sodium bicarbonate is much less soluble than the other salts in the Mixed Na stream. Crystallization of sodium bicarbonate can be mitigated by adding acid

Demonstration of Zero Discharge Desalination (ZDD)

directly to the Mixed Na stream or by reducing bicarbonate alkalinity in the brackish water feed by adding acid and possibly removing CO₂ with an aeration tower.

More research is needed to identify the true limiting current density for safe operations (conservative setpoints have been identified). Most of the scientific literature for limiting current density is based on a two compartment ED system and there is a gap in understanding and interpreting the effect of limiting current density in EDM. This project's work in New Mexico and Colorado, combined with Black & Veatch's efforts in Florida and California, have provided the necessary operational and empirical information to identify conservative setpoints for operation. Careful laboratory experiments aimed at describing the effects of cell velocity, solution chemistry and concentration, as well as NaCl and E-Rinse concentration are needed to fully optimize the EDM and minimize energy consumption.

NaCl recovery has been demonstrated to be feasible and limits for the level of impurities for safe use have been identified. Additional research is needed to determine what level of impurities will cause swelling of the ion exchange membranes between the NaCl and EDM concentrate compartments. Research will also look at whether this effect is limited to a single type of membrane. Full-scale demonstration of NaCl recovery processes, including operating a full-scale EDM system with recovered NaCl should also be performed.

Finally, continuous operation for extended periods of time by utility engineers and water operators will be a primary focus of any future efforts to demonstrate and commercialize this technology.

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APPENDIX A

GRAB SAMPLE DATA

Appendix A1 – Alamogordo, Year 1

Appendix A2 – Alamogordo, Year 2

Appendix A3 – Brighton, Colorado

(Note: La Junta, Colorado, data can be found in (11) and Beverly Hills data can be found in (7))

APPENDIX A1
ALAMOGORDO,
YEAR 1 LOG SHEET DATA

Appendix A1-1.—Alamogordo Year 1 Grab Sample Data

pH	3/15/11	3/16/11	3/17/11	3/18/11	3/21/11	3/22/11
FEED WATER	7.76	6.86	5.73	7.53	7.67	7.55
NF FEED	7.45	7.44	7.04	7.51	7.58	7.5
NF PERMEATE	7.55	7.63	7.3	7.5	7.55	7.4
NF REJECT	4.95	7.45	7.3	7.24	7.12	7.3
EDM FEED	5.28	7.45	7.15	7.09	6.95	7.03
EDM DILUATE	5.4	7.3	7	6.83	6.62	6.75
EDM MIXED CHLORIDE SALTS	2.41	4.7	5.8	5.8	3.11	5.38
EDM MIXED SODIUM SALTS	6.42	8.2	8.11	8.3	7.62	8.06
SODIUM CHLORIDE STREAM	3.01	6.97	7.17	6.83	4.02	6.42
Temperature (C)	3/15/11	3/16/11	3/17/11	3/18/11	3/21/11	3/22/11
FEED WATER	23.2	22.9	23.8	25.8	22.9	22.3
NF FEED	23.6	23.2	23.8	26.4	23.5	22.4
NF PERMEATE	23.5	23.3	23.8	26.5	23.8	22.4
NF REJECT	23.1	23.7	23.9	26.9	23.7	22.5
EDM FEED	24.1	24.1	24.6	27.6	25	23.1
EDM DILUATE	24.4	24.6	24.9	27.8	24.7	23.4
EDM MIXED CHLORIDE SALTS	24.7	24.9	24.3	27.7	25.0	23.7
EDM MIXED SODIUM SALTS	24.2	25.1	24.5	27.6	25.2	23.8
SODIUM CHLORIDE STREAM	24.6	24.7	24.9	28.1	25.2	23.3
Conductivity (µS/cm)	3/15/11	3/16/11	3/17/11	3/18/11	3/21/11	3/22/11
FEED WATER	3012	3448	3409	2704	2675	3248
NF FEED	2286	2815	2788	2137	2067	2790
NF PERMEATE	901.1	1220	1206	769.3	755.7	1144
NF REJECT	4211	4689	4786	3746	3488	4685
EDM FEED	2400	2479	2740	2180	2160	2883
EDM DILUATE	1409	1298	1579	1119	1111	1660
EDM MIXED CHLORIDE SALTS	50900	76650	77240	89450	122000	137900
EDM MIXED SODIUM SALTS	67600	82900	82170	91700	108100	113900
SODIUM CHLORIDE STREAM	50270	50930	47100	46270	55810	55920
Total Hardness (mg/L as CaCO₃)	3/15/11	3/16/11	3/17/11	3/18/11	3/21/11	3/22/11
FEED WATER	1020	1230	1240	850	800	1290
NF FEED	745	1045	970	630	580	960
NF PERMEATE	98	160	164	64	62	166
DI WATER	0.1	0	0	0	0.1	0
NF REJECT	1820	2290	2330	1490	1450	2230
Ca Hardness (mg/L as CaCO₃)	3/15/11	3/16/11	3/17/11	3/18/11	3/21/11	3/22/11
FEED WATER	605	745	770	500	510	730
NF FEED		610	555	370	360	600
NF PERMEATE	6.5	12	10.5	4.8	4.2	9.9
DI WATER	20	0	0	0	0.1	0
NF REJECT	1090	1290	1380	840	870	1200
Mg Hardness (mg/L as CaCO₃)	3/15/11	3/16/11	3/17/11	3/18/11	3/21/11	3/22/11
FEED WATER	415	485	470	350	290	560
NF FEED		435	415	260	220	360
NF PERMEATE	92	148	154	59	58	156
DI WATER	-20	0	0	0	0	0
NF REJECT	730	1000	950	650	580	1030

Appendix A1-1.—Alamogordo Year 1 Grab Sample Data

pH	3/23/11	3/24/11	3/25/11	3/28/11	3/29/11	3/31/11
FEED WATER	7.61	7.58	7.46	7.66	7.29	7.53
NF FEED	7.26	7.29	7.24	7.29	7	7.16
NF PERMEATE	7.24	7.25	7.22	7.24	7.02	7.15
NF REJECT	7.25	6.99	7.1	7.1	6.96	6.97
EDM FEED	6.77	6.89	6.99	6.9	6.82	6.9
EDM DILUATE	6.32	6.5	6.75	6.61	6.48	6.61
EDM MIXED CHLORIDE SALTS	5.26	1.96	2.26	1.95	1.83	1.66
EDM MIXED SODIUM SALTS	8.27	8.38	8.33	8.48	8.39	8.54
SODIUM CHLORIDE STREAM	6.72	3.37	4.39	3.29	3.28	2.92
Temperature (C)	3/23/11	3/24/11	3/25/11	3/28/11	3/29/11	3/31/11
FEED WATER	21.5	21.1	21.6	21.6	21.3	21.1
NF FEED	22.0	21.9	21.2	22.1	21.8	21.5
NF PERMEATE	21.9	21.8	21.2	22.4	21.8	21.6
NF REJECT	22.7	21.8	21.4	22.4	22.1	21.9
EDM FEED	23.1	22.8	22.7	23.6	23.4	22.9
EDM DILUATE	23.1	23	23	24.5	23.9	23.4
EDM MIXED CHLORIDE SALTS	23.1	23.2	22.5	23.9	23.7	23.5
EDM MIXED SODIUM SALTS	23.1	23.1	22.8	23.7	23.8	23.4
SODIUM CHLORIDE STREAM	23.2	23.3	23.3	24.2	24.1	23.6
Conductivity (µS/cm)	3/23/11	3/24/11	3/25/11	3/28/11	3/29/11	3/31/11
FEED WATER	2659	2939	3101	2886	3337	2775
NF FEED	1981	2168	2837	2561	3038	2468
NF PERMEATE	804.7	840	1006	827.1	1081	816
NF REJECT	3881	3976	5202	4776	5459	4620
EDM FEED	2195	2580	3713	3502	3940	3545
EDM DILUATE	1090	1298	2280	2004	2377	1890
EDM MIXED CHLORIDE SALTS	135900	112600	113000	129100	119800	126200
EDM MIXED SODIUM SALTS	111500	96390	93570	113800	111100	112500
SODIUM CHLORIDE STREAM	52440	55960	52400	63050	58700	62150
Total Hardness (mg/L as CaCO₃)	3/23/11	3/24/11	3/25/11	3/28/11	3/29/11	3/31/11
FEED WATER	760	920	1070	910	1230	870
NF FEED	590	670	955	805	1045	765
NF PERMEATE	90	76	100	70	114	60
DI WATER	0	0	0	0	0.2	0
NF REJECT	1500	1700	2210	1800	2500	1840
Ca Hardness (mg/L as CaCO₃)	3/23/11	3/24/11	3/25/11	3/28/11	3/29/11	3/31/11
FEED WATER	530	590	650	640	695	535
NF FEED	350	415	535	460	560	440
NF PERMEATE	5.1	5.4	6.8	4.4	7.5	4.2
DI WATER	0	0	0	0	0	0
NF REJECT	880	960	1340	1130	1455	1030
Mg Hardness (mg/L as CaCO₃)	3/23/11	3/24/11	3/25/11	3/28/11	3/29/11	3/31/11
FEED WATER	230	330	420	270	535	335
NF FEED	240	255	420	345	485	325
NF PERMEATE	85	71	93	65.6	106.5	55.8
DI WATER	0	0	0	0	0.2	0
NF REJECT	620	740	870	670	1045	810

Appendix A1-1.—Alamogordo Year 1 Grab Sample Data

pH	4/4/11	4/5/11	4/6/11	4/7/11	4/8/11
FEED WATER	7.98	7.44	7.25	7.54	7.54
NF FEED	7.62	7.04	6.98	6.79	6.92
NF PERMEATE	7.85	6.99	6.96	6.69	6.83
NF REJECT	7.01	7.02	6.91	6.74	6.75
EDM FEED	6.97	6.93	6.78	6.65	6.82
EDM DILUATE	7.12	6.64	6.38	6.3	6.59
EDM MIXED CHLORIDE SALTS	1.29	1.84	1.79	1.64	2.22
EDM MIXED SODIUM SALTS	7.8	7.93	7.18	7.93	7.06
SODIUM CHLORIDE STREAM	5.51	3.13	3	2.89	3.23
Temperature (C)	4/4/11	4/5/11	4/6/11	4/7/11	4/8/11
FEED WATER	21.1	21.7	22.4	23.5	23.6
NF FEED	21.4	22.2	23.1	24.1	24.7
NF PERMEATE	21.3	22.2	22.9	24.3	24.9
NF REJECT	21.4	22.4	23.3	24.2	24.9
EDM FEED	22.5	23.3	24.4	24.8	25.4
EDM DILUATE	22.9	24.3	24.8	25.3	26
EDM MIXED CHLORIDE SALTS	22.4	24.1	25.0	25.5	26.3
EDM MIXED SODIUM SALTS	22.6	24.1	25.0	25.6	26.2
SODIUM CHLORIDE STREAM	23.1	24.4	25.3	25.2	26.1
Conductivity (µS/cm)	4/4/11	4/5/11	4/6/11	4/7/11	4/8/11
FEED WATER	3000	2857	2902	2918	2991
NF FEED	2632	2724	2818	2762	3403
NF PERMEATE	851.5	871.7	890.2	861.7	1026
NF REJECT	4552	5021	5259	5152	6453
EDM FEED	3751	4271	4429	4407	5767
EDM DILUATE	2191	2404	2591	2645	4181
EDM MIXED CHLORIDE SALTS		124800	104600	110700	114100
EDM MIXED SODIUM SALTS		109000	105600	102500	108900
SODIUM CHLORIDE STREAM		61280	59580	61690	55290
Total Hardness (mg/L as CaCO₃)	4/4/11	4/5/11	4/6/11	4/7/11	4/8/11
FEED WATER	930	960	800	840	1180
NF FEED	790	845	885	850	1145
NF PERMEATE	82	94	74	58	76
DI WATER	0.2	0	0	0	0
NF REJECT	1830	1960	2240	2140	2780
Ca Hardness (mg/L as CaCO₃)	4/4/11	4/5/11	4/6/11	4/7/11	4/8/11
FEED WATER	575	585	570	595	585
NF FEED	445	500	545	480	645
NF PERMEATE	4.8	4.9	4.2	4	3.5
DI WATER	0	0	0	0	0
NF REJECT	1110	1245	1250	1170	1650
Mg Hardness (mg/L as CaCO₃)	4/4/11	4/5/11	4/6/11	4/7/11	4/8/11
FEED WATER	355	375	230	245	595
NF FEED	345	345	340	370	500
NF PERMEATE	77.2	89.1	69.8	54	72.5
DI WATER	0.2	0	0	0	0
NF REJECT	720	715	990	970	1130

Appendix A1-1.—Alamogordo Year 1 Grab Sample Data

T. Alkalinity (mg/L as CaCO₃)	3/15/11	3/16/11	3/17/11	3/18/11	3/21/11	3/22/11
FEED WATER	120	130	35	210	210	215
NF FEED	55	130	95	110	95	110
NF PERMEATE	85	100	90	90	85	90
NF REJECT	100	145	115	110	110	135
Sulfates (mg/L)	3/15/11	3/16/11	3/17/11	3/18/11	3/21/11	3/22/11
FEED WATER	1150	1250	1150	1000	900	1000
NF FEED	1000	1050	800	700	650	1000
NF PERMEATE	14	16	16	12	10	14
EDM FEED	1250	1000	850	950	900	1100
EDM DILUATE	450	450	650	350	300	650
MIXED CL	350	50	650	825	1500	1325
NACL	600	125	150	400	1,150	425
Silica (mg/L)	3/15/11	3/16/11	3/17/11	3/18/11	3/21/11	3/22/11
FEED WATER	18.6	24.8	22.7	22.3	14.3	22.1
NF FEED	13.1	25.9	22.8	24.6	20.7	22
NF PERMEATE	19.9	20.8	21.4	20.9	22.5	23
NF REJECT	25.7	25.1	23.9	27.6	20.9	24.4
EDM FEED	25.4	28.4	25.5	25.5	27.9	23.8
EDM DILUATE	28.8	25	24.3	25.5	26.1	25.1

Appendix A1-1.—Alamogordo Year 1 Grab Sample Data

T. Alkalinity (mg/L as CaCO₃)	3/23/11	3/24/11	3/25/11	3/28/11	3/29/11	3/31/11
FEED WATER	220	230	195	210	200	200
NF FEED	70	75	85	95	70	85
NF PERMEATE	75	65	80	60	55	60
NF REJECT	105	95	125	110	100	125
Sulfates (mg/L)	3/23/11	3/24/11	3/25/11	3/28/11	3/29/11	3/31/11
FEED WATER	950	1150	1450	1300	1300	1150
NF FEED	800	900	1200	1250	1200	900
NF PERMEATE	9	11	15	11	13	8
EDM FEED	1100	1700	2150	1900	2050	1850
EDM DILUATE	500	450	1150	900	1150	850
MIXED CL	2450	1700	1800	1250	1450	475
NACL	1,000	500	500	250	600	625
Silica (mg/L)	3/23/11	3/24/11	3/25/11	3/28/11	3/29/11	3/31/11
FEED WATER	23.6	22.3	16.7	15.9	19	20.9
NF FEED	23.5	23.1	23.7	19.1	21.5	24.2
NF PERMEATE	26.1	23.4	23.2	22.2	21.5	21.7
NF REJECT	22.1	17.5	27.8	27.9	25.7	24.8
EDM FEED	24.9	23.3	21.3	26.9	25.2	27.9
EDM DILUATE	23.7	22.3	20.2	24.6	25.9	29.2

Appendix A1-1.—Alamogordo Year 1 Grab Sample Data

T. Alkalinity (mg/L as CaCO₃)	4/4/11	4/5/11	4/6/11	4/7/11	4/8/11
FEED WATER	210	205	210	210	200
NF FEED	70	110	100	95	95
NF PERMEATE	60	70	65	95	80
NF REJECT	90	135	135	105	120
Sulfates (mg/L)	4/4/11	4/5/11	4/6/11	4/7/11	4/8/11
FEED WATER	1100	1300	1000	850	850
NF FEED	850	1400	1400	1150	1850
NF PERMEATE	12	7	10	12	15
EDM FEED	2100	2500	2400	2300	3600
EDM DILUATE	1000	1200	1450	1100	2200
MIXED CL	2250	1000	900	1325	950
NACL	25	200	500	875	1,325
Silica (mg/L)	4/4/11	4/5/11	4/6/11	4/7/11	4/8/11
FEED WATER	11.1	18.3	22.4	20.1	22.3
NF FEED	19.7	20.5	20.1	24.1	23.6
NF PERMEATE	16.8	22.4	23.1	23.2	22.1
NF REJECT	23.9	23	28	25.1	25.1
EDM FEED	25.8	23.1	26.4	18.7	24.3
EDM DILUATE	25.7	26.5	24	26.6	25.9

Appendix A1-2.—Alamogordo Year 1 Log Sheet Data

	3/15/11	3/16/11	3/17/11	3/18/11	3/21/11	3/22/11
EDM STACK						
Stack Voltage (VDC)	79.7	101.3	90.8	90.7	100.4	100.5
Stack Amperage (amps)	32.6	37.4	35.7	32.7	31.7	38.2
NaCl added (lb)	400	400	400	400	200	200
Totalized Permeate (gal)	17742	19492	20177	20399	20168	20026
Totalized Permeate (1000 gal)	17.742	19.492	20.177	20.399	20.168	20.026
totalized DI Water (gal)	30887	32289	33,298	34193	35,787	36160
Total DI water used (gal)	1225.0	1402	1009	895	1594	373
DI used by Mixed Cl salts (gal)	710.4	483.6	422.4	74.4	14.4	0
DI used by Mixed Na salts (gal)	379.50	309.00	202.50	0.00	0.00	0.00
DI used for salt preparation, etc.	135.10	609.40	384.10	820.60	1579.60	373.00
Total Concentrate flows (gal)	399.47	399.47	399.47	399.47	399.47	452.73
Operational Data						
RO Concentrate Flow Rate (gpm)	7	7	7.2	6.7	6.8	7.2
RO Concentrate Conductivity (µS/cm)	4694	4848	4911	4415	4220	4815
RO Concentrate pH	7.58	7.34	7.26	7.33	7.25	7.29
EDM Feed Conductivity (mS/cm)	6.3	5.8	6.7	5	4.8	6.8
EDM Feed Flow Rate (gpm)	25.6	25.1	25.7	25.6	25.2	24.1
EDM Feed Pressure (psi)	9.5	9.9	9.5	9.6	10.2	10.7
EDM Feed Temperature (°C)	26.2	25.5	25.3	29.1	25.9	24.1
Mixed Sodium Salt Conductivity (mS/cm)	30.9	42.3	41.7	49.4	65.9	69.2
Mixed Sodium Salt Feed Flow Rate (gpm)	22.6	22.7	23.5	22.8	20.1	19.2
Mixed Sodium Salt Feed Pressure (psi)	10.4	11.2	10.3	10.5	11.1	11.7
Mixed Chloride Salt Conductivity (mS/cm)	35.5	49.3	49.7	62.7	94.1	80.8
Mixed Chloride Salt Feed Flow Rate (gpm)	25.3	24.3	25.4	25.3	24.6	24
Mixed Chloride Salt Feed Pressure (psi)	10.4	11.4	10.3	10.4	11.2	11.8
Mixed Chloride Salt Feed Temp (°C)	26.8	26.4	26.2	30	26.8	25
Sodium Chloride Conductivity (mS/cm)	30.8	27.8	28.1	30.2	33	34.2
Sodium Chloride Feed Flow Rate (gpm)	21.3	20.4	21.4	21	20.5	19.6
Sodium Chloride Feed Pressure (psi)	10.6	11.4	10.4	10.8	11.6	12
Electrode Rinse Conductivity (mS/cm)	13	11.8	16.2	15.2	19.9	17.8
Electrode Rinse Feed Flow Rate (gpm)	4.4	4.4	4.4	4.4	4.3	4.4
Electrode Rinse Feed Pressure (psi)	35.8	35.5	36.1	35.6	35.5	35.9
Electrode Rinse Feed Temperature (°C)	25.2	26.4	26.3	29.3	26.4	23.9

Appendix A1-2.—Alamogordo Year 1 Log Sheet Data

	3/23/11	3/24/11	3/25/11	3/28/11	3/29/11	3/31/11
EDM STACK						
Stack Voltage (VDC)	119.9	119.6	119.9	135.3	135.0	150.1
Stack Amperage (amps)	33.6	36.3	49.3	49.9	55.4	49.6
NaCl added (lb)	0	200	320	480	480	480
Totalized Permeate (gal)	20039	20036	29118	29085	29085	29091
Totalized Permeate (1000 gal)	20.039	20.036	29.118	29.085	29.085	29.091
totalized DI Water (gal)	36,342	36547	36879	38317	38785	39684
Total DI water used (gal)	182	205	332	1438	468	899
DI used by Mixed Cl salts (gal)	0	78	243.6	243.6	230.4	199.2
DI used by Mixed Na salts (gal)	0.00	31.50	72.00	72.00	18.00	30.00
DI used for salt preparation, etc.	182.00	95.50	16.40	1122.40	219.60	669.80
Total Concentrate flows (gal)	391.86	433.71	498.39	498.39	595.40	547.85
Operational Data						
RO Concentrate Flow Rate (gpm)	7.1	7.1	9.1	9.7	9.9	10.9
RO Concentrate Conductivity (µS/cm)	4159	4657	5662	5276	5990	5031
RO Concentrate pH	7.18	7.19	7.3	7.15	7.18	7.17
EDM Feed Conductivity (mS/cm)	4.3	5.9	9.9	8.7	10.8	7.1
EDM Feed Flow Rate (gpm)	23.9	23.2	26.4	25.3	25.1	24.2
EDM Feed Pressure (psi)	10.9	11.5	12.7	13.2	13	14
EDM Feed Temperature (°C)	22.6	22.7	22.5	22.3	22.9	22.9
Mixed Sodium Salt Conductivity (mS/cm)	67.6	57	59.1	61.8	48	54.2
Mixed Sodium Salt Feed Flow Rate (gpm)	19.7	19	18.3	18.3	18.1	18
Mixed Sodium Salt Feed Pressure (psi)	11.9	12.3	12.4	13.8	13.1	14.2
Mixed Chloride Salt Conductivity (mS/cm)	81.3	72.9	70.4	68	71.8	68.6
Mixed Chloride Salt Feed Flow Rate (gpm)	23.9	23	22.9	22	22.6	21.5
Mixed Chloride Salt Feed Pressure (psi)	11.9	12.6	12.8	13.8	13.4	14.2
Mixed Chloride Salt Feed Temp (°C)	23.7	23.9	23.6	24.1	24.5	24.4
Sodium Chloride Conductivity (mS/cm)	33.8	32.9	35.3	34.6	35.3	32.7
Sodium Chloride Feed Flow Rate (gpm)	19.7	18.5	17.9	17.1	17	16.2
Sodium Chloride Feed Pressure (psi)	12.1	13.3	13.5	14.4	14	14.9
Electrode Rinse Conductivity (mS/cm)	24	22.3	24.1	20.3	23.9	20
Electrode Rinse Feed Flow Rate (gpm)	4.3	4.5	4.3	4.6	4.6	4.6
Electrode Rinse Feed Pressure (psi)	35.9	36.2	35.6	36.3	35.7	36.1
Electrode Rinse Feed Temperature (°C)	23.5	24.6	24.6	25.2	26.1	25.5

Appendix A1-2.—Alamogordo Year 1 Log Sheet Data

	4/4/11	4/5/11	4/6/11	4/7/11	4/8/11
EDM STACK					
Stack Voltage (VDC)	150.0	180.3	179.8	180.4	180.3
Stack Amperage (amps)	50.3	62.4	64.5	62.3	66.3
NaCl added (lb)	480	480	480	320	0
Totalized Permeate (gal)	41074	18893	33732	35341	35299
Totalized Permeate (1000 gal)	41.074	18.893	33.732	35.341	35.299
totalized DI Water (gal)	41074	41664	42460	43063	43699
Total DI water used (gal)	1390	590	796	603	636
DI used by Mixed Cl salts (gal)	0	129.6	108	60	88.5
DI used by Mixed Na salts (gal)	0.00	37.50	289.20	453.60	378.00
DI used for salt preparation, etc.	1390.00	422.90	398.80	89.40	169.50
Total Concentrate flows (gal)	593.50	599.21	616.33	593.50	608.72
Operational Data					
RO Concentrate Flow Rate (gpm)	11.4	12.6	12.6	12.5	12.8
RO Concentrate Conductivity (μ S/cm)	5176	5456	5740	5569	6874
RO Concentrate pH	7.11	6.93	6.79	6.72	6.7
EDM Feed Conductivity (mS/cm)	9.7	5.3	5.6	5.4	9
EDM Feed Flow Rate (gpm)	23.5	24.3	23.5	23.2	23.4
EDM Feed Pressure (psi)	14.7	14.5	14.8	15	15.4
EDM Feed Temperature ($^{\circ}$ C)	21.7	22.6	23.7	25.7	26.3
Mixed Sodium Salt Conductivity (mS/cm)	49.9	54.5	51.9	54.8	54.3
Mixed Sodium Salt Feed Flow Rate (gpm)	16.7	16.9	16.5	16.7	15.9
Mixed Sodium Salt Feed Pressure (psi)	15.2	14.8	14.7	15	15.6
Mixed Chloride Salt Conductivity (mS/cm)	68.3	65.3	67.2	56.1	66.4
Mixed Chloride Salt Feed Flow Rate (gpm)	19	20.7	20.3	20.6	18.9
Mixed Chloride Salt Feed Pressure (psi)	15.6	14.8	14.8	15	16
Mixed Chloride Salt Feed Temp ($^{\circ}$ C)	23.2	24.6	25.9	27.6	28.5
Sodium Chloride Conductivity (mS/cm)	31.7	34.5	32.5	34	33.3
Sodium Chloride Feed Flow Rate (gpm)	14.7	15.5	15	15.3	14.3
Sodium Chloride Feed Pressure (psi)	16.6	15.4	15.8	16.1	16.4
Electrode Rinse Conductivity (mS/cm)	24.6	22.3	19.9	20.8	23.9
Electrode Rinse Feed Flow Rate (gpm)	4.6	4.6	4.7	5	4.7
Electrode Rinse Feed Pressure (psi)	31.6	36.3	35.8	35.7	35.5
Electrode Rinse Feed Temperature ($^{\circ}$ C)	27.9	28.9	31.5	29	34.3

APPENDIX A2
ALAMOGORDO,
YEAR 2 GRAB SAMPLE DATA

Appendix A2: Alamogordo Year 2 Logsheet Data
EDM System Data

Item	EDM SYSTEM	Tag	units	3/28/13 16:56	3/29/13 10:14
1	EDM-1 Feed Flow	rotameter	gpm	31	31
2	EDM-2 Feed Flow	rotameter	gpm	26	26
3	EDM Total Feed Flow	FIT-2111	gpm	58.3	55.1
4	EDM Feed Pressure	PIT-2111	psi	18.6	18.6
5	EDM Feed Conductivity	AIT-2111	mS/cm	4.802	4.536
6	EDM Feed Temperature	TIT-2111	°F	76.7	76.2
7	EDM Diluate Conductivity	AIT-2110	mS/cm	3.295	3.14
8	EDM Diluate Temperature	TIT-2110	°F	79.6	78.8
9	EDM-1 Mixed Na Flow	rotameter	gpm	41	40
10	EDM-2 Mixed Na Flow	rotameter	gpm	35	35
11	EDM Total Mixed Na Flow	FIT-2113	gpm	76	75
12	EDM Mixed Na Pressure	PIT-2113	psi	19.5	19.5
13	EDM Mixed Na Conductivity	AIT-2113	mS/cm	2.876	25.288
14	EDM Mixed Na Temperature	TIT-2113	°F	78.9	78.1
15	EDM-1 Mixed Cl Flow	rotameter	gpm	40	41
16	EDM-2 Mixed Cl Flow	rotameter	gpm	34	34
17	EDM Total Mixed Cl Flow	FIT-2115	gpm	75.9	74.8
18	EDM Mixed Cl Pressure	PIT-2115	psi	19.5	19.3
19	EDM Mixed Cl Conductivity	AIT-2115	mS/cm	77.021	91.576
20	EDM Mixed Cl Temperature	TIT-2115	°F	79.4	78.7
21	EDM-1 NaCl Flow	rotameter	gpm	36	35
22	EDM-2 NaCl Flow	rotameter	gpm	33	33
23	EDM Total NaCl Flow	FIT-2117	gpm	69	68
24	EDM NaCl Pressure	PIT-2117	psi	19.6	18.7
25	EDM NaCl Conductivity	AIT-2117	mS/cm	45.842	47.532
26	EDM NaCl Temperature	TIT-2117	°F	79.1	78.4
27	EDM Total E-Rinse Flow	FIT-2124	gpm	0	9.5
28	EDM E-Rinse Pressure	PIT-2124	psi	11.9	11.9
29	EDM E-Rinse Conductivity	AIT-2124	mS/cm	23.209	22.743
30	EDM E-Rinse Temperature	TIT-2124	°F	80.1	78.1

Appendix A2: Alamogordo Year 2 Logsheet Data
EDM System Data

Item	4/1/13 9:28	4/1/13 11:15	4/1/13 13:22	4/2/13 14:38	4/3/13 16:40	4/5/13 10:06
1	29	31	31	31	32	32
2	28	26	26	26	25	25
3	55	57.6	56.9	57.6	51.2	54.7
4	20.1	18.5	18.5	18.5	18.8	18.1
5	4.891	4.468	4.494	4.487	4.248	4.331
6	73.4	77.8	79.5	79.8	75.9	76.8
7	3.518	3.034	3.062	3.091	2.887	2.941
8	75	80.3	82.2	82.6	77.6	78.8
9	40	40	40	40	36	36
10	35	3.4	35	35	30	31
11	75	43.4	75	75	66	67
12	21.1	19.4	19.3	19.2	19.4	18.5
13	37.97	87.714	22.328	22.134	47.099	4.642
14	74.3	80.3	81.6	81.9	77.1	78.3
15	37	35	38	38	37	35
16	31	29	32	33	29	27
17	72	66	64.8	69.5	67.4	65.2
18	20.9	18.8	19.2	19.2	19.4	18.5
19	58.393	36.977	78.435	74.271	63.795	90.901
20	75.1	79.6	82.2	82.5	77.8	79
21	41	39	39	39	40	40
22	38	35	35	35	35	35
23	79	74	74	74	75	75
24	20.9	18.9	19	18.9	19.3	18.4
25	45.911	50.985	42.178	48.827	47.389	46.857
26	74.9	80	81.9	82.2	77.6	78.8
27	10.2	9.3	10.2	10.1	9.2	12.3
28	12.2	12	12.1	12.2	12.2	11.9
29	23.178	23.329	23.081	23.553	17.612	22.69
30	72.6	79.4	82.4	82.9	75.8	76.9

Appendix A2: Alamogordo Year 2 Logsheet Data
EDM System Data

Item	4/5/13 11:48	4/8/13 10:51	4/8/13 14:42	4/8/13 17:05	4/9/13 10:49	4/9/13 16:35
1	33	33	33	34	34	34
2	25	24	24	24	24	24
3	56.9	55.2	59.2	53.7	56.3	58
4	18.3	18.5	19.1	19.1	20.4	20.5
5	4.351	4.152	4.703	4.772	4.778	5.314
6	78.1	78.8	80.9	81.3	77.7	78.4
7	2.959	2.768	3.285	3.34	3.469	3.812
8	80.4	80.9	83.3	83.8	79.9	80.7
9	39	38	41	42	42	44
10	30	30	31	31	31	32
11	69	68	72	73	73	76
12	18.3	18.6	19.6	19.7	21.4	21.1
13	4.574	4.702	2.082	2.022	2.144	1.952
14	79.6	80.3	82.7	83.2	79.2	79.9
15	38	35	41	42	42	44
16	29	29	29	30	30	32
17	63.8	68.2	64.3	71.1	76.3	67.6
18	18.6	19.3	19.4	19.6	21.2	20.9
19	82.929	88.842	43.882	38.831	52.395	39.833
20	80.4	81	83.3	83.9	79.9	80.6
21	39	42	42	44	45	45
22	37	38	37	39	37	37
23	76	80	79	83	82	82
24	18.4	18.7	19.6	19.5	20.6	20.7
25	45.49	47.38	46.299	55.103	47.09	45.79
26	80.1	80.7	83.1	83.6	79.7	80.3
27	12.3	11.5	11.3	12.1	11.5	12
28	11.9	11.9	12.2	12.1	12.8	12.9
29	23.397	21.988	22.436	22.028	22.089	21.642
30	79.9	79	83.1	83.7	76.7	78.3

Appendix A2: Alamogordo Year 2 Logsheet Data
EDM System Data

Item	4/10/13 11:16	4/10/13 16:30	4/11/13 11:55	4/11/13 13:42	4/11/13 16:00	4/11/13 17:03
1	34	34	34	34	33	33
2	24	24	25	25	24	25
3	55.8	56.7	57.5	58.1	55	55.4
4	20.7	20.5	21	21.1	21.3	21.5
5	5.192	5.703	4.905	5.134	5.441	5.446
6	77.8	78.3	77.2	77.8	78.7	78.8
7	3.578	3.904	3.324	3.376	3.588	3.665
8	80.2	80.9	79.8	80.6	81.5	81.8
9	43	43	41	41	43	42
10	31	32	31	31	33	32
11	74	75	72	72	76	74
12	21.5	21.4	21.9	21	21.1	21.6
13	1.96	1.935	2.012	1.89	1.969	1.892
14	79.6	80	79.1	79.7	80.8	81.1
15	44	44	43	44	44	44
16	31	30	31	32	33	33
17	69.4	73.3	72	74	71	70.3
18	21.8	21.4	21.4	21.7	21.2	21.2
19	38.175	33.917	43.077	35.427	30.651	31.589
20	80.3	80.7	79.8	80.5	81.5	81.8
21	45	45	43	45	44	43
22	37	37	36	38	37	37
23	82	82	79	83	81	80
24	20.5	21.5	21.4	21.4	21.8	21.5
25	46.034	49.059	47.937	42.579	46.011	45.676
26	79.9	80.4	79.5	80.2	81.2	81.5
27	11.5	10.4	10.4	11.5	11.3	11
28	12.8	13	13.1	13.3	13.4	13.4
29	22.407	20.696	22.249	22.118	19.979	21.11
30	78.9	79.9	79.4	81.1	82.2	82.5

Appendix A2: Alamogordo Year 2 Logsheet Data
EDM System Data

Item	4/17/13 10:45	4/17/13 14:54	4/17/13 16:15	4/18/13 10:30	4/18/13 13:03	4/18/13 16:16
1	31	31	31	31	31	31
2	27	27	27	27	27	27
3	53.3	56.7	56.8	59.4	59.2	60.4
4	14.1	13.9	14.1	14.6	14.8	14.5
5	4.258	4.338	4.453	4.672	4675	4822
6	78.5	80.8	80.1	75.6	77.1	77.7
7	2.995	3.07	3.195	3.476	3.464	3.629
8	80.6	83.2	82.2	77.5	79	80
9	32	32	32	33	34	34
10	30	30	29	29	28	30
11	62	62	61	62	62	64
12	14	13.9	14.1	14.5	14.4	14.6
13	5.441	5.449	44.014	68.785	61.5	61.626
14	80	82.4	81.6	76.9	78.7	79.4
15	30	33	33	33	33	33
16	27	30	29	29	29	29
17	60.3	57.9	57.5	62.5	62	59.1
18	13.9	13.9	14.5	15	14.7	14.7
19	147.415	165.757	147.836	127.936	164.4	168.289
20	80.6	83.1	82.2	77.6	79.3	80.1
21	34	36	36	39	38	39
22	32	34	34	37	36	37
23	66	70	70	76	74	76
24	14	14	14.7	14.4	14.7	14.4
25	47.556	50.533	58.378	53.525	48.154	48.558
26	80.3	82.8	82.1	77.2	78.9	79.8
27	11.9	12.8	0	0	0	0
28	10	10.3	10.3	10.5	10.6	10.9
29	20.398	22.408	20.891	21.777	21.941	22.271
30	79.6	83.2	82	75.7	78.3	79.8

Appendix A2: Alamogordo Year 2 Logsheet Data
EDM System Data

Item	4/19/13 10:41	4/19/13 13:57	4/19/13 17:50	4/19/13 21:50	4/20/13 8:27	4/22/13 11:24
1	32	31	31	31	31	32
2	27	26	26	26	26	26
3	59	55.5	55.3	56.6	56.8	57
4	14.5	14.7	14.6	14.8	14.7	15.3
5	4.826	4.835	4.81	4.99	5265	4.57
6	75.6	77.3	77.9	77.2	76.2	78.6
7	3.554	3.551	3.547	3.744	4.012	3.384
8	77.8	79.9	80.5	79.8	78.7	80.8
9	36	36	36	36	37	36
10	31	30	30	30	31	28
11	67	66	66	66	68	64
12	14.7	14.6	14.9	14.7	14.9	15.6
13	69.887	72.585	76.667	78.785	76.3202	54.411
14	77.2	79.2	79.6	79.1	78.1	80.1
15	35	36	36	36	37	35
16	30	31	31	31	31	28
17	57.7	59	62.3	60.3	64.4	57.2
18	14.5	14.3	14.6	14.9	14.9	15.9
19	175.395	166.434	166.883	158.21	161.506	114.3
20	77.9	79.8	80.4	79.7	78.7	80.8
21	39	39	39	39	39	41
22	37	37	37	37	37	34
23	76	76	76	76	76	75
24	14.9	14.8	15	14.9	14.9	15.2
25	50.045	52.326	48.832	45.666	49.629	45.843
26	77.5	79.5	80.1	79.4	78.3	80.6
27	0	0	0	0	0	0
28	10.8	10.8	10.7	10.8	10.9	10.7
29	23.101	24.974	24.622	22.834	23.376	23.461
30	76.9	79.6	80.1	78.2	78.1	81.7

Appendix A2: Alamogordo Year 2 Logsheet Data
EDM System Data

Item	4/22/13 13:34	4/22/13 16:18	4/22/13 21:45	4/23/13 8:27	4/23/13 11:00	4/23/13 15:22
1	32	33	32	32	28	27
2	26	27	26	26	25	25
3	58	59.9	57.1	54.3	52.9	51.4
4	15.2	15.4	15.7	15.7	15.1	15.1
5	4.635	4.806	5.06	5.539	5.486	5.436
6	79.5	80	78.7	78.9	79.8	81
7	3.471	3.607	3.758	4.049	3.879	3.881
8	81.9	82.5	78.7	81.4	82.5	84
9	36	37	37	37	33	33
10	28	29	29	29	28	29
11	64	66	66	66	61	62
12	15.5	15.5	15.4	15.4	15.2	15.1
13	48.047	44.151	57.981	50.225	48.129	41.159
14	81	81.9	80.4	80.7	81.8	83.2
15	35	35	37	37	36	35
16	28	29	29	30	31	30
17	61.6	60.2	64.7	67.8	63.7	61.6
18	15.6	15.6	15.9	15.7	15.1	15
19	106.453	124.852	138.275	148.59	152.376	122.776
20	81.8	82.5	76	69.1	65.7	78.7
21	41	40	40	41	35	35
22	34	34	35	35	30	30
23	75	74	75	76	65	65
24	15.3	15.2	15.7	15.7	15	15.1
25	47.251	53.126	50.642	46.35	44.718	54.911
26	81.5	82.2	80.8	81	82.2	83.7
27	0	0	0	0	0	0
28	10.7	10.9	11.1	11.2	11.2	11.1
29	23.708	23.688	24.123	23.271	23.261	26.263
30	83.3	84.3	80.9	81.1	82.5	84.9

Appendix A2: Alamogordo Year 2 Logsheet Data
EDM System Data

Item	4/23/13 22:30	4/24/13 8:16	4/25/13 16:52	4/25/13 21:45	4/26/13 8:25	4/30/13 16:00
1	27	31	27	28	28	27
2	25	28	0	0	0	0
3	51.1	54.2	25.3	27.1	27.6	24.4
4	15.2	18	11.4	11.8	12	12.3
5	5.578	5.674	4.851	4.829	4.811	4.865
6	79.5	77.9	79.9	78.7	76.2	83.6
7	4.072	4.332	3.37	3.449	3.47	3.296
8	82.2	80.3	82.3	81.1	78.3	86.2
9	34	38	30	31	33	31
10	29	30	0	0	0	0
11	63	68	30	31	33	31
12	15.2	18.3	11.9	11.9	12.1	12.3
13	45.772	107.644	107.064	106.504	106.79	108.521
14	81.4	79.5	81.6	80.4	77.7	85.5
15	35	38	32	32	30	30
16	30	32	0	0	0	0
17	63.6	68.2	30.9	32.1	33	31.2
18	15.4	18.5	11.4	11.5	12.1	12.6
19	143.2	1639.18	1660.94	1681.75	1547.48	151.409
20	67.4	-4	-4	-4	-4	86.3
21	35	40	30	31	32	33
22	32	35	0	0	0	0
23	67	75	30	31	32	33
24	15.2	18.4	11.6	11.6	12.1	12.6
25	43.347	43.983	52.592	50.878	50.957	55.762
26	81.8	78.9	82	80.8	78	85.9
27	0	0	0	0	0	0
28	11.2	11.8	12	12.1	12.4	12.8
29	27.4	25.191	21.674	21.584	21.518	20.95
30	81.1	78.9	82	79	76.1	86.6

Appendix A2: Alamogordo Year 2 Logsheet Data
EDM System Data

Item	4/30/13 21:51	5/1/13 8:53	5/1/13 13:35	5/1/13 16:41	5/1/13 21:20	5/2/13 7:56
1	27	28	27	27	27	28
2	0	0	0	0	0	0
3	27.1	26.6	26.8	26.1	26.8	27.8
4	12.8	13	13	13.1	13.2	13.6
5	5.144	5.489	5.108	4.932	4.888	5.599
6	82.1	80.1	82.1	85.2	83.1	79.2
7	3.555	4.047	3.622	3.569	3.54	4.299
8	84.8	82.4	84.8	87.7	85.4	81.2
9	31	31	30	32	30	30
10	0	0	0	0	0	0
11	31	31	30	32	30	30
12	13.4	13.1	13.5	13.5	12.9	13.4
13	106.331	105.592	104.483	104.516	115.759	110.73
14	84.1	81.7	84.1	87	84.7	80.4
15	31	33	33	31	31	32
16	0	0	0	0	0	0
17	32.7	32.6	32.5	32.4	32.5	31.5
18	13.2	13.4	13.2	13.5	13.5	13.4
19	160.687	150.749	158.307	163.858	152.579	147.23
20	84.7	82.4	84.8	87.8	85.4	81.2
21	33	34	35	33	34	34
22	0	0	0	0	0	0
23	33	34	35	33	34	34
24	13.4	13.4	13.6	13.5	13.4	13.5
25	53.143	50.966	50.783	51.834	51.481	50.216
26	84.5	82.2	84.5	87.4	85.1	81
27	0	0	0	0	0	0
28	13.2	13.6	13.3	13.4	13.5	13.8
29	21.92	2.809	21.372	21.947	22.007	21.507
30	82.2	80.8	86.5	88.1	83.2	77.5

Appendix A2: Alamogordo Year 2 Logsheet Data
EDM System Data

Item	5/2/13 14:37	5/2/13 21:46	5/29/13 13:50	5/31/13 13:32	5/31/13 15:20	6/3/13 11:33
1	27	27	32	32	33	33
2	0	0	0	0	0	0
3	25.3	26.3	31.4	33	30.4	33.5
4	13.6	13.7	23.3	22.4	22.6	27
5	4.731	4.98	4.735	5.128	5.888	5.336
6	81	78.1	90.1	84.3	85	84.4
7	3.446	3.726	3.412	3.975	4.806	4.184
8	83.1	80.2	92.2	86.2	87.1	86.1
9	34	32	28	29	30	40
10	0	0	0	0	0	0
11	34	32	28	29	30	40
12	14.3	14.3	23.5	22.8	23.1	26.8
13	106.017	106.972	102.069	82.05	81.34	73.546
14	82.4	79.5	91.6	85.6	86.4	86.6
15	31	31	40	39	38	42
16	0	0	0	0	0	0
17	31	31	40.2	36.1	36.7	38
18	14.2	14.2	23.1	22.6	22.9	27.9
19	158.041	154.467	148.588	152.436	156.688	123.665
20	83.1	80.2	92.3	86.5	87.3	86.5
21	34	33	40	41	41	45
22	0	0	0	0	0	0
23	34	33	40	41	41	45
24	13.9	13.8	23.8	22.7	23.2	26.8
25	51.458	53.745	52.834	49.968	50.744	50.667
26	82.9	79.9	91.9	86.1	86.8	86.3
27	0	0	0	0	0	0
28	14.1	14.3	19.3	19.4	19.3	20.7
29	21.859	21.186	22.554	22.221	21.812	22.074
30	82.8	76.7	90.3	86.5	87.7	86.7

Appendix A2: Alamogordo Year 2 Logsheet Data
EDM System Data

Item	6/3/13 14:46	6/4/13 11:10	6/4/13 14:55	6/4/13 16:50	6/10/13 0:00	6/14/13 15:35
1	32	32	33	32	24	25
2	0	0	0	0	29	28
3	30.2	31.5	31.2	32.8	50.3	51.5
4	26.9	24.6	25	25	14.3	14.6
5	5.469	5.076	5.117	5.135	4.771	4.26
6	86	84.9	86.6	87.1	86.6	87.4
7	4.29	3.851	3.939	3.954	3.3	2.833
8	87.9	86.7	88.6	89.1	88.2	89.4
9	42	41	39	37	26	30
10	0	0	0	0	22	32
11	42	41	39	37	48	62
12	26.7	25.6	24.8	25.1	15	14.6
13	112.128	107.087	100.407	104.942	6.482	26.625
14	88.4	87.3	89.2	89.6	88.9	89.8
15	41	41	40	38	25	29
16	0	0	0	0	30	32
17	40	38.1	35.3	35.6	61.3	62.5
18	27.7	24.5	25.4	26.1	14.2	14.4
19	158.744	138.194	160.881	162.308	9.391	104.309
20	88.4	87.2	89.1	89.6	88.7	89.5
21	42	45	41	41	26	30
22	0	0	0	0	30	32
23	42	45	41	41	56	62
24	27.2	25.1	24.6	25.6	14.8	14.6
25	50.71	50.489	54.842	53.837	46.179	48.953
26	88.1	56.9	88.8	89.3	88.6	89.4
27	0	0	0	0	0	0
28	20.8	20.5	20.6	20.7	10.7	11.2
29	21.796	21.389	21.812	21.85	18.954	22.131
30	89.1	86.9	89.7	90.3	88.3	87.6

Appendix A2: Alamogordo Year 2 Logsheet Data
EDM System Data

Item	6/17/13 13:47	6/18/13 11:15	6/18/13 14:15	6/18/13 17:23	6/19/13 10:21	6/19/13 14:26
1	26	25	25	25	25	25
2	32	32	32	32	33	33
3	57.6	54.4	56.7	54.7	56.2	56.2
4	18.1	18.1	18.2	18.1	18.6	18.7
5	4.752	4.758	4.676	4.664	5.768	4.674
6	86.5	85.6	87	89.5	85.3	86.9
7	3.492	3.442	3.394	3.387	4.548	3.566
8	88.3	387.3	88.8	89.6	87.1	88.5
9	35	35	35	34	36	35
10	39	39	39	37	41	40
11	74	74	74	71	77	75
12	17.9	18.8	18.5	18.3	18.4	18.8
13	40.421	35.356	34.717	32.407	49.729	35.894
14	88.7	87.8	89.3	90.1	87.6	89.2
15	30	31	33	32	32	32
16	36	36	37	35	38	37
17	70.8	69.5	67.4	67.2	68.2	72.7
18	18	18.7	18.9	18.2	19.1	18.9
19	148.937	95.903	132.379	136.586	144.53	142.211
20	88.5	87.6	89.1	89.9	87.4	88.9
21	30	35	35	37	37	37
22	34	37	37	40	40	40
23	64	72	72	77	77	77
24	18.2	18.2	18.7	18.1	18.4	19
25	42.585	50.451	50.063	52.332	51.192	50.782
26	88.5	87.5	89.1	89.8	87.3	88.8
27	12.1	12.1	11.6	11.9	11.7	11.6
28	12	12.1	12.2	12.2	12.6	12.7
29	22.343	22.185	21.778	22.079	21.663	22.029
30	87.6	86.1	88.3	89	86.7	88.3

Appendix A2: Alamogordo Year 2 Logsheet Data
EDM System Data

Item	6/20/13 10:06	6/20/13 14:29	6/21/13 8:25	6/21/13 11:13	6/21/13 14:51	6/24/13 13:30
1	25	25	25	25	25	25
2	33	33	33	33	33	33
3	59.6	58.7	56.2	55.2	54.9	59.2
4	18.9	19	19	19	19.1	20.2
5	5.109	4.398	4.721	4.755	4.311	4.49
6	85.7	87.6	85.7	86	88	89.4
7	4.004	3.265	3.607	3.668	3.186	3.357
8	87.2	89.4	87.4	87.7	89.8	91
9	34	33	35	32	34	34
10	39	37	41	37	40	40
11	73	70	76	69	74	74
12	19.4	19.3	19.3	19.7	19.5	20.3
13	49.858	35.769	42.63	38.186	31.172	32.505
14	87.8	90	87.9	88.2	90.4	91.5
15	31	31	32	32	31	30
16	37	37	39	39	37	36
17	70.4	75.3	69.8	67.9	69.6	68.1
18	19.2	19.4	19.2	19.3	19.4	19.9
19	135.681	118.477	112.16	114.691	106.671	124.011
20	87.6	89.8	87.7	88.1	90.2	91.3
21	36	35	36	36	35	35
22	40	40	40	40	38	40
23	76	75	76	76	73	75
24	19	18.7	19.5	19	19.2	20.1
25	49.695	54.81	51.352	56.412	50.983	51.535
26	87.5	89.7	87.7	88	90.1	91.2
27	11	11.7	11	11.7	11.6	11.9
28	13.2	13.1	13.4	13.6	13.7	13.7
29	21.769	22.24	21.85	21.641	22.497	22.498
30	86.3	88.8	86.6	87	89.3	89.8

Appendix A2: Alamogordo Year 2 Logsheet Data
EDM System Data

Item	6/24/13 16:37	6/25/13 8:15	6/26/13 14:22	6/26/13 16:15	6/27/13 8:25	6/27/13 16:47
1	25	24	24	24	24	25
2	33	34	35	35	35	36
3	56.1	59.8	58.6	53.3	53.1	60.9
4	20.1	20.9	19.1	19.3	19.4	20.9
5	4.513	5.072	4.481	4.529	4.122	4.717
6	88.1	85.3	84.5	85.2	81.6	87.5
7	3.373	3.966	3.328	3.423	3.152	3.663
8	89.9	86.9	86.2	86.9	83	89.2
9	34	33	31	31	31	32
10	39	40	40	39	40	42
11	73	73	71	70	71	74
12	20.6	21.3	19.5	19.5	19.4	21
13	34.074	44.841	5.657	5.59	7.869	4.114
14	90.5	87.4	86.8	87.5	83.5	89.8
15	30	30	29	29	28	29
16	36	38	38	36	36	40
17	71.47	72.8	65.9	67.7	67.1	65.7
18	20.5	21.3	19.4	19.5	19.7	21.9
19	121.419	108.339	10.3	10.493	10.037	87.526
20	90.2	87.2	86.3	87.3	83.3	89.6
21	35	36	34	34	33	34
22	40	42	42	40	40	42
23	75	78	76	74	73	76
24	20.1	21.4	19.5	19.2	19.1	21.2
25	53.22	50.84	51.347	51.109	50.888	53.988
26	90.1	87.1	86.5	87.2	83.2	89.5
27	11.1	10.2	11.3	10.4	11.4	11.7
28	13.9	14.2	13.3	13.3	13.3	13.8
29	20.757	22.041	22.216	21.925	23.074	22.153
30	89.3	85.1	86.1	87	81.3	89.2

Appendix A2: Alamogordo Year 2 Logsheet Data
EDM System Data

Item	6/28/13 7:45	6/28/13 15:30	7/8/13 11:45	7/8/13 15:50	7/9/13 8:30	7/9/13 15:25
1	25	25	25	25	25	25
2	37	37	41	44	44	43
3	59.6	60.4	57.5	59.1	56.7	60
4	21.3	21.7	20.8	23.1	24.4	23.5
5	5.064	4.529	4.372	4.478	5.019	4.612
6	85.7	88	88.6	88.5	86.2	87.9
7	4.035	3.49	3.184	3.422	3.962	3.522
8	87.3	89.8	90.2	90.2	87.9	89.7
9	34	33	29	31	34	33
10	43	42	40	43	46	46
11	77	75	69	74	80	79
12	22	22.4	20.9	23.5	24.4	23.5
13	6.586	4.524	107.072	113.791	112.982	116.086
14	87.9	90.4	89.1	90.8	87.6	90.6
15	29	29	27	29	29	29
16	40	40	36	41	42	41
17	64.9	69.2	61.8	60.5	62.5	58.2
18	22.1	22.4	21.1	23.9	24.4	23.8
19	83.575	79.38	135.86	155.386	151.505	154.289
20	87.6	90.2	89.4	90.5	87.3	90.3
21	35	34	34	35	35	34
22	45	43	43	45	45	45
23	80	77	77	80	80	79
24	21.5	22.2	21	23.2	24.2	23.4
25	51.804	51.962	49.62	50.366	56.187	50.274
26	87.6	90.1	90.5	90.6	88.2	90.1
27	11.5	10.4	11.9	11.3	11.2	9.8
28	14.2	14.3	13.8	14.4	15.2	15.2
29	22.63	22.48	22.03	22.073	22.366	22.5
30	85.8	89.3	89.2	89.5	86.4	89.5

Appendix A2: Alamogordo Year 2 Logsheet Data
EDM System Data

Item	7/10/13 8:14	7/10/13 15:50	7/11/13 8:26	7/11/13 17:11	7/12/13 9:35	7/12/13 17:00
1	25	25	24	23	23	24
2	44	43	42	37	40	41
3	61.1	62.6	61.9	55.7	54.4	58.9
4	25.8	25.3	26.1	23.2	25.4	25.8
5	4.739	4.512	4.698	4.283	4.834	4.979
6	86.5	88.4	87.3	84.7	86.7	88.1
7	3.7	3.42	3.601	3.104	3.779	3.89
8	88.2	90.4	89.1	86.1	88.5	90.1
9	33	32	32	26	31	33
10	46	45	45	34	44	45
11	79	77	77	60	75	78
12	25.9	25.4	26.9	24	26	26.2
13	111.462	109.754	14.51	59.952	110.689	110.215
14	87.9	91.3	88.7	86.5	88.3	90.7
15	27	26	25	24	24	26
16	40	37	36	34	36	38
17	61.4	54.3	54.8	53.3	56.9	52.4
18	26.2	25.1	25.9	23.5	25.6	26.1
19	160.345	155.008	157.954	72.414	163.23	158.909
20	87.4	91	88.1	86.2	87.9	90.5
21	36	34	34	31	30	33
22	47	45	46	40	42	43
23	83	79	80	71	72	76
24	26.2	25.4	25.9	23.8	25.5	26.1
25	50.234	49.27	49.715	49.742	53.212	50.133
26	88.5	90.7	89.4	86.5	88.8	90.5
27	8.6	9.5	9	10.6	10.4	0
28	16.2	16.1	16.6	15.8	16.6	16.3
29	23.02	22.095	23.013	22.031	22.261	23.54
30	86.8	89.9	87.6	84.9	87.5	89.5

Appendix A2: Alamogordo Year 2 Logsheet Data
EDM System Data

Item	7/13/13 8:40	7/13/13 16:25	7/14/13 7:53
1	23	23	22
2	39	39	38
3	55	60.5	60.6
4	25.7	25.2	27.2
5	5.07	4.73	6.662
6	86.2	89.8	85.7
7	4.003	3.846	5.929
8	88	87.9	87.4
9	31	31	29
10	43	42	42
11	74	73	71
12	26.4	25.1	27.5
13	111.88	109.63	110.773
14	87.4	90.7	86.8
15	22	22	19
16	31	32	29
17	49	54	48
18	26.8	25.8	27.8
19	160.561	158.008	152.118
20	87.1	90.4	86.3
21	30	29	28
22	40	40	38
23	70	69	66
24	26.4	26	27.9
25	48.552	52.455	52.447
26	88.3	90.2	87.7
27	0	0	0
28	17	17.4	18.5
29	24.68	27.03	28.965
30	86.4	89.5	85.3

Appendix A2: Alamogordo Year 2 Logsheet Data
(Mix Na Mix Cl Specific Gravity)

date	Mixed Cl		Mixed Na	
	sp gr	cond	sp gr	cond
3/28/2013 14:30		1.09	1.17	
3/28/2013 17:20		1.09	1.18	
3/29/2013 10:00		1.094	1.139	
3/29/2013 15:38		1.083	1.116	
3/29/2013 15:58		1.09	1.168	
4/1/2013 11:03		1.04	1.083	
4/1/2013 11:30		1.06	1.113	90
4/1/2013 12:00		1.08	120 1.13	100
4/1/2013 12:12		1.084	126 1.138	103
4/1/2013 12:22			1.14	105.5
4/1/2013 12:32		1.09	130 1.13	100
4/1/2013 12:52		1.092	135 1.14	105.5
4/1/2013 13:02			1.13	102
4/1/2013 13:12		1.098	141.5 1.14	106.5
4/1/2013 13:16			1.13	
4/1/2013 13:32		1.102	145 1.14	107
4/1/2013 13:52		1.1102	147 1.14	107.5
4/1/2013 14:12		1.11	1.14	107
4/1/2013 14:32		1.115	151 1.14	107
4/1/2013 14:52		1.115	152.6	
4/1/2013 15:12		1.1152		
4/1/2013 15:17		1.1152	1.132	105.5
4/1/2013 15:34		1.154	1.134	
4/1/2013 15:57			1.126	
4/5/2013 10:00		1.062	111 1.092	88
4/5/2013 10:50		1.088	141 1.13	111.5
4/5/2013 11:17		1.092	149.5 1.134	112.6
4/5/2013 11:37		1.098	154.2 1.14	114.8
4/5/2013 11:46			1.096	91.5
4/8/2013 10:45		1.054	102.2 1.084	86.03
4/8/2013 11:15		1.078	129 1.1152	102

Appendix A2: Alamogordo Year 2 Logsheet Data
(Mix Na Mix Cl Specific Gravity)

date	Mixed Cl		Mixed Na	
	sp gr	cond	sp gr	cond
4/8/2013 11:40			1.125	110
4/8/2013 11:50	1.09	143.5	1.132	112.5
4/8/2013 12:35	1.098		1.13	
4/8/2013 12:52	1.11	157	1.13	110
4/8/2013 13:27	1.11	160.8	1.14	115.8
4/8/2013 13:58	1.11	162.2	1.132	111.5
4/8/2013 14:24			1.13	112.5
4/8/2013 14:40	1.11	164.2	1.14	115.3
4/8/2013 14:45			1.13	111.3
4/8/2013 15:06	1.11	165	1.13	111.1
4/8/2013 15:52			1.128	110.4
4/8/2013 16:15	1.112	166.7	1.128	111.6
4/9/2013 9:05	1.06	104.4	1.078	80.76
4/9/2013 9:53			1.12	105.1
4/9/2013 10:15			1.134	111
4/9/2013 10:21	1.09	144.9		
4/9/2013 11:04			1.128	108.2
4/9/2013 11:15	1.104	157		
4/9/2013 11:52	1.108	161.6		
4/9/2013 12:10	1.11	160.8	1.128	108.9
4/9/2013 13:35	1.112	165.5	1.14	114.2
4/9/2013 14:23			1.128	109.7
4/9/2013 14:40	1.114	166.7		
4/9/2013 15:40	1.114	167.3	1.138	113.1
4/9/2013 16:45	1.114	167.3	1.136	111.6
4/10/2013 9:22	1.066	112	1.092	87.2
4/10/2013 9:53			1.12	104.1
4/10/2013 10:25			1.142	114.9
4/10/2013 10:38	1.098	151.5	1.126	109.1
4/10/2013 11:07			1.132	110.9
4/10/2013 12:28	1.112	163.9	1.14	114.8

Appendix A2: Alamogordo Year 2 Logsheet Data
(Mix Na Mix Cl Specific Gravity)

date	Mixed Cl		Mixed Na	
	sp gr	cond	sp gr	cond
4/10/2013 13:28	1.112	165.9	1.13	109
4/10/2013 15:25	1.116	166.9	1.138	113.5
4/10/2013 16:30	1.116	168	1.13	110.5
4/11/2013 10:20	1.052	92.2	1.068	70.37
4/11/2013 10:57			1.114	94.71
4/11/2013 11:01	1.076	128.2		
4/11/2013 11:42	1.092	146.8	1.138	113.3
4/11/2013 11:46			1.128	108.7
4/11/2013 13:21	1.11	162.2	1.13	
4/11/2013 14:41	1.114	165.7	1.138	115.3
4/11/2013 16:18	1.114	167.6	1.13	111.3
4/17/2013 9:10	1.048	88.66	1.066	71.1
4/17/2013 9:35			1.088	87.9
4/17/2013 10:11	1.08	133.2	1.116	103
4/17/2013 10:28	1.086	140.2	1.13	110
4/17/2013 10:53	1.092	147.2	1.128	110.7
4/17/2013 11:41	1.102	156.6	1.13	111
4/17/2013 12:45	1.116	162.9		
4/17/2013 14:10	1.11	164.8	1.128	111.3
4/17/2013 14:50	1.11	165.3	1.132	113.8
4/17/2013 15:10	1.112	166.4		
4/17/2013 16:30	1.098	152.6	1.128	111.5
4/18/2013 9:10			1.066	71.1
4/18/2013 9:50	1.072	123.1	1.106	96.4
4/18/2013 10:21	1.084	138.5	1.128	110.1
4/18/2013 11:16	1.088	153.1	1.13	111.1
4/18/2013 12:36	1.106	161	1.124	108
4/18/2013 13:20	1.11	162.8	1.122	107
4/18/2013 16:20	1.11	166.1	1.126	110
4/19/2013 9:10		100.5	1.08	81.1
4/19/2013 9:55			1.12	105.5

Appendix A2: Alamogordo Year 2 Logsheet Data
(Mix Na Mix Cl Specific Gravity)

date	Mixed Cl		Mixed Na	
	sp gr	cond	sp gr	cond
4/19/2013 10:08	1.084	140.4	1.13	110
4/19/2013 11:10	1.098	156.5	1.134	114.2
4/19/2013 13:50	1.11	163.7	1.134	113.1
4/19/2013 17:20	1.112	165	1.132	112
4/19/2013 17:50	1.114	165.3	1.13	109.7
4/19/2013 21:40	1.108	160	1.132	110.8
4/20/2013 8:30	1.11	160	1.128	107.8
4/22/2013 10:10	1.06	110	1.088	86.1
4/22/2013 10:40			1.114	102
4/22/2013 11:02	1.086	139.2	1.128	108.2
4/22/2013 13:55	1.108	161.5	1.132	111.5
4/22/2013 16:36	1.11	163.6	1.13	111
4/22/2013 21:11	1.11	163	1.132	110
4/23/2013 7:46	1.106	158.5	1.13	110
4/23/2013 9:36	1.112	161.4	1.13	109.1
4/23/2013 11:25	1.112	163.1	1.132	110.2
4/23/2013 15:45	1.112	163	1.13	109.8
4/23/2013 22:13			1.12	108
4/23/2013 22:35			1.12	100
4/24/2013 8:38	1.114	162.5	1.12	103.1
4/25/2013 12:21			1.092	86
4/25/2013 13:15			1.124	103
4/25/2013 15:01			1.124	103
4/25/2013 16:32	1.108	159.3	1.128	107.5
4/25/2013 17:00			1.134	110
4/25/2013 21:20	1.112	163	1.126	105.6
4/26/2013 8:55	1.104	155.2	1.124	104.3
4/30/2013 14:14			1.108	94.6
4/30/2013 14:39			1.124	103.5
4/30/2013 14:56			1.13	107.5
4/30/2013 15:17	1.092	142.7	1.14	110.9

Appendix A2: Alamogordo Year 2 Logsheet Data
(Mix Na Mix Cl Specific Gravity)

date	Mixed Cl		Mixed Na	
	sp gr	cond	sp gr	cond
4/30/2013 16:30	1.11	152.9	1.13	109.1
4/30/2013 21:34	1.112	162	1.128	106.4
5/1/2013 9:36	1.108	160.7	1.13	109.5
5/1/2013 13:53	1.106	157.8	1.128	109.5
5/1/2013 17:00	1.104	158.1	1.126	109.2
5/1/2013 21:47	1.101	157.2	1.12	105.2
5/1/2013 21:57			1.124	107.1
5/2/2013 14:53	1.1	153.7	1.126	106.2
5/2/2013 21:34	1.102	153.4	1.116	100.7
5/29/2013 14:18	1.1	154.6	1.132	110.4
5/31/2013 9:48	1.05	92.04	1.068	70.28
5/31/2013 11:00			1.126	102.1
5/31/2013 12:30			1.128	
5/31/2013 13:36	1.1102	154	1.13	107.2
5/31/2013 15:38	1.106	160	1.122	105.6
6/3/2013 11:50	1.088	141.9	1.13	109.5
6/3/2013 14:57	1.102	159.2	1.13	107.6
6/4/2013 11:25	1.086	141	1.126	107.8
6/4/2013 14:55	1.104	162.4	1.12	105.1
6/4/2013 17:00	1.108	164.6	1.128	
6/14/2013 15:45	1.09	138.4	1.132	111.4
6/17/2013 11:20			1.132	111
6/17/2013 11:37			1.14	114
6/17/2013 11:40	1.092	143	1.13	111
6/17/2013 11:55			1.13	
6/17/2013 12:55	1.1	147	1.132	110
6/17/2013 14:15	1.104	155.7	1.136	112.8
6/18/2013 11:02	1.09	139.2	1.13	108.1
6/18/2013 14:30	1.106	153.2	1.138	113.7
6/19/2013 10:50	1.11	159.7	1.138	114.3
6/19/2013 14:40	1.106	158.9	1.13	112.7

Appendix A2: Alamogordo Year 2 Logsheet Data
(Mix Na Mix Cl Specific Gravity)

date	Mixed Cl		Mixed Na	
	sp gr	cond	sp gr	cond
6/20/2013 10:33	1.104	153.3	1.12	105.3
6/20/2013 15:05	1.108	157.3	1.12	104.8
6/21/2013 10:56	1.108	151.5	1.13	107.9
6/21/2013 15:05	1.11	158.4	1.14	115.5
6/24/2013 13:45	1.088	150.8	1.12	107.8
6/24/2013 16:35	1.102	157.3	1.128	111.8
6/25/2013 8:30	1.119	159.1	1.128	109.7
6/26/2013 14:38	1.1	153.6	1.126	109.2
6/26/2013 16:30	1.104	158.2	1.13	112.2
6/27/2013 8:30	1.106	159.9	1.122	107.5
6/27/2013 12:08	1.108	159	1.132	111
6/27/2013 17:00	1.11	161.8	1.128	111.1
6/27/2013 19:45	1.108	160	1.132	112.2
6/28/2013 8:06	1.108	161.4	1.126	110.1
6/28/2013 15:45	1.108	159.3	1.128	112.4
7/8/2013 12:00	1.09	142.6	1.122	108.3
7/8/2013 16:05	1.1	158.6	1.126	110.8
7/9/2013 8:43	1.104	158.9	1.122	110.3
7/9/2013 15:40	1.1	158.6	1.126	112.5
7/10/2013 8:33	1.102	157.7	1.126	110
7/10/2013 16:05	1.102	157.7	1.128	111.6
7/11/2013 8:49	1.104	156.4	1.126	109.4
7/12/2013 9:30	1.109	164.6	1.124	109.7
7/12/2013 17:15	1.106	163	1.12	109.4
7/13/2013 8:48	1.108	157	1.128	105.5
7/13/2013 16:48	1.108	158.8	1.124	108.3

Item	Location	Tag	units	3/28/2013 16:40	3/29/2013 10:08	4/1/2013 9:25	4/1/2013 11:15
1	20-µm Cartridge filters	(gauges)	psi	0/0	0/0	0/0	0/0
2	5-µm Cartridge filters-inlet	PIT-2007	psi	48	48.1	48.1	47.5
3	5-µm Cartridge filters-outlet	PIT-2007B	psi	46	46.1	46.2	46.2
4	High pressure pump feed Pressure	(gauge)	psi	46	46	46	46
5	NF Feed Pressure	PIT-2008	psi	107	107	109	106
6	NF Reject Pressure	PIT-2008R	psi	68	67	68	66
7	NF Permeate Pressure	PIT-2009	psi	0.7	0.6	0.7	0.6
8	Feed Flow	FIT-2001	gpm	51.6	49.7	53.8	47.6
9	Feed pH	AIT-2003		7.15	7.08	6.54	6.59
10	Tank Level	LIT-2003	in	20.68	20.653	20.736	20.542
11	NF Feed pH	AIT-2008A		7.17	7.05	6.63	6.65
12	NF Feed Cond	AIT-2008	µS/cm	3,666	3,571	3,494	3,368
13	NF Feed Temp	TIT-2008	°F	68.21	68.39	68.58	70.38
14	NF Permeate Flow	FIT-2009P	gpm	43.2	40.86	42.33	41.6
15	NF Permeate Conductivity	AIT-2009P	µS/cm	1,412	1,325	1,322	1,297
16	NF Reject Flow	FIT-2009R	gpm	28.66	29.16	28.63	28.69
17	NF Reject Conductivity	AIT-2009R1	µS/cm	8,609	7,803	8,081	7,804
18	NF Reject pH	AIT-2009R2		7.36	7.29	6.91	6.93
19	NF Reject Temp	TIT-2009R2	°F	67.78	67.97	67.95	69.83
20	Product Flow	FIT-2011	gpm	60.5	63.8	64.1	62.7

Item	4/1/2015 13:22	4/3/2013 16:35	4/5/2013 10:00	4/5/2013 11:45	4/8/2013 10:47	4/8/2013 14:31	4/8/2013 17:05	4/9/2013 10:28
1	0/0	0/0	0/0	0/0	0/0	0/0	0/0	0/0
2	47.8	48.1	48.5	48.4	48.3	48	48.3	48.3
3	46	46.5	46.7	46.6	46.4	46.3	46.8	46.8
4	45	46	46	46	45	45	45	45
5	106	103	101	102	103	100	98	98
6	66	65	63	64	64	63	62	62
7	0.6	0.6	0.5	0.5	0.5	0.5	0.5	0.5
8	46.7	47.6	42.5	43.1	49.5	42	43.5	43.5
9	6.64	6.62	6.51	6.54	6.53	6.54	6.57	6.57
10	20.653	20.597	20.542	20.542	20.736	20.68	20.597	20.597
11	6.66	6.66	6.6	6.6	6.5	6.5	6.53	6.53
12	3,366	3,316	3,369	3,383	3,341	3,466	3489.3	3,489
13	71.03	69.93	69.99	70.6	71.7	72.78	73.35	73.35
14	41.96	40.1	38.67	40.58	39.59	39.59	39.85	39.85
15	1,323	1,256	1,316	70	72	1,385	1360.9	1,361
16	28.53	27.52	28.22	27.59	27.75	27.92	27.99	27.99
17	7,819	7,645	7,326	7,643	7,543	7,801	7899.6	7,900
18	6.91	6.86	6.85	6.82	6.75	6.78	6.81	6.81
19	70.65	69.19	69.38	70.13	71.24	72.38	72.9	72.9
20	61.5	62	61.8	65	59.4	65.5	61.9	61.9

Item	4/9/2013 16:40	4/10/2013 11:22	4/10/2013 16:35	4/11/2013 11:59	4/11/2013 13:43	4/11/2013 16:11	4/11/2013 17:04
1	0/0	0/0	0/0	0/0	0/0	0/0	0/0
2	48.5	48.3	48.3	48.3	48.5	48.3	48.4
3	46.8	46.6	46.5	46.8	46.7	46.8	46.8
4	47	47	46	47	46	46	46
5	100	103	104	105	106	104	106
6	64	67	67	67	68	68	69
7	0.5	0.5	0.5	0.5	0.5	0.5	0.5
8	54.5	51.6	42	57	47.6	44.2	55
9	6.49	6.37	6.46	6.47	6.73	6.65	6.57
10	20.68	20.68	20.761	20.763	20.678	20.602	20.681
11	6.51	6.42	6.55	6.49	6.64	6.48	6.45
12	3,445	3,571	3,674	3,518	3,481	3,566	3,602
13	71	70.08	70.45	69.24	69.29	70.15	70.02
14	39.06	40.62	39.78	39.51	39.87	39.42	39.18
15	1,339	1,400	1,411	1,371	1,353	1,377	1,392
16	27.89	28.13	29.06	28.99	28.66	28.42	28.79
17	7,823	7,946	8,210	7,863	7,806	8,070	8,055
18	6.69	6.84	6.82	6.82	6.85	6.68	6.76
19	70.19	69.81	69.61	68.75	68.9	69.59	69.7
20	61.4	61.1	62.7	60.7	64.7	61.4	

Item	4/17/2013 10:40	4/17/2013 14:25	4/17/2013 16:20	4/18/2013 10:10	4/18/2013 13:00	4/18/2013 16:16	4/19/2013 10:45	4/19/2013 14:01	4/19/2013 17:40
1	0/0	0/0	0/0	0/0	0/0	0/0	0/0	0/0	0/0
2	48.3	48.4	48	48.5	48.1	48.3	48.3	48.2	48.1
3	46.8	46.9	46.5	46.9	47	46.3	46.6	46.5	46.5
4	46	46	45	48	46	46	46	46	46
5	100	98	100	103	104	104	107	107	106
6	64	62	64	66	67	67	69	70	68
7	0.6	0.6	0.5	0.6	0.5	0.6	0.5	0.5	0.5
8	47.2	48.3	44.2	53.1	52.2	47.5	51.5	47.9	46.6
9	6.57	6.8	6.7	6.16	6.25	6.86	6.37	6.23	6.21
10	20.547	20.736	20.542	20.68	20.81	20.739	20.763	20.655	20.543
11	6.47	6.5	6.6	6.34	6.33	6.58	6.26	6.27	6.31
12	3338.6	3314.9	3382	3429.4	3386.5	3407.1	3525.1	3563.6	3558
13	71.59	73.2	73.19	69.36	70.35	70.63	68.68	69.63	69.96
14	39.48	39.87	40.87	39.79	40.3	40.19	40.1	39.3	39.98
15	1388.3	1363.5	1367	1361	1316	1336.6	1388.2	1391.8	1393.3
16	28.16	28.09	27.42	28.4	28.2	28.42	28.83	29.26	29.03
17	7467.8	7501.8	7600.8	7670	7682.1	7700.8	7867.7	8063.4	8017.9
18	6.7	6.76	6.91	6.83	6.64	6.86	6.64	6.55	6.52
19	71.13	72.67	72.57	68.86	69.76	70.03	68.23	69.11	69.56
20	55.4	53.9	50.6	54.6	51.8	53.7	51		54.5

Item	4/19/2013 21:45	4/20/2013 8:25	4/22/2013 11:24	4/22/2013 13:38	4/22/2013 16:14	4/22/2013 21:40	4/23/2013 8:14	4/23/2013 11:05
1	0/0	0/0	0/0	0/0	0/0	0/0	0/0	0/0
2	48.6	48.1	48.4	48.3	48.3	47.8	48.4	47.9
3	46.8	46.4	46.4	46.6	46.5	46.2	46.5	46.5
4	46	46	46	45	45	46	46	45
5	106	109	104	103	104	107	107	108
6	69	72	68	67	67	70	70	70
7	0.5	0.5	0.5	0.5	0.6	0.5	0.6	0.5
8	47.4	54.6	48.1	56.2	57.5	48.7	43.5	56.1
9	6.36	6.25	6.69	6.13	6.29	6.98	6.58	6.5
10	20.73	20.68	20.736	20.761	20.736	20.736	20.658	20.764
11	6.26	6.24	6.54	6.27	6.3	6.46	5.97	6.17
12	3605.1	3651.9	3333.2	3350.4	3418.7	3408	3601.3	3612.4
13	69.52	68.67	71.23	71.8	72.08	71.46	71.53	72.66
14	39.9	39.71	39.68	40.42	39.22	40.46	40.7	39.14
15	1398.4	1423.5	1367.7	1340.2	1354.6	1363.6	1389	1426.4
16	28.89	28.95	28.76	28.52	28.72	28.96	29.3	29.23
17	7984.3	8080.7	7531.5	7627.4	7671.1	7702.3	8024.3	8070.3
18	6.55	6.6	6.96	6.67	6.57	6.84	6.24	6.33
19	68.91	68.46	70.87	71.27	71.76	70.97	71.02	72.06
20	51.6	52.5	56.1	52.3	53.3	51.7	58.3	50.8

Item	4/23/2013 15:30	4/23/2013 22:27	4/24/2013 8:20	4/25/2013 16:57	4/25/2013 21:45	4/26/2013 8:32	4/30/2013 16:10	4/30/2013 21:47
1	0/0	0/0	0/0	0/0	0/0	0/0	0/0	0/0
2	48	48.1	47.9	51.5	51.5	51.6	51.2	51.4
3	46.2	46.1	46	50.8	50.7	51	50.7	50.6
4	45	45	46	50	50	50	49	50
5	106	109	112	64	64	62	65	67
6	70	72	73	45	45	44	46	47
7	0.5	0.5	0.5	0.4	0.4	0.1	0.4	0.4
8	45.4	41.8	57.1	20.9	21.3	19.8	27.4	15.6
9	6.54	7.58	6.47	6.72	6.87	5.44	6.58	6.52
10	20.597	20.648	20.373	20.902	20.927	20.819	20.93	20.93
11	6.15	7.14	6.21	6.18	6.19	6.1	6.07	6.12
12	3596.3	3614.4	3635.9	3417.4	3342	3360.4	3404.1	3550.7
13	73.45	72.58	71.23	72.96	72.4	70.84	75.73	75.02
14	40.31	42.2	40.39	21.87	21.45	20.41	22.19	21.94
15	1415.2	1472.6	1397.4	1072	1175.8	943.77	952.97	1116.2
16	29.2	30.07	30.04	15.03	14.53	14.4	14.56	14.83
17	8046.7	7949.02	8124.6	7754.5	7558	7451.7	7710.7	7962.9
18	6.29	7.24	6.31	6.18	6.12	6.33	6.17	6.04
19	72.91	71.78	70.5	72.37	71.63	69.92	75.46	74.3
20	48.7	52.7	51.1	51.9	52.1	51.4	54.2	54.3

Item	5/1/2013 9:00	5/1/2013 13:35	5/1/2013 16:45	5/1/2013 21:25	5/2/2013 14:45	5/2/2013 21:42	5/29/2013 13:54	5/31/2013 13:37	5/31/2013 15:22
1	0/0	0/0	0/0	0/0	0/0	0/0	0/0	0/0	0/0
2	51.2	51.5	51.4	51.3	51.5	51.5	50.7	51.0	51.1
3	50.6	50.6	50.4	50.7	50.9	51.7	50.3	50.7	50.7
4	50	49	49	50	50	50	49.0	49.0	49.0
5	70	69	69	70	68	70	60.0	59.0	61.0
6	46	42	40	37	12	8	42.0	41.0	43.0
7	0.4	0.4	0.4	0.4	0.4	0.4	0.5	0.4	0.4
8	44.6	43.4	40	32	37.6	20.1	37.9	39.1	38.3
9	6.49	6.83	6.61	6.3	6.88	7.64	5.5	6.9	6.2
10	20.902	20.73	20.847	20.926	20.846	20.819	20.905	21.063	21.013
11	6.09	6.5	6.28	6.55	6.48	7.22	6.0	5.7	5.8
12	3574.5	3473.4	3345.9	3231.4	3286.7	3322.3	2959.7	3292.8	3524.8
13	73.99	75.79	77.06	75.85	74.63	72.4	83.5	77.6	78.4
14	21.87	22.26	21.72	22.01	20.29	20.13	25.6	21.6	21.0
15	1118.8	1167.3	1140.2	1124	757.65	964.2	1067.8	1228.8	1293.3
16	14.5	13.76	12.72	12.35	12.52	12.96	15.9	16.0	15.9
17	8149	7824.4	7751.4	7522.4	6030.4	5817.2	6439.3	7081.0	7111.9
18	6.1	6.61	6.48	6.55	6.65	7.29	5.8	6.0	5.8
19	73.38	75.26	76.66	75	73.4	71.56	82.9	77.2	77.9
20	56	55.9	54		51.4	51.5	54.1	0.0	55.3

Item	6/3/2013 0:00	6/3/2013 14:50	6/4/2013 11:15	6/18/2013 14:21	6/18/2013 17:23	6/19/2013 10:22	6/19/2013 14:30	6/20/2013	6/20/2013 14:54
1	0/0	0/0	0/0	0/0	0/0	0/0	0/0	0/0	0/0
2	51.0	51.1	50.8	47.7	47.7	47.3	48.8	48.9	48.9
3	50.4	50.6	50.4	46.5	46.5	46.2	47.8	47.9	47.9
4	50.0	50.0	50.0	45.0	45.0	45.0	46.0	46.0	46.0
5	60.0	61.0	62.0	93.0	92.0	99.0	83.0	85.0	83.0
6	42.0	42.0	42.0	58.0	50.0	63.0	52.0	54.0	52.0
7	0.4	0.4	0.4	0.5	0.5	0.5	0.4	0.4	0.4
8	44.9	36.0	30.4		52.2	45.5	42.9	40.6	43.4
9	6.2	6.3	7.0	7.1	6.7	7.2	6.9	7.2	5.8
10	20.901	21.013	20.934	20.846	20.736	20.517	20.763	20.763	20.761
11	6.0	6.1	6.0	6.2	6.1	6.3	6.1	6.2	6.3
12	3405.5	3389.5	3179.7	3089.8	3091.6	3371.5	3143.7	3277.8	3105.4
13	79.3	79.7	79.4	81.8	82.2	80.5	81.9	81.1	82.0
14	21.4	21.3	22.9	40.0	40.6	39.8	34.5	35.0	35.7
15	1281.0	1294.8	1167.1	1109.0	1106.4	1168.2	1125.9	1170.9	1141.6
16	15.8	15.4	16.0	28.3	28.6	30.0	27.4	28.3	27.5
17	7619.0	7746.3	7594.1	7226.5	7150.2	7971.4	6920.1	7284.9	6826.9
18	5.9	6.0	6.0	6.3	6.5	6.4	6.4	6.4	6.3
19	78.2	79.1	78.5	81.4	81.9	80.0	81.5	80.7	81.7
20	59.0	61.5	60.9	64.7		66.6	0.0	0.0	65.9

Item	6/21/2013 9:28	6/21/2013 11:10	6/21/2013 14:54	6/24/2013 13:30	6/24/2013 16:30	6/25/2013 8:16	6/26/2013 14:30	6/26/2013 16:21	6/27/2013 8:27
1	0/0	0/0	0/0	0/0	0/0	0/0	0/0	0/0	0/0
2	48.7	48.6	48.7	48.7	48.7	48.7	48.9	48.8	48.7
3	47.9	47.8	48.0	47.6	47.9	47.7	48.0	48.2	47.9
4	47.0	47.0	47.0	46.0	48.0	48.0	47.0	47.0	48.0
5	86.0	85.0	82.0	81.0	83.0	86.0	85.0	84.0	89.0
6	55.0	54.0	52.0	51.0	52.0	55.0	54.0	53.0	57.0
7	0.4	0.4	0.5	0.4	0.4	0.4	0.5	0.4	0.3
8	45.0	43.4	41.0	45.3	45.4	43.0	44.3	40.5	41.2
9	3.8	5.7	7.1	7.1	5.7	6.5	6.6	7.0	7.1
10	20.763	20.730	20.822	20.819	20.818	20.761	20.763	20.653	20.756
11	6.3	6.2	6.1	6.1	6.1	6.2	6.1	6.1	6.2
12	3168.0	3166.7	3043.9	3130.2	3187.0	3272.0	3120.5	3150.6	3258.0
13	80.9	81.2	82.0	84.0	82.5	80.7	79.7	80.2	77.6
14	35.6	35.5	34.7	35.0	34.9	33.9	34.6	35.4	35.1
15	1124.5	1133.4	1125.0	1156.5	1137.6	1163.1	1085.2	1112.2	1077.7
16	27.8	28.0	27.9	27.4	27.7	28.4	28.6	28.1	29.0
17	7070.3	7074.9	6734.6	6885.0	6880.9	7246.7	6829.2	6802.1	7177.5
18	6.3	6.4	6.3	6.4	6.4	6.4	6.5	6.5	6.3
19	80.0	80.6	81.9	83.6	82.1	80.0	79.0	79.9	77.1
20	63.2	61.4	65.5	66.3	63.4	65.4	51.8	50.4	52.4

Item	6/27/2013 16:56	6/28/2013 7:55	6/28/2013 15:35	7/8/2013 11:50	7/8/2013 15:54	7/9/2013 8:32	7/9/2013 15:30	7/10/2013 8:19	7/10/2013 15:54
1	0/0	0/0	0/0	0/0	0/0	0/0	0/0		
2	48.7	48.9	48.8	48.7	48.9	48.6	48.7	48.7	48.3
3	48.0	47.7	48.0	47.9	48.0	47.4	47.7	47.8	47.4
4	47.0	47.0	47.0	47.0	47.0	47.0	47.0	47.0	47.0
5	85.0	87.0	85.0	81.0	83.0	86.0	84.0	86.0	86.0
6	54.0	56.0	54.0	51.0	53.0	55.0	54.0	55.0	56.0
7	0.5	0.5	0.4	0.5	0.4	0.3	0.4	0.4	0.4
8	44.8	41.7	45.2	40.7	43.1	40.0	41.0	42.9	45.2
9	5.1	5.6	7.1	6.1	6.2	6.2	5.9	6.3	6.9
10	20.768	20.603	20.763	20.7	20.8	20.8	20.9	20.8	20.9
11	6.3	6.5	6.5	6.4	6.5	6.4	6.4	6.4	6.4
12	3159.1	3274.9	3109.0	3131.3	3146.5	3327.5	3142.3	3146.3	3242.5
13	81.8	80.9	82.4	84.0	83.2	81.2	82.1	81.6	82.4
14	34.8	35.2	35.3	35.4	34.7	34.7	34.6	35.0	34.9
15	15.4	1133.9	1103.4	1127.6	1122.9	1167.9	1111.0	1083.9	1154.0
16	28.4	28.8	28.2	28.1	27.8	28.8	28.4	28.9	29.0
17	7006.7	7227.9	6849.2	6828.1	6908.0	7210.9	6868.1	6927.3	7074.2
18	6.2	6.6	6.5	6.5	6.5	6.6	6.6	6.6	6.6
19	81.4	80.3	82.0	83.4	82.9	80.5	81.7	80.9	81.7
20	54.0	50.5	51.9	0.0	70.5	63.3	58.3	60.3	61.3

Item	7/11/2013 8:34	7/11/2013 17:10	7/12/2013 9:35	7/12/2013 17:07	7/13/2013 8:42	7/13/2013 16:30	7/14/2013 8:00
1							
2	48.8	48.5	48.7	48.6	48.5	48.4	48.6
3	47.9	47.5	47.5	47.5	47.4	47.2	47.2
4	47.0	47.0	47.0	47.0	48.0	48.0	48.0
5	87.0	86.0	87.0	88.0	90.0	89.0	91.0
6	56.0	55.0	56.0	57.0	58.0	58.0	58.0
7	0.5	0.4	0.3	0.5	0.3	0.4	0.4
8	40.3	45.6	42.0	42.7	42.1	41.9	59.9
9	7.1	2.9	5.9	7.1	7.1	7.1	6.8
10	20.8	20.8	20.8	20.9	20.9	20.8	20.7
11	6.4	6.6	6.5	6.5	6.4	6.3	6.5
12	3199.3	3128.7	3244.1	3275.9	3319.1	3333.4	3183.6
13	81.9	80.6	81.3	81.9	81.8	82.0	78.6
14	34.5	35.4	34.4	34.8	35.5	34.6	35.1
15	1120.7	1108.7	1122.0	1163.0	1116.4	1112.5	1246.9
16	28.6	28.4	29.3	29.2	29.4	29.5	29.8
17	6926.0	6678.7	7032.4	7147.9	7234.0	7219.5	6823.7
18	6.5	7.4	6.5	6.6	6.6	6.6	6.7
19	81.4	80.0	80.6	81.3	80.2	81.5	78.5
20	58.4	54.1	61.4	61.9	59.8	48.6	57.2

Appendix A2: Alamogordo Year 2 Logsheet Data

NaCl Usage Data

	Tank A					
	Water Level	Salt Level	Bags salt added	Δ salt height (in)	Salt used (lb)	in use?
4/9/2013 8:20	11.75	27.75		0		yes
4/9/2013 9:42	10	29.25		0	1.5	48.73 yes
4/9/2013 10:56	11.75	30		0	0.75	24.36 yes
4/9/2013 15:00	15.5	34.5		0	4.5	146.19 yes
4/9/2013 16:00	15	35.5		0	1	32.49 yes
4/9/2013 17:27	18	36.875		0	1.375	44.67 yes
4/10/2013 8:30	15.5	36.625		0	-0.25	-8.12 yes
4/10/2013 10:45	17.125	39.25		0	2.625	85.28 yes
4/10/2013 11:05	9.75	12.625	20			yes
4/10/2013 13:10	13.625	19	0	6.375	207.10	yes
4/10/2013 13:20	10.75	19	0	0	0.00	yes
4/10/2013 15:20	14.25	21	0	2	64.97	yes
4/10/2013 15:50	12.25	21.5	0	0.5	16.24	yes
4/10/2013 16:48	13.75	22.375	0	0.875	28.43	yes
4/11/2013 9:50	13.75	23	0	0.625	20.30	yes
4/11/2013 12:00	13.75	24.5	0	1.5	48.73	yes
4/11/2013 15:15	16	27	0	2.5	81.22	yes
4/11/2013 17:10	15.75	27.625	0	0.625	20.30	yes
4/17/2013 8:52	unknown	31.625			4	129.95 no
4/17/2013 10:35						no
4/17/2013 16:15				0	0.00	no
4/18/2013 8:43	27	35				yes
4/18/2013 9:00	15	17.25		-17.75		yes
4/18/2013 11:08	9.625	17.75		0.5	16.24	yes
4/18/2013 16:53	17.5	26.5		8.75	284.25	yes
4/19/2013 8:30	17.5	26				yes
4/19/2013 11:10	9	29.25		3.25	105.58	yes
4/19/2013 16:15	14	34.5		5.25	170.55	yes
4/19/2013 16:30	8.5	15.625		-18.875	-613.18	yes
4/19/2013 21:22	11.5	23.25		7.625	247.71	no
4/20/2013 9:15	8.5	23.25		0	0.00	no
4/22/2013 9:30		25			0.00	yes
4/22/2013 10:24	15.25	27		2	64.97	yes
4/22/2013 14:25	9.75	32		5	162.43	yes
4/22/2013 16:00	8.75	16	14		0.00	yes
4/22/2013 21:51	10	20.5		4.5	146.19	yes
4/23/2013 7:54	8.75	32.875	0	12.375	402.02	yes
4/23/2013 10:35	10.25	36.5	0	3.625	117.76	yes
4/23/2013 13:15	14.75	39	0	2.5	81.22	yes
4/23/2013 13:35	13.5	39	0	0	0.00	yes
4/23/2013 15:50	16.5	41.5	0	2.5	81.22	yes

Appendix A2: Alamogordo Year 2 Logsheet Data

NaCl Usage Data

	Tank A					in use?
	Water Level	Salt Level	Bags salt added	Δ salt height (in)	Salt used (lb)	
4/23/2013 16:35	14.5	41.75	0	0.25	8.12	
4/23/2013 4:45	11.5	41.75	0	0	0.00	no
4/23/2013 22:25	14	44.375	0	2.625	85.28	no
4/24/2013 7:15	17.625	50.5	0	6.125	198.98	no
4/24/2013 8:52	20	52.5	0	2	64.97	no
4/24/2013 9:35	9.75	14	30		0.00	no
4/25/2013 10:50	9.5	18.375	0	4.375	142.13	no
4/25/2013 16:30	14.375	21.625	0	3.25	105.58	no
4/25/2013 21:30	16	24	0	2.375	77.15	no
4/26/2013 8:15	17.5	27.25	0	3.25	105.58	no
4/30/2013 14:10	20.75	29.125	0	1.875	60.91	no
4/30/2013 16:40	23	30.5	13	1.375	44.67	no
4/30/2013 16:40	11.75	13	0		0.00	no
4/30/2013 21:25	14	15.625	0	2.625	85.28	no
5/1/2013 6:45	16.625	18.75	0	3.125	101.52	no
5/1/2013 14:02	17.75	21.75	1	3	97.46	no
5/1/2013 17:00	18.25	22.25	2	0.5	16.24	no
5/1/2013 21:52	15.25	24	3	5.25	170.55	no
5/2/2013 9:54	11	27	1	8.25	268.01	
5/2/2013 15:30	11.75	29	2	2	64.97	
5/2/2013 20:00	8.375	30.375	3	1.375	44.67	
5/13/2013 8:00	8.75	32	4	1.625	52.79	
5/13/2013 11:50		32		0	0.00	no
5/29/2013 9:35		33		1	32.49	no
5/29/2013 14:20		34.75		1.75	56.85	no
5/31/2013 8:55		35.25		0.5	16.24	no
5/31/2013 15:42		38.25		3	97.46	no
6/3/2013 9:15		40	20	1.75	56.85	no
6/3/2013 10:00		41.5		1.5	48.73	no
6/3/2013 15:06		43.25		3.25	105.58	no
6/3/2013 15:40		18			0.00	no
6/4/2013 8:45		18.75		0.75	24.36	no
6/4/2013 11:30		19.5		1.5	48.73	no
6/4/2013 15:15		21.25		2.5	81.22	no
6/4/2013 17:10		22.25		1	32.49	no
6/14/2013 16:10		39		16.75	544.15	no
6/17/2013 9:30		27.75		-11.25	-365.47	no
6/17/2013 14:16		31.75			0.00	no
6/17/2013 16:45		33.75		2	64.97	no
6/18/2013 8:50		34.5		0.75	24.36	no
6/18/2013 11:20		37.25		2.75	89.34	no

Appendix A2: Alamogordo Year 2 Logsheet Data

NaCl Usage Data

Tank A						
Water Level	Salt Level	Bags salt added	Δ salt height (in)	Salt used (lb)	in use?	
6/18/2013 14:41	39.75		2.5	81.22	no	
6/18/2013 15:10	39.75		2.5	81.22	no	
6/18/2013 15:10	14.625			0.00	no	
6/19/2013 11:00	31.25		16.625	540.08	no	
6/19/2013 14:46	34.5		3.25	105.58	no	
6/19/2013 16:00	35.25		0.75	24.36	no	
6/19/2013 16:00	16			0.00	no	
6/20/2013 10:40	28.5		12.5	406.08	no	
6/20/2013 15:10	32		3.5	113.70	no	
6/20/2013 15:31	14			0.00		
6/21/2013 8:13	25.75		11.75	381.71	no	
6/21/2013 11:20	28		2.25	73.09		
6/21/2013 15:10	31.5		3.5	113.70	no	
6/24/2013 10:05	32.75		1.25	40.61	no	
6/24/2013 12:00	36		3.25	105.58	no	
6/24/2013 14:38	36.625		3.875	125.88	no	
6/24/2013 14:38	14.625			0.00		
6/25/2013 8:34	27.75		13.125	426.38	no	
6/26/2013 10:15	29.25		1.5	48.73	no	
6/26/2013 14:43	33.5		4.25	138.07	no	
6/26/2013 15:11	33.75		0.25	8.12	no	
6/26/2013 15:30	13.5			0.00		
6/26/2013 16:35	14.25		0.75	24.36	yes	
6/27/2013 8:46	25.785		11.535	374.73	yes	
6/27/2013 17:12	32		6.215	201.90	yes	
6/28/2013 8:10	43		11	357.35	yes	
6/28/2013 15:53	48		5	162.43	yes	
7/8/2013 9:05	48.5		0.5	16.24	yes	
7/8/2013 12:03	51.75		3.25	105.58	yes	
7/8/2013 16:10	55		6.5	211.16	yes	
7/8/2013 16:30	34			0.00		
7/9/2013 8:55	46		12	389.84		
7/9/2013 15:45	50		4	129.95		
7/9/2013 16:30	45			0.00		
7/9/2013 19:05	45.5		0.5	16.24	yes	
7/9/2013 20:15	46.375		1.375	44.67	yes	
7/10/2013 8:40	55		8.625	280.19	yes	
7/10/2013 13:54	55		8.625	280.19	yes	
7/10/2013 14:15	13.625			0.00	yes	
7/10/2013 16:09	16.5		2.875	93.40	yes	
7/11/2013 8:54	33.375		16.875	548.21	yes	

Appendix A2: Alamogordo Year 2 Logsheet Data

NaCl Usage Data

Tank A						
Water Level	Salt Level	Bags salt added	Δ salt height (in)	Salt used (lb)	in use?	
7/11/2013 16:49		37		3.625	117.76	yes
7/11/2013 17:00		25			0.00	yes
7/11/2013 20:05		28.5		3.5	113.70	yes
7/12/2013 8:45		42		13.5	438.56	yes
7/12/2013 16:30		48		6	194.92	yes
7/12/2013 16:52		13			0.00	yes
7/13/2013 8:33		31.5		18.5	601.00	yes
7/13/2013 9:12		12.125			0.00	yes
7/13/2013 17:20		21		8.875	288.32	yes
7/14/2013 8:00		35.5		14.5	471.05	yes

Appendix A2: Alamogordo Year 2 Logsheet Data

NaCl Usage Data

Tank B					
Water Level	Salt Level	Bags salt added	Δ salt height (in)	Salt used (lb)	in use?
12.5	24		0		yes
11	24.5		0	0.5	16.24 yes
12.5	24.5		0	0	0.00 yes
16.75	26.5		0	2	64.97 yes
16.75	27		0	0.5	16.24 yes
16.75	27.25		0	0.25	8.12 yes
17.5	27.25		0	0	0.00 yes
19.5	27.75		0	0.5	16.24 yes
10.25	21.5		5		yes
13.75	21.625		0	0.125	4.06 yes
10.875	21.625		0	0	0.00 yes
14.5	22.25		0	0.625	20.30 yes
12.25	22.5		0	0.25	8.12 yes
14	23.125		0	0.625	20.30 yes
14	23		0	-0.125	-4.06 yes
13.5	24.5		0	1.5	48.73 yes
15.75	26.75		0	2.25	73.09 yes
15.625	28.5		0	1.75	56.85 yes
19.25	29		0	0.5	16.24 yes
28	31.5				0.00 yes
unknown	38.5		0	7	227.40 yes
29.5	40				
15	17.25			-22.75	
32.5	19			1.75	56.85
					0.00
14	21.25				0.00
14	21.25			0	0.00 no
14	21.25			0	0.00 no
11	17			-4.25	-138.07 no
11	17			0	0.00 yes
below salt	25.75			8.75	284.25 yes
	27				0.00 no
15.5	28			1	32.49 no
10	30			2	64.97 no
8.75	16	12			0.00 no
10	19			3	97.46 no
11.5	20			1	32.49 no
13.75	20			0	0.00 no
18.5	21			1	32.49 no
15.5	21			0	0.00 no
19.5	22.75			1.75	56.85 no

Appendix A2: Alamogordo Year 2 Logsheet Data
NaCl Usage Data

Tank B					
Water Level	Salt Level	Bags salt added	Δ salt height (in)	Salt used (lb)	in use?
16	16		6	-6.75	-219.28
14.25	16			0	0.00 yes
18.375	20.875		0	4.875	158.37 yes
22	25.75		0	4.875	158.37 yes
24.75	26.5		0	0.75	24.36 yes
9.5	26.75		0	0.25	8.12 yes
9.5	26.625		0		0.00 yes
13.75	27.25		0	0.625	20.30 yes
16	29.875		0	2.625	85.28 yes
16.5	31.25		0	1.375	44.67 yes
20	32.5		0	1.25	40.61 yes
22.25	33.5	14		1	32.49 yes
14	16.625		0		0.00 yes
13.75	17.375		0	0.75	24.36 yes
15.5	19.875		0	2.5	81.22 yes
17.5	22.75		1	2.875	93.40 yes
15.75	24		2	1.25	40.61 yes
14.75	25.5		3	5.625	182.74 yes
10.25	30			10.125	328.92
11.5	31.75			1.75	56.85
8.625	33.375			1.625	52.79
8.5	33.75			0.375	12.18
	33.75			0	0.00 yes
	34.25			0.5	16.24 yes
	34.75			0.5	16.24 yes
	34.75			0	0.00 yes
	35.5			0.75	24.36 yes
	35.5			0	0.00 yes
	35.75			0.25	8.12 yes
	35.75			0.25	8.12 yes
	18				0.00 yes
	19			1	32.49 yes
	19			1	32.49 yes
	19.5			0.5	16.24 yes
	19.25				0.00 yes
	26.75			7.5	243.65 yes
	26.5			-0.25	-8.12 yes
	28.75			2.25	73.09 yes
	29.5			0.75	24.36 yes
	29.75			0.25	8.12 yes
	30			0.25	8.12 yes

Appendix A2: Alamogordo Year 2 Logsheet Data
NaCl Usage Data

Tank B					
Water Level	Salt Level	Bags salt added	Δ salt height (in)	Salt used (lb)	in use?
	32		2	64.97	yes
	31.75		1.75	56.85	yes
	14.75			0.00	yes
	22.25		7.5	243.65	yes
	23.5		1.25	40.61	yes
	23.75		0.25	8.12	yes
	14			0.00	yes
	21.25		7.25	235.53	yes
	23		1.75	56.85	yes
	16			0.00	
	23		7	227.40	yes
	24		1	32.49	yes
	24.75		0.75	24.36	yes
	25.25		0.5	16.24	yes
	26		0.75	24.36	yes
	26.25		1	32.49	yes
	14.5			0.00	
	21		6.5	211.16	yes
	21.75		0.75	24.36	yes
	22.75		1	32.49	yes
	23		0.25	8.12	yes
	14.75			0.00	
	15		0.25	8.12	no
	20		5	162.43	no
	23.5		3.5	113.70	no
	28.75		5.25	170.55	no
	31.75		3	97.46	no
	32.5		0.75	24.36	no
	31.75			0.00	no
	34		1.5	48.73	no
	34.5		0.5	16.24	no
	40.75		6.25	203.04	no
	44		3.25	105.58	no
	32			0.00	no
	33.25		1.25	40.61	no
	33.75		1.75	56.85	no
	38.25		4.5	146.19	no
	42		8.25	268.01	no
	42			0.00	no
	43		1	32.49	no
	43		0	0.00	no

Appendix A2: Alamogordo Year 2 Logsheet Data
 NaCl Usage Data

Tank B					
Water Level	Salt Level	Bags salt added	Δ salt height (in)	Salt used (lb)	in use?
	43			0	0.00 no
	43			0	0.00 no
	43			0	0.00 no
	43			0	0.00 no
	43			0	0.00 no
	43			0	0.00 no
	43			0	0.00 no
	43			0	0.00 no
	43			0	0.00 no
	43			0	0.00 no

Appendix A2: Alamogordo Year 2 Logsheet Data

NaCl Usage Data

Tank Filling - For calculating lb/inch

		# 40-lb bags	tank height	lb/inch	avg lb/inch
4/8/2013	Tank A	18	21	34.3	34.0
	Tank B	23	27.25	33.8	
4/18/2013	Tank A	14	17.75	31.5	31.6
	Tank B	18	22.75	31.6	
4/19/2013	Tank A	16	18.875	33.9	35.8
	Tank B	4	4.25	37.6	
6/3/2013	tank A	20	25.25	31.7	32.7
	tank B	15	17.75	33.8	
6/17/2013	tank A	9	12.125	29.7	
6/18/2013	tank A	20	26.375	30.3	31.6
	tank B	14	17	32.9	
6/19/2013	tank A	16	19.25	33.2	33.0
	tank B	8	9.75	32.8	
6/20/2013	tank A	14	18	31.1	32.7
	tank B	6	7	34.3	
6/24/2013	tank A	18	22	32.7	31.7
	tank B	9	11.75	30.6	
6/26/2013	tank A	16	20.25	31.6	32.8
	tank B	7	8.25	33.9	
7/8/2013	tank A	16	21	30.5	
7/9/2013	tank A	5	6	33.3	33.3
	tank B	10	12	33.3	
7/10/2013	tank A	34	41.375	32.9	
7/11/2013	tank A	10	12	33.3	
7/12/2013	tank A	29	35	33.1	
7/13/2013	tank A	15	19.375	31.0	
avg lb/inch - entire			32.5		

Appendix A2: Alamogordo Year 2 Logsheet Data
NaCl Usage Data

Tank Filling - For calculating lb/inch

# 40-lb bags	tank height	lb/inch	avg lb/inch
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Appendix A2: Alamogordo Year 2 Logsheet Data
NaCl Usage Data

Tank Filling - For calculating lb/inch

# 40-lb bags	tank height	lb/inch	avg lb/inch
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Appendix A2: Alamogordo Year 2 Logsheet Data
NaCl Usage Data

Tank Filling - For calculating lb/inch

# 40-lb bags	tank height	lb/inch	avg lb/inch
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Mixed Cl					
Mixed Cl	inches	duration (min)	volume (gal)	dilutions?	flow (gpm)
3/29/2013	6.5	60	32.5	no	0.54
3/29/2013	4.625	48	23.125	yes	0.48
3/29/2013	7	48	35	yes	0.73
3/29/2013	11.75	75	58.75	no	0.78
4/1/2013	6.375	60	31.875	yes	0.53
4/1/2013	11.625	90	58.125	yes	0.65
4/5/2013	6	60	30	no	0.50
4/5/2013	6.375	60	31.875	no	0.53
4/8/2013	11	120	55	no	0.46
4/9/2013	6.375	60	31.875	no	0.53
4/9/2013	6.25	60	31.25	yes	0.52
4/10/2013	5.625	60	28.125	no	0.47
4/10/2013	6.5	60	32.5	yes	0.54
4/10/2013	6.875	60	34.375	no	0.57
4/11/2013	9.75	60	48.75	no	0.81
4/11/2013	9.75	62	48.75	no	0.79
4/11/2013					
4/17/2013	9.625	63	48.125	no	0.76
4/17/2013	10.125	62	50.625	no	0.82
4/17/2013	18	121	90	no	0.74
4/18/2013	10.5	60	52.5	no	0.88
4/18/2013	11	61	55	with NaCl	0.90
4/18/2013	10.875	61	54.375	no	0.89
4/19/2013	11.375	61	56.875	no	0.93
4/19/2013	11.375	70	56.875	with NaCl	0.81
4/19/2013	10.125	62	50.625	with NaCl	0.82
4/22/2013	16	108	80	w/NaCl	0.74
4/22/2013	9	56	45	w/NaCl	0.80
4/23/2013	15	80	75	w/NaCl	0.94
4/23/2013	11.5	70	57.5	w/o NaCl	0.82
4/23/2013	4.75	31	23.75	w/o NaCl	0.77
4/24/2013	20.125	118	100.625	w/ NaCl	0.85
4/25/2013					
4/25/2013	3.5	62	17.5	w/ NaCl	0.28
4/26/2013	2.5	63	12.5	w/o NaCl	0.20
4/30/2013	3.625	66	18.125	w/o NaCl	0.27
4/30/2013	16.5	294	82.5	w/ NaCl	0.28
5/1/2013	3.625	64	18.125	w/ NaCl	0.28
5/1/2013	5.375	117	26.875	w/ NaCl	0.23
5/2/2013	13.75	256	68.75	w/ NaCl	0.27
5/2/2013					
5/31/2013	3.5	69	17.5	no	0.25

Mixed Cl					
Mixed Cl	inches	duration (min)	volume (gal)	dilutions?	flow (gpm)
5/31/2013	5.25	110	26.25	no	0.24
6/3/2013	2.875	62	14.375	no	0.23
6/3/2013	3.625	102	18.125	no	0.18
6/4/2013	7.625	136	38.125	no	0.28
6/4/2013	8.625	174	43.125	no	0.25
6/4/2013	3.125	64	15.625	no	0.24
6/4/2013	4.5	86	22.5	no	0.26
6/17/2013	3.75	78	18.75	no	0.24
6/17/2013	5.875	95	29.375	no	0.31
6/18/2013	10.25	114	51.25	no	0.45
6/18/2013	9.25	121	46.25	no	0.38
6/18/2013	4.25	69	21.25	no	0.31
6/19/2013	8.875	250	44.375	no	0.18
6/20/2013	12.375	167	61.875	no	0.37
6/20/2013	16.125	297	80.625	no	0.27
6/20/2013				no	
6/24/2013	5.5	71	27.5	no	0.39
6/24/2013	8.375	122	41.875	no	0.34
6/26/2013	3.75	107	18.75	no	0.18
6/26/2013	7	73	35	no	0.48
6/26/2013	6.75	60	33.75	no	0.56
6/26/2013	4.75	55	23.75	no	0.43
6/26/2013	5.875	65	29.375	no	0.45
7/8/2013	4.25	120	21.25	no	0.18
7/8/2013				no	
7/8/2013				no	
7/9/2013	5.875	63	29.375	no	0.47
7/9/2013	10.125	113	50.625	no	0.45
7/10/2013	6.625	67	33.125	no	0.49
7/10/2013	6	60	30	no	0.50
7/10/2013				no	
7/12/2013	3.125	40	15.625	no	0.39
7/12/2013	6.625	76	33.125	no	0.44
7/12/2013	3	40	15	no	0.38
7/12/2013	4.25	62	21.25	no	0.34
7/13/2013	6.625	80	33.125	no	0.41
7/13/2013	8.625	103	43.125	no	0.42
7/13/2013	7.875	100	39.375	no	0.39
7/13/2013	4.875	60	24.375	no	0.41

Mixed Na				
inches	duration	volume	dilutions?	M. Na flow
6	60	30	no	0.50
4.625	48	23.125	yes	0.48
8	41	40	yes	0.98
6	60	30	no	0.50
11	75	55	yes	0.73
9	60	45	yes	0.75
5.625	60	28.125	no	0.47
11.75	60	58.75	yes	0.98
7.625	60	38.125	yes	0.64
8.875	60	44.375	no	0.74
7.875	60	39.375	yes	0.66
9	60	45	no	0.75
8.625	60	43.125	yes	0.72
8.25	60	41.25	yes	0.69
9.625	68	48.125	yes	0.71
9.625	60	48.125	yes	0.80
9.25	63	46.25	yes	0.73
8.625	60	43.125	yes	0.72
9.5	60	47.5	yes	0.79
10.625	60	53.125	yes	0.89
9	62	45	yes	0.73
7.75	60	38.75	yes	0.65
8.375	60	41.875	yes	0.70
15.75	106	78.75	yes	0.74
12	76	60	yes	0.79
15.125	108	75.625	yes	0.70
8.5	63	42.5	yes	0.67
10.875	75	54.375	yes	0.73
7.625	49	38.125	yes	0.78
11.5	83	57.5	yes	0.69
5.625	70	28.125	yes	0.40
4	62	20	yes	0.32
4.75	63	23.75	yes	0.38
22.75	296	113.75	yes	0.38
4.25	73	21.25	yes	0.29
4.875	64	24.375	yes	0.38
10.375	108	51.875	yes	0.48
6.75	98	33.75	yes	0.34

Mixed Na				
inches	duration	volume	dilutions?	M. Na flow
5	73	25	yes	0.34
3.625	62	18.125	yes	0.29
5.5	102	27.5	yes	0.27
			yes	
12.375	174	61.875	yes	0.36
4.125	64	20.625	yes	0.32
6.5	86	32.5	yes	0.38
18	135	90	yes	0.67
12	95	60	yes	0.63
			yes	
16.125	138	80.625	yes	0.58
9.25	66	46.25	yes	0.70
9.25	133	46.25	yes	0.35
9.125	70	45.625	yes	0.65
4.5	36	22.5	yes	0.63
10.75	93	53.75	yes	0.58
7.75	61	38.75	yes	0.64
			yes	
6.75	59	33.75	yes	0.57
8.625	73	43.125	yes	0.59
7.5	60	37.5	yes	0.63
6.125	55	30.625	yes	0.56
			yes	
11.125	100	55.625	yes	0.56
7.125	60	35.625	yes	0.59
9.5	83	47.5	yes	0.57
20	176	100	yes	0.57
			yes	
7.5	67	37.5	yes	0.56
6.25	60	31.25	yes	0.52
21.375	187	106.875	yes	0.57
4.75	45	23.75	yes	0.53
4.625	40	23.125	yes	0.58
			yes	
			yes	
			yes	
11.625	103	58.125	yes	0.56
10.125	100	50.625	yes	0.51
7	60	35	yes	0.58

Appendix A2: Alamogordo Year 2 Logsheet Data
Remote Data - Single Stack Data

EDM SYSTEM			4/25/2013	
Location			Day/Time (MST)-->	10:00pm 4/26/2013
Tag			10:00 PM	5:30 AM
units				
RO Unit	NF Feed pH	AIT-2008A	6.3	6.2
	NF Permeate Flow	FIT-2009P	gpm	21.1 21
	NF Permeate Conductivity	AIT-2009P	μS/cm	1345 1061
	NF Reject Flow	FIT-2009R	gpm	14.6 14.9
	NF Reject Conductivity	AIT-2009R1	μS/cm	7530 7674
EDM Units	EDM-1 reference		volt	159 159
	EDM-1 current		amps	54 54
	EDM Total Feed Flow	FIT-2111	gpm	25 24
	EDM Feed Pressure	PIT-2111	psi	7.8 7.9
	EDM Feed Conductivity	AIT-2111	μS/cm	4892 5099
	EDM Feed Temperature	TIT-2111	°F	76 74
	EDM Diluate Conductivity	AIT-2110		
	EDM Diluate Temperature			
	EDM Total Mixed Na Flow	FIT-2113	gpm	0 0
	EDM Mixed Na Pressure	PIT-2113	psi	7.9
Tank 1	EDM Mixed Na Conductivity	AIT-2113	μS/cm	106900 108400
	EDM Mixed Na Temperature	TIT-2113	°F	78 76
	EDM Total Mixed Cl Flow	FIT-2115	gpm	27 27
	EDM Mixed Cl Pressure	PIT-2115	psi	8.1 8.1
	EDM Mixed Cl Conductivity	AIT-2115	μS/cm	-200 -200
	EDM Mixed Cl Temperature	TIT-2115	°F	78 76
	EDM Total NaCl Flow	FIT-2117	gpm	0 0
	EDM NaCl Pressure	PIT-2117	psi	7.7 8.05
	EDM NaCl Conductivity	AIT-2116	μS/cm	49000 50000
	EDM NaCl Temperature	TIT-2117	°F	78 76
Tank 2	EDM Total E-Rinse Flow	FIT-2124	gpm	0 0
	EDM E-Rinse Pressure	PIT-2124	psi	8 8.2
	EDM E-Rinse Conductivity	AIT-2123	μS/cm	20000 21000
	EDM E-Rinse Temperature	TIT-2124	°F	76 72
	High level switch on TK-2141	LSHH-2141	gray/red	gray gray

Appendix A2: Alamogordo Year 2 Logsheet Data
Remote Data - Single Stack Data

		5/1/2013	5/1/2013	5/1/2013	6/3/2013 actual	6/3/2013 HMI
EDM SYSTEM						
Location		11:12 PM	11:18 PM	11:21 PM	5:00 PM	5:00 PM
RO Unit	NF Feed pH	6.6	6.7	6.6		
	NF Permeate Flow	22.5	24.5	23		
	NF Permeate Conductivity	1076	1056	1274		
	NF Reject Flow	12.1	12.6	12.3		
	NF Reject Conductivity	7522	7754	7631		
EDM Units	EDM-1 reference			154	124	127
	EDM-1 current			52	52	53
	EDM Total Feed Flow			24	32	30
	EDM Feed Pressure			8.7	27.3	17.9
	EDM Feed Conductivity			5073	6306	6268
Tank 1	EDM Feed Temperature			80	86.5	86
	EDM Diluate Conductivity				5416	4286
	EDM Diluate Temperature				88.3	-
	EDM Total Mixed Na Flow			0	0	0
	EDM Mixed Na Pressure			8.7	28.4	18.1
	EDM Mixed Na Conductivity			110000	107.926	107700
	EDM Mixed Na Temperature			82	88.9	88
	EDM Total Mixed Cl Flow			27	36.9	40
	EDM Mixed Cl Pressure			9.2	27.1	18.8
	EDM Mixed Cl Conductivity			151600	161.6	161100
Tank 2	EDM Mixed Cl Temperature			82	88.8	88
	EDM Total NaCl Flow			0	0	0
	EDM NaCl Pressure			8.7	27.6	18.1
	EDM NaCl Conductivity			49000	50925	53000
	EDM NaCl Temperature			82	88.6	88
DI System	EDM Total E-Rinse Flow			0	0	0
	EDM E-Rinse Pressure			8.9	21.2	14
	EDM E-Rinse Conductivity			21000	20.906	20000
	EDM E-Rinse Temperature			80	89.7	89
High level switch on TK-2141				gray		grey

Appendix A2: Alamogordo Year 2 Logsheet Data
Remote Data - Single Stack Data

		6/4/2013	6/4/2013	6/4/2013	6/4/2013	6/4/2013
EDM SYSTEM		actual	HMI	6/4/2013	6/4/2013	6/4/2013
Location		4:00 PM	4:00 PM	7:30 PM	9:00 PM	10:30PM
RO Unit	NF Feed pH	6.03	5.9	5.9	6	6
	NF Permeate Flow	22.87	22.8	23.2	23.1	22.7
	NF Permeate Conductivity	1140.6	1129	1109	1089	1085
	NF Reject Flow	16.44	15.9	16.2	16.4	16
	NF Reject Conductivity	7683.6	7673	7653	7664	7695
EDM Units	EDM-1 reference	150	153	153	153	153
	EDM-1 current	54	55	56	55	55
	EDM Total Feed Flow	29.4	29	30	31	28
	EDM Feed Pressure	24.9	16.6	16.7	17.2	17.4
	EDM Feed Conductivity	5138	5108	5176	5159	5274
Tank 1	EDM Feed Temperature	87	86	86	85	85
	EDM Diluate Conductivity	3944	2428	2472	2491	2611
	EDM Diluate Temperature	89 -				
	EDM Total Mixed Na Flow	0	0	0	0	0
	EDM Mixed Na Pressure	25.5	16.6	17	17.4	17.3
	EDM Mixed Na Conductivity	103700	106200	114700	91700	34000
	EDM Mixed Na Temperature	89.6	89	88	88	87
	EDM Total Mixed Cl Flow	38.9	37	36	39	35
	EDM Mixed Cl Pressure	25	17.1	17.3	17.5	17.7
	EDM Mixed Cl Conductivity	162300	161900	162700	163900	164400
Tank 2	EDM Mixed Cl Temperature	89.5	88	89	88	87
	EDM Total NaCl Flow	0	0	0	0	0
	EDM NaCl Pressure	25.6	17	17.5	17	18.1
	EDM NaCl Conductivity	50543	56000	49000	49000	49000
	EDM NaCl Temperature	89.2	88	88	88	87
DI System	EDM Total E-Rinse Flow	0	0	0	0	0
	EDM E-Rinse Pressure	20.7	13.8	13.9	14	14.1
	EDM E-Rinse Conductivity	21321	20000	20000	21000	20000
	EDM E-Rinse Temperature	90.3	90	89	87	86
High level switch on TK-2141			grey	grey	grey	gray

Appendix A2: Alamogordo Year 2 Logsheet Data
Remote Data - Single Stack Data

EDM SYSTEM		6/5/2013	6/5/2013	6/17/2013	
RO Unit	Location	12:05 AM	5:30am	8:06 PM	
	NF Feed pH	6	5.9	6.2	
	NF Permeate Flow	22.1	22.7	40.4	
	NF Permeate Conductivity	1075	821	1099	
	NF Reject Flow	16.2	16.2	28.9	
	NF Reject Conductivity	7715	7048	7263	
EDM Units	EDM-1 reference	153	153	121	
	EDM-1 current	55	50	52	
	EDM Total Feed Flow	30	29	126	
	EDM Feed Pressure	17.5	17.9	53	
	EDM Feed Conductivity	5355	4915	4755	
	EDM Feed Temperature	85	83	86	
	Tank 1	EDM Diluate Conductivity	2653	2163	3470
		EDM Diluate Temperature			
		EDM Total Mixed Na Flow	0	0	0
		EDM Mixed Na Pressure	17.5-18.2	17.7 - 18.6	18.1
EDM Mixed Na Conductivity		31400	17700	35500	
EDM Mixed Na Temperature		87	85	88	
EDM Total Mixed Cl Flow		35	0	68	
EDM Mixed Cl Pressure		17.5	1.6	18.8	
EDM Mixed Cl Conductivity		165300	162900	134200	
EDM Mixed Cl Temperature		87	85	88	
Tank 2	EDM Total NaCl Flow	0	0	0	
	EDM NaCl Pressure	17.4	1.4	18.6	
	EDM NaCl Conductivity	56000	51000	50000	
	EDM NaCl Temperature	87	84	88	
	EDM Total E-Rinse Flow	0	0	11	
	EDM E-Rinse Pressure	14.2	2.2	12.2	
DI System	EDM E-Rinse Conductivity	21000	20000	21000	
	EDM E-Rinse Temperature	86	82	87	
	High level switch on TK-2141	grey		gray	

Appendix A2: Alamogordo Year 2 Logsheet Data
Remote Data - 2-Stack Data

EDM SYSTEM		Day/Time (MST)-->		4/22/2013	4/22/2013
Location		Tag	units	4:30 PM	9:00 PM
RO Unit	NF Feed pH	AIT-2008A		6.7	6.6
	NF Permeate Flow	FIT-2009P	gpm	40.3	40.2
	NF Permeate Conductivity	AIT-2009P	μS/cm	1364	1371
	NF Reject Flow	FIT-2009R	gpm	28.6	28.7
	NF Reject Conductivity	AIT-2009R1	μS/cm	7534	7637
EDM Units	EDM-1 reference		volt	159	159
	EDM-1 current		amps	61	61
	EDM-2 reference		volt	155	155
	EDM-2 current		amps	47	47
Tank 1	EDM Total Feed Flow	FIT-2111	gpm	58	56
	EDM Feed Pressure	PIT-2111	psi	10.2	10.3
	EDM Feed Conductivity	AIT-2111	μS/cm	4767	4867
	EDM Feed Temperature	TIT-2111	°F	78	77
	EDM Diluate Conductivity	AIT-2110	μS/cm		
	EDM Total Mixed Na Flow	FIT-2113	gpm	0	0
	EDM Mixed Na Pressure	PIT-2113	psi	10.2	10.3
	EDM Mixed Na Conductivity	AIT-2113	μS/cm	44900	56700
	EDM Mixed Na Temperature	TIT-2113	°F	80	79
	EDM Total Mixed Cl Flow	FIT-2115	gpm	60	59
	EDM Mixed Cl Pressure	PIT-2115	psi	10.4	10.2
	EDM Mixed Cl Conductivity	AIT-2115	mS/cm	126500	138700
	EDM Mixed Cl Temperature	TIT-2115	°F	80	79
	EDM Total NaCl Flow	FIT-2117	gpm	0	0
DI System	EDM NaCl Pressure	PIT-2117	psi	10.3	10.2
	EDM NaCl Conductivity	AIT-2116	mS/cm	53000	50000
	EDM NaCl Temperature	TIT-2117	°F	83	79
	EDM Total E-Rinse Flow	FIT-2124	gpm	0	0
	EDM E-Rinse Pressure	PIT-2124	psi	7.2	7.3
	EDM E-Rinse Conductivity	AIT-2123	mS/cm	22000	22000
	EDM E-Rinse Temperature	TIT-2124	°F	83	80
	High level switch on TK-2141	LSHH-2141	gray/red	gray	gray

Appendix A2: Alamogordo Year 2 Logsheet Data
Remote Data - 2-Stack Data

EDM SYSTEM		4/22/2013	4/23/2013	4/23/2013	4/23/2013
Location		10:30 PM	3:00 AM	6:00 AM	11:00 PM
RO Unit	NF Feed pH	5.9	6	6	7.1
	NF Permeate Flow	39.7	41.8	41.1	38.8
	NF Permeate Conductivity	1355	1362	1352	1471
	NF Reject Flow	28.6	28.4	29.1	29.6
	NF Reject Conductivity	7777	7856	7839	7981
EDM Units	EDM-1 reference	168	168	168	174
	EDM-1 current	63	64	64	64
	EDM-2 reference	160	160	160	165
	EDM-2 current	48	49	49	49
Tank 1	EDM Total Feed Flow	54	53	52	49
	EDM Feed Pressure	10.3	10.4	10.4	10.1
	EDM Feed Conductivity	5017	5213	5282	2617
	EDM Feed Temperature	77	77	77	77
	EDM Diluate Conductivity				
	EDM Total Mixed Na Flow	0	0	0	0
	EDM Mixed Na Pressure	10.6	10.4	10.4	10.3
	EDM Mixed Na Conductivity	55300	53500	55300	53000
	EDM Mixed Na Temperature	80	80	79	80
	EDM Total Mixed Cl Flow	63	61	66	61
	EDM Mixed Cl Pressure	10.4	10.7	10.7	10.2
	EDM Mixed Cl Conductivity	137900	133000	153100	127800
	EDM Mixed Cl Temperature	80	79	79	80
	EDM Total NaCl Flow	0	0	0	0
Tank 1	EDM NaCl Pressure	10.4	10.6	10.7	10.3
	EDM NaCl Conductivity	52000	48000	45000	47000
	EDM NaCl Temperature	80	79	79	80
	EDM Total E-Rinse Flow	0	0	0	0
	EDM E-Rinse Pressure	7.4	7.4	7.4	7.4
	EDM E-Rinse Conductivity	22000	22000	22000	27000
	EDM E-Rinse Temperature	80	79	79	79
	High level switch on TK-2141	gray	gray	gray	gray

Appendix A2: Alamogordo Year 2 Logsheet Data
Remote Data - 2-Stack Data

EDM SYSTEM		4/24/2013	4/24/2013	6/18/2013	6/18/2013
Location		2:30 AM	3:40 AM	8:06 PM	10:00 PM
RO Unit	NF Feed pH	7.1	7.1	6.2	6.1
	NF Permeate Flow	41.2	40.9	40.4	39.7
	NF Permeate Conductivity	1492	1489	1099	89
	NF Reject Flow	29.5	29.9	28.9	29.4
	NF Reject Conductivity	8019	8087	7263	7345
EDM Units	EDM-1 reference	174	174	121	126
	EDM-1 current	64	63	52	54
	EDM-2 reference	165	165	126	121
	EDM-2 current	48	48	53	52
Tank 1	EDM Total Feed Flow	52	53		54
	EDM Feed Pressure	10.2	10.2		18.2
	EDM Feed Conductivity	2844	5770	4755	4833
	EDM Feed Temperature	76	76	86	86
	EDM Diluate Conductivity			3470	3571
	EDM Total Mixed Na Flow	0	0	0	0
	EDM Mixed Na Pressure	10.5	11	18.1	18.4
	EDM Mixed Na Conductivity	68600	72000	35500	37700
	EDM Mixed Na Temperature	79	79	88	88
	EDM Total Mixed Cl Flow	62	0	68	74
	EDM Mixed Cl Pressure	10.1	10.2	18.8	18.2
	EDM Mixed Cl Conductivity	19600	199600	134200	135300
	EDM Mixed Cl Temperature	79	79	88	88
	EDM Total NaCl Flow	0	0	0	0
Tank 1	EDM NaCl Pressure	10.5	10.1	18.6	18.9
	EDM NaCl Conductivity	42000	45000	50000	49000
	EDM NaCl Temperature	79	79	88	88
	EDM Total E-Rinse Flow	0	0	11	9
	EDM E-Rinse Pressure	10.5	10.3	12.2	12.2
	EDM E-Rinse Conductivity		28000	21000	19000
	EDM E-Rinse Temperature		78	87	86
	High level switch on TK-2141		gray	gray	grey

Appendix A2: Alamogordo Year 2 Logsheet Data
Remote Data - 2-Stack Data

EDM SYSTEM		6/19/2013	6/19/2013	6/19/2013	6/19/2013
Location		12:35 AM	5:34 AM	7:52 PM	9:47 PM
RO Unit	NF Feed pH	6.1	6.1	6.2	6.2
	NF Permeate Flow	40.3	39.5	35.5	34.8
	NF Permeate Conductivity	1106	1106	1112	1116
	NF Reject Flow	29.1	29.2	27.6	27.6
	NF Reject Conductivity	7346	7403	7164	7131
EDM Units	EDM-1 reference	121	121	121	121
	EDM-1 current	52	51	48	48
	EDM-2 reference	126	126	113	113
	EDM-2 current	54	54	47	47
Tank 1	EDM Total Feed Flow	51	56	54	54
	EDM Feed Pressure	18.2	18.5	18.6	18.7
	EDM Feed Conductivity	4898	5022	4778	4865
	EDM Feed Temperature	85	85	86	85
	EDM Diluate Conductivity	3644	3754	3671	3750
	EDM Total Mixed Na Flow	0	0	0	0
	EDM Mixed Na Pressure	18.6	19.3	18.6	18.9
	EDM Mixed Na Conductivity	38000	45900	40300	42200
	EDM Mixed Na Temperature	88	88	88	19
	EDM Total Mixed Cl Flow	66	74	71	71
Tank 1	EDM Mixed Cl Pressure	18.2	19.1	18.7	19
	EDM Mixed Cl Conductivity	135800	135000	146400	150300
	EDM Mixed Cl Temperature	87	87	88	87
	EDM Total NaCl Flow	0	0	0	0
	EDM NaCl Pressure	18.7	18.8	18.8	19.1
	EDM NaCl Conductivity	50000	50000	50000	50000
	EDM NaCl Temperature	87	87	88	87
DI System	EDM Total E-Rinse Flow	11	11	11	11
	EDM E-Rinse Pressure	12.3	12.4	12.8	12.8
	EDM E-Rinse Conductivity	20000	20000	21000	20000
	EDM E-Rinse Temperature	86	86	87	86
High level switch on TK-2141		grey	gray	gray	gray

Appendix A2: Alamogordo Year 2 Logsheet Data
Remote Data - 2-Stack Data

EDM SYSTEM		6/19/2013	6/20/2013	6/20/2013	6/20/2013
Location		11:58 PM	6:28 PM	8:15 PM	8:20 PM
RO Unit	NF Feed pH	6.3	6.1	6.2	6.1
	NF Permeate Flow	34.8	35	34.7	39.9
	NF Permeate Conductivity	1116	1081	1058	1037
	NF Reject Flow	27.9	27.4	27.5	28.9
	NF Reject Conductivity	7155	6659	6696	6968
EDM Units	EDM-1 reference	121	133	133	133
	EDM-1 current	48	48	47	49
	EDM-2 reference	113	124	124	124
	EDM-2 current	47	47	47	49
Tank 1	EDM Total Feed Flow	56	55	55	54
	EDM Feed Pressure	18.6	18.5	18.5	19
	EDM Feed Conductivity	4894	4276	4273	4573
	EDM Feed Temperature	85	87	86	86
	EDM Diluate Conductivity	3796	3169	3156	3399
	EDM Total Mixed Na Flow	0	0	0	0
	EDM Mixed Na Pressure	19	18.5	18.3	19.5
	EDM Mixed Na Conductivity	50000	33800	35000	38500
	EDM Mixed Na Temperature	87	89	89	88
	EDM Total Mixed Cl Flow	66	68	68	67
Tank 1	EDM Mixed Cl Pressure	19.2	18.9	18.6	19.2
	EDM Mixed Cl Conductivity	144700	121400	118500	126200
	EDM Mixed Cl Temperature	87	89	88	88
	EDM Total NaCl Flow	0	0	0	0
	EDM NaCl Pressure	19	18.5	18.6	18.8
	EDM NaCl Conductivity	50000	49000	50000	49000
	EDM NaCl Temperature	87	89	88	88
DI System	EDM Total E-Rinse Flow	11	11	11	11
	EDM E-Rinse Pressure	12.8	13	13.1	13.2
	EDM E-Rinse Conductivity	21000	21000	20000	21000
	EDM E-Rinse Temperature	86	89	88	88
High level switch on TK-2141		gray	gray	gray	gray

Appendix A2: Alamogordo Year 2 Logsheet Data
Remote Data - 2-Stack Data

EDM SYSTEM		6/20/2013	6/20/2013	6/20/2013	6/21/2013	6/24/2013
	Location	9:04 PM	9:54 PM	11:03 PM	2:24 AM	7:51 PM
RO Unit	NF Feed pH	6.2	6.1	6.1	6.2	6.2
	NF Permeate Flow	39.7	35.1	34.6	34.9	34.9
	NF Permeate Conductivity	1068	1061	1062	1048	1126
	NF Reject Flow	29.4	27.4	27.6	27.6	27.6
	NF Reject Conductivity	7355	6738	6694	6775	6876
EDM Units	EDM-1 reference	133	133	133	133	128
	EDM-1 current	52	47	47	47	47
	EDM-2 reference	124	124	124	124	126
	EDM-2 current	52	47	47	47	49
Tank 1	EDM Total Feed Flow	56	55	57	57	55
	EDM Feed Pressure	18.9	18.5	18.7	18.7	20.2
	EDM Feed Conductivity	5028	4363	4331	4369	4483
	EDM Feed Temperature	3826	86	85	85	86
	EDM Diluate Conductivity	3830	3240	3255	3300	3368
	EDM Total Mixed Na Flow	0	0	0	0	0
	EDM Mixed Na Pressure	19	19.1	18.7	18.7	21
	EDM Mixed Na Conductivity	42800	38600	40900	42400	36100
	EDM Mixed Na Temperature	88	88	88	87	89
	EDM Total Mixed Cl Flow	67	65	64	65	67
Tank 1	EDM Mixed Cl Pressure	18.7	18.7	18.9	19.1	20.2
	EDM Mixed Cl Conductivity	126200	118200	120000	118500	113600
	EDM Mixed Cl Temperature	88	88	87	87	89
	EDM Total NaCl Flow	0	0	0	0	0
	EDM NaCl Pressure	18.8	19.1	18.9	19.1	20.6
	EDM NaCl Conductivity	49000	50000	49000	48000	49000
	EDM NaCl Temperature	88	88	87	87	88
DI System	EDM Total E-Rinse Flow	11	11	11	11	11
	EDM E-Rinse Pressure	13.3	13.3	13.3	13.4	13.9
	EDM E-Rinse Conductivity	21000	21000	21000	21000	21000
	EDM E-Rinse Temperature	87	87	87	86	88
	High level switch on TK-2141	gray	gray	gray	gray	gray

Appendix A2: Alamogordo Year 2 Logsheet Data
Remote Data - 2-Stack Data

EDM SYSTEM		6/25/2013	6/25/2013	6/26/2013	6/26/2013	6/27/2013
	Location	12:15 AM	6:25 AM	7:23 PM	9:42 PM	12:00 AM
RO Unit	NF Feed pH	6.1	6	6.3	6.3	6.3
	NF Permeate Flow	34.9	35.2	34.5	34.5	34.8
	NF Permeate Conductivity	1130	1141	1092	1048	984
	NF Reject Flow	28	28.2	28.7	28.4	28.2
	NF Reject Conductivity	7037	7126	7039	6935	6844
EDM Units	EDM-1 reference	128	128	118	118	118
	EDM-1 current	47	48	44	44	42
	EDM-2 reference	126	126	127	127	127
	EDM-2 current	50	51	51	50	49
Tank 1	EDM Total Feed Flow	55	54	55	55	56
	EDM Feed Pressure	20.6	20.7	18.8	18.8	19.1
	EDM Feed Conductivity	4803	5014	4737	4781	4708
	EDM Feed Temperature	85	84	84	83	83
	EDM Diluate Conductivity	3686	3923	3643	3705	3655
	EDM Total Mixed Na Flow	0	0	0	0	0
	EDM Mixed Na Pressure	20.6	20.6	19.5	19.2	18.9
	EDM Mixed Na Conductivity	42500	45900	5700	6300	6800
	EDM Mixed Na Temperature	87	86	87	85	85
	EDM Total Mixed Cl Flow	67	65	65	64	65
Tank 1	EDM Mixed Cl Pressure	20.3	21	19.4	19.1	19.8
	EDM Mixed Cl Conductivity	114600	108900	10300	10100	10000
	EDM Mixed Cl Temperature	87	86	86	85	85
	EDM Total NaCl Flow	0	0	0	0	0
	EDM NaCl Pressure	20.6	20.9	19	18.9	19.4
	EDM NaCl Conductivity	49000	49000	49000	49000	50000
	EDM NaCl Temperature	87	86	86	85	85
DI System	EDM Total E-Rinse Flow	11	11	11	11	11
	EDM E-Rinse Pressure	14	14.1	13.1	13.2	13.1
	EDM E-Rinse Conductivity	21000	21000	21000	21000	21000
	EDM E-Rinse Temperature	85	84	86	84	83
	High level switch on TK-2141	gray	gray	gray	gray	gray

Appendix A2: Alamogordo Year 2 Logsheet Data
Remote Data - 2-Stack Data

EDM SYSTEM			
	Location		
RO Unit	NF Feed pH		
	NF Permeate Flow	6/27/13 mass balance - 12:15 am	
	NF Permeate Conductivity	flow	cond
	NF Reject Flow	perm	35.5 970
	NF Reject Conductivity	reject	28.2 6803
EDM Units	EDM-1 reference	NF feed	63.7 3043
	EDM-1 current	Σ in	193839.1
	EDM-2 reference	Σ out	226279.6
	EDM-2 current	difference	15%
Tank 1	EDM Total Feed Flow		
	EDM Feed Pressure	SW	37.2 2619.855
	EDM Feed Conductivity	EDM dil	26.5 3637
	EDM Feed Temperature	NF feed	63.7 3043
	EDM Diluate Conductivity		
	EDM Total Mixed Na Flow	I think that we might lose the tank tonight	
	EDM Mixed Na Pressure		
	EDM Mixed Na Conductivity		
	EDM Mixed Na Temperature		
	EDM Total Mixed Cl Flow		
Tank 1	EDM Mixed Cl Pressure		
	EDM Mixed Cl Conductivity		
	EDM Mixed Cl Temperature		
	EDM Total NaCl Flow		
	EDM NaCl Pressure		
	EDM NaCl Conductivity		
	EDM NaCl Temperature		
DI System	EDM Total E-Rinse Flow		
	EDM E-Rinse Pressure		
	EDM E-Rinse Conductivity		
	EDM E-Rinse Temperature		
	High level switch on TK-2141		

Appendix A2: Alamogordo Year 2 Logsheet Data
Remote Data - 2-Stack Data

EDM SYSTEM		6/27/2013	6/27/2013		
	Location	3:00 AM	3:12 AM		
RO Unit	NF Feed pH	6.2	6.2		
	NF Permeate Flow	35	34.8	6/27/13 mass balance - 3:14 am	
	NF Permeate Conductivity	882	899	flow	cond
	NF Reject Flow	28.4	28.4	perm	35.2 872
	NF Reject Conductivity	6576	6508	reject	28 6447
EDM Units	EDM-1 reference	118	118	NF feed	63.2 2809
	EDM-1 current	41	40	Σ in	177528.8
	EDM-2 reference	127	127	Σ out	211210.4
	EDM-2 current	47	47	difference	17%
Tank 1	EDM Total Feed Flow	53	44		
	EDM Feed Pressure	19.3	14.2	SW	47.2 2618.831
	EDM Feed Conductivity	4527	4590	EDM dil	16 3370
	EDM Feed Temperature	82	82	NF feed	63.2 2809
	EDM Diluate Conductivity	3487	3387		
	EDM Total Mixed Na Flow	0	0	I think that we might lose the tank	
	EDM Mixed Na Pressure	19.6	14.6		
	EDM Mixed Na Conductivity	6800	6800		
	EDM Mixed Na Temperature	84	84		
	EDM Total Mixed Cl Flow	65	50		
Tank 1	EDM Mixed Cl Pressure	19.7	14.2		
	EDM Mixed Cl Conductivity	9900	9600		
	EDM Mixed Cl Temperature	84	84		
	EDM Total NaCl Flow	0	0		
	EDM NaCl Pressure	19.4	14.2		
	EDM NaCl Conductivity	51000	51000		
	EDM NaCl Temperature	84	84		
DI System	EDM Total E-Rinse Flow	11	11		
	EDM E-Rinse Pressure	13.3	11.8		
	EDM E-Rinse Conductivity	21000	21000		
	EDM E-Rinse Temperature	82	82		
	High level switch on TK-2141	gray	gray		

Appendix A2: Alamogordo Year 2 Logsheet Data
Remote Data - 2-Stack Data

	EDM SYSTEM	6/27/2013		
	Location	5:32 AM		
RO Unit	NF Feed pH	6.1		
	NF Permeate Flow	34.8 6/27/13 mass balance - 5:39 am		
	NF Permeate Conductivity	811	flow	cond
	NF Reject Flow	27.8 perm	35	821
	NF Reject Conductivity	6234 reject	28	6237
EDM Units	EDM-1 reference	118 NF feed	63	2724
	EDM-1 current	38 Σ in	171612	
	EDM-2 reference	127 Σ out	203371	
	EDM-2 current	44 difference	17%	
Tank 1	EDM Total Feed Flow	47		
	EDM Feed Pressure	14.3 SW	47	2553.447
	EDM Feed Conductivity	4418 EDM dil	16	3225
	EDM Feed Temperature	80 NF feed	63	2724
	EDM Diluate Conductivity	3230		
	EDM Total Mixed Na Flow	0		
	EDM Mixed Na Pressure	14		
	EDM Mixed Na Conductivity	6800		
	EDM Mixed Na Temperature	82		
	EDM Total Mixed Cl Flow	50		
	EDM Mixed Cl Pressure	14.5		
	EDM Mixed Cl Conductivity	9700		
	EDM Mixed Cl Temperature	82		
	EDM Total NaCl Flow	0		
	EDM NaCl Pressure	14.4		
EDM NaCl Conductivity	51000			
EDM NaCl Temperature	82			
DI System	EDM Total E-Rinse Flow	11		
	EDM E-Rinse Pressure	11.7		
	EDM E-Rinse Conductivity	20000		
	EDM E-Rinse Temperature	80		
	High level switch on TK-2141	gray		

Appendix A2: Alamogordo Year 2 Logsheet Data
Remote Data - 2-Stack Data

EDM SYSTEM				6/27/2013	6/27/2013	
	Location			6:52 PM	10:16 PM	
RO Unit	NF Feed pH			6.2	6.3	
	NF Permeate Flow	6/27/13 mass balance - 7:24 am		35.1	35.3	
	NF Permeate Conductivity	flow	cond	1064	1088	
	NF Reject Flow	perm	35 954	28.2	28.6	
	NF Reject Conductivity	reject	28 6563	6970	6975	
EDM Units	EDM-1 reference	NF feed	63 2962	128	128	
	EDM-1 current	Σ in	186606	46	46	
	EDM-2 reference	Σ out	217154	126	126	
	EDM-2 current	difference	15%	50	49	
Tank 1	EDM Total Feed Flow			54	56	
	EDM Feed Pressure	SW	47 2752.298	21	21.3	
	EDM Feed Conductivity	EDM dil	16 3578	4725	4835	
	EDM Feed Temperature	NF feed	63 2962	87	85	
	EDM Diluate Conductivity			3697	3802	
	EDM Total Mixed Na Flow			0	0	
	EDM Mixed Na Pressure			21.1	21.3	
	EDM Mixed Na Conductivity			4400	4800	
	EDM Mixed Na Temperature			89	88	
	EDM Total Mixed Cl Flow			64	65	
	EDM Mixed Cl Pressure			21.3	21.8	
	EDM Mixed Cl Conductivity			88100	86400	
	EDM Mixed Cl Temperature			89	87	
	EDM Total NaCl Flow			0	0	
	Tank 1	EDM NaCl Pressure			21.3	21.1
EDM NaCl Conductivity				49000	49000	
EDM NaCl Temperature				89	87	
EDM Total E-Rinse Flow				11	11	
EDM E-Rinse Pressure				13.6	13.8	
EDM E-Rinse Conductivity				21000	21000	
EDM E-Rinse Temperature				88	87	
DI System		High level switch on TK-2141			gray	gray

Appendix A2: Alamogordo Year 2 Logsheet Data
Remote Data - 2-Stack Data

EDM SYSTEM		7/8/2013	7/8/2013	7/9/2013	7/9/2013	7/9/2013	
	Location	8:37 PM	10:38	2:04	10:15 PM	11:30 PM	
RO Unit	NF Feed pH	6.5	6.4	6.4	6.4	6.4	
	NF Permeate Flow	35.4	35.3	35.3	34.9	35.1	
	NF Permeate Conductivity	1126	1133	1113	1111	1089	
	NF Reject Flow	28.5	28.8	28.6	28.4	28.5	
	NF Reject Conductivity	7131	7152	7234	6880	6972	
EDM Units	EDM-1 reference	128	128	128	135	136	
	EDM-1 current	47	47	47	47	48	
	EDM-2 reference	126	126	126	133	133	
	EDM-2 current	50	50	51	50	51	
Tank 1	EDM Total Feed Flow	58	60	59	58	56	
	EDM Feed Pressure	23.6	23.7	23.7	25.2	25	
	EDM Feed Conductivity	4794	4887	4998	4646	4663	
	EDM Feed Temperature	86	86	86	87	86	
	EDM Diluate Conductivity	3735	3820	3940	3579	3618	
	EDM Total Mixed Na Flow	0	0	0	0	0	
	EDM Mixed Na Pressure	24.2	24.1	24.4	25	25.2	
	EDM Mixed Na Conductivity	113900	115400	116100	110700	112200	
	EDM Mixed Na Temperature	89	88	88	89	89	
	EDM Total Mixed Cl Flow	63	61	56	58	61	
	EDM Mixed Cl Pressure	23.7	24.4	24	25.7	25.5	
	EDM Mixed Cl Conductivity	149800	150000	150400	158100	158400	
	EDM Mixed Cl Temperature	89	88	87	89	89	
Tank 1	EDM Total NaCl Flow	0	0	0	0	0	
	EDM NaCl Pressure	24.2	24.1	24.5	25.8	25.1	
	EDM NaCl Conductivity	48000	48000	51000	49000	52000	
	EDM NaCl Temperature	89	88	87	89	89	
	DI System	EDM Total E-Rinse Flow	10	11	10	10	10
		EDM E-Rinse Pressure	14.6	14.7	14.9	15.6	15.6
		EDM E-Rinse Conductivity	20000	21000	21000	21000	21000
EDM E-Rinse Temperature		87	87	86	87	87	
	High level switch on TK-2141	gray	gray	gray	gray	gray	

Appendix A2: Alamogordo Year 2 Logsheet Data
Remote Data - 2-Stack Data

EDM SYSTEM		7/10/2013	7/10/2013	7/11/2013	7/11/2013	7/12/2013
	Location	8:40 PM	10:50 PM	2:01	9:21 PM	12:13 AM
RO Unit	NF Feed pH	6.3	6.4	6.4	6.4	6.4
	NF Permeate Flow	34.7	34.6	34.5	34.8	35
	NF Permeate Conductivity	1112	1119	1099	1110	1130
	NF Reject Flow	28.4	28.5	28.4	28.9	29
	NF Reject Conductivity	6883	6880	6880	6850	6982
EDM Units	EDM-1 reference	144	144	144	134	134
	EDM-1 current	49	49	48	47	47
	EDM-2 reference	141	141	141	136	136
	EDM-2 current	52	52	52	51	51
Tank 1	EDM Total Feed Flow	59	58	59	58	56
	EDM Feed Pressure	25.8	25.9	26	25	23
	EDM Feed Conductivity	4513	4546	4505	4462	4826
	EDM Feed Temperature	87	87	87	86	85
	EDM Diluate Conductivity	3413	3446	3436	3355	3709
	EDM Total Mixed Na Flow	0	0	0	0	0
	EDM Mixed Na Pressure	25.7	26.4	26.3	25.6	23.8
	EDM Mixed Na Conductivity	111000	111500	111800	110200	110400
	EDM Mixed Na Temperature	89	89	89	88	88
	EDM Total Mixed Cl Flow	59	55	51	53	50
	EDM Mixed Cl Pressure	26.1	25.8	25.8	25.6	22.9
	EDM Mixed Cl Conductivity	155400	155700	156200	156600	162700
	EDM Mixed Cl Temperature	89	89	89	88	87
	EDM Total NaCl Flow	0	0	0	0	0
	EDM NaCl Pressure	26.3	26.2	26.1	25	23.2
EDM NaCl Conductivity	49000	48000	51000	47000	49000	
EDM NaCl Temperature	89	89	89	88	87	
DI System	EDM Total E-Rinse Flow	10	10	10	10	10
	EDM E-Rinse Pressure	16.2	16.4	16.4	16.1	15.8
	EDM E-Rinse Conductivity	21000	22000	21000	17000	16000
	EDM E-Rinse Temperature	88	88	88	87	86
	High level switch on TK-2141	gray	gray	gray	gray	gray

Appendix A2: Alamogordo Year 2 Logsheet Data
Remote Data - 2-Stack Data

EDM SYSTEM		7/12/2013	7/12/2013	7/13/2013	7/13/2013	7/13/2013
	Location	7:30 PM	9:30 PM	1:30	7:44	6:21 PM
RO Unit	NF Feed pH	6.5	6.4	6.4	6.4	6.4
	NF Permeate Flow	35.1	35.2	34.4	34.9	34.7
	NF Permeate Conductivity	1140	1147	1154	1155	1112
	NF Reject Flow	29	28.9	29.6	29.4	29.6
	NF Reject Conductivity	7128	7148	1154	1155	7293
EDM Units	EDM-1 reference	154	154	154		154
	EDM-1 current	49	48	48		46
	EDM-2 reference	151	151	151		146
	EDM-2 current	54	54	54		52
Tank 1	EDM Total Feed Flow	56	58	57		55
	EDM Feed Pressure	24.6	25.1	25.4		25.4
	EDM Feed Conductivity	4943	4978	5043		4821
	EDM Feed Temperature	86	86	85		87
	EDM Diluate Conductivity	3872	3907	3986		3974
	EDM Total Mixed Na Flow	0	0	0		0
	EDM Mixed Na Pressure	25	25	25.7		26.1
	EDM Mixed Na Conductivity	111800	109500	110700		110800
	EDM Mixed Na Temperature	89	89	88		90
	EDM Total Mixed Cl Flow	53	52	47		44
	EDM Mixed Cl Pressure	25.1	25.5	25.9		26.2
	EDM Mixed Cl Conductivity	158900	159200	160000		15850
	EDM Mixed Cl Temperature	89	88	88		89
	EDM Total NaCl Flow	0	0	0		0
	Tank 1	EDM NaCl Pressure	24.6	25.2	25.5	
EDM NaCl Conductivity		47000	47000	159900		47000
EDM NaCl Temperature		89	88	87		89
EDM Total E-Rinse Flow		0	0	0		0
EDM E-Rinse Pressure		16.1	16.1	16.4		17.5
EDM E-Rinse Conductivity		22000	23000	22000		25000
EDM E-Rinse Temperature		88	87	86		89
High level switch on TK-2141		gray	gray	gray		gray

Appendix A2: Alamogordo Year 2 Logsheet Data
Remote Data - 2-Stack Data

EDM SYSTEM		7/13/2013	7/14/2013		
	Location	9:24 PM	1:31		
RO Unit	NF Feed pH	6.3	6.3		
	NF Permeate Flow	34.4	34.8	7/14/13 mass balance - 1:36 am	
	NF Permeate Conductivity	1126	1137	flow	cond
	NF Reject Flow	29.9	30.2	perm	34.8 1137
	NF Reject Conductivity	7518	8024	reject	30.2 8024
EDM Units	EDM-1 reference	154	154	NF feed	65 3680
	EDM-1 current	45	44	Σ in	239200
	EDM-2 reference	146	146	Σ out	281892.4
	EDM-2 current	52	52	difference	16%
Tank 1	EDM Total Feed Flow	54	60		
	EDM Feed Pressure	25.7	26.1	SW	37.2 2652.446
	EDM Feed Conductivity	5260	5840	EDM dil	27.8 5055
	EDM Feed Temperature	86	86	NF feed	65 3680
	EDM Diluate Conductivity	4440	5055		
	EDM Total Mixed Na Flow	0	0		
	EDM Mixed Na Pressure	25.8	26.6		
	EDM Mixed Na Conductivity	109900	111700		
	EDM Mixed Na Temperature	89	88		
	EDM Total Mixed Cl Flow	42	41		
	EDM Mixed Cl Pressure	26.2	26.8		
	EDM Mixed Cl Conductivity	158800	149800		
	EDM Mixed Cl Temperature	89	88		
	EDM Total NaCl Flow	0	0		
DI System	EDM NaCl Pressure	26.6	27		
	EDM NaCl Conductivity	51000	149800		
	EDM NaCl Temperature	89	88		
	EDM Total E-Rinse Flow	0	0		
	EDM E-Rinse Pressure	17.6	17.9		
	EDM E-Rinse Conductivity	27000	27000		
	EDM E-Rinse Temperature	87	86		
	High level switch on TK-2141	gray	gray		

Appendix A2: Alamogordo Year 2 Logsheet Data

Voltage Current Data

Date	Time	EDM1			EDM2		
		volts	amps	#quads	volts	amps	#quads
3/25/2013	13:07	160	53	100	150	49	95
3/25/13	14:13	160	62	100	149	54	95
3/25/13	15:58	160	63	100	149	54	95
3/25/13	17:00	160	64	100	149	55	95
3/26/13	10:00	160	41	100	150	40	95
3/26/13	10:45	160	63	100	150	56	95
3/26/13	11:25	160	63	100	150	55	95
3/26/13	12:45	160	65	100	150	56	95
3/26/13	13:15	160	63	100	150	55	95
3/26/13	13:55	160	58	100	150	50	95
3/26/13	14:32	160	60	100	150	51	95
3/27/13	9:26	160	49	100	150	48	95
3/27/13	10:31	160	63	100	150	54	95
3/27/13	11:11	160	64	100	150	54	95
3/27/13	11:43	160	64	100	150	55	95
3/27/13	11:46	160	64	100	150	55	95
3/28/13	8:07	162	56	100	152	49	95
3/28/13	8:16	162	57	100	152	49	95
3/28/13	10:13	160	62	100	150	53	95
3/28/13	14:40	160	62	100	150	52	95
4/1/13	9:15	160	57	100	150	50	95
4/1/13	10:08	160	59	100	150	50	95
4/1/13	11:15	160	60	100	150	50	95
4/1/13	12:00	160	62	100	150	52	95
4/1/13	13:05	160	63	100	150	52	95
4/1/13	14:24	160	64	100	150	52	95
4/3/13	16:33	160	54	100	150	45	95
4/3/13	16:47	160	55	100	150	45	95
4/3/13	17:01	160	56	100	149	46	95
4/3/13	17:03	150	54	100	140	43	95
4/3/13	17:20	160	57	100	140	45	95
4/5/13	9:54	160	58	100	151	47	95
4/5/13	11:00	160	61	100	151	48	95
4/5/13	11:36	160	62	100	151	49	95
4/8/13	9:50	160	59	100	150	47	95
4/8/13	10:46	160	58	100	150	47	95
4/8/13	10:55	150	56	100	140	45	95
4/8/13	12:45	150	61	100	140	49	95
4/8/13	14:20	150	62	100	140	49	95
4/8/13	14:55	150	63	100	140	49	95
4/8/13	16:48	150	63	100	140	49	95
4/9/13	8:18	150	61	100	140	57	95

Appendix A2: Alamogordo Year 2 Logsheet Data

Voltage Current Data

Date	Time	EDM1			EDM2		
		volts	amps	#quads	volts	amps	#quads
4/9/13	9:42	150	59	100	140	46	95
4/9/13	10:28	150	60	100	140	47	95
4/9/13	11:27	150	61	100	140	48	95
4/9/13	12:27	150	62	100	140	48	95
4/9/13	14:50	150	62	100	140	48	95
4/9/13	16:43	150	63	100	140	48	95
4/10/13	8:21	160	54	100	150	32	95
4/10/13	9:54	160	62	100	150	49	95
4/10/13	11:15	160	64	100	150	50	95
4/10/13	12:31	160	65	100	150	50	95
4/10/13	13:44	160	65	100	150	50	95
4/10/13	14:27	160	66	100	150	50	95
4/10/13	15:25	160	65	100	150	49	95
4/10/13	16:40	160	66	100	150	49	95
4/11/13	9:40	160	42	100	150	35	95
4/11/13	9:50	175	56	100	160	45	95
4/11/13	11:25	175	65	100	160	48	95
4/11/13	12:04	175	66	100	160	49	95
4/11/13	12:50	175	66	100	160	49	95
4/11/13	13:38	175	66	100	160	48	95
4/11/13	14:41	175	67	100	160	49	95
4/11/13	16:00	175	67	100	160	49	95
4/11/13	17:08	175	68	100	160	50	95
4/17/13	8:30	155	36	100	155	31	95
4/17/13	10:05	155	57	100	155	46	95
4/17/13	10:30	155	58	100	155	47	95
4/17/13	15:05	155	61	100	155	48	95
4/17/13	15:05	155	61	98	155	48	95
4/18/13	8:38	155	50	98	155	42	95
4/18/13	9:20	155	58	98	155	46	95
4/18/13	9:56	155	61	98	155	47	95
4/18/13	12:12	155	61	98	154	48	95
4/18/13	13:30	155	62	98	154	48	95
4/18/13	16:42	155	62	98	155	48	95
4/19/13	8:27	155	51	98	155	43	95
4/19/13	9:27	155	61	98	155	50	95
4/19/13	9:27	165	63	98	155	61	95
4/19/13	10:29	165	64	98	165	51	95
4/19/13	14:51	165	65	98	165	50	95
4/19/13	16:38	165	65	98	164	51	95
4/19/13	16:38	170	68	98	164	51	95
4/19/13	18:01	170	67	98	164	49	95

Appendix A2: Alamogordo Year 2 Logsheet Data

Voltage Current Data

Date	Time	EDM1			EDM2		
		volts	amps	#quads	volts	amps	#quads
4/19/13	21:15	170	66	98	164	49	95
4/19/13	22:00	170	66	98	164	49	95
4/20/13	8:21	170	66	98	164	49	95
4/22/13	9:05	155	45	98	155	37	95
4/22/13	11:30	155	59	98	155	46	95
4/22/13	13:27	155	59	98	155	46	95
4/22/13	16:33	155	60	98	155	46	95
4/22/13	21:11	155	59	98	155	45	95
4/22/13	22:00	155	60	98	155	46	95
4/22/13	22:05	160	61	98	155	46	95
4/22/13	22:10	164	63	98	160	47	95
4/23/13	8:30	164	65	98	160	49	95
4/23/13	8:38	170	67	98	165	51	95
4/23/13	9:36	170	66	98	165	50	95
4/23/13	10:55	170	65	98	165	51	95
4/23/13	12:41	170	66	98	165	51	95
4/23/13	15:20	170	66	98	165	51	95
4/23/13	22:34	170	64	98	165	49	95
4/24/13	7:24	170	63	98	165	48	95
4/24/13	8:30	170	62	98	166	49	95
4/25/13	10:50	156	39	98			
4/25/13	11:50	156	51	98			
4/25/13	15:13	156	57	98			
4/25/13	16:19	156	57	98			
4/25/13	16:51	156	56	98			
4/25/13	21:35	156	54	98			
4/25/13	21:52	156	54	98			
4/26/13	8:20	156	51	98			
4/30/13	16:37	154	57	98			
4/30/13	21:29	155	58	98			
4/30/13	21:38	160	59	98			
5/1/13	7:00	150	59	98			
5/1/13	8:48	150	56	98			
5/1/13	11:55	150	57	98			
5/1/13	13:55	150	54	98			
5/1/13	16:04	150	54	98			
5/1/13	16:30	150	53	98			
5/1/13	21:16	150	50	98			
5/2/13	9:47	160	51	98			
5/2/13	14:37	160	49	98			
5/2/13	15:38	160	48	98			
5/2/13	21:41	160	47	98			

Appendix A2: Alamosordo Year 2 Logsheet Data

Voltage Current Data

Date	Time	EDM1			EDM2		
		volts	amps	#quads	volts	amps	#quads
5/3/13							
5/29/13	9:31	150	34	98			
5/29/13	9:41	150	43	98			
5/29/13	14:15	150	56	98			
5/31/13	8:52	150	48	98			
5/31/13	11:38	150	58	98			
5/31/13	11:39	125	50	98			
5/31/13	12:20	125	45	98			
5/31/13	12:21	133	49	98			
5/31/13	15:22	125	51	98			
6/3/13	9:00	150	44	98			
6/3/13	10:00	139	50	98			
6/3/13	11:37	139	53	98			
6/3/13	14:50	139	54	98			
6/3/13	16:53	124	50	98			
6/4/13	8:45	150	41	98			
6/4/13	11:10	150	53	98			
6/4/13	11:30	150	54	98			
6/4/13	14:53	150	54	98			
6/4/13	16:47	150	55	98			
6/10/13	16:00	129	34	91	130	33	94
6/10/13	16:30	129	48	91	130	46	94
6/14/13	15:32	130	52	91	130	49	94
6/17/13	9:12	111	43	91	111	42	94
6/17/13	9:26	120	50	91	126	50	94
6/17/13	13:45	115	51	91	124	52	94
6/18/13	11:02	118	52	91	125	53	94
6/18/13	14:14	118	51	91	125	52	94
6/19/13	10:12	118	53	91	112	52	94
6/19/13	14:26	118	47	91	112	46	94
6/20/13	10:00	118	47	91	112	47	94
6/20/13	11:00	130	50	91	123	50	94
6/20/13	11:23	130	48	91	123	48	94
6/20/13	15:09	130	48	91	123	48	94
6/21/13	8:05	130	48	91	123	48	94
6/21/13	9:00	125	46	91	125	47	94
6/21/13	10:42	125	47	91	125	49	94
6/21/13	11:53	125	45	91	125	46	94
6/21/13	12:52	137	49	91	130	50	94
6/21/13	14:51	137	48	91	130	48	94
6/24/13	9:16	125	33	91	125	34	94
6/24/13	9:30	130	37	91	130	37	94

Appendix A2: Alamogordo Year 2 Logsheet Data

Voltage Current Data

Date	Time	EDM1			EDM2		
		volts	amps	#quads	volts	amps	#quads
6/24/13	10:30	130	44	91	130	44	94
6/24/13	11:20	113	43	91	113	43	94
6/24/13	13:45	113	46	91	113	46	94
6/24/13	16:30	120	47	91	120	47	94
6/25/13	8:10	125	47	91	125	50	94
6/25/13	8:36	130	48	91	131	51	94
6/26/13	10:13	130	31	91	131	34	94
6/26/13	10:18	113	31	91	110	32	94
6/26/13	10:21	113	33	91	130	39	94
6/26/13	14:22	113	43	91	130	50	94
6/26/13	16:32	115	44	91	126	49	94
6/27/13	8:20	115	43	91	126	50	94
6/27/13	11:20	120	44	91	125	48	94
6/27/13	11:40	130	48	91	125	50	94
6/27/13	12:13	130	47	91	125	49	94
6/27/13	16:21	125	45	91	125	48	94
6/27/13	16:46	125	46	91	125	49	94
6/27/13	18:03	125	45	91	125	48	94
6/28/13	7:45	125	46	91	125	49	94
6/28/13	8:20	142	50	91	131	50	94
6/28/13	15:30	130	45	91	135	49	94
7/8/13	11:45	125	44	91	125	49	94
7/8/13	16:00	125	45	91	125	49	94
7/9/13	8:20	136	50	91	123	49	94
7/9/13	10:35	125	42	91	123	44	94
7/9/13	12:00	136	49	91	149	54	94
7/9/13	14:50	136	46	91	125	47	94
7/9/13	15:25	136	48	91	125	48	94
7/9/13	20:15	132	47	91	132	49	94
7/10/13	8:38	132	46	91	132	49	94
7/10/13	10:13	140	49	91	132	50	94
7/10/13	10:46	132	47	91	132	50	94
7/10/13	15:44	144	49	91	140	51	94
7/11/13	8:24	141	48	91	140	52	94
7/11/13	16:35	131	35	91	131	37	94
7/12/13	9:52	155	49	91	142	49	94
7/12/13	11:30	124	49	91	126	50	94
7/12/13	13:56	152	47	91	152	51	94
7/12/13	13:56	135	43	91	135	47	94
7/12/13	16:28	148	48	91	145	51	94
7/12/13	17:20	150	48	91	150	53	94
7/13/13	8:31	150	47	91	150	53	94

Appendix A2: Alamogordo Year 2 Logsheet Data

Voltage Current Data

Date	Time	EDM1			EDM2		
		volts	amps	#quads	volts	amps	#quads
7/13/13	11:50	140	44	91	145	51	94
7/13/13	16:25	145	45	91	145	51	94
7/13/13	16:43	145	45	91	145	51	94

Table A2-1.—Grab Samples - Conductivity

	Source	NF Feed	NF Perm	NF Conc	EDM Feed	EDM Diluate	Mix Na	Mix Cl	NaCl	E-Rinse	RO/DI
Date/Time	μS/cm	μS/cm	μS/cm	μS/cm	μS/cm	μS/cm	mS/cm	mS/cm	mS/cm	mS/cm	μS/cm
3/28/2013 17:03	3478	3396	1327	7993	4494	3124	115.5	128.9	41.6	20.7	100
4/5/2013 10:25	3538	3343	1362	7795	4474	3030	108	133.9	47.6	22.66	68.9
4/8/2013 16:40	3567	3514	1374	8007	4942	3493	115	166.4	50.11	21.72	127.5
4/9/2013 10:33	3490	3449	1372	7999	4997	3578	109.6	150	49.46	21.88	158.8
4/9/2013 16:37	3493	3536	1355	8265	5343	3980	111.6	167.3	48.12	21.3	120.9
4/10/2013 11:17	3598	3562	1385	8271	5152	3731	109.4	158.5	46.7	21.84	64.85
4/10/2013 13:30	3615	3614	1397	8312	5267	3889	114	165.8	47.08	20.94	72.32
4/10/2013 16:30	3607	3652	1394	8463	5454	4057	110.5	168.3	48.17	20.92	67
4/11/2013 12:05	3478	3460	1346	8025	4921	3508	109.6	153.5	45.53	20.87	54.55
4/11/2013 14:47	3608	3556	1404	8216	5106	3609	111.7	166.4	43.83	21.33	77.35
4/11/2013 16:22	3584	3583	1399	8286	5200	3780	109.7	167.3	43.56	21.11	55.85
4/17/2013 10:57	3489	3351	1422	7707	4508	3137	112.4	150.5	43.51	20.6	129
4/17/2013 14:35	3374	3361	1366	7702	4648	3232	111.2	165.3	45.24	22.13	77.66
4/17/2013 16:34	3451	3382	1375	7862	4702	3336	108.8	150.3	46.38	20.91	114.2
4/18/2013 10:31	3444	3426	1369	7879		3509	113.4	144.3	44.71	21.8	44.82
4/18/2013 13:00	3457	3414	1330	7920	4950	3638	112	163.2	46.17	21.38	96.42
4/18/2013 14:11	3545	3553	1403	8175	5099	3706	111	164.4	45.84	24.2	128.5
4/18/2013 16:21	3459	3445	1364	8010	5032	3716	110.5	166.4	47.42	21.82	90.6
4/19/2013 10:52	3759	3525	1403	8065	5090	3702	110.8	153.3	47.39	22.6	57.27
4/19/2013 17:35	3635	3553	1392	8198	5084	3698	112.8	165.3	46.25	23.88	133.8
4/19/2013 21:30	3590	3591	1402	8244	5181	3839	108.8	158.9	44.97	22.87	67.8
4/22/2013 11:06	3375	3357	1362	7739	4799	3472	110.8	139.2	47.12	22.54	125.9
4/22/2013 13:40	3350	3361	1344	7787	4920	3655	110.5	161	47.33	23.19	112.6
4/22/2013 16:23	3343	3390	1354	7851	5134	3704	112.9	163.4	47.22	22.81	55.62
4/22/2013 21:20	3303	3440	1357	7871	5088	3835	109.6	163.6	46.86	23.13	120.2

	Source	NF Feed	NF Perm	NF Conc	EDM Feed	EDM Diluate	Mix Na	Mix Cl	NaCl	E-Rinse	RO/DI
Date/Time	μS/cm	μS/cm	μS/cm	μS/cm	μS/cm	μS/cm	mS/cm	mS/cm	mS/cm	mS/cm	μS/cm
4/23/2013 8:07	3514	3573	1389	8230	5394	4096	106.5	159.2	44.6	22.24	278.3
4/23/2013 11:10	3576	3621	1404	8318	5360	3917	107.8	162.3	43.92	21.52	73.4
4/23/2013 15:31	3571	3603	1416	8304	5476	3879	108.4	162.8	44.77	23.98	71.75
4/24/2013 8:23	3446	3614	1368	8319	5706	4294	107.2	162.2	45.04	28.87	131.2
4/26/2013 8:39	3391	3374	1408	7672	5129	3519	106	154	51.58	24.33	111.1
4/30/2013 16:16	3709	3434	1479	7984	5203	3358	108.2	151.9	52.4	23.5	68.2
5/1/2013 9:13	3407	3536	1467	8351	5497	4100	109.5	159.7	53.83	21.46	107.3
5/1/2013 13:40	3455	3448	1514	8038	5222	3794	109.1	156	53.73	21.27	53.53
5/1/2013 16:50	3285	3289	1395	8061	5053	3571	108.6	156.9	53.06	21.34	53.69
5/1/2013 21:30	3152	3178	1342	8093	4911	3609	108.9	156.8	52.57	21.14	158.8
5/2/2013 14:54	3228	3220	1369	7525	4710	3442	106.2	153.7	52.7	21.29	39.77
5/29/2013 13:59	2873	2923	1036	7879	4844	3445	107.9	153.6	52.27	22.2	196.7
5/31/2013 13:50	3136	3300	1185	7886	5198	4033	109.1	157.7	51.29	21.7	108.3
5/31/2013 15:25	3244	3533	1248	8300	6879	4850	109.6	160.6	51.9	21.49	74.69
6/3/2013 11:42	3185	3374	1242	8075	5494	4238	109.4	141.9	51.08	21.56	95.26
6/3/2013 14:53	3168	3426	1253	8121	5508	4403	109.5	159.2	50.69	23.06	46.82
6/4/2013 11:16	3035	3189	1150	7859	5577	3985	107.8	141	51.82	25.25	156.5
6/4/2013 14:55	3010	3194	1158	7849	5165	3947	105.7	102.4	50.58	27.18	65.81
6/4/2013 16:55	3000	3186	1152	7873	5092	4083	109.3	164.6	52.57	21.22	115
6/14/2013 15:45	3053	2941	1147	7930	4350	2882	111.4	138.4	48.74	20.93	61.02
6/17/2013 14:03	3077	3146	1098	7504	4838	3572	110.9	153.2	53.91	21.32	33.2
6/18/2013 11:02	3126	3162	1113	7481	4699	3435	108.1	139.2	49.96	29	39.04
6/18/2013 14:24	3058	3085	1131	7407	4631	3397	113.7	153.2	51.02	27.89	48.14
6/19/2013 10:50	3106	3386	1134	8130	5703	4581	113.4	152.9	49.75	21.25	72.03
6/19/2013 14:32	3065	3138	1088	7198	4649	3571	112.7	158.9	52.02	25.2	124.1

	Source	NF Feed	NF Perm	NF Conc	EDM Feed	EDM Diluate	Mix Na	Mix Cl	NaCl	E-Rinse	RO/DI
Date/Time	μS/cm	μS/cm	μS/cm	μS/cm	μS/cm	μS/cm	mS/cm	mS/cm	mS/cm	mS/cm	μS/cm
6/20/2013 10:35	3111	3301	1131	7484	5029	3985	104.3	153.6	51.53	21.35	112.7
6/20/2013 14:55	3123	3099	1110	7038	4485	3308	104.8	157.3	50.38	21.12	180.4
6/21/2013 11:10	3064	3165	1102	7282	4788	3692	107.9	151.5	50.16	21.56	108.2
6/21/2013 15:09	3051	3039	1095	6958	4271	3199	115.5	158.4	50.48	21.583	82.15
6/24/2013 13:34	3102	3117	1147	7148	4507	3422	107.8	150.8	49.44	22.69	142.7
6/24/2013 16:35	3063	3110	1096	6857	4524	3351	111.8	157.3	52.26	21.59	124.7
6/25/2013 8:20	3113	3268	1162	7397	5042	3983	109.7	159.1	50.94	21.09	49.81
6/26/2013 14:30	3101	3113	1093	6994	4457	3355	109.2	153.6	52.12	20.35	67.39
6/26/2013 16:23	3121	3131	1094	7085	4614	3441	112.2	158.2	52.86	22.2	57.02
6/27/2013 8:31	3019	3238	1080	7187	5118	4056	107.5	150.8	51.87	22.9	82.38
6/27/2013 16:59	3070	3154	1144	7247	4710	3680	111.1	161.58	52.4	22.79	61.9
6/28/2013 7:58	3074	3278	1151	7286	5809	4035	110.1	161.4	49.83	20.75	
6/28/2013 15:37	3048	3119	1113	7022	4612	3477	112.4	159.3	50.09	23.03	61.64
7/8/2013 11:51	3196	3087	1195	6719	4397	3216	108.3	142.6	50.45	25.4	34
7/8/2013 15:57	3121	3132	1123	7101	4491	3425	110.8	158.2	50.19	21.84	73.45
7/9/2013 8:34	3098	3296	1140	7353	4940	3913	110.3	158.9	50.05	22.36	95.97
7/9/2013 15:34	3066	3126	1096	7071	4530	3455	112.6	158.6	51.08	22.52	160.8
7/10/2013 8:22	3009	3131	1104	7022	4342	3692	110.3	157.7	50.05	23.56	116.9
7/10/2013 15:57	3134	3194	1143	7154	4833	3407	111.6	157.7	50.33	22.45	
7/11/2013 8:45	3080	3196	1115	7177	4612	3584	109.4	156.4	48.34	25.73	78.01
7/12/2013 9:30	3101	3220	1132	7186	4829	3749	109.7	164.6	49.68	22.33	103.1
7/12/2013 17:10	3154	3270	1158	7292	4929	3810	109.4	163	49.97	23.64	
7/13/2013 11:30	3028	3150	1124	7088	4447	3518	101.3	147.3	49.22	24.44	201.8
7/13/2013 16:40	3070	3307	1104	7334	4748	3781	109.2	156.1	55.61	28.38	100.6

Table A2-1.—Grab Samples - pH

	Source	NF Feed	NF Perm	NF Conc	EDM Feed	EDM Diluate	Mix Na	Mix Cl	NaCl	E-Rinse	RO/DI
Date/Time	μS/cm	μS/cm	μS/cm	μS/cm	μS/cm	μS/cm	mS/cm	mS/cm	mS/cm	mS/cm	μS/cm
4/5/2013 10:25	7.37	6.79	6.8	6.93	6.69	6.9	7.92	4.13	6.1	11.1	6.17
4/8/2013 16:40	7.46	6.76	6.71	6.88	6.62	6.39	7.88	3.62	4.31	11.43	5.74
4/9/2013 10:33	7.5	6.8	6.91	7.04	6.52	6.37	7.89	2.54	4.22	11.74	6.05
4/9/2013 16:37	7.53	6.92	6.93	6.96	6.56	6.27	7.5	2.71	3.92	12.2	6.51
4/10/2013 11:17	7.49	6.77	6.78	6.83	6.6	6.27	7.77	2.32	3.6	12.07	6.53
4/10/2013 13:30	7.39	6.74	6.83	6.82	6.49	6.27	8.03	2.58	3.82	12.12	6.01
4/10/2013 16:30	7.43	6.87	6.75	6.93	6.79	6.33	8	2.39	3.95	12.08	6.31
4/11/2013 12:05	7.5	6.77	6.75	6.87	6.57	6.37	8.11	2.2	3.5	11.79	5.68
4/11/2013 14:47	7.45	6.68	6.85	7	6.53	6.33	7.74	2.23	3.54	11.73	5.87
4/11/2013 16:22	7.43	6.89	6.91	6.92	6.54	6.35	7.83	2.2	3.46	11.63	5.71
4/17/2013 10:57	7.5	6.96	7.02	6.86	6.8	6.43	8.18	2.7	5.85	11.5	7.03
4/17/2013 14:35	7.54	6.64	6.73	6.98	6.41	6.39	7.91	2.76	4.04	11.55	6.66
4/17/2013 16:34	7.49	6.96	6.96	7.07	6.5	6.33	7.85	2.53	4.58	11.7	6.83
4/18/2013 10:31	7.49	7.01	6.95	6.98	6.37	6.36	8.13	2.52	3.83	11.42	6.84
4/18/2013 13:00	7.49	6.96	6.6	6.99	6.47	6.33	7.95	2.63	3.85	11.8	6.36
4/18/2013 14:11	7.08	6.27	6.28	6.35	6.45	6.21	7.48	2.34	3.73	12.15	5.77
4/18/2013 16:21	7.44	6.69	6.9	6.78	6.85	6.56	7.81	2.7	3.89	11.74	6.23
4/19/2013 10:52	7.48	6.66	6.66	6.77	6.31	6.1	8.08	2.28	3.44	12	5.94
4/19/2013 17:35	7.07	6.24	6.2	6.34	6.16	5.94	7.62	2.26	4.12		6.43
4/19/2013 21:30	7.07	6.26	6.24	6.36	6.16	5.94	7.59	2.25	3.8	12.33	5.64
4/22/2013 11:06	7.08	6.54	6.54	6.5	6.58	6.46	7.89	2.04	3.47	12.32	6.41
4/22/2013 13:40	7.13	6.38	6.42	6.27	6.64	6.31	7.51	2.15	3.5	12.33	6.02
4/22/2013 16:23	7.14	6.56	6.56	6.58	6.73	6.33	7.53	2.12	3.57	12.34	5.78
4/22/2013 21:20	7.09	6.47	6.62	6.51	6.42	6.28	7.6	2.02	3.49	12.46	6.56

	Source	NF Feed	NF Perm	NF Conc	EDM Feed	EDM Diluate	Mix Na	Mix Cl	NaCl	E-Rinse	RO/DI
Date/Time	μS/cm	μS/cm	μS/cm	μS/cm	μS/cm	μS/cm	mS/cm	mS/cm	mS/cm	mS/cm	μS/cm
4/23/2013 8:07	7.07	6.26	6.23	6.41	5.53	5.27	7.11	2.11	3.4	12.24	6.22
4/23/2013 11:10	7.26	6.26	6.27	6.32	6.07	5.89	6.85	1.98	3.52	12.33	5.64
4/23/2013 15:31	7.25	6.26	6.25	6.3	6.09	5.88	7.01	1.99	4.1	12.37	5.81
4/24/2013 8:23	7.43	6.38	6.28	6.42	6.19	5.97	7.58	2.08	3.78	12.54	6.01
4/25/2013 21:37	7.32	6.42	6.42	6.24							
4/26/2013 8:39	7.38	6.38	6.31	6.25	6.15	5.73	7.94	2.2	3.73	10.2	5.99
4/30/2013 16:16	7.27	6.22	6.22	6.2	5.84	5.64	8.12	2.03	3.6	11.51	5.7
5/1/2013 9:13	7.23	6.5	6.15	6.16	5.71	5.5	8.01	2.05	3.47	10.43	6.02
5/1/2013 13:40	7.36	6.65	6.68	6.64	6.24	6.01	7.97	2.06	3.52	10.17	5.66
5/1/2013 16:50	7.36	6.77	6.56	6.51	6.12	5.99	8.03	2.04	3.49	10.27	5.74
5/1/2013 21:30	7.41	6.72	6.67	6.55	6.16	5.95	8	1.99	3.44	10.59	6.24
5/2/2013 14:54	7.42	6.72	6.71	6.64	6.24	5.99	8.1	2.01	3.57	10.81	5.73
5/29/2013 13:59	7.65	6.12	6.03	6.05	5.42	5.3	7.91	2	3.15	9.79	7.11
5/31/2013 13:50	7.06	6.08	5.97	5.82	5.64	5.48	7.41	2.1	3.5	9.7	5.81
5/31/2013 15:25	7.39	5.86	6.36	5.75	6.02	5.78	7.52	2.14	3.5	9.55	5.93
6/3/2013 11:42	7.43	6.05	6.15	6.04	5.6	5.53	8.47	2.16	3.2	9.84	6.06
6/3/2013 14:53	7.46	6.1	6.04	6	5.67	5.6	6.99	2.17	3.55	9.68	5.56
6/4/2013 11:16	7.5	6.1	6.06	6.1	5.58	5.36	7.17	2.05	3.19	9.74	6.15
6/4/2013 14:55	7.51	6.02	6.25	6.02	5.62	5.72	6.79	1.98	3.25	9.74	5.69
6/4/2013 16:55	7.54	6.12	6.07	5.94	5.62	5.42	6.62	1.78	3.15	9.93	6.15
6/14/2013 15:45	7.54	6.44	6.5	6.52	6.07	5.92	7.7	3.85	5.95	11.04	5.58
6/17/2013 14:03	7.54	6.31	6.41	6.48	5.98	5.86	7.36	5.3	6.03	10.79	5.33
6/18/2013 11:02	7.48	6.33	6.46	6.53	6.14	6.11	7.56	5.52	6.03	10.75	5.39
6/18/2013 14:24	7.5	6.29	6.34	6.38	6.06	5.82	7.38	5.28	5.94	10.69	5.6
6/19/2013 10:50	7.56	6.26	6.23	6.42	6.01	6.29	6.38	5.27	5.84	10.7	6.05

	Source	NF Feed	NF Perm	NF Conc	EDM Feed	EDM Diluate	Mix Na	Mix Cl	NaCl	E-Rinse	RO/DI
Date/Time	μS/cm	μS/cm	μS/cm	μS/cm	μS/cm	μS/cm	mS/cm	mS/cm	mS/cm	mS/cm	μS/cm
6/19/2013 14:32	6.99	6.21	6.35	6.28	5.97	5.7	7.26	5.211	5.84	10.3	6.1
6/20/2013 10:35	7.46	6.21	6.29	6.28	5.9	5.82	6.99	5.23	5.88	10.68	6.02
6/20/2013 14:55	7.48	6.31	6.63	6.3	6.03	5.85	7.66	5.24	5.89	10.68	6.03
6/21/2013 11:10	7.54	6.41	6.32	6.49	6.04	5.94	6.01	5.07	5.98	10.79	5.8
6/21/2013 15:09	7.55	6.25	6.258	6.33	5.91	5.67	7.87	5.16	5.99	10.71	5.89
6/24/2013 13:34	6.59	6.33	6.29	6.3	6.07	5.82	7.44	5.65	6.02	10.52	6.08
6/24/2013 16:35	7.59	6.258	6.32	6.36	6.07	5.81	7.33	5.04	5.89	10.6	6.64
6/25/2013 8:20	7.58	6.31	6.33	6.37	6.1	5.87	7.39	5.12	5.96	10.477	5.63
6/26/2013 14:30	7.37	6.02	6.13	6.25	5.9	5.71	7.02	3.13	5.1	10.47	6.08
6/26/2013 16:23	7.47	6.17	6.18	6.18	6.03	5.85	7.19	3.66	5.49	10.77	6.35
6/27/2013 8:31	7.53	6.17	6.47	6.35	6.11	5.85	7.42	5.42	5.586	10.75	6.64
6/27/2013 16:59	7.59	6.2	6.18	6.1	5.93	5.69	6.82	5.058	5.91	10.54	6.14
6/28/2013 7:58	7.63	6.5	6.41	6.41	6.28	6.04	7.53	5.3	6.05	10.7	
6/28/2013 15:37	7.6	6.49	6.38	6.37	6.13	5.9	7.83	5.51	6.32	9.75	5.92
7/8/2013 11:51	7.58	6.2	6.36	6.31	5.96	5.87	6.7	2.59	3.81	7.98	6.13
7/8/2013 15:57	7.62	6.51	6.38	6.39	6.15	5.92	7.38	3.08	5.38	9.89	6.92
7/9/2013 8:34	7.65	6.36	6.44	6.6	6.22	5.99	7.36	5.1	5.87	10.28	6
7/9/2013 15:34	7.6	6.42	6.33	6.49	6.16	5.95	7.478	4.83	6.01	10.05	6.32
7/10/2013 8:22	7.65	6.44	6.37	6.53	6.18	6.01	7.64	4.91	6.08	9.14	6.33
7/10/2013 15:57	7.54	6.52	6.44	6.43	6.21	5.92	7.97	4.91	6.21	10.34	
7/11/2013 8:45	7.64	6.44	6.39	6.44	6.09	5.74	7.98	4.17	6.16	8.87	6.03
7/12/2013 9:30	7.28	6.29	6.21	6.3	6.25	5.92	7.69	4.79	6.15	9.95	6.46
7/12/2013 17:10	7.57	6.4	6.37	6.56	6.25	6.06	7.74	5.02	6.25	8.18	
7/13/2013 11:30	7.46	6.43	6.58	6.53	6.06	5.88	7.95	5.31	6.24	6.86	5.96
7/13/2013 16:40	7.42	6.37	6.39	6.53	6.13	5.91	8.22	5.48	6.4	6.82	6.07

Table A2-1. Grab Samples - Temperatures

	Source	NF Feed	NF Perm	NF Conc	EDM Feed	EDM Diluate	Mix Na	Mix Cl	NaCl	E-Rinse	RO/DI
4/5/2013 10:25	19.1	20.8	20.9	21	24	25.8	25.5	25.7	26	25.5	19.8
4/8/2013 16:40	20.9	22.8	23.2	23.4	26.2	27.8	28.1	28.1	27.8	27.9	23.7
4/9/2013 10:33	18.9	20.8	20.5	20.5	23.8	25.2	25.7	25	25.4	24.4	20.3
4/9/2013 16:37	19.4	20.9	21.2	21.3	24.3	26	26.2	26.1	26	25	21.3
4/10/2013 11:17	19.2	20.7	20.9	21	24.1	26	25.6	26.1	26	25.4	21.7
4/10/2013 13:30	19.2	20.8	21.2	20.9	24.3	25.9	26.2	26	26.1	25.5	21.5
4/10/2013 16:30	19	20.6	20.9	20.9	24.2	26.1	25.5	25	26.1	26.1	20.6
4/11/2013 12:05	18.5	20.2	20.2	20.4	24.3	25.4	25.4	25.4	25.3	25	19.2
4/11/2013 14:47	18.8	20.6	20.7	21.3	24.9	25.9	26.5	26.4	26.3	26.9	21.6
4/11/2013 16:22	19.2	20.6	20.8	20.9	24.8	25.5	26.5	26.9	26.8	27.7	21.4
4/17/2013 10:57	20.2	21.9	22.6	22.4	25.4	25.9	26	26.2	26.4	26.2	23
4/17/2013 14:35	21.6	22.7	22.5	23.5	25.6	27.3	27.5	27.5	27.5	27.8	22.7
4/17/2013 16:34	21.4	22.4	22.5	22.7	26	27.1	27.3	27.3	27	27.3	23.2
4/18/2013 10:31	18.6	20	20.7	20.3	23.1	23.6	24.6	24.5	24.6	24	20.5
4/18/2013 13:00	19	20.4	20.5	20.9	24.3	25.2	25.1	25.6	25.6	25.3	20.6
4/18/2013 14:11	18.6	20.7	20.5	21.4	24.7	25.8	25.7	25.7	25.5	25.2	21.1
4/18/2013 16:21	19.1	20.8	20.8	21.1	25	26	25.1	26	26	26.3	21.6
4/19/2013 10:52	18.2	20	20	20.4	23.1	24.4	24.7	24.2	24.5	24.4	20.6
4/19/2013 17:35	19.3	20.7	20.9	20.7	24.9	25.9	26.2	26.3	26.3	26.4	20.7
4/19/2013 21:30	18.3	20.1	20.3	20.3	24.3	25.6	25.9	25.9	25.7	25.4	20.3
4/22/2013 11:06	20.9	21.5	21.6	21.9	25.3	26.2	25.9	25.7	26.2	27	23.2
4/22/2013 13:40	21.1	22	22	22.7	26	27.1	26.8	26.8	27	27.5	22.9
4/22/2013 16:23	21.3	22.3	22.1	22.4	26.4	27.3	27.6	27.5	27.4	28	22.2
4/22/2013 21:20	19.5	21.4	21.3	21.3	25.9	26.6	26.7	26.8	26.6	26.8	22.7
4/23/2013 8:07	19.4	21.1	21.4	21.2	24.7	26.1	26.2	26.7	26.6	26.7	22.5

	Source	NF Feed	NF Perm	NF Conc	EDM Feed	EDM Diluate	Mix Na	Mix Cl	NaCl	E-Rinse	RO/DI
4/23/2013 11:10	20.2	22.3	22.3	22.3	26	26.9	27.6	27.4	27.32	26.9	22.6
4/23/2013 15:31	20.6	22.3	22.5	23.5	27	28	28.2	28.2	28.1	28.3	23.6
4/24/2013 8:23	18.1	20.8	21.2	21	25.2	26.3	26.3	26.4	26.1	25.4	21.4
4/25/2013 9:37	20.1	21.7	21.8	21.8							
4/26/2013 8:39	19.4	20.9	21	20.8	24	24.7	25.1	24.7	24.7	23.8	20.9
4/30/2013 16:16	23.2	24.4	24.1	25.1	28.3	29.2	29.5	29.4	29	29.1	25.3
5/1/2013 9:13	21.5	22.8	22.9	23	26.2	27.2	27.1	27.1	27.4	27	23.3
5/1/2013 13:40	23.4	24	24	25	27.9	29	29	28.9	28.8	29.2	24.7
5/1/2013 16:50	23.4	24.7	25.1	25.7	29.2	30.3	30.3	30.2	30.1	30.4	26.3
5/1/2013 21:30	22.2	23.5	23.5	23.7	27.7	28.7	28.4	27.9	27.8	27.4	24.2
5/2/2013 14:54	21.4	22.9	23	23.1	26.8	27.8	27.7	27.5	27.4	27.4	23.8
5/29/2013 13:59	27.1	28	28.2	28.2	30.8	32.5	32.6	32.7	32.6	31.5	26.2
5/31/2013 13:50	23.6	25.2	25.2	25.2	26.3	29.4	29.6	29.7	29.7	30	25.7
5/31/2013 15:25	24.9	25.4	25.5	26.3	29.1	29.6	30.1	30.1	30	29.9	24.8
6/3/2013 11:42	23.8	25.7	25.6	25.7	29.4	30.4	30.4	30.2	30.3	30.4	25.5
6/3/2013 14:53	24.6	26.2	26.1	26.3	30.6	31.4	31.1	31.3	31.5	32.1	27.1
6/4/2013 11:16	24.3	26.3	25.8	25.9	29.6	30.6	30.1	30.5	30	30.1	26.3
6/4/2013 14:55	24.5	26.8	26.3	27.2	30.7	31.7	31.7	31	31.8	32.3	27.1
6/4/2013 16:55	25.7	26.3	26.4	26.6	31.1	32	32	31.9	32.2	32.7	26.5
6/14/2013 15:45	23.8	26.3	26.5	26.7	30.7	31.8	31.1	29.9	31.6	30.7	27.4
6/17/2013 14:03	25.2	27.3	27	27.9	30.5	31.5	31.5	31.4	31.2	30.7	28.3
6/18/2013 11:02	24.7	26.7	26.7	27.3	29.7	30.8	30.1	30.6	30.5	30	27.5
6/18/2013 14:24	25.4	27.3	27.4	28	30.8	31.8	31.4	31	31.7	31.5	28.8
6/19/2013 10:50	24.8	26.2	26.9	26.7	30	30.9	30.9	31	30.9	30.8	28
6/19/2013 14:32	25.6	27.4	27.5	28.1	30.9	31.7	31.7	31.6	31.7	31.1	29.1
6/20/2013 10:35	25.1	27	26.8	27.4	30.2	31	31.1	31.1	31	30.3	28.3

	Source	NF Feed	NF Perm	NF Conc	EDM Feed	EDM Diluate	Mix Na	Mix Cl	NaCl	E-Rinse	RO/DI
6/20/2013 14:55	26	27.5	27.5	28.6	31.2	32	31.6	31.9	32.2	32	29
6/21/2013 11:10	25.1	26.7	26.9	27	30.1	30.8	30.9	30.4	31	30.8	28.1
6/21/2013 15:09	26.4	27.4	27.4	27.6	31.4	32.1	32.2	32.1	32.4	32	28.3
6/24/2013 13:34	26.8	28.2	28.5	28.6	32	32.8	32.5	32.4	32.6	31.8	28.9
6/24/2013 16:35	26.8	27.7	27.7	27.7	31.3	32.1	32.1	32	32.3	31.5	29.2
6/25/2013 8:20	24.6	26	26.3	26.4	29.7	30.6	30.3	30.1	30.4	29.1	26.8
6/26/2013 14:30	24.5	26	26.1	26.2	29.4	30.4	30.2	30.2	30.4	30.4	25.8
6/26/2013 16:23	24.5	26.3	26.5	27.3	30	30.7	30.9	30.8	30.9	30.9	26.2
6/27/2013 8:31	22.7	24.4	24.7	24.3	27.6	28.6	28.4	28.4	28.5	27.6	25.5
6/27/2013 16:59	28.1	27.6	27.5	28.3	31.2	32.1	32.1	32	32	32.2	29.1
6/28/2013 7:58	24.4	26.4	26.5	26.2	29.6	30.8	30.4	30.6	30.5	29.4	
6/28/2013 15:37	27	27.9	27.7	27.9	31.4	32.3	32.4	32.4	32.5	32.3	28.6
7/8/2013 11:51	26	27.8	28.2	28.5	31.4	32.3	32.3	32.2	32.2	31	27.4
7/8/2013 15:57	26.9	27.7	27.9	28.6	31.6	32.5	32.5	32.5	32.5	32	29.3
7/9/2013 8:34	24.4	26.5	26.5	26.6	29.9	31.1	30.7	29.3	30.8	29.5	27.4
7/9/2013 15:34	26.4	27.6	27.6	27.7	31.5	32.4	32.4	32.3	32.5	32.4	28.9
7/10/2013 8:22	24.9	26.6	26.8	26.9	26.3	30.9	31	30.9	31.2	29.8	27.7
7/10/2013 15:57	25.7	27.4	27.4	27.6	31.7	32.6	32.4	32.6	32.8	32.4	
7/11/2013 8:45	25.2	27.1	27.1	27.3	30.8	31.5	31.3	31.1	31.6	30.2	28
7/12/2013 9:30	24.7	26.6	26.8	26.7	30.6	31.6	31.6	31.1	31.3	30.3	27.9
7/12/2013 17:10	25.6	27.1	27.3	27.9	30.8	32.4	32.5	32.5	32.6	32.3	
7/13/2013 11:30	25.2	26.9	27.1	27.6	30.1	31.1	31.3	31.9	31.5	31.4	27.2
7/13/2013 16:40	24.9	27.1	27.3	27.3	30.7	32.1	32.5	32.6	32.6	32.3	29

APPENDIX A3
Brighton,
Colorado

Appendix A3-1 Brighton, Colorado Grab Sample Data

Date	Time	RO perm pH	NF perm pH	NF conc pH	RO conc pH	NF/RO conc pH	EDM-feed pH	
10/1/13	12:51		5.8	5.11	4.64	7.32	5.46	5.35
10/2/13	13:30		5.02	4.91	4.43	7.2	5.36	5.11
10/3/13	8:45		6	4.88	4.18	7.32	5.29	4.96
10/3/13	19:45						5.59	5.38
10/4/13	13:45		5.76	5.03	4.47	7.51	5.42	5.14
10/5/13	12:30		5.66	5.16	4.65	7.41	5.56	5.24
10/6/13	13:30		5.87	5.12	4.66	7.5	5.49	5.18
10/7/13	14:00		5.66	4.58	3.94	7.47	5.17	4.77
10/8/13	14:00		5.61	5.07	4.4	7.49	5.53	5.23
10/9/13	14:40		5.86	4.63	3.86	7.55	5.33	4.85
10/10/13	20:30		5.6	4.9	4.26	7.45	5.46	5.1
10/11/13	14:00		6.07	4.96	4.19	7.51	5.5	5.14
10/12/13	14:15		5.71	5.02	4.36	7.57	5.56	5.2
10/15/13	16:00		5.58	5.17	4.89	7.46	5.53	5.25
10/15/13	20:00				5	7.49	5.53	5.36
10/16/13	13:30		5.5	5.04	4.54	7.54	5.55	5.31
10/17/13	16:00		5.51	4.95	4.32	7.39	5.43	5.1
10/18/13	14:00		5.9	4.94	4.28	7.52	5.48	5.1
10/22/13	14:15		5.71	5.26	4.97	7.48	5.64	5.46
10/23/13	16:30		5.65	5.01	4.49	7.54	5.63	5.36
10/24/13	16:20		5.48	4.94	4.32	7.53	5.54	5.17
10/25/13	16:15		5.48	4.86	4.18	7.52	5.44	5.09

Date	Time	RO perm µS/cm	NF perm µS/cm	NF conc µS/cm	RO conc µS/cm	NF/RO conc µS/cm	EDM-feed µS/cm
10/1/13	12:51	56.5	784.1	4495	5269	4916	3353
10/2/13	13:30	52.94	757.7	4435	5186	4830	3294
10/3/13	8:45	52.71	728.9	4441	4970	4690	3246
10/3/13	19:45					4805	3298
10/4/13	13:45	59.1	775.6	4644	5225	4947	3409
10/5/13	12:30	71.6	882.4	5281	5498	5372	3825
10/6/13	13:30	64.07	818.9	5246	5232	5174	3690
10/7/13	14:00	62	788.2	4979	4972	4902	3511
10/8/13	14:00	68	796.3	5240	4969	5042	3603
10/9/13	14:40	72	782.8	4883	4741	4770	3462
10/10/13	20:30	66.2	875.6	5646	4948	5359	3965
10/11/13	14:00	56.5	902	5780	5320	5505	4039
10/12/13	14:15	54.7	869.9	5793	5421	5543	4062
10/15/13	16:00	58.7	925.6	5050	5151	5127	3624
10/15/13	20:00			5100	5250	5144	3679
10/16/13	13:30	62	902.6	5423	5069	5262	3823
10/17/13	16:00	61	892.6	5646	4859	5204	3855
10/18/13	14:00	81	857.1	5559	4623	4985	3743
10/22/13	14:15	56	962.3	5105	5267	5151	3714
10/23/13	16:30	57	922	5502	5178	5291	3863
10/24/13	16:20	54	932.9	5786	5230	5419	4007
10/25/13	16:15	52	925.5	5873	5236	5479	4095

Appendix A3-1 Brighton, Colorado Grab Sample Data

Date	Time	EDM-diluate pH	E-Rinse pH	NaCl pH	M. Na pH	M. Cl pH	Plant raw pH
10/1/13	12:51	4.99		10.55	5.85	7.19	4.61
10/2/13	13:30	4.75		8.83	5.48	7.03	2.76
10/3/13	8:45	4.61		9.19	5.58	7.05	2.61
10/3/13	19:45	5.12				7.23	
10/4/13	13:45	4.85		10.34	5.65	7.2	2.94
10/5/13	12:30	4.99		10.98	5.62	7.27	3.09
10/6/13	13:30	4.89		10.3	5.49	7.36	2.95
10/7/13	14:00	4.36		8.46	3.49	6.99	2.53
10/8/13	14:00	4.85		8.23	5.79	7.6	2.95
10/9/13	14:40	4.25		8.51	3.49	7.63	2.21
10/10/13	20:30	4.7		8.14	3.97	7.84	2.6
10/11/13	14:00	4.71		8.18	4.16	7.95	2.66
10/12/13	14:15	4.75		9.11	4.8	8.3	2.75
10/15/13	16:00	5.09		8.68	5.08	7.2	3.38
10/15/13	20:00						
10/16/13	13:30	4.99		8.6	5.85	7.5	4.21
10/17/13	16:00	4.75		8.67	5.08	7.75	2.73
10/18/13	14:00	4.71		9.3	3.91	7.75	2.78
10/22/13	14:15	5.28		10.42	5.73	7.23	4.9
10/23/13	16:30	5.08		9.59	5.87	7.84	3.73
10/24/13	16:20	4.73		9.75	4.58	8.14	2.67
10/25/13	16:15	4.61		9.78	3.77	8.02	2.71

Date	Time	EDM-diluate μS/cm	E-Rinse mS/cm	NaCl mS/cm	M. Na mS/cm	M. Cl mS/cm	Plant raw μS/cm
10/1/13	12:51	2639		28.08	46.82	47.53	125.5
10/2/13	13:30	2581		34.22	48.03	52.42	125.7
10/3/13	8:45	2588		30.67	48.17	56.37	117.7
10/3/13	19:45	2603				59.82	
10/4/13	13:45	2753		28.27	45.8	61.4	118.3
10/5/13	12:30	3147		28.57	48.9	58.57	124.4
10/6/13	13:30	3024		30.01	48.54	60.61	118.9
10/7/13	14:00	2885		32.53	48.38	62.4	122.7
10/8/13	14:00	2998		43.51	48.11	60.11	115.7
10/9/13	14:40	2884		31.43	46.61	60.29	119
10/10/13	20:30	3338		34.56	46.7	55.7	119.1
10/11/13	14:00	3424		33.1	46.7	58	116.6
10/12/13	14:15	3416		30.44	52.1	57.71	102.8
10/15/13	16:00	2924		37.19	47.07	60.24	115.8
10/15/13	20:00						
10/16/13	13:30	3177		41.38	47.53	66.33	119.6
10/17/13	16:00	3269		37.54	46.77	63.28	113.5
10/18/13	14:00	3212		22.52	46.71	61.45	114.4
10/22/13	14:15	3038		20.99	45.76	60.08	113.2
10/23/13	16:30	3203		23.73	46.26	61.23	116.3
10/24/13	16:20	3391		23.3	48.69	65.68	119.1
10/25/13	16:15	3485		21.14	50.21	54.97	97.24

Appendix A3-1 Brighton, Colorado Grab Sample Data

Date	Time	Total Alkalinity (mg/L CaCO ₃)		
		RO conc	NF/RO conc	M. Na
10/1/13	12:51	1300	134	2260
10/2/13	13:30		145	2220
10/3/13	8:45	1120	97	2280
10/3/13	19:45		185	184
10/4/13	13:45	1190	118	2560
10/5/13	12:30	1190	154	2700
10/6/13	13:30	1200	130	2530
10/7/13	14:00	1120	69	2000
10/8/13	14:00	1090	131	
10/9/13	14:40	1440	81	2860
10/10/13	20:30	1150	128	2520
10/11/13	14:00	1250	142	3360
10/12/13	14:15	1340	158	3690
10/13/13	16:00	1180	162	2920
10/15/13	20:00		159	
10/16/13	13:30	1110	120	3280
10/17/13	16:00	1120	139	2900
10/18/13	14:00	1000	110	2800
10/22/13	14:15	1180	192	2400
10/23/13	16:30	1200	184	3380
10/24/13	16:20	1080	152	3540
10/25/13	16:15	1100	129	2760

Appendix A3-1 Brighton, Colorado Grab Sample Data

Date		Specific Gravity		
Date	Time	M. Na	M. Cl	
10/1/13	10/1/13	12:51	1.032	1.074
10/2/13	10/2/13	13:30	1.038	1.076
10/3/13	10/3/13	8:45	1.042	1.068
10/3/13	10/3/13	19:45	1.046	
10/4/13	10/4/13	13:45	1.048	1.07
10/5/13	10/5/13	12:30	1.044	1.076
10/6/13	10/6/13	13:30	1.046	1.07
10/7/13	10/7/13	14:00	1.048	1.07
10/8/13	10/8/13	14:00	1.046	1.066
10/9/13	10/9/13	14:40	1.046	1.066
10/10/13	10/10/13	20:30	1.04	1.07
10/11/13	10/11/13	14:00	1.044	1.066
10/12/13	10/12/13	14:15	1.044	1.046
10/13/13	10/13/13	16:00	1.046	1.066
10/15/13	10/16/13	13:30	1.052	1.07
10/16/13	10/17/13	16:00	1.048	1.066
10/17/13	10/18/13	14:00	1.046	1.064
10/18/13	10/22/13	14:15	1.044	1.066
10/22/13	10/23/13	16:30	1.048	1.068
10/23/13	10/24/13	16:20	1.05	1.068
10/24/13	10/25/13	16:15	1.04	1.052
10/25/13				

Appendix A3-1 Brighton, Colorado Grab Sample Data

Date	Time	Silica (mg/L)					
		NF perm	NF conc	RO conc	NF/RO conc	EDM-feed	EDM-diluate
10/1/13	12:51	93.8	127.6	102.8	112	110.8	117.8
10/2/13	13:30	98.2	115.4	100.2	106.8	110	112
10/3/13	8:45	96.8	117.4	89.4	102.6	103.4	97.4
10/3/13	19:45				107.4	111.4	116.6
10/4/13	13:45				105.6	101.6	105.6
10/5/13	12:30	106.8	120	96.4	109.8	100.4	114
10/6/13	13:30		109.8			106.4	99.4
10/7/13	14:00		99.2		89.2	86.2	98.8
10/8/13	14:00	56.2	106.8	84.8	101.16	107.2	101.4
10/9/13	14:40						
10/10/13	20:30						
10/11/13	14:00	98.8	100.2	91.4	106.8	100.6	101.6
10/12/13	14:15		109.8	96.4		103.2	
10/13/13	7:25		123.6	92.6		105.6	
10/15/13	20:00		129.8	96		120	
10/15/13	22:00		128	81.2		113.6	
10/16/13	8:20	98.4	120.6	99.4		110	
10/16/13	13:30	84.8	114.4			99.8	108.6
10/17/13	16:00		122.2	62	105		
10/18/13	14:00		103.6	82.4	91.8		
10/22/13	14:15					114.2	
10/23/13	16:30					105	
10/24/13	16:20		127.8			106.2	
10/25/13	16:15		105.2			108	

Appendix A3-2 Brighton Log Sheet Data
 NF and EDM Stream Conductivity

Date	Time	NF feed μS/cm	NF perm μS/cm	NF conc μS/cm	NF/RO conc μS/cm	EDM-feed μS/cm	EDM-diluate μS/cm
10/1/13	12:51	2688	733	4571	4899	3424.22	2612
10/2/13	13:15	2632	707	4521	4854	3372	2555
10/2/13	17:50	2601	694	4454	4820	3336	2522
10/2/13	20:55	2650	709	4508	4841	3387	2607
10/3/13	8:24	2637	662	4525	4696	3315	2554
10/3/13	18:30	2646	693	4531	4927	3402	2562
10/4/13	11:50	2740	709	4691	4966	3504	2682
10/4/13	16:50	2695	675	4669	4772	3419	2675
10/4/13	19:40	2781	696	4772	4803	3477	2740
10/5/13	8:25	3134	782	5211	5358	3956	3156
10/5/13	12:15	3202	801	5382	5387	3939	3119
10/5/13	17:45	3149	785	5312	5391	3940	3106
10/6/13	13:15	3072	748	5298	5196	3791	2987
10/6/13	15:30	2947	724	5136	4965	3635	2845
10/6/13	20:05	2942	736	5060	4938	3627	2918
10/7/13	11:05	2952	742	5055	4919	3630	2902
10/7/13	13:45	2929	729	5042	4883	3589	2853
10/7/13	16:30	2956	752	5047	5038	3670	2918
10/7/13	20:15	3077	756	5343	5120	3787	3023
10/8/13	7:30	3140	765	5369	5205	3919	3140
10/8/13	10:30	3185	769	5562	5240	3900	3095
10/8/13	13:35	3082	738	5455	5185	3793	3019
10/8/13	16:30	3051	773	5415	5162	3774	2971
10/8/13	20:00	3109	744	5437	5199	3838	3095
10/9/13	8:50	3207	756	5553	5242	3906	3118
10/9/13	14:30	2970	738	5195	4866	3592	2874
10/9/13	17:20	2938	716	5038		3610	2904
10/10/13	8:30	2956	703	5046	4866	3665	3001
10/10/13	13:45	3270	774	5715	5590	4208	3393
10/10/13	17:50	3284	790	5738	5422	4032	3224
10/10/13	20:15	3410	808	5890	5491	4141	3361
10/11/13	8:30	3391	813	5859	5449	4113	3993
10/11/13	1:50	3473	826	5976	5585	4202	3413
10/11/13	17:40	3468	814	5958	5559	4187	3386
10/11/13	20:30	3518	822	5958	5583	4234	3453
10/12/13	10:30	3567	833	6043	5679	4282	3477
10/12/13	13:45	3513	798	5976	5667	4250	3437
10/12/13	17:00	3504	810	5971	5651	4216	3415
10/12/13	19:20	3504	793	6088	5710	4288	3455
10/13/13	7:25	3445	843	5993	5284	4014	3304
10/15/13	15:15	3045	877	5234	5229	3798	3012
10/15/13	20:00	3027	823	5221	5240	3804	3030
10/15/13	22:05	3126	893	5346	5304	3890	3082
10/16/13	6:45	3278	866	5534	5390	4014	3248
10/16/13	8:10	3323	789	5628	5440	4092	3313
10/16/13	13:10	3288	880	5656	5385	3982	3170
10/16/13	17:10	3244	847	5685	5429	4002	3211

Appendix A3-2 Brighton Log Sheet Data
 NF and EDM Stream Conductivity

Date	Time	Condition	M. Na mS/cm	M. Cl mS/cm	NaCl mS/cm	E-Rinse mS/cm
10/1/13	12:51	M.Na-45mS	43.4	115.42	49.4	31.8
10/2/13	13:15	M.Na-50 mS	47.81	115.25	51.3	37.1
10/2/13	17:50	M.Na-50 mS	52.29	117.32	49.7	31.3
10/2/13	20:55	M.Na-55 mS	53.32	116.71	50.3	32.2
10/3/13	8:24	M.Na-55mS	52.76	117.05	51	34.5
10/3/13	18:30	M.Na-60 mS	58.81	119.33	48.7	32.2
10/4/13	11:50	M.Na-60 mS	59.25	119.06	50	32.4
10/4/13	16:50	M.Na-60 mS	58.88	114.7	48.1	33.8
10/4/13	19:40	M.Na-60 mS	50.84	94.64	51.1	34.1
10/5/13	8:25	M.Na-60 mS	58.75	105.64	50.2	31.2
10/5/13	12:15	M.Na-60 mS	57.72	117.21	50.3	32.3
10/5/13	17:45	M.Na-60 mS	58.09	114.98	50.9	35.7
10/6/13	13:15	M.Na-60 mS	58.28	114.7	50.1	31.9
10/6/13	15:30	M.Na-60 mS	58.46	116.57	50	32.3
10/6/13	20:05	M.Na-60 mS	57.61	115.16	49.8	33.7
10/7/13	11:05	M.Na-60 mS	60.3	113.58	50.6	37
10/7/13	13:45	M.Na-60 mS	59.3	116.75	50	32
10/7/13	16:30	M.Na-60 mS	59.32	115.94	47.5	32.7
10/7/13	20:15	M.Na-60 mS	59.72	117.86	50.3	33.9
10/8/13	7:30	M.Na-60 mS	57.51	115.92	51.3	36.4
10/8/13	10:30	M.Na-60 mS	57.29	118.03	49.8	38.4
10/8/13	13:35	M.Na-60 mS	56.67	119.08	48.5	39.3
10/8/13	16:30	M.Na-60 mS	58.27	114.42	49.3	31.6
10/8/13	20:00	M.Na-60 mS	59.19	114.72	49	33
10/9/13	8:50	M.Na-60 mS	57.8	118.52	49.8	36.6
10/9/13	14:30	M.Na-60 mS	60.72	112.92	50.8	30.8
10/9/13	17:20	M.Na-60 mS	57.7	117.59	48.7	31.3
10/10/13	8:30	M.Na-60 mS	58.96	117.59	50.3	37.9
10/10/13	13:45	M.Na-60 mS	58.57	114.64	51.2	36.2
10/10/13	17:50	M.Na-60 mS	5748	115.78	50.5	33.9
10/10/13	20:15	M.Na-60 mS	5724	115.88	51.8	34.5
10/11/13	8:30	M.Na-60 mS	58.51	113.42	50.3	31.6
10/11/13	1:50	M.Na-60 mS	57.89	119.68	51.5	33
10/11/13	17:40	M.Na-60 mS	59.73	114.93	52.2	33.8
10/11/13	20:30	M.Na-60 mS	60.73	114.17	51.6	30.8
10/12/13	10:30	M.Na-60 mS, M.Cl-100 mS	58.44	96.09	53.5	32.1
10/12/13	13:45	M.Na-60 mS, M.Cl-100 mS	58.99	96.98	55.6	31.9
10/12/13	17:00	M.Na-60 mS, M.Cl-100 mS	57.59	95.8	54.1	32.2
10/12/13	19:20	M.Na-60 mS, M.Cl-100 mS	57.72	98.31	55.6	32.9
10/13/13	7:25	M.Na-60 mS, M.Cl-100 mS	58.02	96.35	55.6	36.4
10/15/13	15:15	M.Na-65 mS, M.Cl-120 mS	63.29	116.31	50.9	36.6
10/15/13	20:00	M.Na-65 mS, M.Cl-120 mS	63.74	119.28	51.4	33.4
10/15/13	22:05	M.Na-65 mS, M.Cl-120 mS	63.38	114.24	51.2	33.8
10/16/13	6:45	M.Na-65 mS, M.Cl-120 mS	63.04	116.35	51.1	36.6
10/16/13	8:10	M.Na-65 mS, M.Cl-120 mS	62.13	116.48	51.5	36.9
10/16/13	13:10	M.Na-65 mS, M.Cl-120 mS	65.27	115.63	50	37
10/16/13	17:10	M.Na-65 mS, M.Cl-120 mS	66.1	119.63	49.5	31.5

Appendix A3-2 Brighton Log Sheet Data
 NF and EDM Stream Conductivity

Date	Time	NF feed μS/cm	NF perm μS/cm	NF conc μS/cm	NF/RO conc μS/cm	EDM-feed μS/cm	EDM-diluate μS/cm
10/17/13	8:00	3642	911	6172	5722	4364	3597
10/17/13	12:20	3687	919	6365	5733	4439	3628
10/17/13	14:30	3373	838	5957	5265	4056	3293
10/17/13	16:00	3351	826	5857	5258	4035	3294
10/17/13	17:30	3364	841	5868	5270	4050	3331
10/17/13	19:55	3351	841	5850	5260	4035	3291
10/18/13	7:15	3490	842	6015	5325	4162	3430
10/18/13	13:30	3404	865	5939	5162	4059	3331
10/18/13	15:25	3396	864	5929	5145	4050	3342
10/18/13	17:25	3400	853	5926	5168	4079	3364
10/18/13	20:20	3521	872	6078	5238	4185	3488
10/19/13	7:55	3615	932	6181	5265	4258	3550
10/22/13	13:05	3194	903	5383	5413	3993	3191
10/22/13	14:00	3189	916	5383	5411	3954	3140
10/22/13	16:00	3203	917	5401	5417	3994	3202
10/22/13	16:20	3225	915	5410	5409	3979	3169
10/22/13	20:25	3333	925	5562	5476	4063	3253
10/23/13	5:00	3508	972	6010	5682	4320	3529
10/23/13	9:05	3602	978	6155	5739	4345	3544
10/23/13	10:50	3579	983	6128	5716	4331	3543
10/23/13	13:40	3557	971	6106	5686	4297	3463
10/23/13	16:15	3374	921	5899	5622	4162	3336
10/23/13	18:10	3427	897	5940	5630	4188	3375
10/23/13	20:30	3427	871	5849	5591	4165	3381
10/24/13	8:20	3615	955	6244	5798	4407	3592
10/24/13	9:50	3633	912	6203	5772	4367	3583
10/24/13	15:30	3589	881	6191	5762	4330	3528
10/24/13	17:30	3670	950	6281	5797	4429	3655
10/25/13	10:00	3876	980	6529	5916	4574	3763
10/25/13	10:22	3242	851	5938	5597	4266	3508
10/25/13	14:10	3670	910	6263	5779	4397	3594
10/25/13	16:00	3657	852	6245	5799	4400	6310
10/25/13	17:50	3679	924	6280	5799	4410	3626
10/25/13	20:30	3746	908	6343	5826	4472	3719
10/26/13	10:40	3858	910	6477	5874	4560	3758

Appendix A3-2 Brighton Log Sheet Data
NF and EDM Stream Conductivity

Date	Time	Condition	M. Na mS/cm	M. Cl mS/cm	NaCl mS/cm	E-Rinse mS/cm
10/17/13	8:00	M.Na-65 mS, M.Cl-120 mS	62.26	117.13	49	35.9
10/17/13	12:20	M.Na-65 mS, M.Cl-120 mS	63.37	114.24	51	32.9
10/17/13	14:30	M.Na-65 mS, M.Cl-120 mS	64.7	118.21	51.2	33.4
10/17/13	16:00	M.Na-65 mS, M.Cl-120 mS	65.98	119.45	50.6	33.9
10/17/13	17:30	M.Na-65 mS, M.Cl-120 mS	65.87	114.39	52.4	31.8
10/17/13	19:55	M.Na-65 mS, M.Cl-120 mS	65.07	119.51	51.6	32.2
10/18/13	7:15	M.Na-65 mS, M.Cl-120 mS	64.15	120.4	48.8	31.1
10/18/13	13:30	M.Na-65 mS, M.Cl-120 mS	64.27	119.34	49.4	22
10/18/13	15:25	M.Na-65 mS, M.Cl-120 mS	62.21	115.53	50.6	23.1
10/18/13	17:25	M.Na-65 mS, M.Cl-120 mS	62.77	118.98	50.1	24
10/18/13	20:20	M.Na-65 mS, M.Cl-120 mS	62.37	113.52	50.3	25
10/19/13	7:55	M.Na-65 mS, M.Cl-120 mS	63.47	114.58	51.7	28
10/22/13	13:05	M.Na-65 mS, M.Cl-120 mS	62.96	113.94	49.2	23.9
10/22/13	14:00	M.Na-65 mS, M.Cl-120 mS	64.66	120.02	50.2	24
10/22/13	16:00	M.Na-65 mS, M.Cl-120 mS	61.12	118.77	50	24.3
10/22/13	16:20	M.Na-65 mS, M.Cl-120 mS	65.02	113.94	51.3	24.4
10/22/13	20:25	M.Na-65 mS, M.Cl-120 mS	62.61	119.54	50.5	24.4
10/23/13	5:00	M.Na-65 mS, M.Cl-120 mS	63.77	112.68	50.8	24.7
10/23/13	9:05	M.Na-65 mS, M.Cl-120 mS	64.75	119.82	50.5	22.9
10/23/13	10:50	M.Na-65 mS, M.Cl-120 mS	62	118.2	50	23.4
10/23/13	13:40	M.Na-65 mS, M.Cl-120 mS	62.5	116.43	49.7	24.5
10/23/13	16:15	M.Na-65 mS, M.Cl-120 mS	62.21	113.48	51.6	24.5
10/23/13	18:10	M.Na-65 mS, M.Cl-120 mS	64.41	114.41	51	23.9
10/23/13	20:30	M.Na-65 mS, M.Cl-120 mS	62.47	115.28	50.8	24.6
10/24/13	8:20	M.Na-65 mS, M.Cl-120 mS	63.31	114.27	51.3	22.7
10/24/13	9:50	M.Na-65 mS, M.Cl-120 mS	63.39	117.98	49	23.1
10/24/13	15:30	M.Na-65 mS, M.Cl-120 mS	63.01	114.84	50	23.9
10/24/13	17:30	M.Na-65 mS, M.Cl-120 mS	61.79	119.01	51.3	21.1
10/25/13	10:00	M.Na-55 mS, M.Cl-110 mS	53.13	106.57	51.1	25.4
10/25/13	10:22	M.Na-55 mS, M.Cl-110 mS	53.02	104.77	50.9	21.8
10/25/13	14:10	M.Na-55 mS, M.Cl-100 mS	52.38	94.19	50.5	23.6
10/25/13	16:00	M.Na-55 mS, M.Cl-100 mS	56.64	96.66	48.4	22.6
10/25/13	17:50	M.Na-55 mS, M.Cl-100 mS	53.41	99.62	49.8	23.4
10/25/13	20:30	M.Na-55 mS, M.Cl-100 mS	52.35	94.21	52	24.3
10/26/13	10:40	M.Na-55 mS, M.Cl-100 mS	55.01	96.46	49.7	25.3

Appendix A3-2 Brighton Log Sheet Data
NF and EDM Flows

Date	Time	NF perm	1st stage NF	NF conc	RO conc	NF/RO conc
			perm	(rotameter)	(rotameter)	
			gpm	gpm	gpm	gpm
10/1/13	12:51	5	3.8	4	5.2	9.1
10/2/13	13:15	5	3.8	4	5.2	9.2
10/2/13	17:50			4	5.2	9.2
10/2/13	20:55	5	3.8	4	5.2	9.2
10/3/13	8:24	5	3.8	4	5.1	9
10/3/13	18:30	5	3.8	4.1	5.1	9.2
10/4/13	11:50	5	3.8	4	5.2	9.2
10/4/13	16:50	5	3.8	4	5.2	9.2
10/4/13	19:40	5	3.8	4	5.2	9.1
10/5/13	8:25	5	3.75	4	5.1	9.2
10/5/13	12:15	5	3.75	4	5.2	9.3
10/5/13	17:45	5	3.8	4	5.2	9.2
10/6/13	13:15	5	3.8	4	5.2	9.2
10/6/13	15:30	5	3.8	4.1	5.2	9.2
10/6/13	20:05	5	3.8	4	5.1	9.2
10/7/13	11:05	5	3.8	4	5.2	9.2
10/7/13	13:45	5	3.8	4	5.2	9.2
10/7/13	16:30	5	3.8	4	5.2	9.2
10/7/13	20:15	5	3.8	3.9	5.1	9.15
10/8/13	7:30	5	3.8	4.1	5	9.35
10/8/13	10:30	5	3.8	4	5	9.25
10/8/13	13:35	5	3.85	4	5.1	9.25
10/8/13	16:30	5	3.9	4	5	9.15
10/8/13	20:00	5	3.8	4	5.1	9.15
10/9/13	8:50	5	3.8	4	5	9.1
10/9/13	14:30	5	3.8	4	5	9.25
10/9/13	17:20	5	3.8	4.1	5	9.15
10/10/13	8:30	5	3.75	4.15	5	9.15
10/10/13	13:45	5	3.8	4	5.2	9.15
10/10/13	17:50	5	3.8	4	5	9.05
10/10/13	20:15	5	3.8	4	5	9.15
10/11/13	8:30	5	3.8	4	5	9.1
10/11/13	1:50	5	3.8	4	5	9.15
10/11/13	17:40	5	3.8	4	5	9.15
10/11/13	20:30	5	3.75	4	5	9.15
10/12/13	10:30	5	3.75	4	5	9.15
10/12/13	13:45	5	3.8	4.1	5	9.2
10/12/13	17:00	5	3.8	4.1	5	9.15
10/12/13	19:20	5	3.8	4	5	9.15
10/13/13	7:25	5	3.9	4.1	4.9	9.05
10/15/13	15:15	5	3.8	3.9	5.2	9.15
10/15/13	20:00	5	3.8	3.9	5.2	9.2

Appendix A3-2 Brighton Log Sheet Data
NF and EDM Flows

Date	Time	EDM-feed gpm	M. Na gpm	M. Cl gpm	NaCl gpm	E-Rinse gpm
10/1/13	12:51	31.8	26.8	26.5	27.8	5.2
10/2/13	13:15	32.1	26.7	26.6	27	4.9
10/2/13	17:50	32.1	26.7	26	27.5	5
10/2/13	20:55	32.2	26.7	26.2	27.2	5
10/3/13	8:24	31.5	26.6	26.4	27.3	5
10/3/13	18:30	32.1	27.1	26.3	27.4	5.1
10/4/13	11:50	31.7	26.7	26.2	27.3	5.1
10/4/13	16:50	32	26.8	26.2	27.3	5.1
10/4/13	19:40	31.8	26.7	26.4	27.4	5
10/5/13	8:25	31.4	26.8	26.4	27.4	5
10/5/13	12:15	31.5	27	26.2	27.5	5.1
10/5/13	17:45	31.6	27.3	26.4	27.2	5
10/6/13	13:15	31.6	26.7	26.4	27.7	5
10/6/13	15:30	31.8	26.6	26	27.2	4.9
10/6/13	20:05	31.8	27	26.2	27.1	4.9
10/7/13	11:05	31.7	27.2	26.4	27.3	4.9
10/7/13	13:45	31.6	27.1	26.5	27.3	5
10/7/13	16:30	31.8	26.6	26.4	27.1	4.9
10/7/13	20:15	31.3	26.8	26.4	27.1	4.9
10/8/13	7:30	31.6	27.1	26.4	27.2	4.8
10/8/13	10:30	31.9	27.1	26.3	27.7	4.9
10/8/13	13:35	31.8	26.8	26.2	27	4.9
10/8/13	16:30	31.7	27.6	26.3	26.8	5
10/8/13	20:00	31.6	27.6	26.3	27	4.9
10/9/13	8:50	31.5	26.7	26.5	27.3	4.9
10/9/13	14:30	32.1	27.2	26.6	27	4.9
10/9/13	17:20	31.9	26.8	26.6	26.8	4.9
10/10/13	8:30	31.7	26.7	26.4	27.1	4.9
10/10/13	13:45	31.7	27.1	26.1	27.1	4.9
10/10/13	17:50	31.5	26.6	26.2	27.4	4.9
10/10/13	20:15	31.9	27.1	26.5	27.4	4.9
10/11/13	8:30	31.7	26.9	26.3	27.1	4.9
10/11/13	1:50	31.7	27.1	26.5	27.4	4.9
10/11/13	17:40	31.7	27.1	26.5	27.4	4.9
10/11/13	20:30	31.7	27.3	26.6	26.6	4.8
10/12/13	10:30	31.6	27.4	26.4	26.9	4.9
10/12/13	13:45	31.6	27	26.3	27.2	4.9
10/12/13	17:00	31.7	27	26.3	27.6	4.7
10/12/13	19:20	31.6	26.9	26.1	27.5	4.7
10/13/13	7:25	31.9	26.4	26.1	27.1	4.6
10/15/13	15:15	31.5	26.5	26	26.7	4.9
10/15/13	20:00	31.4	26.8	26	27.2	4.9

Appendix A3-2 Brighton Log Sheet Data
NF and EDM Flows

Date	Time	silica purge	silica purge	Δ height (in)	NF F
		gph	gpm		time (min)
10/1/13	12:51	4	0.067		
10/2/13	13:15	4	0.067		
10/2/13	17:50	4	0.067		
10/2/13	20:55	4	0.067		
10/3/13	8:24	3	0.050		
10/3/13	18:30	3	0.050		
10/4/13	11:50	3	0.050		
10/4/13	16:50	3	0.050		
10/4/13	19:40	3	0.050	3.500	130
10/5/13	8:25	3	0.050	1.5	120
10/5/13	12:15	3	0.050		
10/5/13	17:45	3	0.050	1.250	75
10/6/13	13:15	3	0.050	1.000	60
10/6/13	15:30	3	0.050		
10/6/13	20:05	3	0.050		
10/7/13	11:05	2.5	0.042		
10/7/13	13:45	2.5	0.042		
10/7/13	16:30	2.5	0.042		
10/7/13	20:15	2.5	0.042	3.125	140
10/8/13	7:30	2.5	0.042		
10/8/13	10:30	2.5	0.042	1.5	120
10/8/13	13:35	2.5	0.042		
10/8/13	16:30	2.5	0.042		
10/8/13	20:00	2.5	0.042		
10/9/13	8:50	1.25	0.021		
10/9/13	14:30	1.25	0.021		
10/9/13	17:20	2	0.033		
10/10/13	8:30	2	0.033		
10/10/13	13:45	3	0.050		
10/10/13	17:50	3.25	0.054		
10/10/13	20:15	3.25	0.054	2.5	166
10/11/13	8:30	3.25	0.054		
10/11/13	1:50	3.25	0.054		
10/11/13	17:40	3.25	0.054		
10/11/13	20:30	3.25	0.054	2.125	390
10/12/13	10:30	3.25	0.054		
10/12/13	13:45	3.25	0.054		
10/12/13	17:00	3.25	0.054		
10/12/13	19:20	3.25	0.054		
10/13/13	7:25	3.25	0.054		
10/15/13	15:15	3.25	0.054		
10/15/13	20:00	4	0.067		

Appendix A3-2 Brighton Log Sheet Data
 NF and EDM Flows

Date	Time	Used Tank overflowing?
10/1/13	12:51	
10/2/13	13:15	
10/2/13	17:50	
10/2/13	20:55	
10/3/13	8:24	
10/3/13	18:30	
10/4/13	11:50	
10/4/13	16:50	
10/4/13	19:40	
10/5/13	8:25	
10/5/13	12:15	
10/5/13	17:45	
10/6/13	13:15	
10/6/13	15:30	
10/6/13	20:05	
10/7/13	11:05	
10/7/13	13:45	
10/7/13	16:30	
10/7/13	20:15	
10/8/13	7:30	yes
10/8/13	10:30	no
10/8/13	13:35	no
10/8/13	16:30	no
10/8/13	20:00	no
10/9/13	8:50	yes
10/9/13	14:30	no
10/9/13	17:20	no
10/10/13	8:30	no
10/10/13	13:45	no
10/10/13	17:50	no
10/10/13	20:15	no
10/11/13	8:30	no
10/11/13	1:50	no
10/11/13	17:40	no
10/11/13	20:30	no
10/12/13	10:30	yes
10/12/13	13:45	no
10/12/13	17:00	no
10/12/13	19:20	no
10/13/13	7:25	no
10/15/13	15:15	no
10/15/13	20:00	no

Appendix A3-2 Brighton Log Sheet Data
NF and EDM Flows

Date	Time	NF perm	1st stage NF	NF conc	RO conc	NF/RO conc
			perm	(rotameter)	(rotameter)	
			gpm	gpm	gpm	gpm
10/15/13	22:05	5	3.8	3.95	5.2	9.15
10/16/13	6:45	5	3.75	4	5	9.15
10/16/13	8:10	5	3.8	4.1	5	9.15
10/16/13	13:10	5	3.8	4	5	9.15
10/16/13	17:10	5	3.9	3.9	5.1	9.15
10/17/13	8:00	5	3.75	4	4.9	9.05
10/17/13	12:20	5	3.9	4.1	5	9.25
10/17/13	14:30	5	3.8	3.9	5.2	9.25
10/17/13	16:00	5	3.8	3.9	5.1	9.15
10/17/13	17:30	5	3.8	3.9	5.1	9.2
10/17/13	19:55	5	3.8	3.9	5.1	9.15
10/18/13	7:15	5	3.8	4	5.1	9.15
10/18/13	13:30	5	3.8	4	5.1	9.15
10/18/13	15:25	5	3.8	3.9	5.1	9.15
10/18/13	17:25	5	3.8	4	5	9.15
10/18/13	20:20	5	3.8	4	5	9.15
10/19/13	7:55	5	3.8	4	5.1	9.15
10/22/13	13:05	5	3.8	4	5.1	9.15
10/22/13	14:00	5	3.8	4	5	9.15
10/22/13	16:00	5	3.8	4	5	9.25
10/22/13	16:20	5	3.75	4	5	9.15
10/22/13	20:25	5	3.75	4.1	4.9	9.2
10/23/13	5:00	5	3.8	4	5	9.15
10/23/13	9:05	5	3.75	4	5	9.15
10/23/13	10:50	5	3.8	4	5	9.15
10/23/13	13:40	5	3.8	4	5.1	9.15
10/23/13	16:15	5	3.8	4	5	9.15
10/23/13	18:10	5	3.8	3.9	5.1	9.15
10/23/13	20:30	5	3.8	4	5	9.15
10/24/13	8:20	5	3.8	4	5	9.15
10/24/13	9:50	5	3.8	4.1	4.9	9.25
10/24/13	15:30	5	3.8	4.1	5	9.2
10/24/13	17:30	5	3.8	4.1	4.9	9.15
10/25/13	10:00	5	3.75	4.15	4.8	9.2
10/25/13	10:22	5	3.8	4	4.9	9.05
10/25/13	14:10	5	3.8	4	4.9	9.15
10/25/13	16:00	5	3.8	4	4.9	9.2
10/25/13	17:50	5	3.8	4.1	5	9.2
10/25/13	20:30	5	3.8	4.1	4.9	9.2
10/26/13	10:40	5	3.8	4.15	4.9	9.2

Appendix A3-2 Brighton Log Sheet Data
NF and EDM Flows

Date	Time	EDM-feed gpm	M. Na gpm	M. Cl gpm	NaCl gpm	E-Rinse gpm
10/15/13	22:05	31.4	26.8	26	27.7	4.9
10/16/13	6:45	31.3	26.6	25.9	26.9	4.9
10/16/13	8:10	31.1	26.9	25.8	26.5	4.9
10/16/13	13:10	31.2	26.4	25.9	27	4.9
10/16/13	17:10	31.4	26.5	26.1	27.2	4.9
10/17/13	8:00	31	26.4	25.7	27.2	4.8
10/17/13	12:20	31.1	26.5	25.9	26.9	4.9
10/17/13	14:30	31.4	26.8	25.7	26.8	4.9
10/17/13	16:00	31.3	26.5	26	27.1	4.9
10/17/13	17:30	31.3	26.3	25.9	26.9	4.9
10/17/13	19:55	31.1	26.9	25.8	27.5	4.9
10/18/13	7:15	31.3	26.5	25.9	27.4	4.9
10/18/13	13:30	31.4	26.8	26	27	4.9
10/18/13	15:25	31	26.3	25.6	27.1	4.9
10/18/13	17:25	31	26	25.9	26.5	4.9
10/18/13	20:20	31.1	25.7	25.8	27.1	4.9
10/19/13	7:55	31.4	26.2	25.7	27.1	4.8
10/22/13	13:05	31.4	26.3	26.1	24.8	5
10/22/13	14:00	30.7	26.8	26	24.5	5
10/22/13	16:00	31.2	26.1	25.7	26.7	5
10/22/13	16:20	31.2	26.7	25.7	27.3	4.9
10/22/13	20:25	31.3	26.9	25.5	27	5
10/23/13	5:00	31.3	26.1	25.7	27	5
10/23/13	9:05	31.2	26.5	25.7	26.7	4.9
10/23/13	10:50	30.8	27.1	25.9	26.7	5
10/23/13	13:40	31.1	26.1	25.7	27	5
10/23/13	16:15	31	26.8	26	27	4.9
10/23/13	18:10	31.2	26.5	25.8	27	4.9
10/23/13	20:30	31.7	26.4	25.8	26.7	4.9
10/24/13	8:20	31.2	26.3	25.7	26.5	4.9
10/24/13	9:50	30.8	26.4	25.6	26.5	4.9
10/24/13	15:30	30.7	26.4	25.9	26.9	4.9
10/24/13	17:30	31.2	26.5	25.8	27.1	4.9
10/25/13	10:00	31.1	26.4	25.7	27.1	4.9
10/25/13	10:22	31.5	26.4	25.5	26.7	4.9
10/25/13	14:10	30.7	26.3	25.7	27.8	4.9
10/25/13	16:00	31	26.3	25.8	27.1	4.9
10/25/13	17:50	30.7	26.5	25.6	27.2	4.9
10/25/13	20:30	31.2	26.4	25.5	27.2	4.9
10/26/13	10:40	31	26.5	25.4	27.1	4.8

Appendix A3-2 Brighton Log Sheet Data
NF and EDM Flows

Date	Time	silica purge	silica purge	Δ height (in)	NF F
		gph	gpm		time (min)
10/15/13	22:05	4	0.067		
10/16/13	6:45	4	0.067	2	80
10/16/13	8:10	4	0.067		
10/16/13	13:10	4	0.067		
10/16/13	17:10	4	0.067		
10/17/13	8:00	4	0.067	5.25	180
10/17/13	12:20	4	0.067		
10/17/13	14:30	4	0.067		
10/17/13	16:00	4	0.067		
10/17/13	17:30	4	0.067	1.25	73
10/17/13	19:55	4	0.067	1.4	135
10/18/13	7:15	4	0.067		
10/18/13	13:30	4	0.067		
10/18/13	15:25	4	0.067	1.6	120
10/18/13	17:25	4	0.067		
10/18/13	20:20	4	0.067		
10/19/13	7:55	4	0.067		
10/22/13	13:05	4	0.067		
10/22/13	14:00	4	0.067		
10/22/13	16:00	4	0.067		
10/22/13	16:20	4	0.067		
10/22/13	20:25	4	0.067		
10/23/13	5:00	4	0.067		
10/23/13	9:05	4	0.067	1	60
10/23/13	10:50	4	0.067		
10/23/13	13:40	4	0.067		
10/23/13	16:15	4	0.067		
10/23/13	18:10	4	0.067		
10/23/13	20:30	4	0.067	1.375	150
10/24/13	8:20	4	0.067		
10/24/13	9:50	4	0.067		
10/24/13	15:30	4	0.067		
10/24/13	17:30	4	0.067		
10/25/13	10:00	4	0.067		
10/25/13	10:22	4	0.067		
10/25/13	14:10	4	0.067		
10/25/13	16:00	4	0.067	1	120
10/25/13	17:50	4	0.067		
10/25/13	20:30	4	0.067	2	180
10/26/13	10:40	4	0.067	0.75	30

Appendix A3-2 Brighton Log Sheet Data
 NF and EDM Flows

Date	Time	sed Tank overflowing?
10/15/13	22:05	no
10/16/13	6:45	yes
10/16/13	8:10	no
10/16/13	13:10	no
10/16/13	17:10	no
10/17/13	8:00	yes
10/17/13	12:20	no
10/17/13	14:30	no
10/17/13	16:00	no
10/17/13	17:30	no
10/17/13	19:55	no
10/18/13	7:15	yes
10/18/13	13:30	yes (overflowed from 11-1:30)
10/18/13	15:25	no
10/18/13	17:25	no
10/18/13	20:20	no
10/19/13	7:55	no
10/22/13	13:05	no
10/22/13	14:00	no
10/22/13	16:00	no
10/22/13	16:20	no
10/22/13	20:25	no
10/23/13	5:00	yes
10/23/13	9:05	yes
10/23/13	10:50	no
10/23/13	13:40	yes (from ~12-1:40)
10/23/13	16:15	no
10/23/13	18:10	no
10/23/13	20:30	no
10/24/13	8:20	yes
10/24/13	9:50	no
10/24/13	15:30	no
10/24/13	17:30	no
10/25/13	10:00	no
10/25/13	10:22	no
10/25/13	14:10	no
10/25/13	16:00	no
10/25/13	17:50	no
10/25/13	20:30	no
10/26/13	10:40	yes

Appendix A3-2 Brighton Log Sheet Data
EDM NaCl Usage

	Condition	NaCl level	NaCl after addn	NaCl added (#bags)	NaCl (lb/in)
9/30/13	M.Na-45 mS	25			
10/1/13	M.Na-45 mS	30.25			
10/1/13	M.Na-50 mS	31.25			
10/2/13	M.Na-50 mS	34.5			
10/2/13	M.Na-50 mS	36	17.5	16	34.59
10/2/13	M.Na-55mS	17.75			
10/2/13	M.Na-55mS	18.25			
10/3/13	M.Na-55mS	21.5			
10/4/13	M.Na-60 mS	26.5			
10/4/13	M.Na-60 mS	27.5			
10/5/13	M.Na-60 mS	31.5			
10/5/13	M.Na-60 mS	33.5	13.75	16	32.41
10/6/13	M.Na-60 mS	19.75			
10/6/13	M.Na-60 mS	20.5			
10/7/13	M.Na-60 mS	25			
10/7/13	M.Na-60 mS	26	19.75	6	38.40
10/8/13	M.Na-60 mS	23.75			
10/8/13	M.Na-60 mS	25	16	8	35.56
10/10/13	M.Na-60 mS	26			
10/11/13	M.Na-60 mS	36	21	13	34.67
10/11/13	M.Na-60 mS	22			
10/11/13	M.Na-60 mS	23			
10/12/13	M.Na-60 mS, M.Cl-100 mS	29.75			
10/12/13	M.Na-60 mS, M.Cl-100 mS	31.5			
10/13/13	M.Na-60 mS, M.Cl-100 mS	35.25			
10/15/13	M.Na-65 mS, M.Cl-120 mS	37.125	19.25	14	31.33
10/15/13	M.Na-65 mS, M.Cl-120 mS	20.5			
10/16/13	M.Na-65 mS, M.Cl-120 mS	23.5			
10/16/13	M.Na-65 mS, M.Cl-120 mS	25	20	4	32.00
10/17/13	M.Na-65 mS, M.Cl-120 mS	24.75			
10/17/13	M.Na-65 mS, M.Cl-120 mS	26.75	17.25	8	33.68
10/18/13	M.Na-65 mS, M.Cl-120 mS	22.25			
10/22/13	M.Na-65 mS, M.Cl-120 mS	27.75	22.75	4	32.00
10/22/13	M.Na-65 mS, M.Cl-120 mS	23.5			
10/23/13	M.Na-65 mS, M.Cl-120 mS	27.75			
10/23/13	M.Na-65 mS, M.Cl-120 mS	28.25	16.5	10	34.04
10/23/13	M.Na-65 mS, M.Cl-120 mS	17.25			
10/24/13	M.Na-65 mS, M.Cl-120 mS	21.75			
10/24/13	M.Na-65 mS, M.Cl-120 mS	23.75			
10/25/13	M.Na-55 mS, M.Cl-100 mS	29.75			
10/26/13	M.Na-55 mS, M.Cl-100 mS	34.125			

Appendix A3-2 Brighton Log Sheet Data
EDM Concentrate Overflow (Waste) Flows

Date	Time	Condition	M. Na			M. Cl		
			gal	time	gpm	gal	time	gpm
10/1/13	12:51	M.Na-45 mS	27	60	0.45	21	150	0.14
10/1/13	5-6 pm	M.Na-50mS	25	60	0.42	9	60	0.15
10/2/13	am	M.Na-50mS	24.5	60	0.41	9	60	0.15
10/3/13	4-5 pm	M.Na-55 mS	22.5	60	0.38	17	120	0.14
10/3/13		M.Na-60 mS	20	60	0.33	16	120	0.13
10/4/13	2-4 pm	M.Na-60 mS	19	60	0.32	17	120	0.14
10/5/13	12-2 pm	M.Na-60 mS	20	60	0.33	19	120	0.16
10/6/13	1-3 pm	M.Na-60 mS	20	60	0.33	17.5	120	0.15
10/7/13	2-4 pm	M.Na-60 mS	19	60	0.32	17	120	0.14
10/8/13	10am-1pm	M.Na-60 mS	19	60	0.32	17.5	120	0.15
10/8/13	4-5 pm	M.Na-60 mS	17.5	60	0.29	9	60	0.15
10/9/13	8-10am	M.Na-60 mS	19.2	60	0.32	17.3	120	0.14
10/9/13	10-11am	M.Na-60 mS	19.2	60	0.32	17.3	120	0.14
10/9/13	2:30-5pm	M.Na-60 mS	20.0	64	0.31	12.0	95	0.13
10/10/13	12-2pm	M.Na-60 mS	19	60	0.32	17	120	0.14
10/10/13	5:50 PM	M.Na-60 mS	19	60	0.32	17	120	0.14
10/11/13	4-6pm	M.Na-60 mS	21	61	0.34	18.5	120	0.15
10/12/13	10:30-12	M.Na-60 mS, M.Cl-100 mS	19.5	60	0.33	24	120	0.20
10/15/13	4-6pm	M.Na-65 mS, M.Cl-120 mS	19	60	0.32	17.5	122	0.14
10/16/13	1-3pm	M.Na-65 mS, M.Cl-120 mS	21	70	0.30	17	121	0.14
10/17/13	3-5pm	M.Na-65 mS, M.Cl-120 mS	21	72	0.29	15.5	120	0.13
10/18/13	2-4pm	M.Na-65 mS, M.Cl-120 mS	15	60	0.25	16	120	0.13
10/22/13	2-4pm	M.Na-65 mS, M.Cl-120 mS	18.5	60	0.31	19.5	121	0.16
10/23/13	4-6pm	M.Na-65 mS, M.Cl-120 mS	27.5	90	0.31	12.5	90	0.14
10/24/13	1-3pm	M.Na-65 mS, M.Cl-120 mS	25	88	0.28	15	88	0.17
10/25/13	4-6pm	M.Na-55 mS, M.Cl-100 mS	29	86	0.34	16.5	86	0.19

Appendix A3-2 Brighton Log Sheet Data
NF and EDM Pressures

Date	Time	2nd stage	2nd stage	NF perm	NF conc (final	EDM-feed
		CF-inlet	CF-outlet		stage)	
		psi	psi	psi	psi	psi
10/1/13	12:51	36	32.8	2.2	28.4	17.1
10/2/13	13:15	36.7	35.9	2.1	21.2	17
10/2/13	17:50	37.1	35.5	2.1	21	17
10/2/13	20:55	37.5	35.9	2.1	21.2	17
10/3/13	8:24	37.5	35.9	2.1	21.3	17
10/3/13	18:30	39.1	36.3	2.1	21.7	17
10/4/13	11:50	40.3	37.7	2.1	22.6	17.5
10/4/13	16:50	40.1	37.3	2	22.4	17.1
10/4/13	19:40	40.3	37.7	2	22.6	17.1
10/5/13	8:25	40.4	39.6	2.1	24.4	17.1
10/5/13	12:15	39.5	38.9	2.1	24.2	17.1
10/5/13	17:45	39	38.5	2	24	17
10/6/13	13:15	41.5	39.1	2	24.3	17.1
10/6/13	15:30	40.8	38.4	2	23.5	17.1
10/6/13	20:05	42	37.6	2.1	23	17.1
10/7/13	11:05	41.8	37.5	2.2	22.9	17.1
10/7/13	13:45	41.9	37.4	2.1	22.8	17
10/7/13	16:30	42	37.5	2.1	23	17
10/7/13	20:15	43	38.7	2.1	24.4	17.1
10/8/13	7:30	41.7	39.7	2	24.7	17.1
10/8/13	10:30	41.9	40	2	25.5	17.1
10/8/13	13:35	41.4	39.4	2	25.1	17
10/8/13	16:30	41.5	38.9	2	24.8	17
10/8/13	20:00	42	39.3	2.1	25.1	17
10/9/13	8:50	42.5	39.9	2	25.4	17.1
10/9/13	14:30	40.2	37.5	2	23	17
10/9/13	17:20	40.5	37.9	2.1	23	17
10/10/13	8:30	40.8	38.1	2	23.4	17.1
10/10/13	13:45	40.9	40.6	2	26.3	17
10/10/13	17:50	40.7	40.5	2	26.2	17
10/10/13	20:15	41.3	41.1	2.1	26.7	17.1
10/11/13	8:30	41.1	40.9	2.1	26.4	17.1
10/11/13	1:50	41.3	41.1	2.2	26.8	17
10/11/13	17:40	41.4	41.1	2.2	26.9	17.1
10/11/13	20:30	41.3	41.1	2.1	26.8	17.1
10/12/13	10:30	41.5	41.3	2	26.9	17
10/12/13	13:45	41.3	41	2.2	26.7	17
10/12/13	17:00	41.2	40.9	2	26.7	17.2
10/12/13	19:20	45.4	42.4	2.1	28	17.2
10/13/13	7:25	44.9	41.8	1.9	26.7	17.4
10/15/13	15:15	39.6	39.3	2.1	24.7	17.6
10/15/13	20:00	39.9	39.2	2.1	24.6	17.7

Appendix A3-2 Brighton Log Sheet Data
NF and EDM Pressures

Date	Time	M. Na psi	M. Cl psi	NaCl psi	E-Rinse psi
10/1/13	12:51	16.5	16.7	16.5	17.4
10/2/13	13:15	16.2	16.8	16.6	16.7
10/2/13	17:50	16.3	16.8	16.6	16.9
10/2/13	20:55	16.3	18.8	16.6	17
10/3/13	8:24	16.3	16.9	16.7	16.9
10/3/13	18:30	16.4	16.8	16.6	17
10/4/13	11:50	16.5	16.7	16.9	17.3
10/4/13	16:50	16.4	16.8	16.7	17.1
10/4/13	19:40	16.4	16.6	16.7	17.2
10/5/13	8:25	16.5	16.8	16.7	17.1
10/5/13	12:15	16.4	16.8	16.6	17.1
10/5/13	17:45	16.4	16.7	16.6	17.1
10/6/13	13:15	16.5	16.8	16.7	17
10/6/13	15:30	16.5	16.8	16.8	16.9
10/6/13	20:05	16.5	16.8	16.7	16.8
10/7/13	11:05	16.6	16.8	16.7	16.9
10/7/13	13:45	16.4	16.7	16.7	17.1
10/7/13	16:30	16.5	16.7	16.7	17
10/7/13	20:15	16.5	16.8	16.9	17.1
10/8/13	7:30	16.5	16.8	16.8	16.9
10/8/13	10:30	16.5	16.7	16.7	17.1
10/8/13	13:35	16.4	16.7	16.6	17
10/8/13	16:30	16.5	16.6	16.6	17.1
10/8/13	20:00	16.5	16.7	16.7	17.1
10/9/13	8:50	16.6	16.8	16.7	17.1
10/9/13	14:30	16.5	16.6	16.7	17
10/9/13	17:20	16.5	16.7	16.7	16.9
10/10/13	8:30	16.7	16.8	16.8	17.2
10/10/13	13:45	16.6	16.6	16.7	17.2
10/10/13	17:50	16.5	16.7	16.7	17.3
10/10/13	20:15	16.6	16.7	16.8	17.3
10/11/13	8:30	16.6	16.7	16.8	17.3
10/11/13	1:50	16.6	16.7	16.7	17.3
10/11/13	17:40	16.7	16.6	16.7	17.2
10/11/13	20:30	16.7	16.7	16.7	17.3
10/12/13	10:30	16.6	16.5	16.8	17.4
10/12/13	13:45	16.7	16.4	16.7	17.5
10/12/13	17:00	16.9	16.6	17	17.1
10/12/13	19:20	16.9	16.7	17	17.2
10/13/13	7:25	17	16.7	17.1	17.2
10/15/13	15:15	17.3	17.2	17.4	17.5
10/15/13	20:00	17.4	17.3	17.4	17.5

Appendix A3-2 Brighton Log Sheet Data
NF and EDM Pressures

Date	Time	2nd stage	2nd stage	NF perm	NF conc (final	EDM-feed
		CF-inlet	CF-outlet		stage)	
		psi	psi	psi	psi	psi
10/15/13	22:05	40.2	39.5	2	24.8	17.7
10/16/13	6:45	40.5	40.3	2.2	25.6	17.8
10/16/13	8:10	41.8	41.6	2.1	26.4	17.8
10/16/13	13:10	40.8	40.6	2.2	26	17.7
10/16/13	17:10	40.6	40.3	2	26.1	17.7
10/17/13	8:00	42	41.7	2.1	27.3	17.7
10/17/13	12:20	44	43.7	2	28.8	17.7
10/17/13	14:30	43	40.9	1.9	26.7	17.7
10/17/13	16:00	43	40.9	2.1	26.6	17.7
10/17/13	17:30	43.1	41.1	2	26.8	17.7
10/17/13	19:55	43.2	41.1	2.1	26.7	17.7
10/18/13	7:15	43.9	41.8	1.9	27.2	17.9
10/18/13	13:30	43.2	41.2	1.9	26.7	17.8
10/18/13	15:25	43.2	41.2	2.1	26.8	17.8
10/18/13	17:25	43.2	41.2	2	26.8	17.8
10/18/13	20:20	43.2	42.2	2.1	27.7	17.9
10/19/13	7:55	43.3	42.3	2.1	27.7	17.9
10/22/13	13:05	39.7	39.5	2.1	24.8	17.7
10/22/13	14:00	39.7	39.5	2.1	24.7	17.8
10/22/13	16:00	39.7	39.4	2	24.9	17.8
10/22/13	16:20	39.7	39	2	24.6	17.9
10/22/13	20:25	40	39.9	2.2	25.2	17.9
10/23/13	5:00	42.2	42	2.1	27.3	17.8
10/23/13	9:05	42.5	42.2	2.2	27.7	17.9
10/23/13	10:50	42.2	41.9	2.1	27.6	17.8
10/23/13	13:40	42.1	41.8	2.2	27.6	17.8
10/23/13	16:15	41.5	41.2	2.1	27	17.7
10/23/13	18:10	43.6	41.1	2.2	26.9	17.8
10/23/13	20:30	43.4	40.9	2.1	26.5	17.8
10/24/13	8:20	43.2	42.9	2.1	28.3	17.9
10/24/13	9:50	43.1	42.8	2.1	28	17.9
10/24/13	15:30	42.5	42.3	2	27.8	17.8
10/24/13	17:30	42.6	42.4	2.1	28	17.8
10/25/13	10:00	43.9	43.5	2.1	28.9	18
10/25/13	10:22	42.4	42.1	1.8	27.1	17.9
10/25/13	14:10	42.4	42.1	2.1	27.7	17.9
10/25/13	16:00	42.4	42.1	2.1	27.7	17.8
10/25/13	17:50	42.6	42.4	2.1	27.9	17.9
10/25/13	20:30	42.8	42.6	2.1	28	17.9
10/26/13	10:40	43.2	43	2	28.4	18

Appendix A3-2 Brighton Log Sheet Data
NF and EDM Pressures

Date	Time	M. Na psi	M. Cl psi	NaCl psi	E-Rinse psi
10/15/13	22:05	17.4	17.3	17.5	17.6
10/16/13	6:45	17.4	17.5	17.4	17.5
10/16/13	8:10	17.4	17.4	17.5	17.7
10/16/13	13:10	17.4	17.3	17.3	17.5
10/16/13	17:10	17.4	17.3	17.3	17.5
10/17/13	8:00	17.4	17.3	17.4	17.2
10/17/13	12:20	17.4	17.2	17.4	17.5
10/17/13	14:30	17.4	17.3	17.4	17.5
10/17/13	16:00	17.4	17.3	17.4	17.4
10/17/13	17:30	17.4	17.3	17.4	17.5
10/17/13	19:55	17.4	17.3	17.4	17.5
10/18/13	7:15	17.5	17.4	17.5	17.6
10/18/13	13:30	17.5	17.4	17.5	17.4
10/18/13	15:25	17.5	17.3	17.5	17.4
10/18/13	17:25	17.5	17.4	17.6	17.5
10/18/13	20:20	17.6	17.4	17.6	17.5
10/19/13	7:55	17.5	17.4	17.6	17.1
10/22/13	13:05	17.4	17.3	16.5	17.4
10/22/13	14:00	17.6	17.4	16.6	17.4
10/22/13	16:00	17.5	17.4	17.5	17.3
10/22/13	16:20	17.6	17.3	17.6	17.3
10/22/13	20:25	17.6	17.5	17.5	17.4
10/23/13	5:00	17.5	17.4	17.6	17.4
10/23/13	9:05	17.6	17.3	17.6	17.4
10/23/13	10:50	17.5	17.4	17.5	17.5
10/23/13	13:40	17.5	17.3	17.5	17.4
10/23/13	16:15	17.4	17.2	17.4	17.3
10/23/13	18:10	17.5	17.3	17.5	17.5
10/23/13	20:30	17.5	17.3	17.5	17.5
10/24/13	8:20	17.5	17.4	17.6	17.5
10/24/13	9:50	17.6	17.4	17.5	17.6
10/24/13	15:30	17.5	17.3	17.5	17.5
10/24/13	17:30	17.5	17.4	17.5	17.7
10/25/13	10:00	17.6	17.4	17.7	17.7
10/25/13	10:22	17.6	17.4	17.7	17.7
10/25/13	14:10	17.5	17.1	17.5	17.7
10/25/13	16:00	17.6	17.1	17.5	17.8
10/25/13	17:50	17.6	17.2	17.6	17.8
10/25/13	20:30	17.6	17.2	17.6	17.9
10/26/13	10:40	17.7	17.4	17.7	

Appendix A3-2 Brighton Log Sheet Data
NF and EDM Temperatures

Date	Time	NF/RO conc °C	EDM-feed °C	EDM-diluate °C	M. Na °C	M. Cl °C	E-Rinse °C
10/1/13	12:51	20.5	21.2		22	22.2	23.3
10/2/13	13:15	21.1	21.7	22.3	22.4	22.7	24
10/2/13	17:50	21	21.8	22.3	22.5	22.58	23.9
10/2/13	20:55	20.4	21	21.6	21.9	21.9	22.4
10/3/13	8:24	20.2	20.6	21.1	21.3	21.4	21.2
10/3/13	18:30	20.3	20.9	21.4	21.7	22	22.7
10/4/13	11:50	19.1	19.5	19.9	19.9	20	19.2
10/4/13	16:50	19.3	19.8	20.2	20.4	20.4	19.9
10/4/13	19:40	19	19.2	19.6	19.7	19.7	17.8
10/5/13	8:25	18.7	19	19.4	19.5	19.5	17.9
10/5/13	12:15	19.9	20.4	20.9	21	21.2	21.5
10/5/13	17:45	20	20.7	21.2	21.2	21.4	21.2
10/6/13	13:15	20.3	20.9	21.4	21.4	21.6	21.7
10/6/13	15:30	20.6	21.3	21.8	22	22.1	22.8
10/6/13	20:05	20.1	20.6	21	21.3	21.3	21
10/7/13	11:05	20.1	20.5	21	21.1	21.3	21.1
10/7/13	13:45	20.9	21.4	21.9	22	22.3	23.2
10/7/13	16:30	21.1	21.8	22.3	22.6	22.8	24.2
10/7/13	20:15	20.2	20.8	21.3	21.5	21.5	21.9
10/8/13	7:30	19.9	20.4	20.9	21.1	21.1	21.4
10/8/13	10:30	20.3	21	21.5	21.8	21.8	22.8
10/8/13	13:35	21.3	22	22.5	22.7	22.9	24.2
10/8/13	16:30	21.4	22.3	22.9	23	23.4	24.7
10/8/13	20:00	20.4	21.1	21.7	21.7	22	22.1
10/9/13	8:50	19.9	20.5	21	21	21.2	21.8
10/9/13	14:30	21.3	22.1	22.7	22.8	23	23.7
10/9/13	17:20	20.9	21.7	22.2	22.3	22.6	23.4
10/10/13	8:30	18.8	20.3	20.8	20.7	20.9	21.5
10/10/13	13:45	20.3	21.1	21.7	21.9	22	22.8
10/10/13	17:50	20.1	21	21.6	21.7	22	22.5
10/10/13	20:15	19.5	20.2	20.7	20.7	20.9	20.8
10/11/13	8:30	19.9	20.6	21.1	21.2	21.3	21.4
10/11/13	1:50	20.5	21.3	21.8	21.9	22.1	22.6
10/11/13	17:40	20.4	21.2	21.8	22	22.3	23
10/11/13	20:30	20.2	20.7	21.3	21.3	21.5	21.3
10/12/13	10:30	20	20.7	21.3	21.4	21.6	21.7
10/12/13	13:45	20.7	21.6	22.1	22.3	22.4	22.3
10/12/13	17:00	20.5	21.4	22	22.1	22.2	22.5
10/12/13	19:20	20.1	21	21.6	21.6	21.8	22.1
10/13/13	7:25	19.7	20.4	20.9	20.8	21	20.5
10/15/13	15:15	20.1	20.4	20.9	21	21.2	21.3
10/15/13	20:00	19.8	20.2	20.7	20.7	21	20.7

Appendix A3-2 Brighton Log Sheet Data
NF and EDM Temperatures

Date	Time	NF/RO conc °C	EDM-feed °C	EDM-diluate °C	M. Na °C	M. Cl °C	E-Rinse °C
10/15/13	22:05	19.6	20	20.4	20.4	20.7	20.3
10/15/13	6:45	19.2	19.5	19.9	19.8	20	18.7
10/15/13	8:10	19.2	19.4	19.8	19.7	19.9	18.8
10/15/13	13:10	20.4	20.9	21.4	21.5	21.6	
10/15/13	17:10	20.6	21.2	21.7	21.8	22	22.1
10/17/13	8:00	19.5	20.1	20.6	20.6	20.9	20.9
10/17/13	12:20	20.3	20.9	21.4	21.7	22	23.1
10/17/13	14:30	20.5	21.1	21.6	21.8	22	22.6
10/17/13	16:00	20.5	21.1	21.6	21.8	22	22.5
10/17/13	17:30	20.2	20.7	21.3	21.4	21.6	21.7
10/17/13	19:55	20	20.6	21.2	21.3	21.5	21.7
10/18/13	7:15	19.2	19.7	19.2	20.1	20.3	19.9
10/18/13	13:30	19.9	20.4	21	21.1	21.3	21.4
10/18/13	15:25	20.1	20.6	21.1	21.3	21.5	22.1
10/18/13	17:25	20	20.5	21.1	21.2	21.4	22
10/18/13	20:20	19.2	19.7	20.1	20.1	20.4	20.1
10/19/13	7:55	19.1	19.6	20	20	20.3	19.9
10/22/13	13:05	20.6	21	21.5	21.6	21.8	22.5
10/22/13	14:00	20.7	21.2	21.6	21.8	21.9	22.8
10/22/13	16:00	20.5	21.1	21.6	21.7	21.9	22.7
10/22/13	16:20	20.5	21	21.5	21.8	21.9	22.6
10/22/13	20:25	19.7	20.2	20.7	20.7	20.9	20.8
10/23/13	5:00	19.3	19.8	20.3	20.3	20.5	20.2
10/23/13	9:05	19.4	20	20.5	20.6	20.8	21.1
10/23/13	10:50	19.7	20.2	20.8	20.9	21.1	21.7
10/23/13	13:40	20.4	21	21.6	21.8	22	23.1
10/23/13	16:15	20.5	21.4	22	22.2	22.3	23.4
10/23/13	18:10	20.1	20.9	21.5	21.7	21.8	22.3
10/23/13	20:30	19.9	20.6	21.2	21.3	21.5	21.7
10/24/13	8:20	19.3	20	20.5	20.6	20.8	
10/24/13	9:50	19.4	20	20.6	20.6	20.9	21
10/24/13	15:30	20.5	21.3	21.9	22.1	22.3	23.5
10/24/13	17:30	20.3	21.1	21.6	21.8	22	22.3
10/25/13	10:00	19.4	19.9	20.4	20.4	20.6	20.8
10/25/13	10:22	19.2	20	20.6	20.7	20.9	20.9
10/25/13	14:10	20.6	21.2	21.7	21.8	22	22.8
10/25/13	16:00	20.6	21.3	21.8	22	22.1	22.8
10/25/13	17:50	20.1	20.8	21.3	21.4	21.6	21.8
10/25/13	20:30	19.7	20.3	20.6	20.9	21.1	21.4
10/26/13	10:40	19.4	20.5	20.5	20.5	20.6	20.7

APPENDIX B

ONLINE DATA

Appendix B1 – Alamogordo, Year 1

Appendix B2 – Alamogordo, Year 2

Appendix B3 – Brighton, Colorado

APPENDIX B1
ALAMOGORDO,
YEAR 1 ONLINE DATA

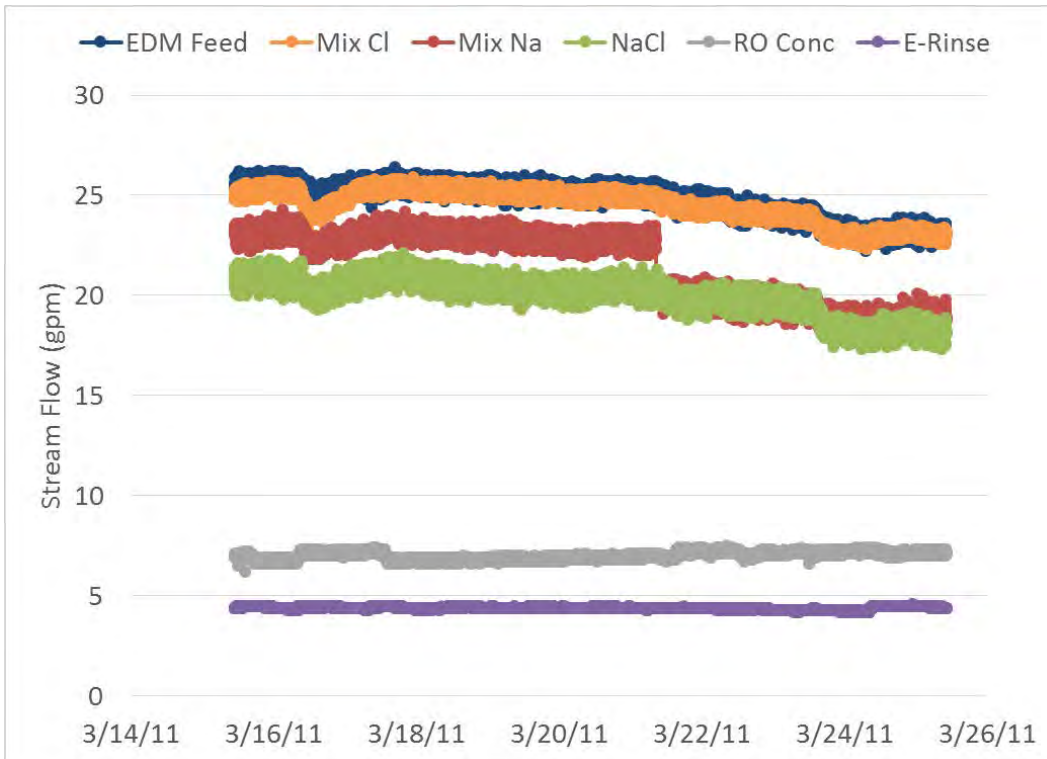


Figure B1-1.—NF concentrate and EDM stream flows (14 gpm NF permeate).

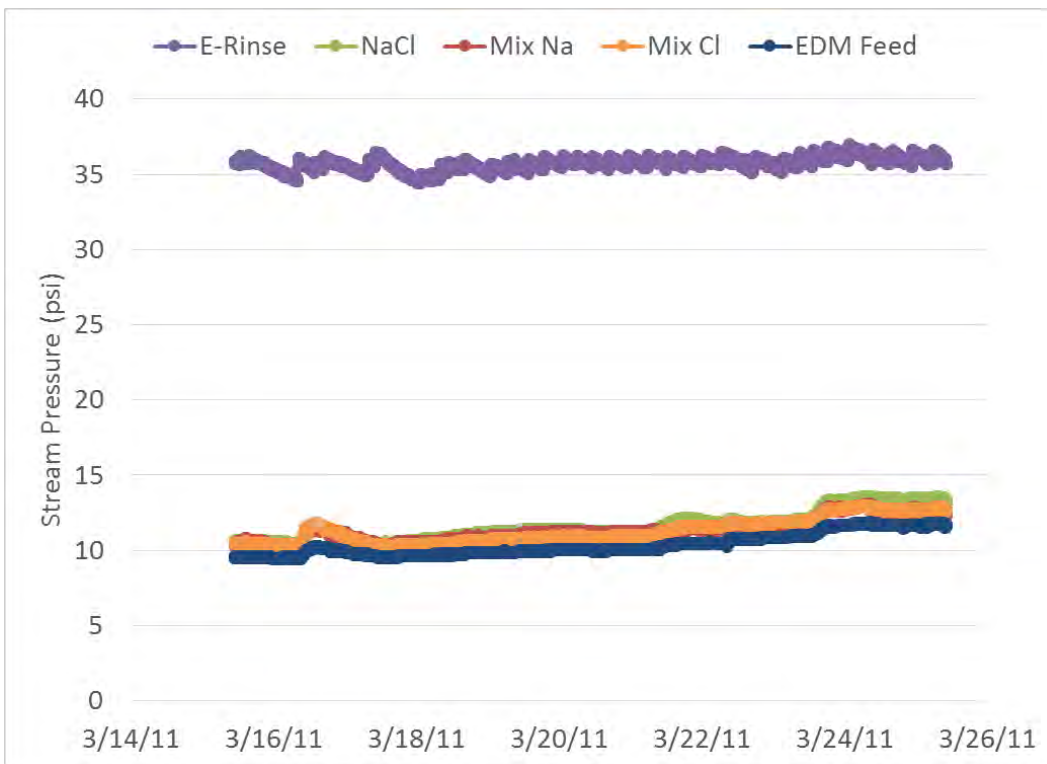


Figure B1-2.—EDM stream pressures (14 gpm NF permeate).

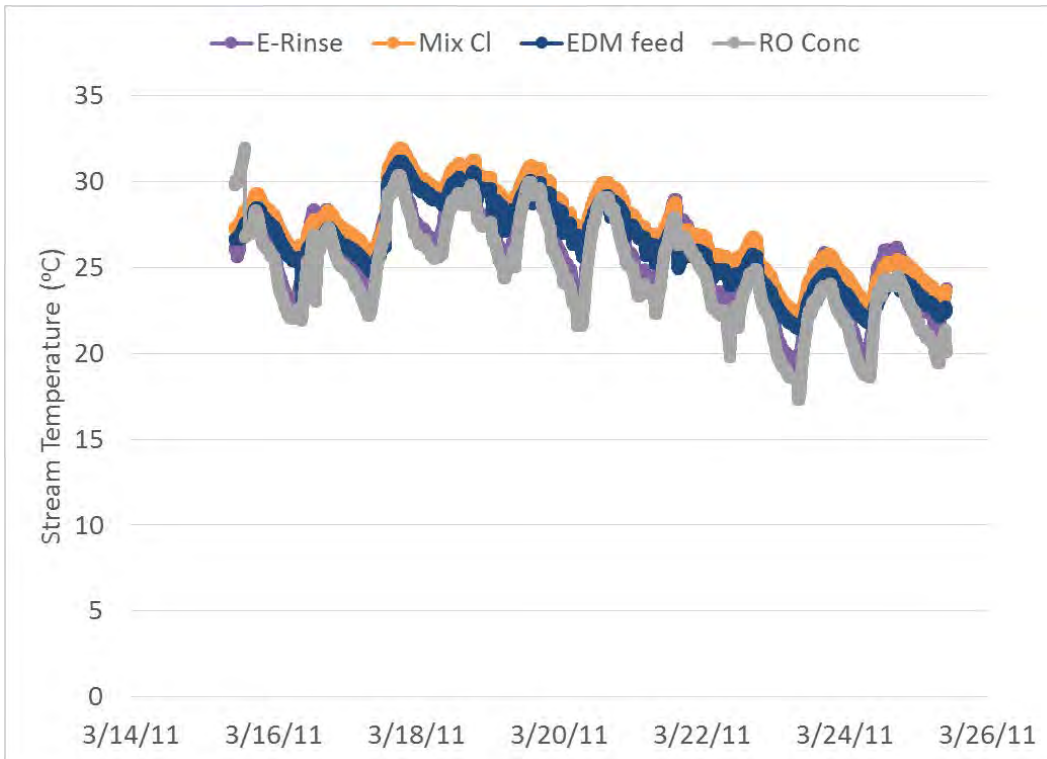


Figure B1-3.—EDM stream temperatures (14 gpm NF permeate).

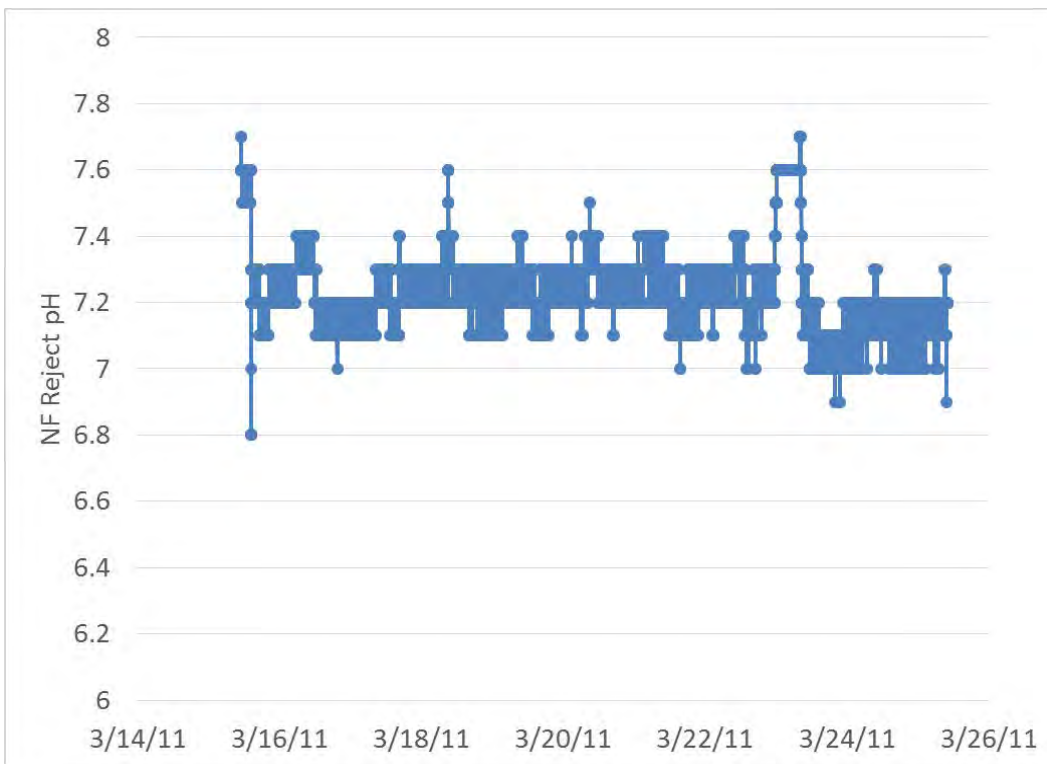


Figure B1-4.—NF reject pH (14 gpm NF permeate).

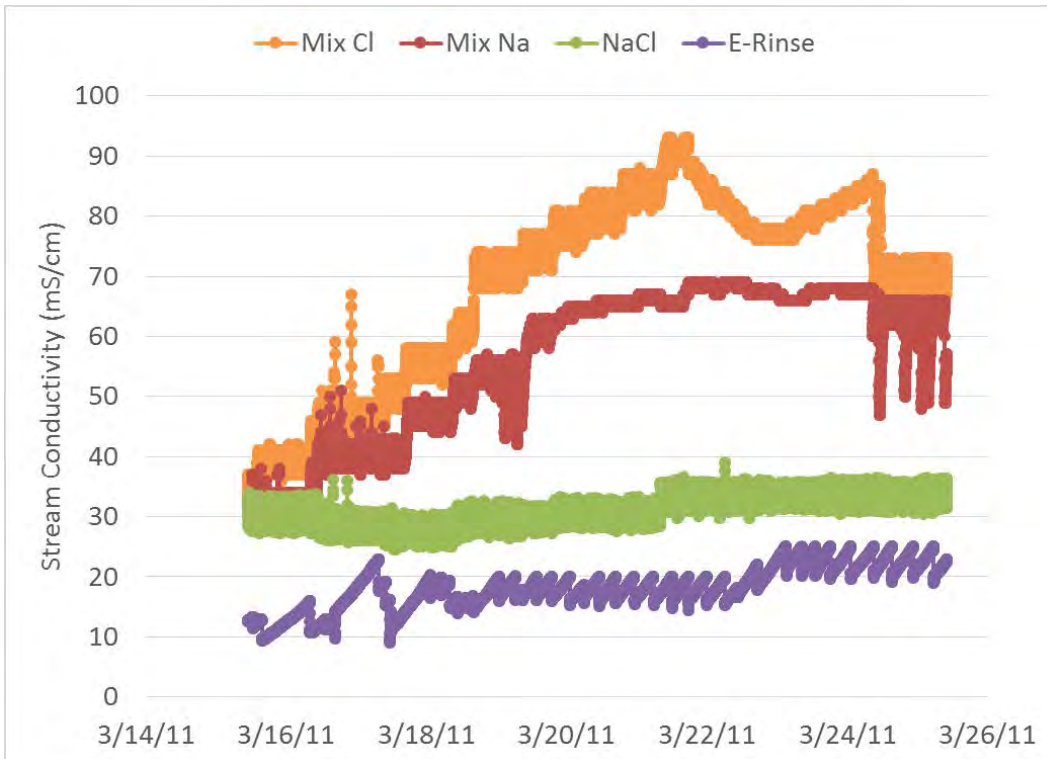


Figure B1-5.—High concentration EDM stream conductivity (14 gpm NF permeate).
(note: Mixed Cl & Mixed Na are not calibrated or scaled)

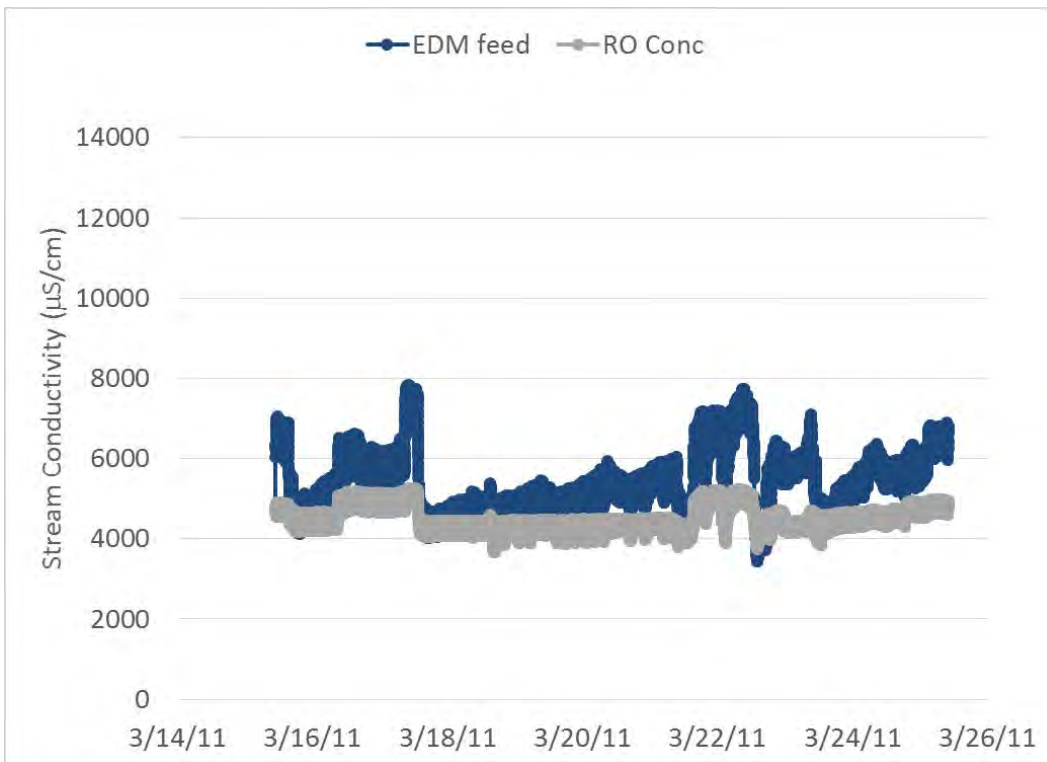


Figure B1-6.—Low concentration EDM stream conductivity (14 gpm NF permeate)

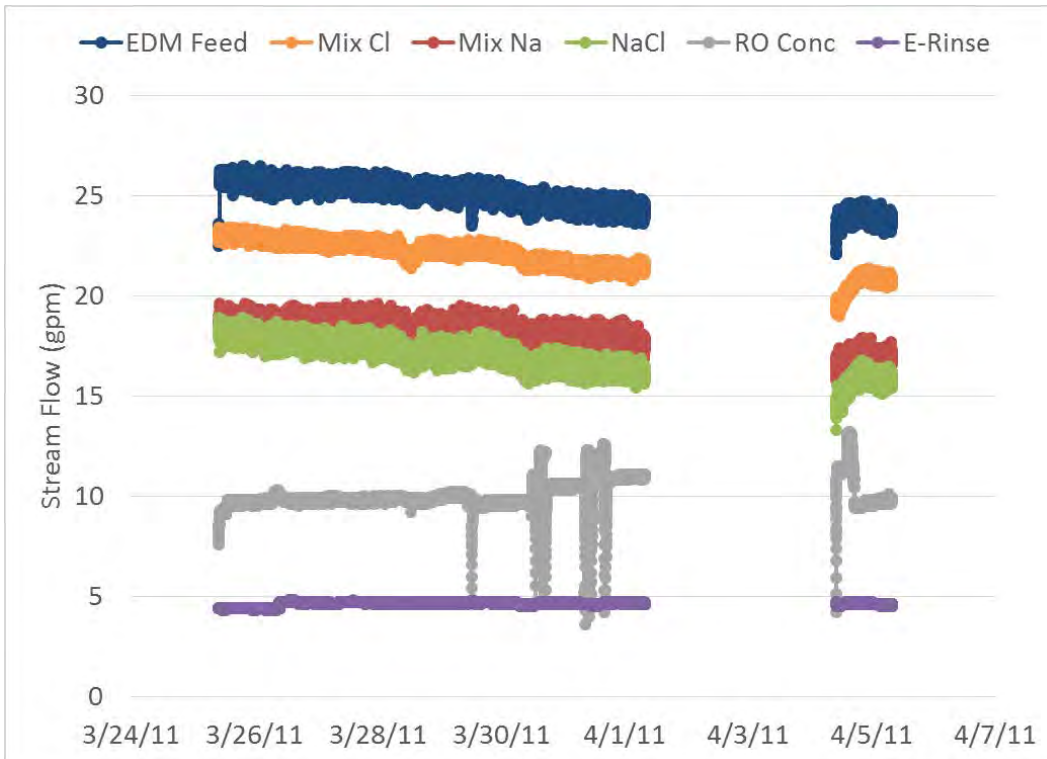


Figure B1-7.—EDM stream flows (20 gpm NF permeate).

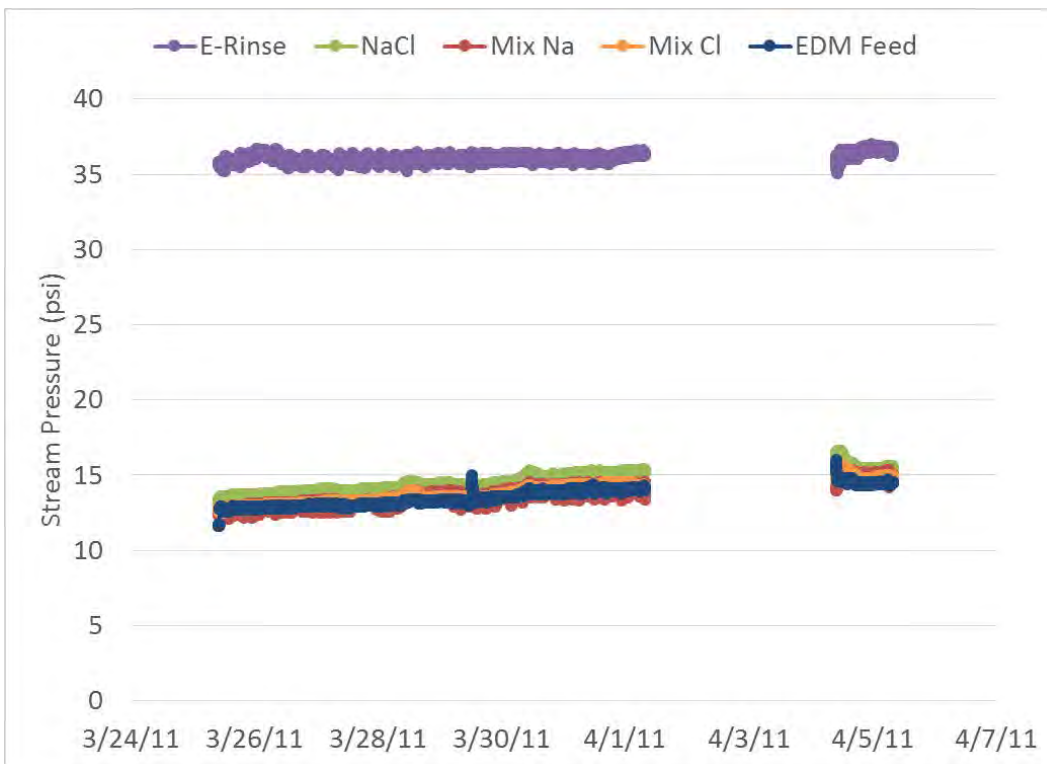


Figure B1-8.—EDM stream pressures (20 gpm NF permeate).

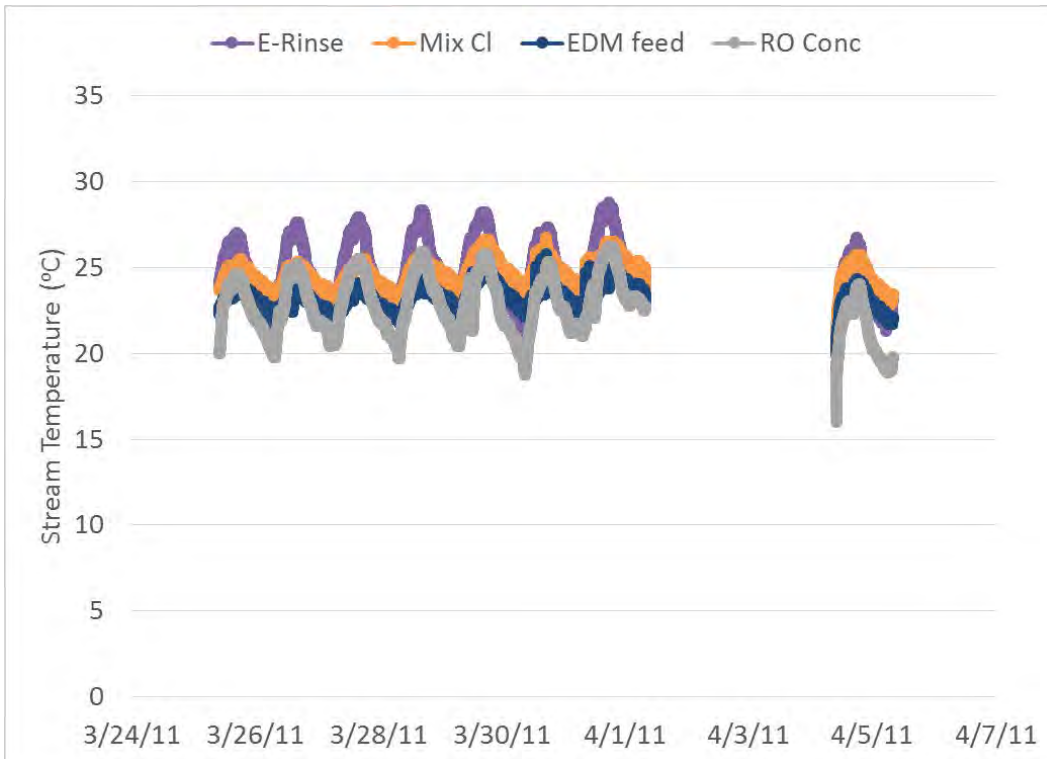


Figure B1-9.—EDM stream temperatures (20 gpm NF permeate).

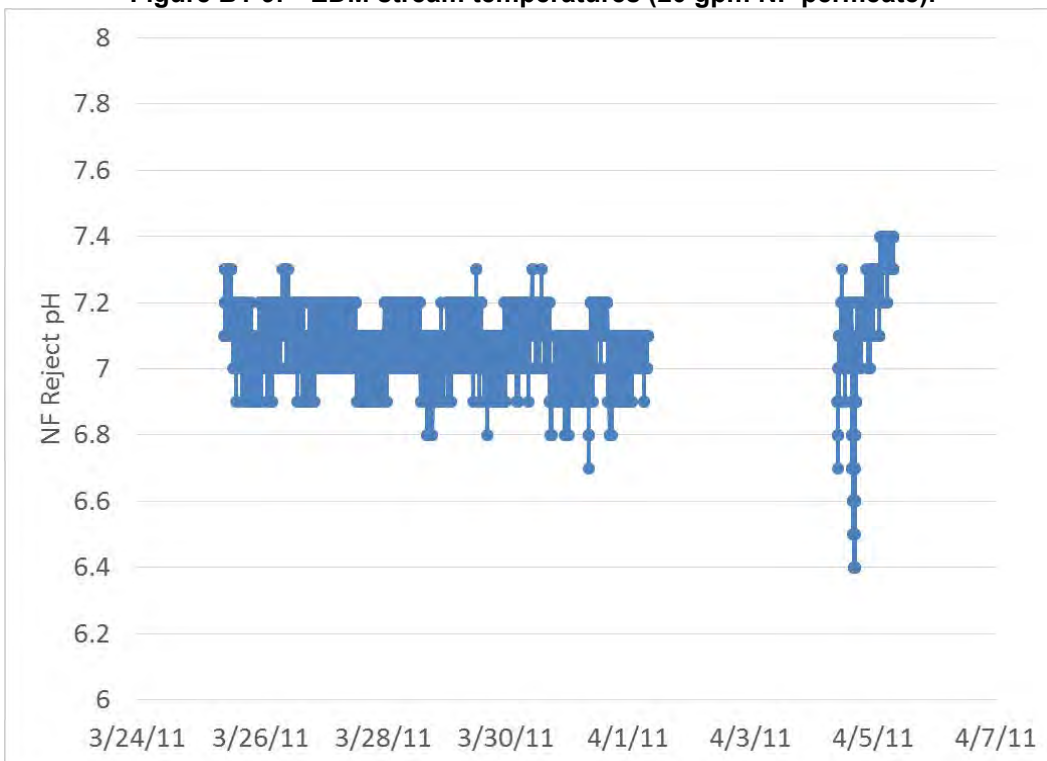


Figure B1-10.—NF concentrate pH (20 gpm NF permeate).

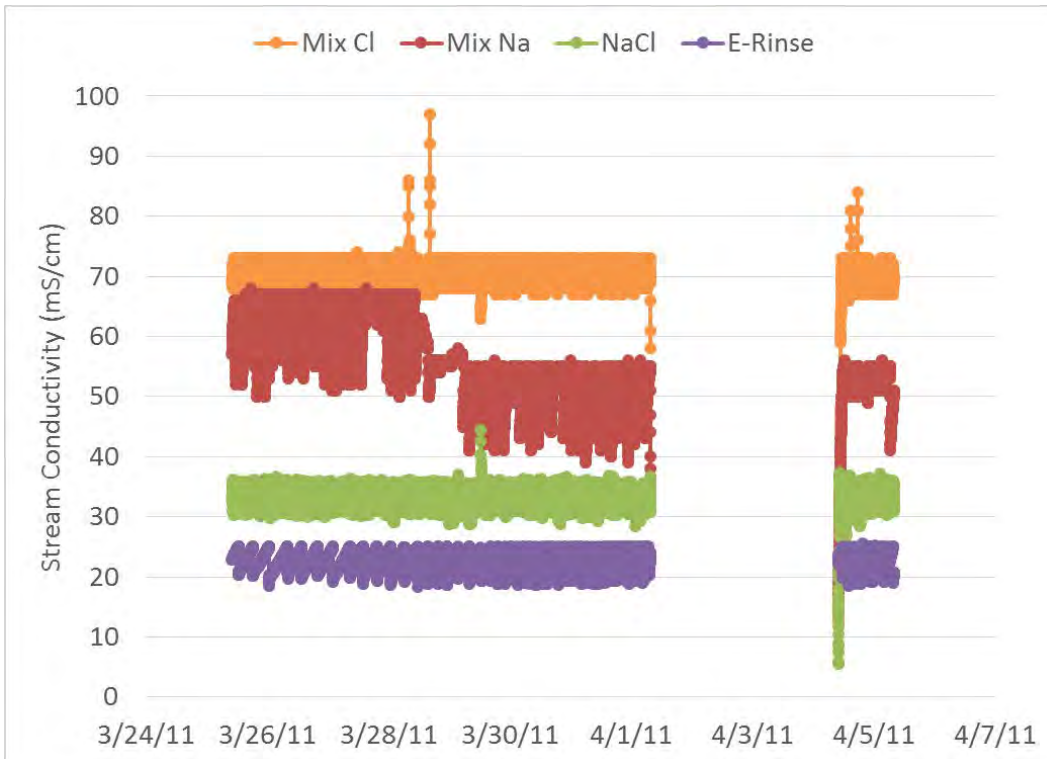


Figure B1-11.—High concentration EDM stream conductivity (20 gpm NF permeate).
(note: Mixed Cl & Mixed Na are not calibrated or scaled)

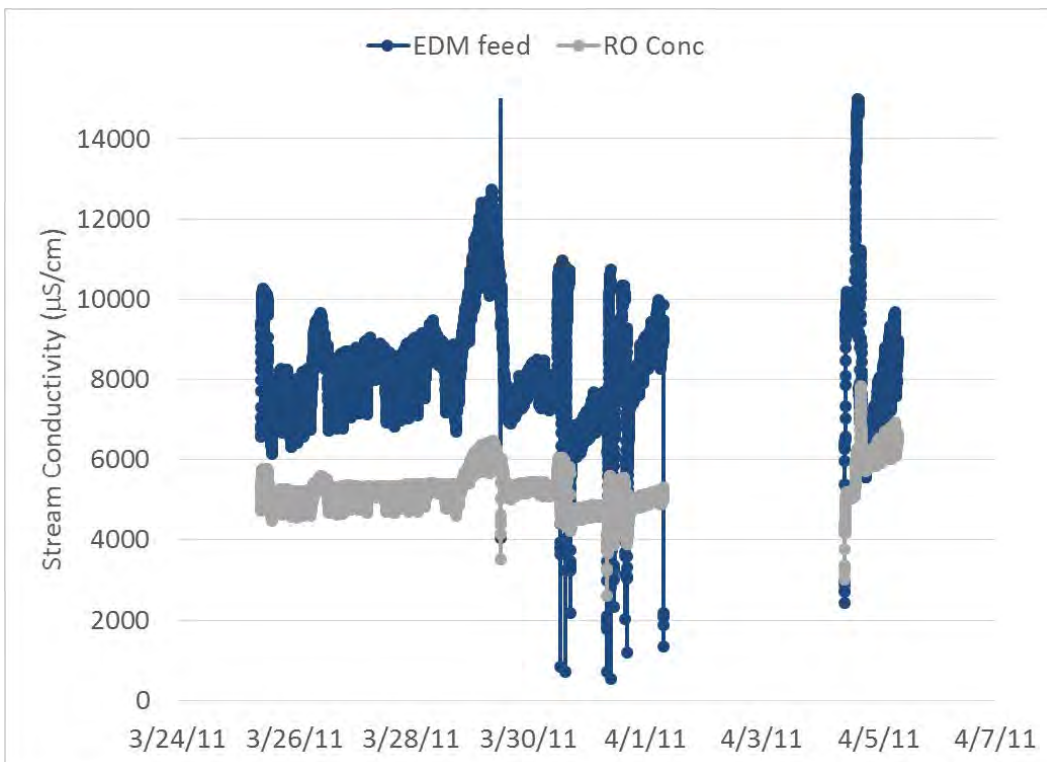


Figure B1-12.—Low concentration EDM stream conductivity (20 gpm NF permeate).

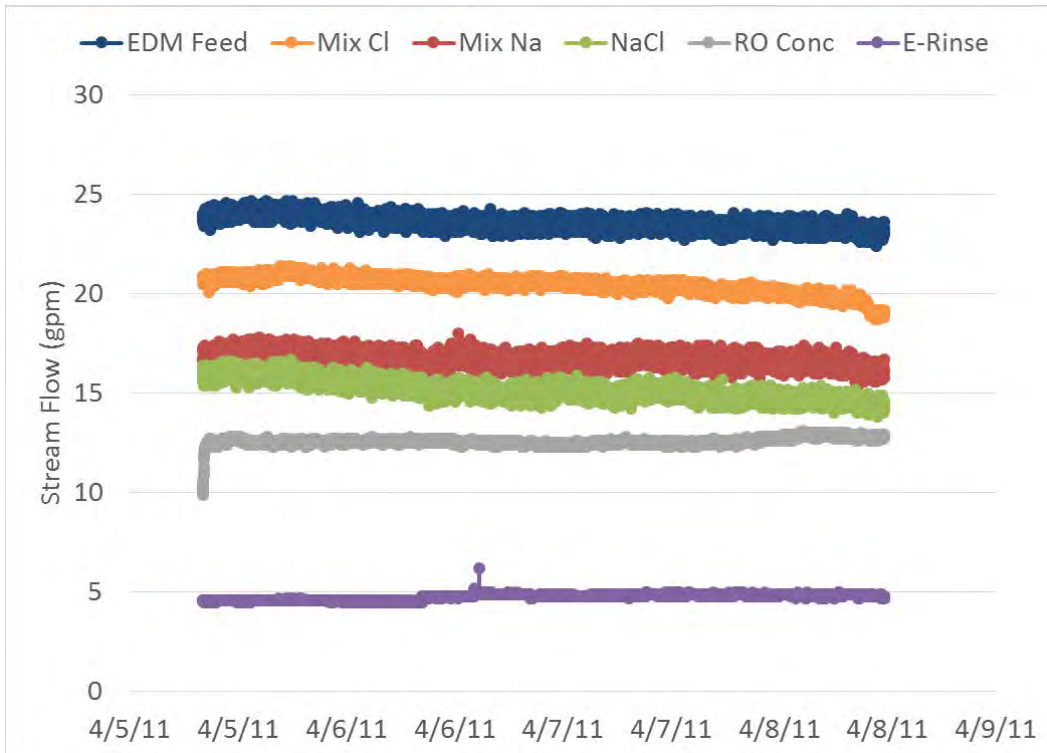


Figure B1-13.—EDM stream flows (25 gpm NF permeate).

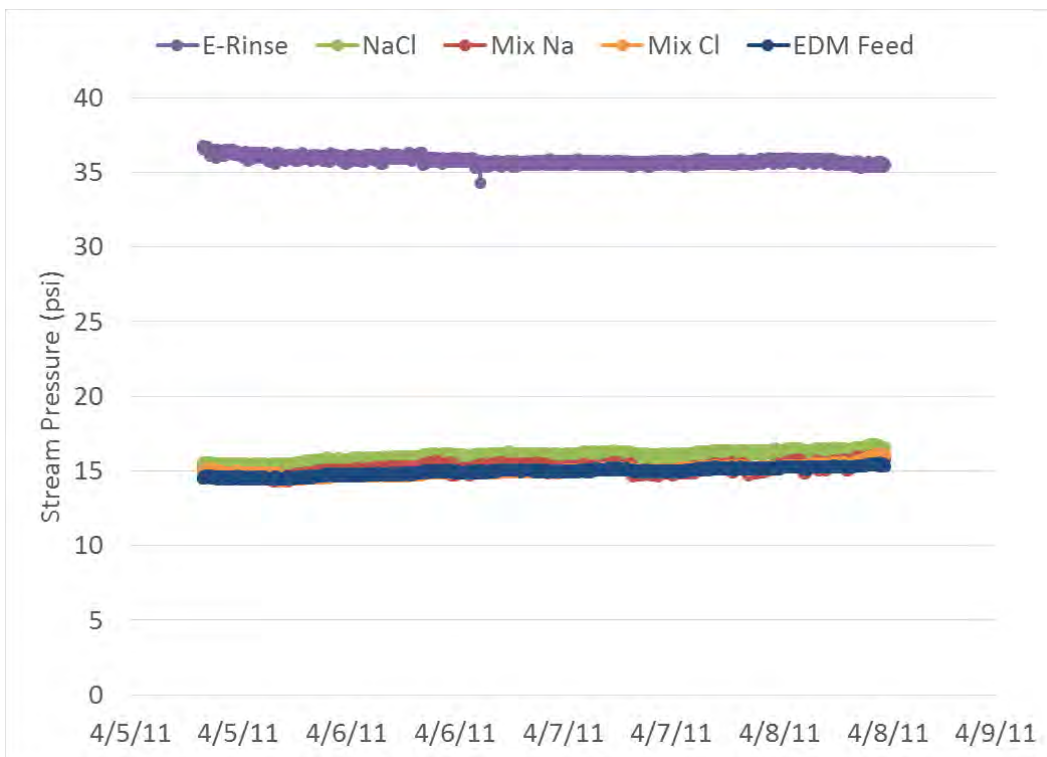


Figure B1-14.—EDM stream pressures (25 gpm NF permeate).

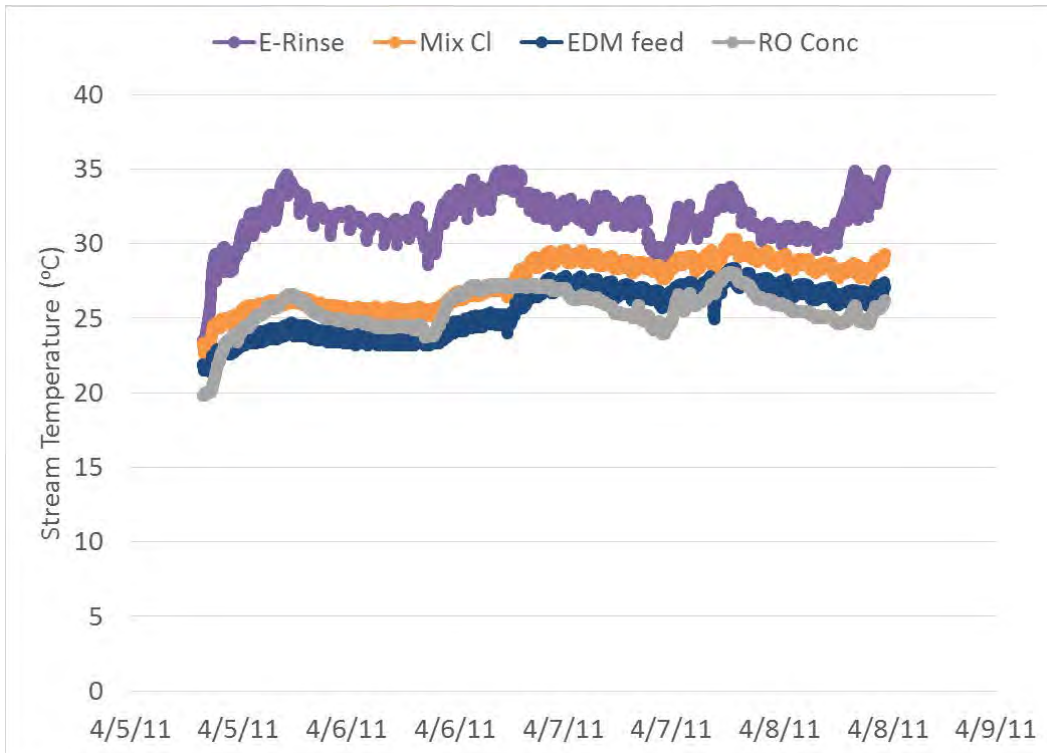


Figure B1-15.—EDM stream temperatures (25 gpm NF permeate).

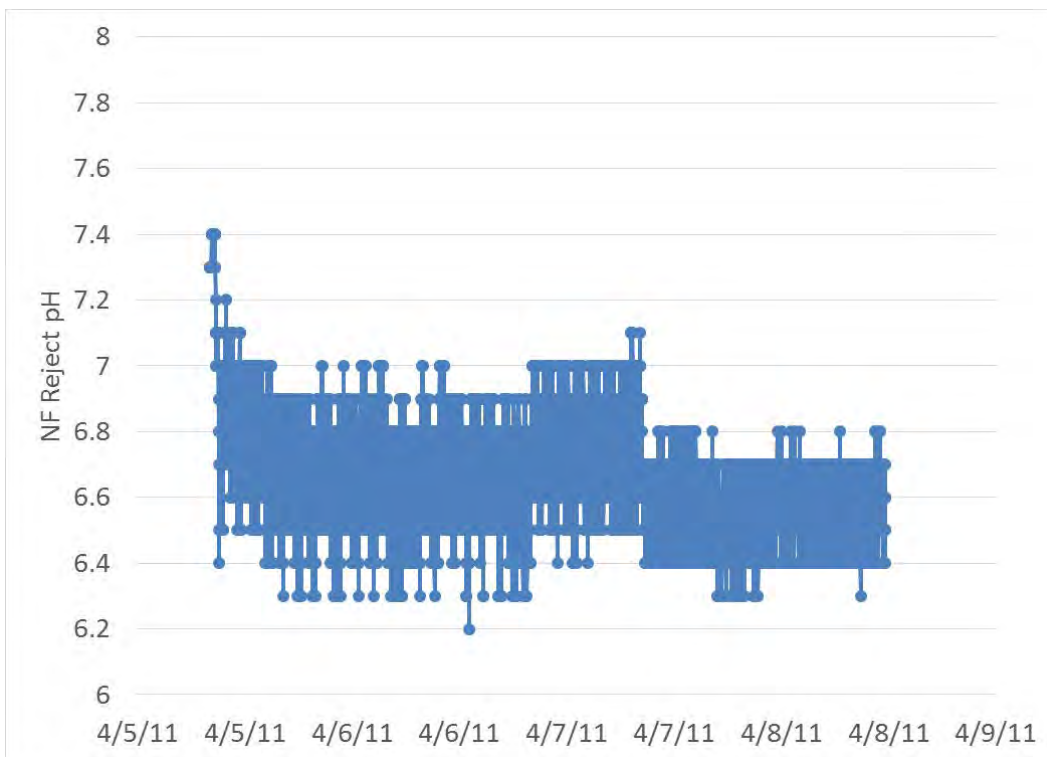


Figure B1-16.—NF reject pH (25 gpm NF permeate).

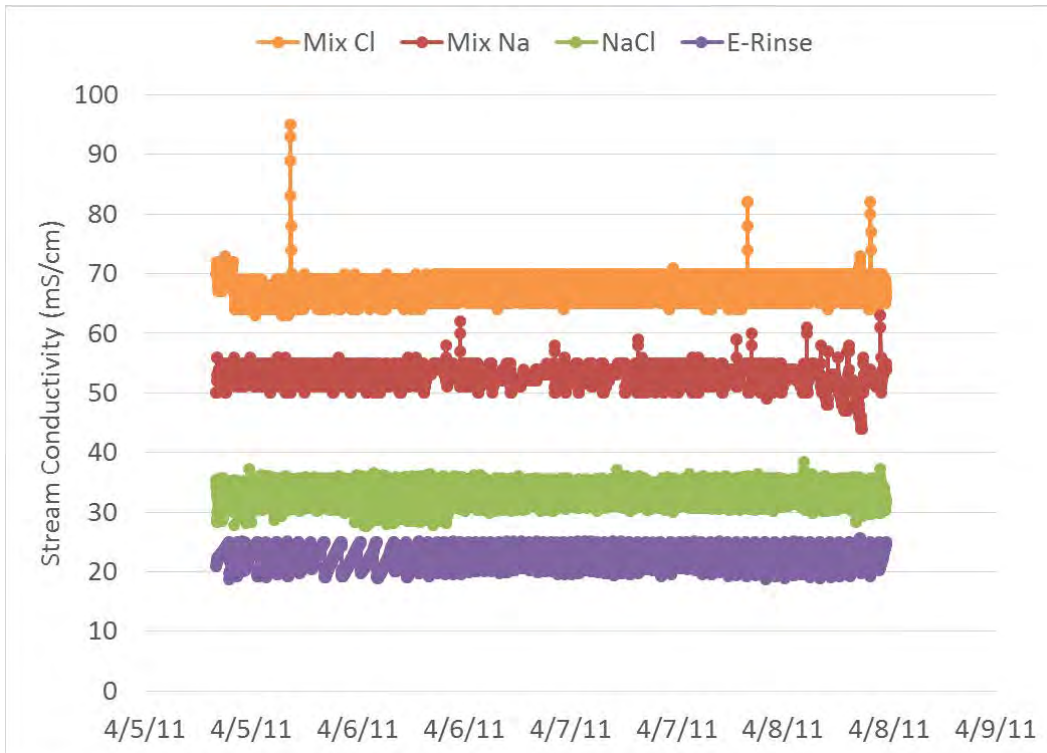


Figure B1-17.—High concentration EDM stream conductivity (25 gpm NF permeate).
(note: Mixed Cl & Mixed Na are not calibrated or scaled)

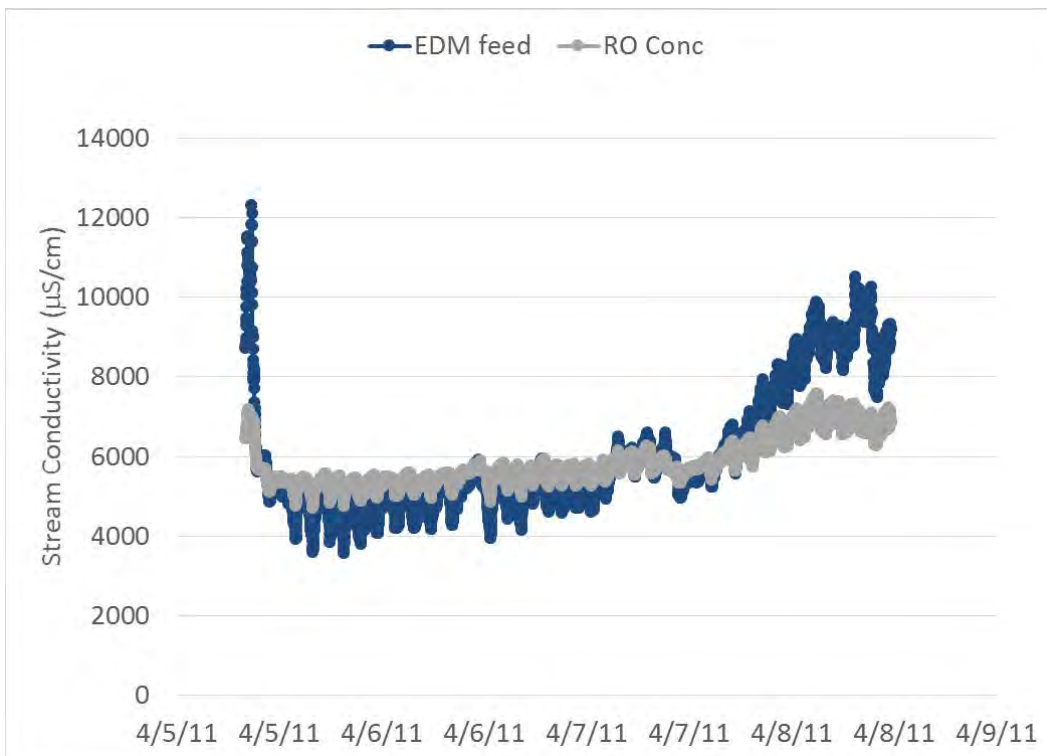


Figure B1-18.—Low concentration EDM stream conductivity (25 gpm NF permeate).

APPENDIX B2
ALAMOGORDO,
YEAR 2 ONLINE DATA

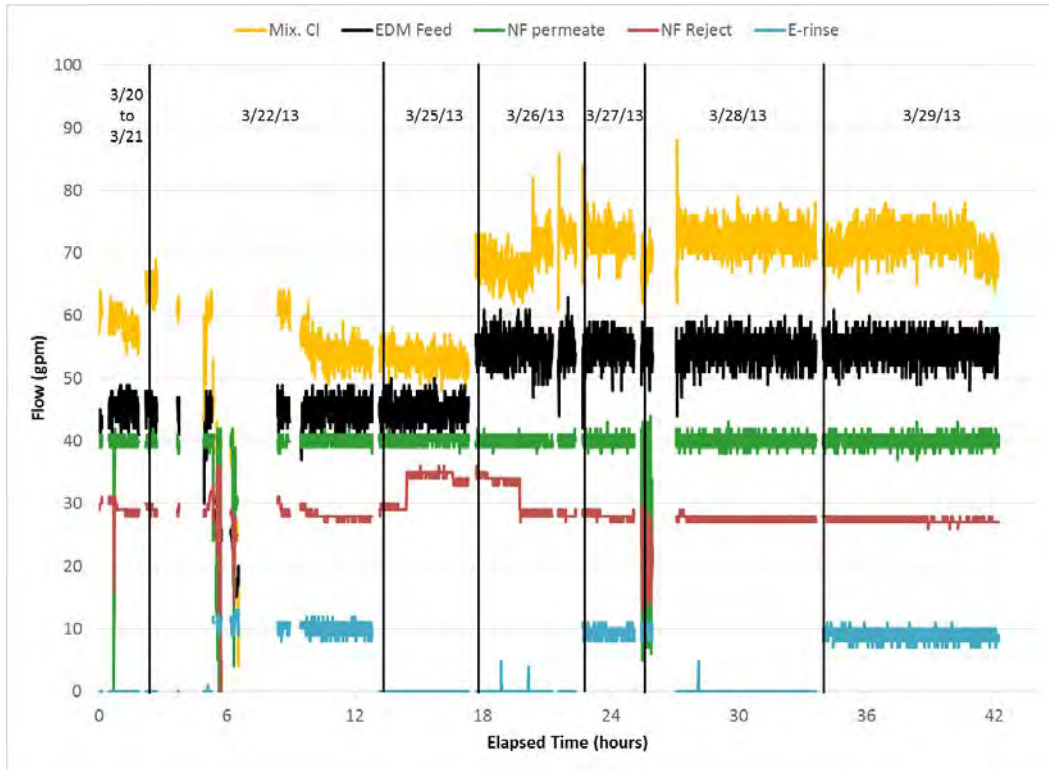


Figure B2-19.—EDM stream flows (preliminary experiments-1).

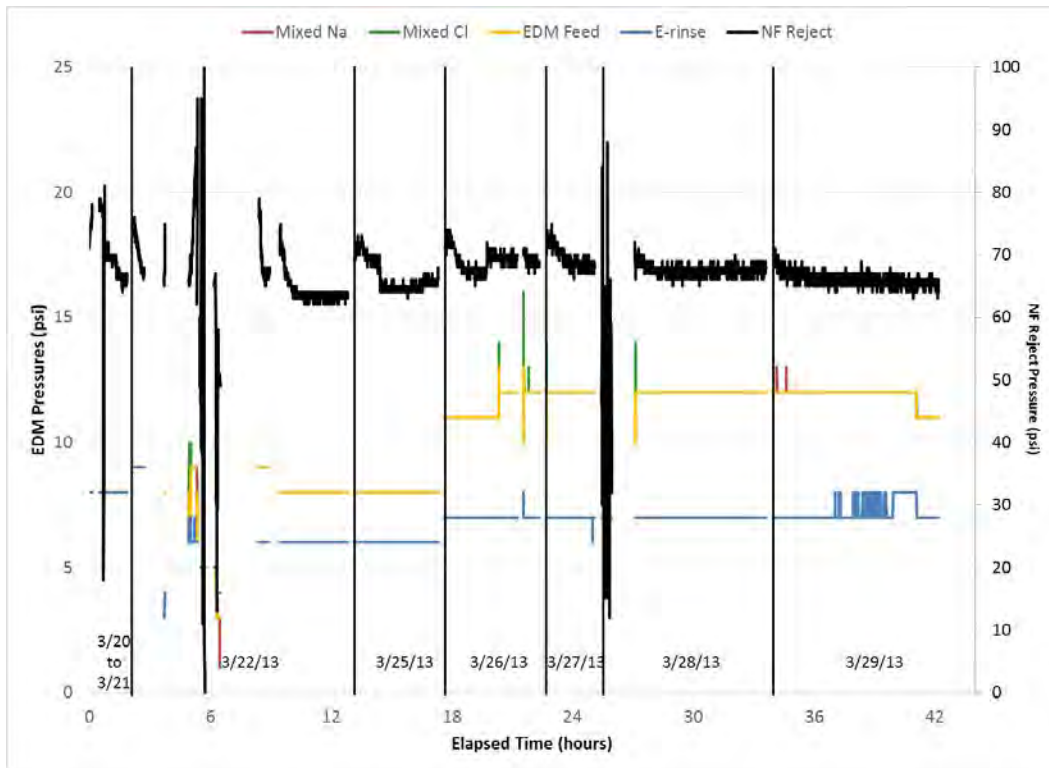


Figure B2-20.—EDM and NF reject stream pressures (preliminary experiments-1).

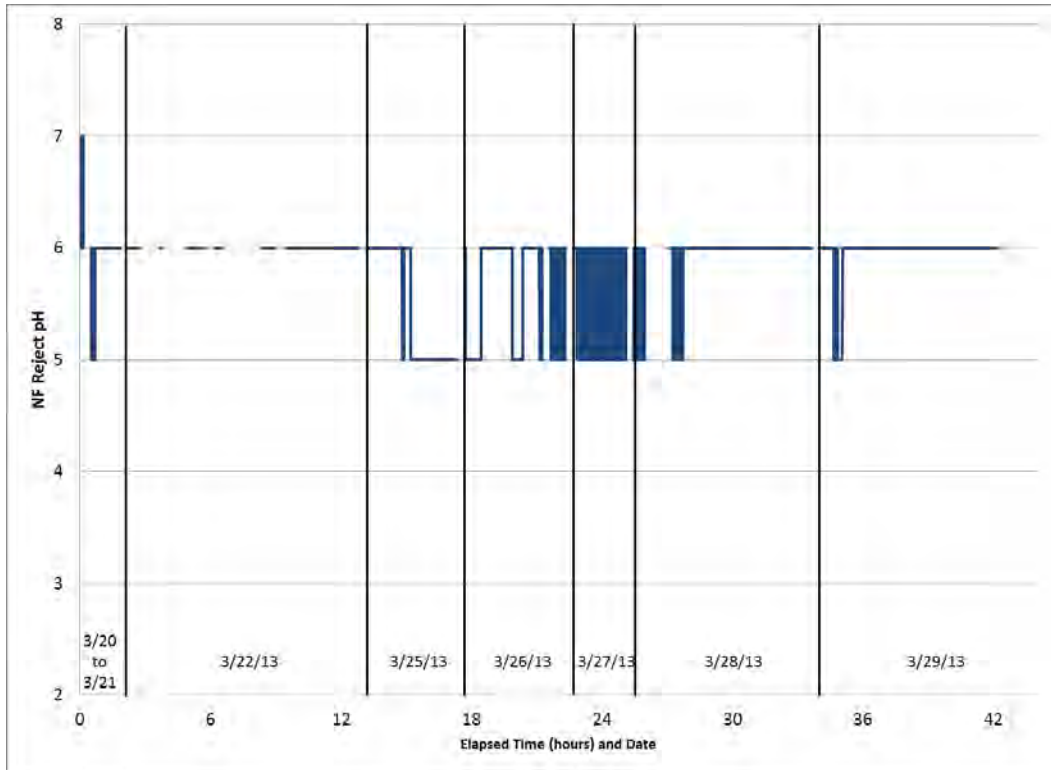


Figure B2-21.—NF reject pH (preliminary experiments-1).

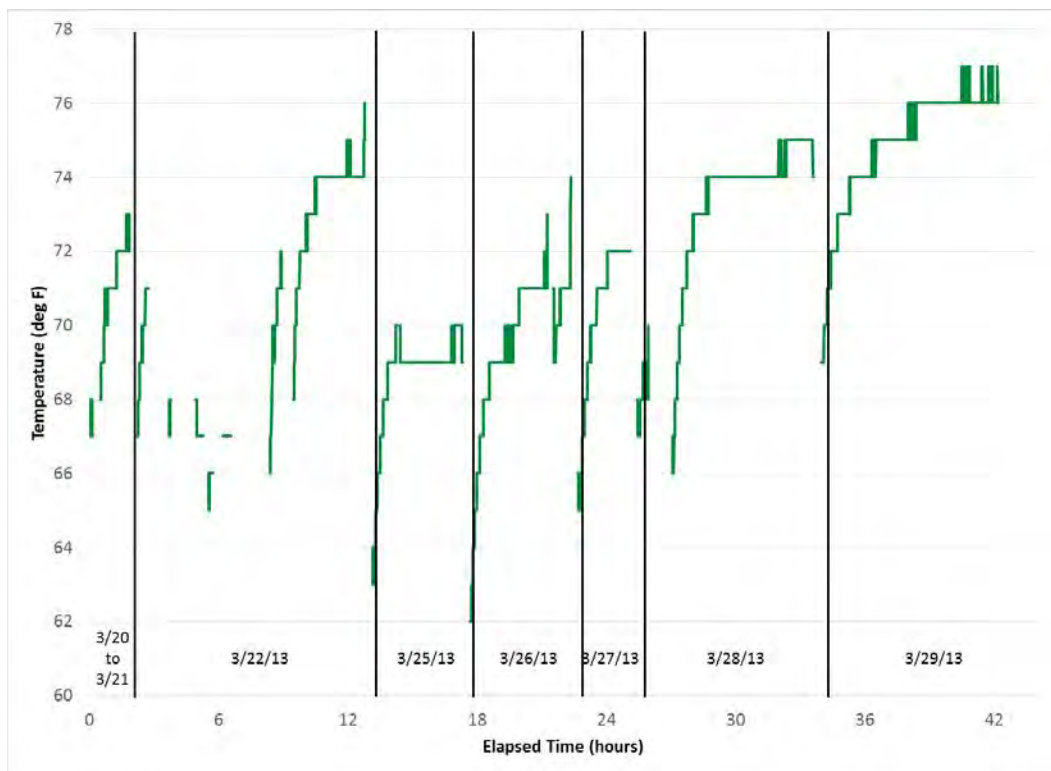


Figure B2-22.—EDM electrode rinse temperature (preliminary experiments-1).

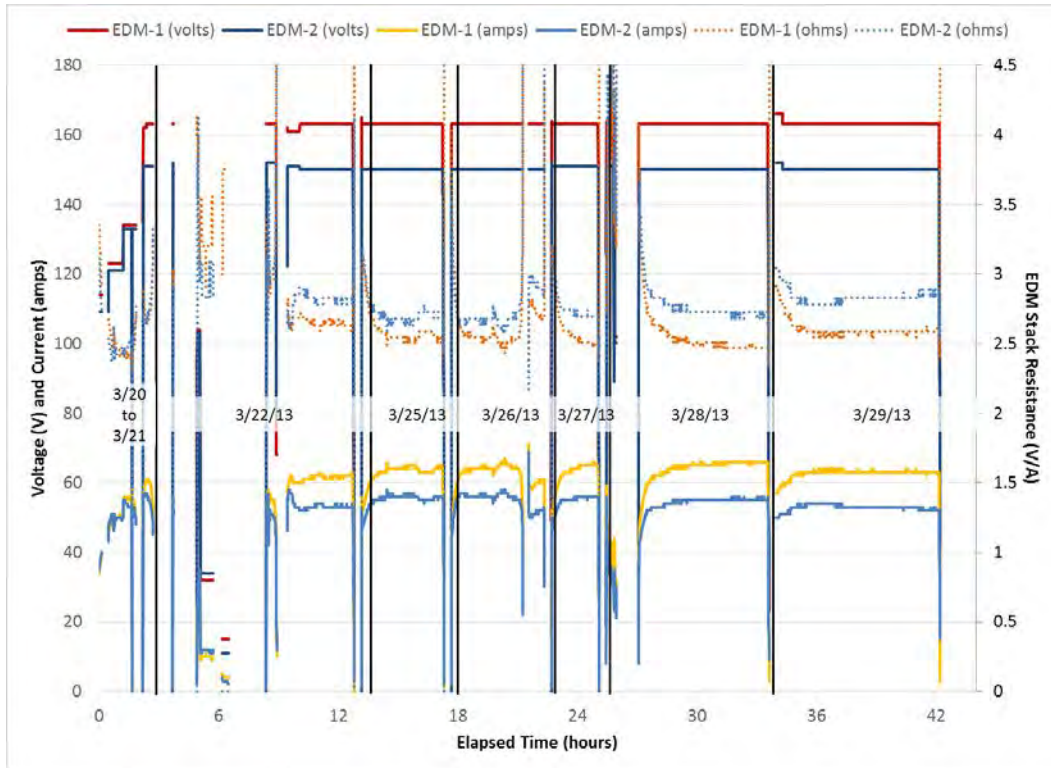


Figure B2-23.—EDM voltage, current, stack resistance (preliminary experiments-1).

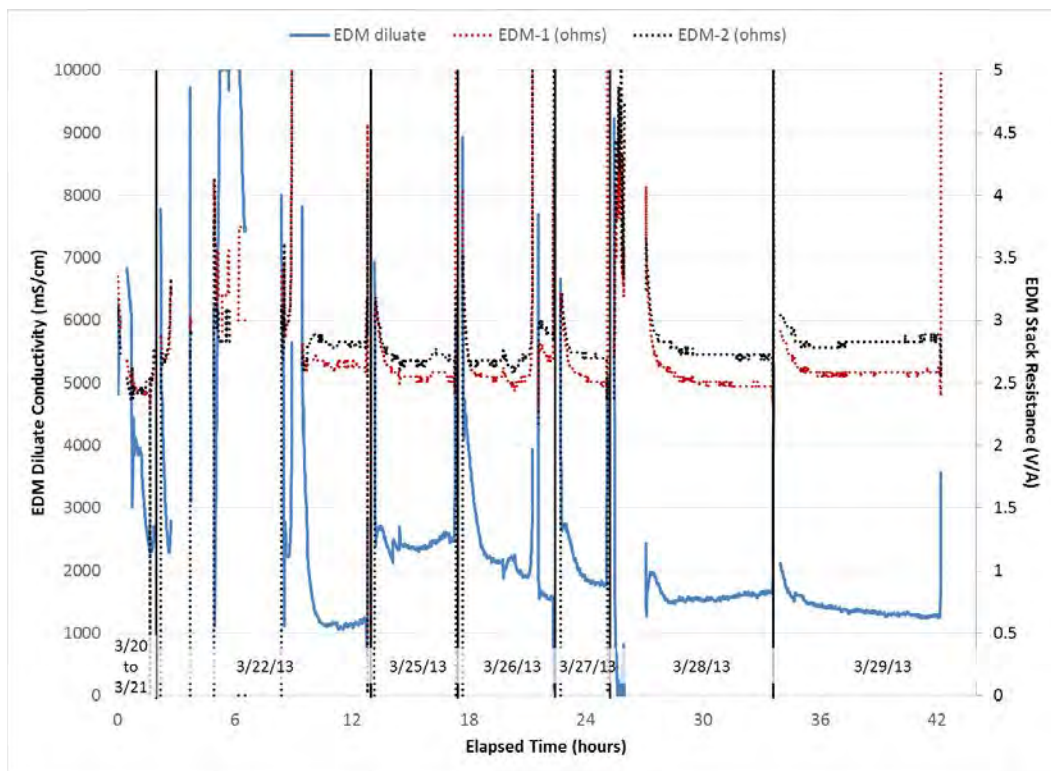


Figure B2-24.—EDM diluate conductivity and stack resistance (preliminary experiments-1).

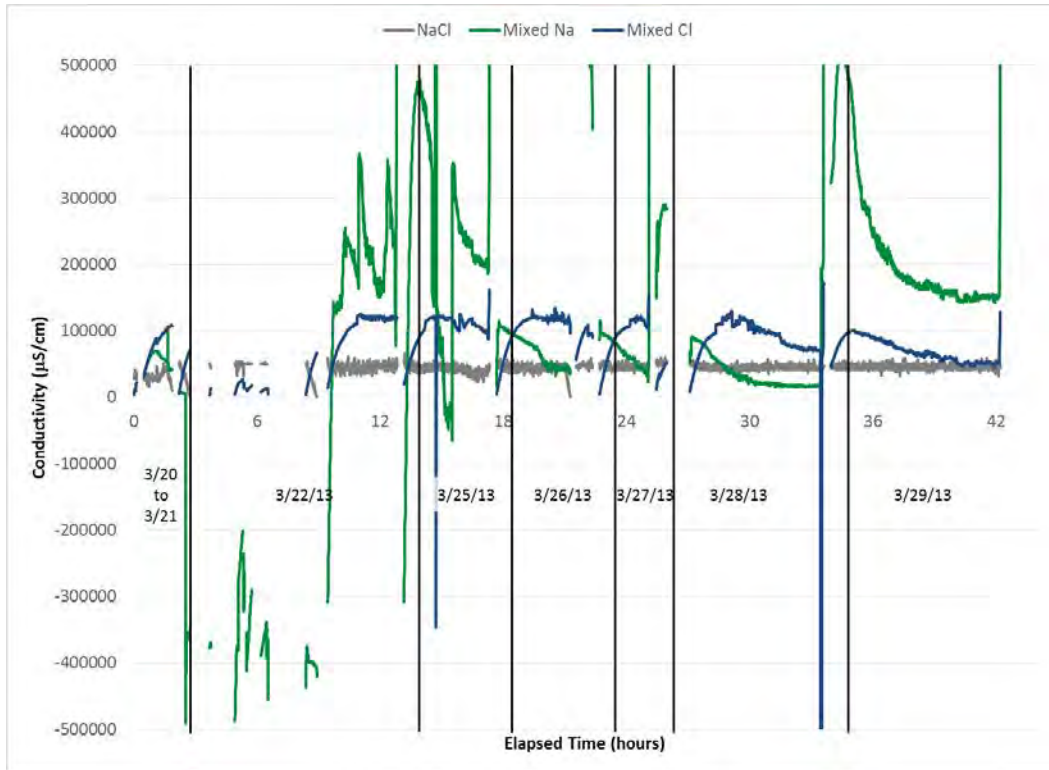


Figure B2-25.—High concentration EDM stream conductivity (preliminary experiments-1). (note: Mixed Cl & Mixed Na are not calibrated or scaled)

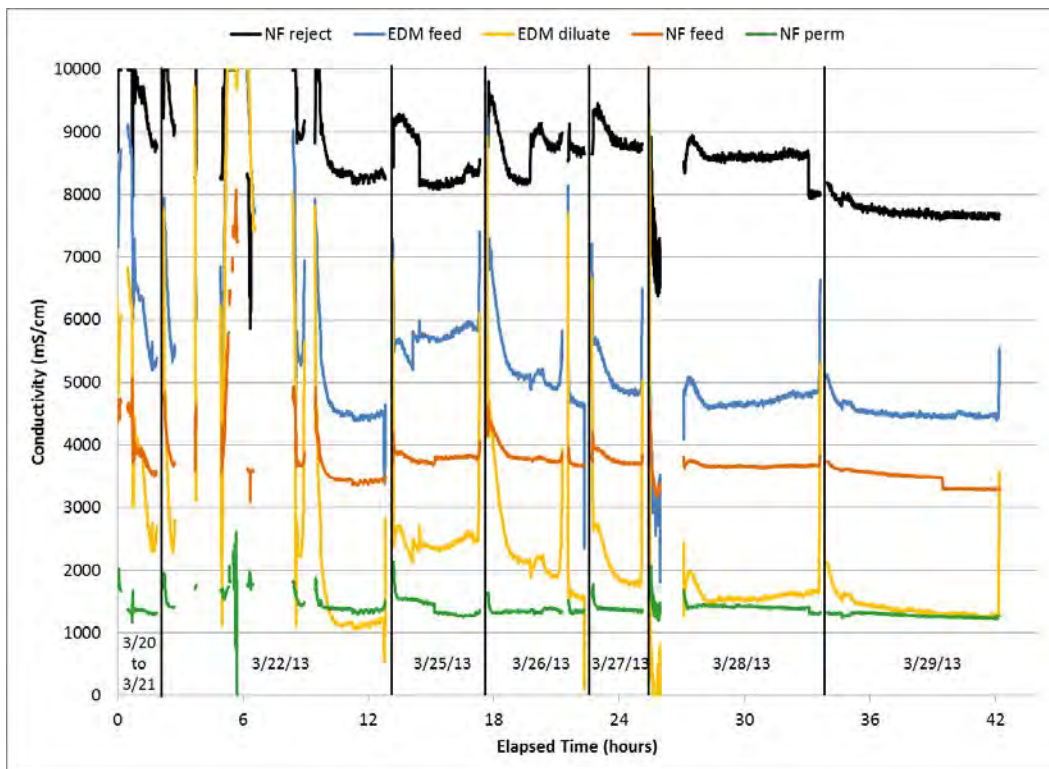


Figure B2-26.—Low concentration EDM stream conductivity (preliminary experiments-1).

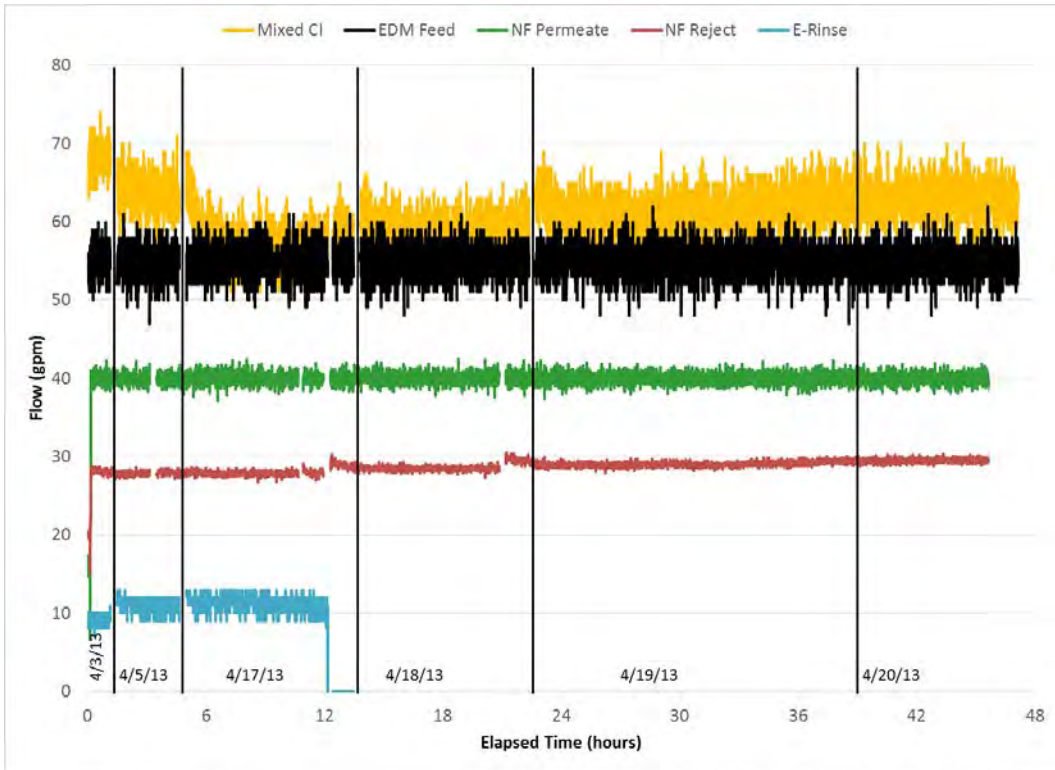


Figure B2-27.—EDM stream flows (preliminary experiments-2).

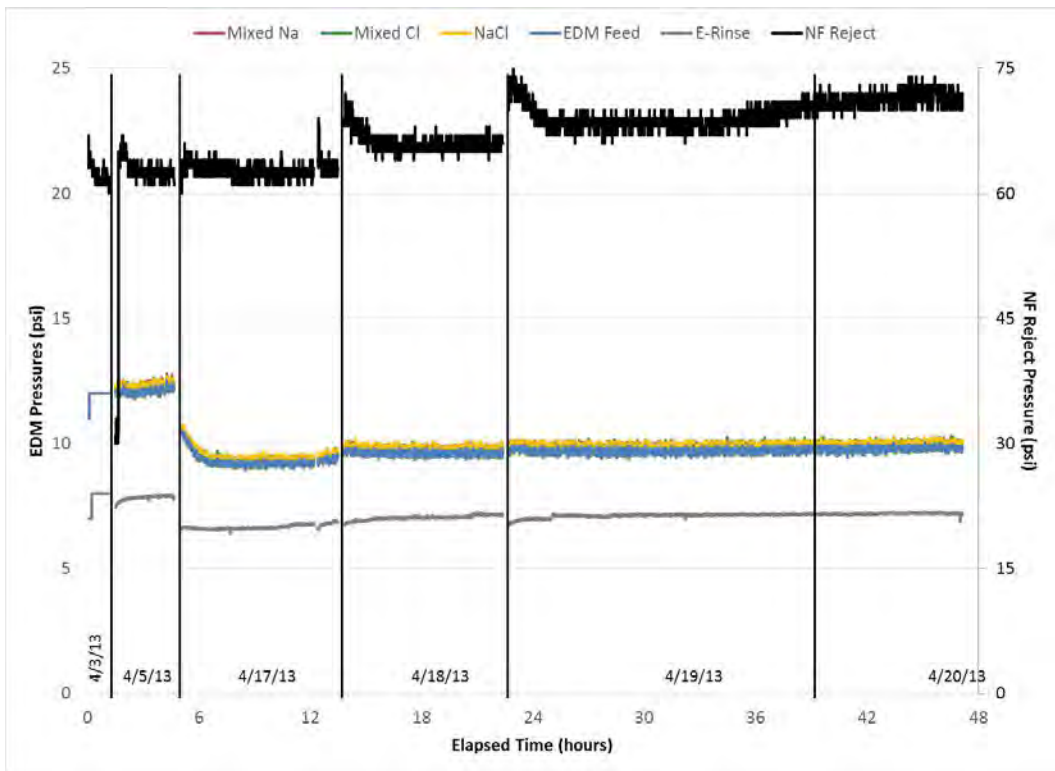


Figure B2-28.—EDM and NF reject stream pressures (preliminary experiments-2).

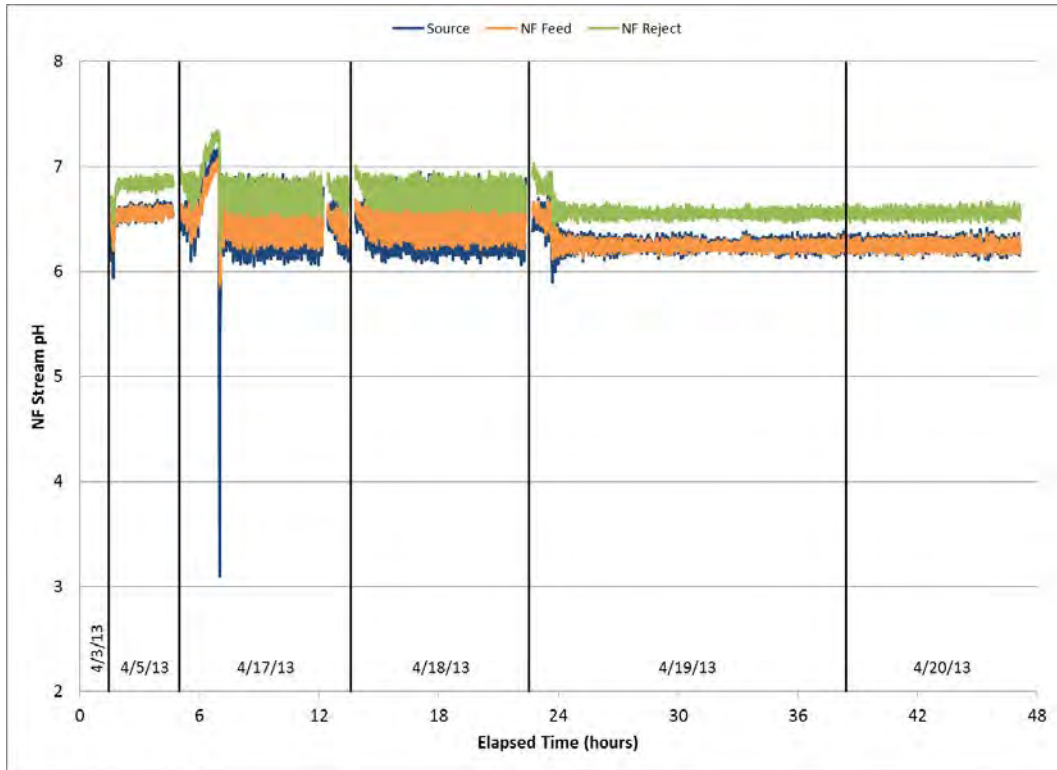


Figure B2-29.—NF stream pH (preliminary experiments-2).

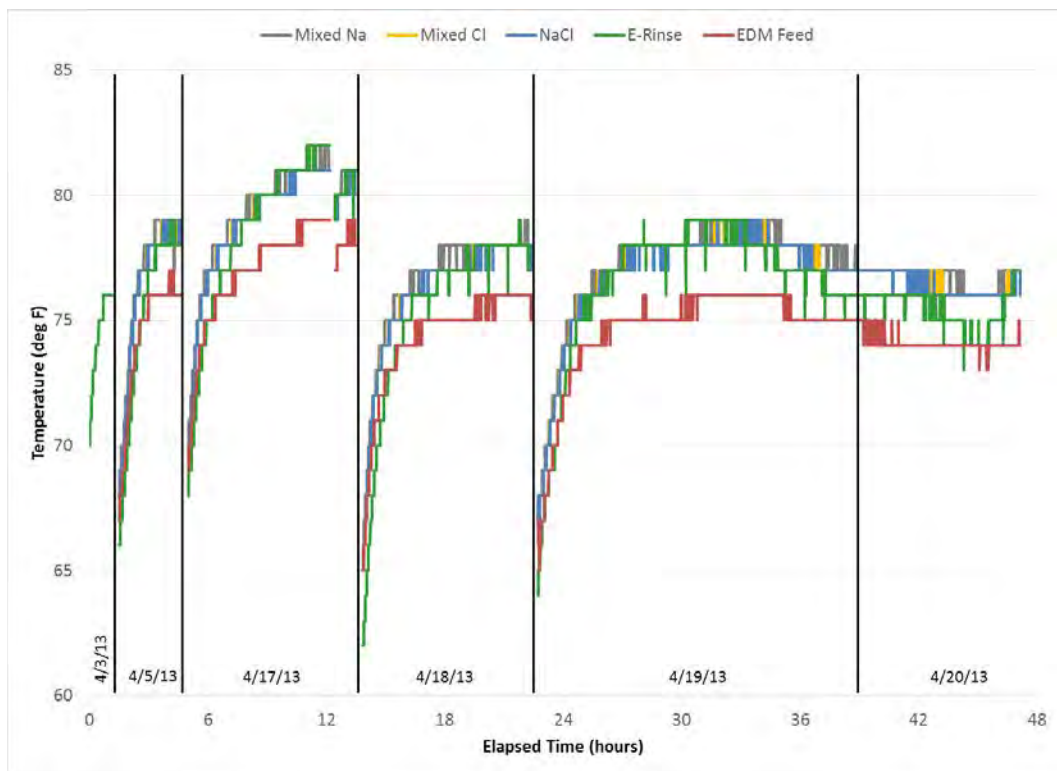


Figure B2-30.—EDM stream temperature (preliminary experiments-2).

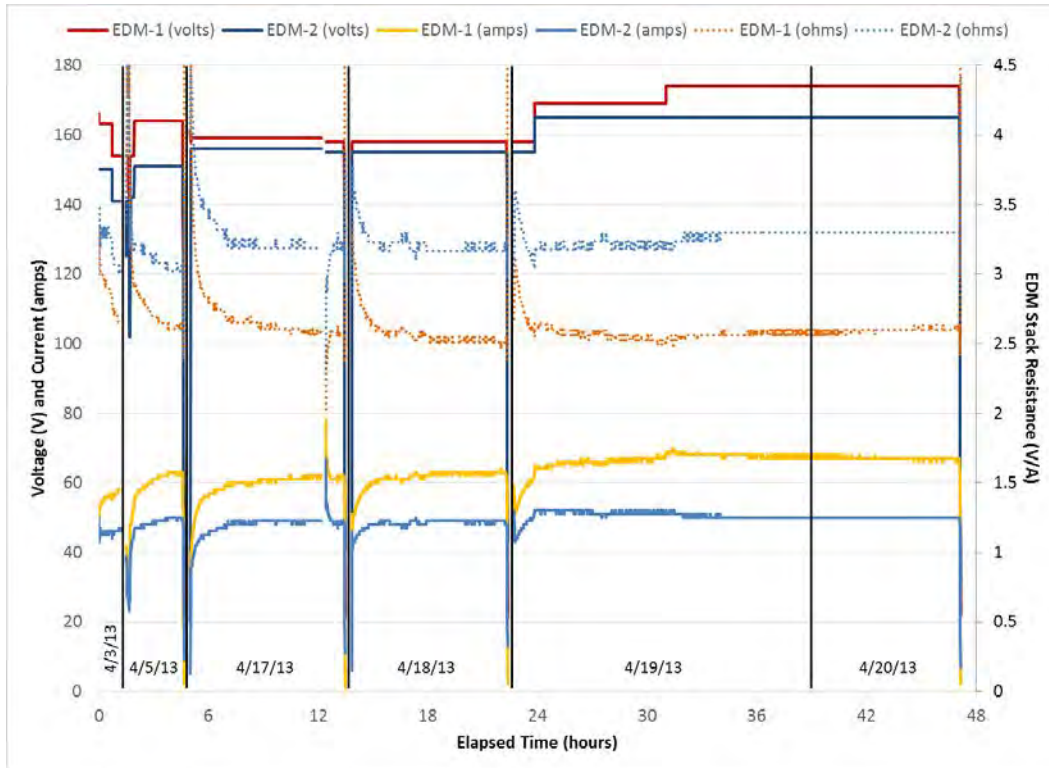


Figure B2-31.—EDM voltage, current, stack resistance (preliminary experiments-2).

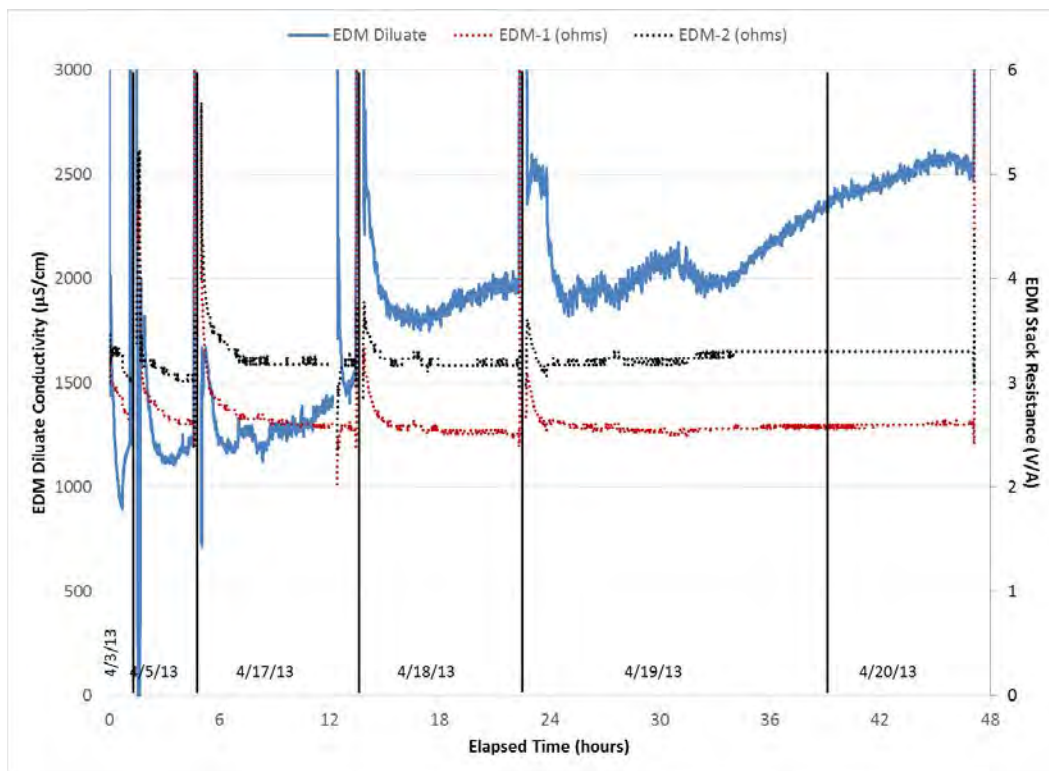


Figure B2-32.—EDM diluate conductivity and stack resistance (preliminary experiments-2).

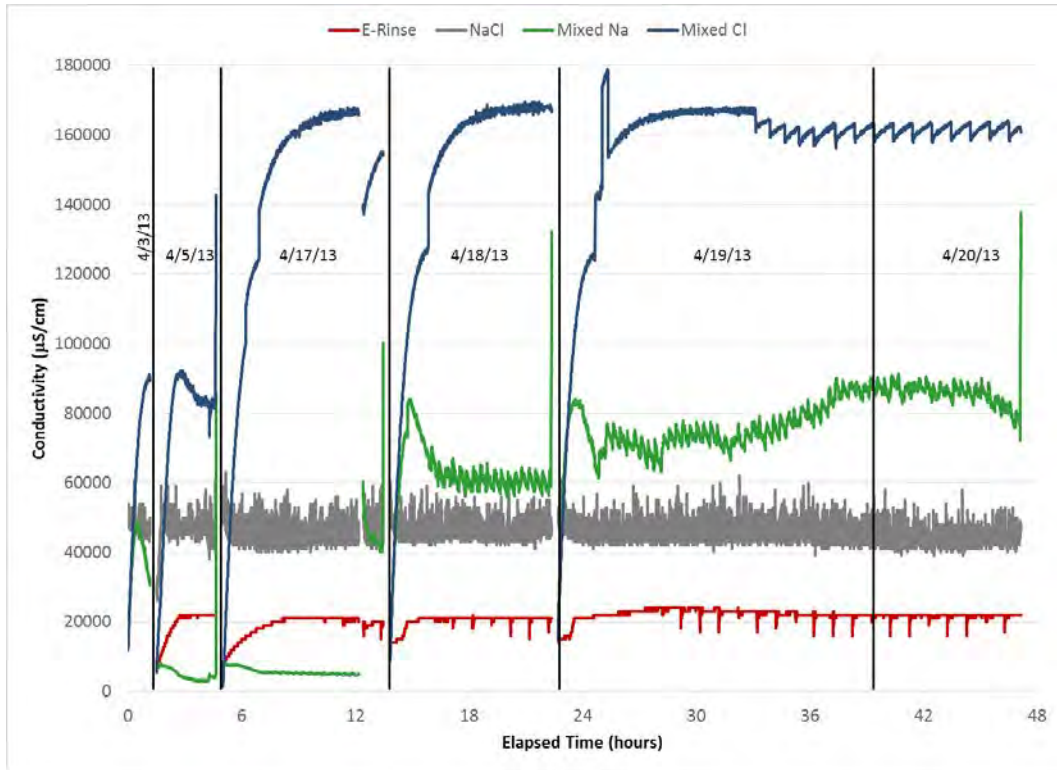


Figure B2-33.—High concentration EDM stream conductivity (preliminary experiments-2).
(note: Mixed Cl & Mixed Na are not calibrated or scaled)

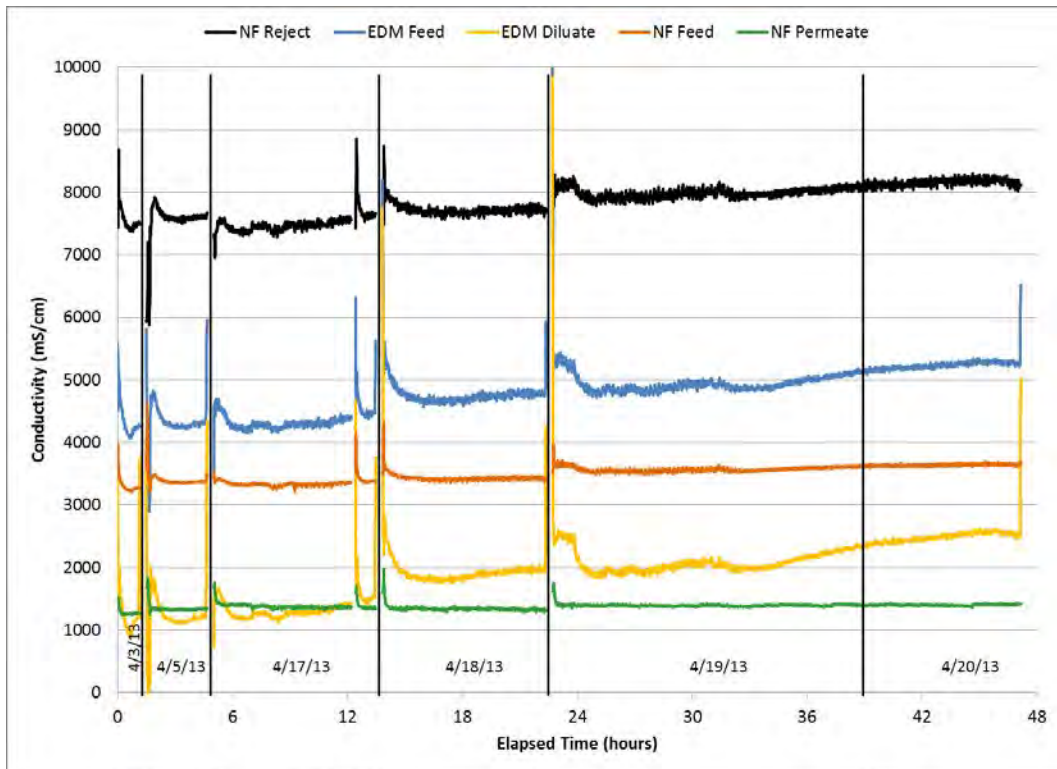


Figure B2-34.—Low concentration EDM stream conductivity (preliminary experiments-2).

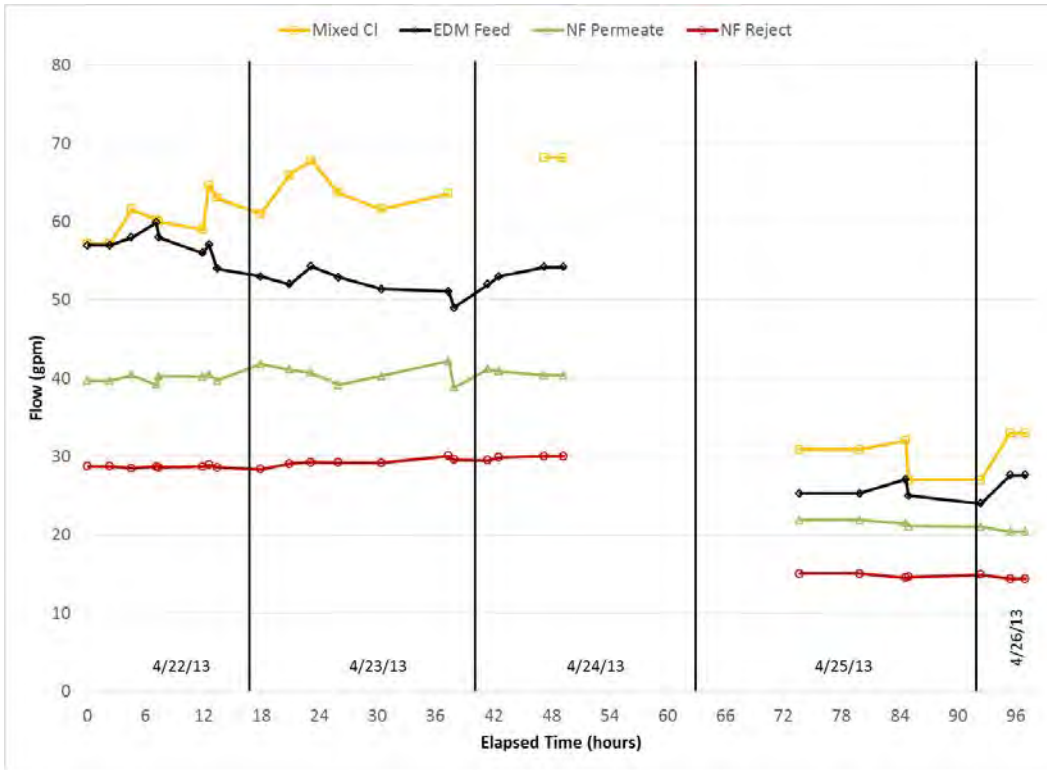


Figure B2-35.—EDM stream flows (experiments 1a + 1b).
(Datalogger failed during this period. Data was recorded on logsheets from online meters, probes, and gauges.)

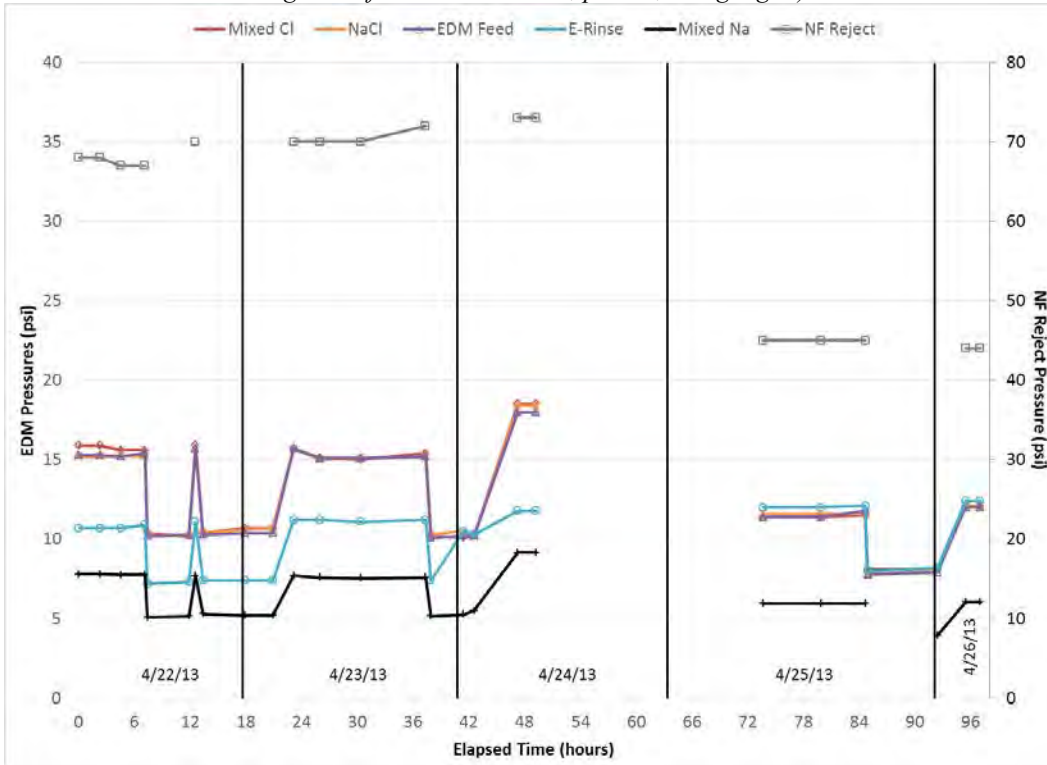


Figure B2-36.—EDM and NF Reject Stream Pressures (Experiments 1a + 1b)
(Datalogger failed during this period. Data was recorded on logsheets from online meters, probes, and gauges.)

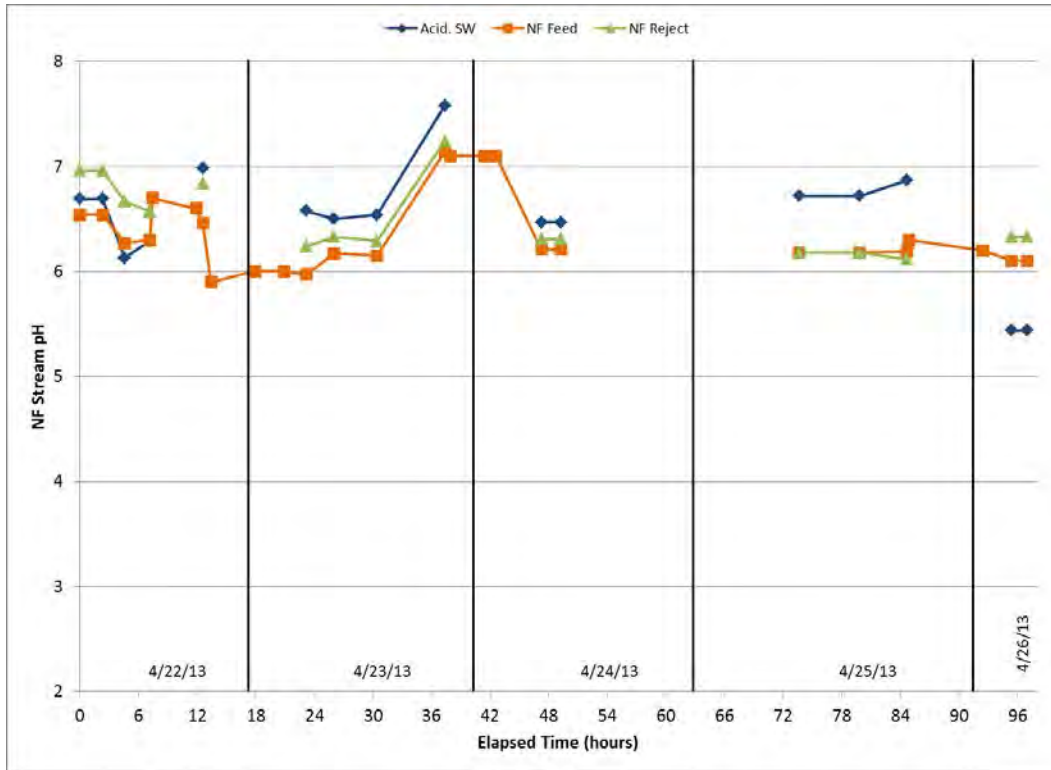


Figure B2-37.—NF stream pH (experiments 1a + 1b).
Datalogger failed during this period. Data was recorded on logsheets from online meters, probes, and gauges.

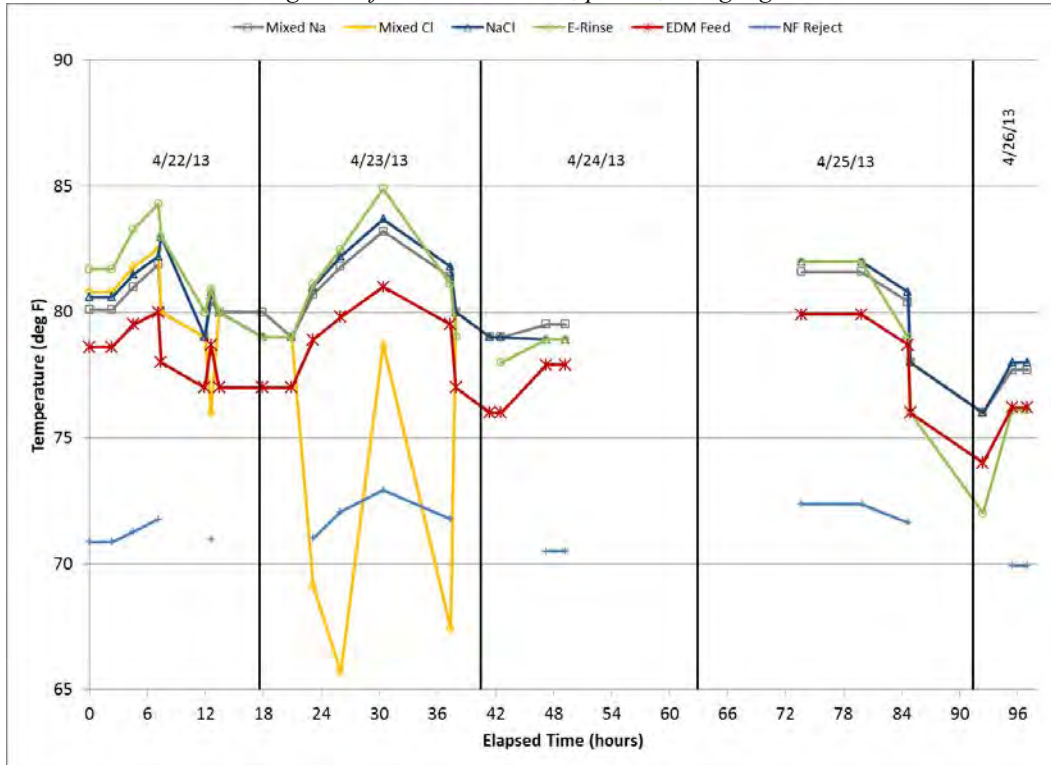


Figure B2-38.—EDM stream temperature (experiments 1a + 1b).
Datalogger failed during this period. Data was recorded on logsheets from online meters, probes, and gauges.

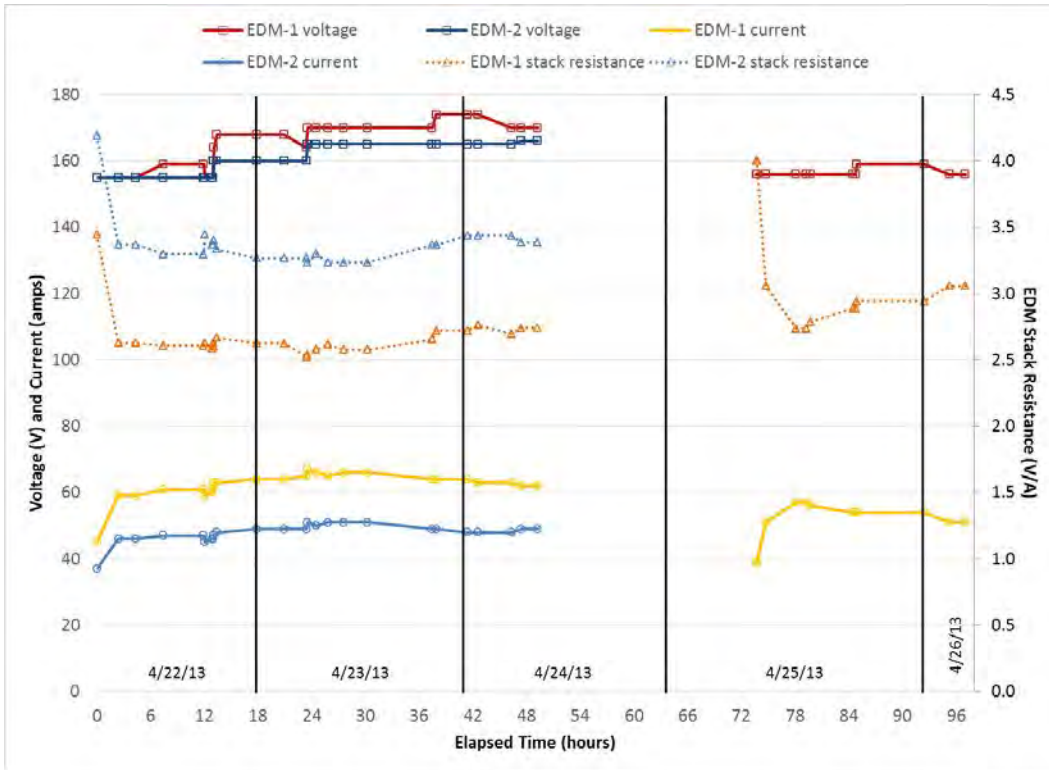


Figure B2-39.—EDM voltage, current, stack resistance (experiments 1a + 1b).

Datalogger failed during this period. Data was recorded on logsheets from online meters, probes, and gauges.

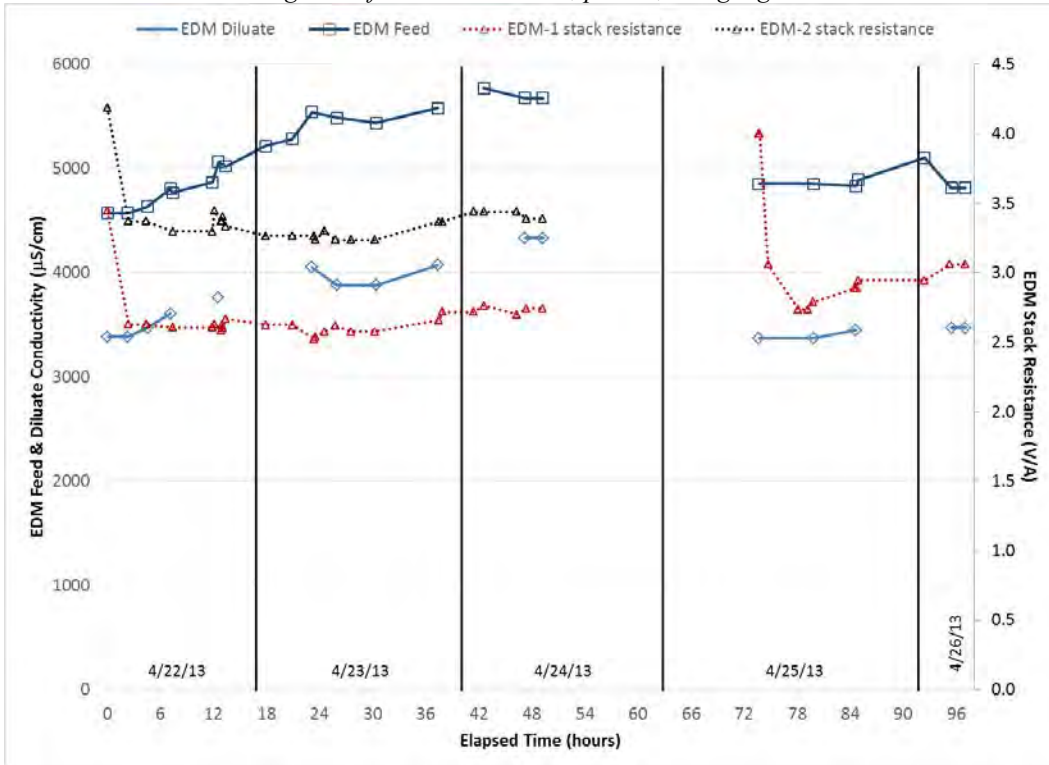


Figure B2-40.—EDM diluate conductivity and stack resistance (experiments 1a + 1b).

Datalogger failed during this period. Data was recorded on logsheets from online meters, probes, and gauges.

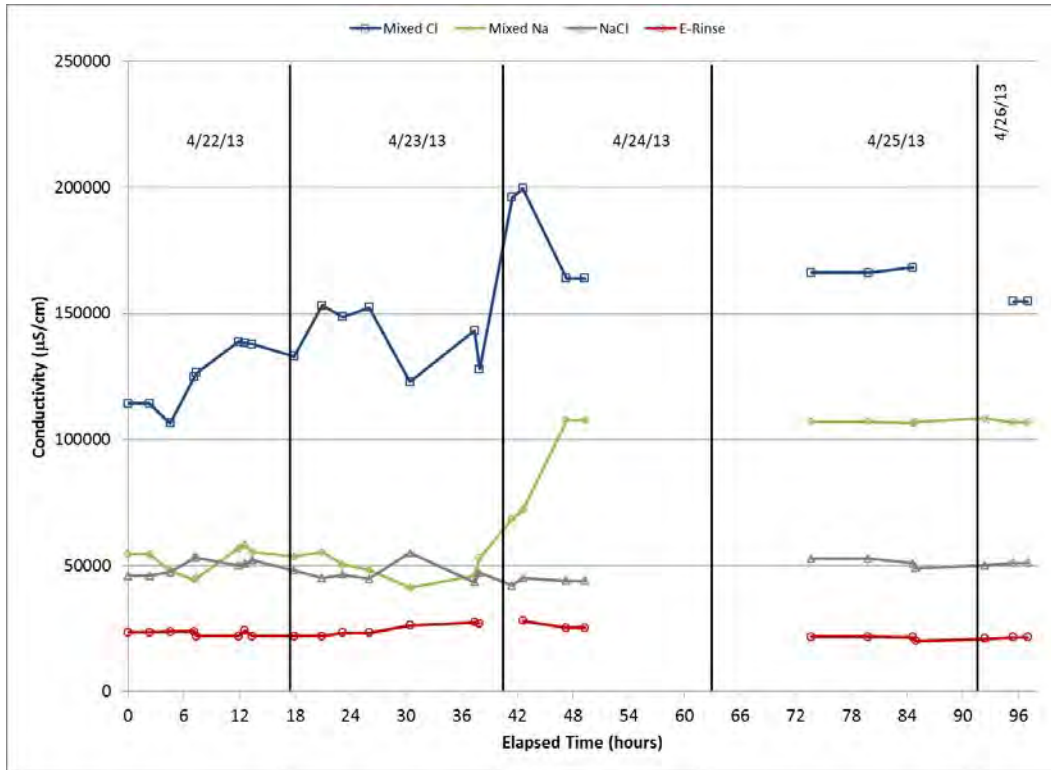


Figure B2-41.—High concentration EDM stream conductivity (experiments 1a + 1b).
(note: Mixed Cl & Mixed Na are not calibrated or scaled)
Datalogger failed during this period. Data was recorded on
logsheets from online meters, probes, and gauges.

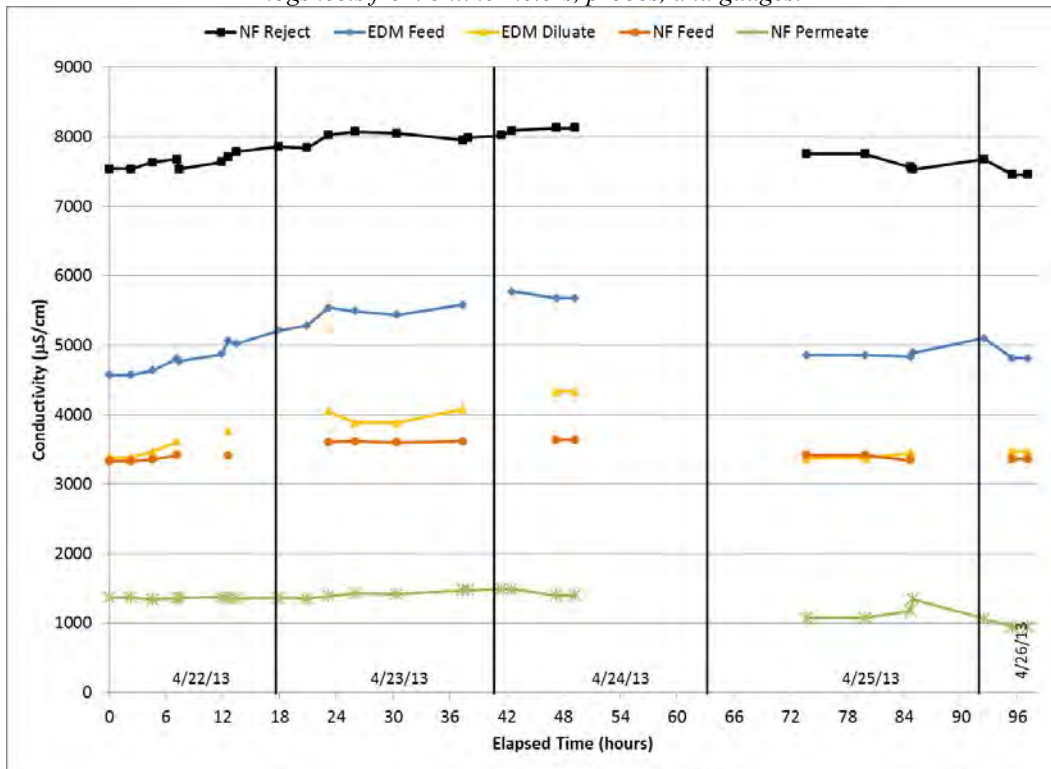


Figure B2-42.—Low concentration EDM stream conductivity (experiments 1a + 1b).
Datalogger failed during this period. Data was recorded on
logsheets from online meters, probes, and gauges.

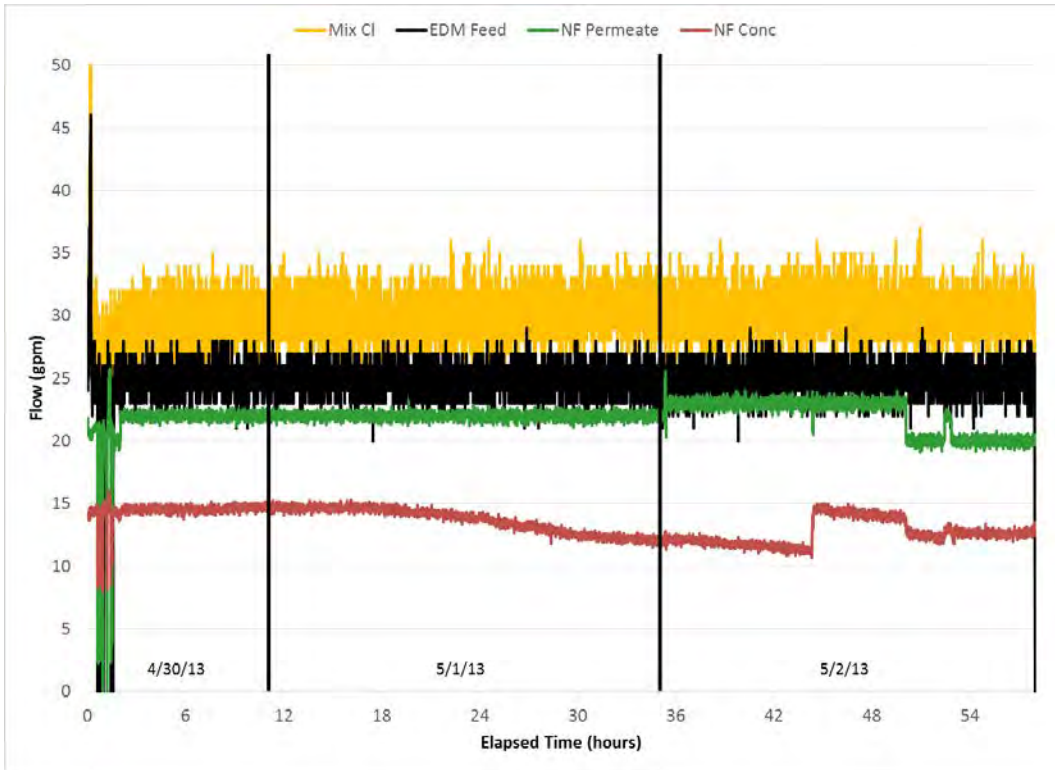


Figure B2-43.—EDM stream flows (experiment 2).

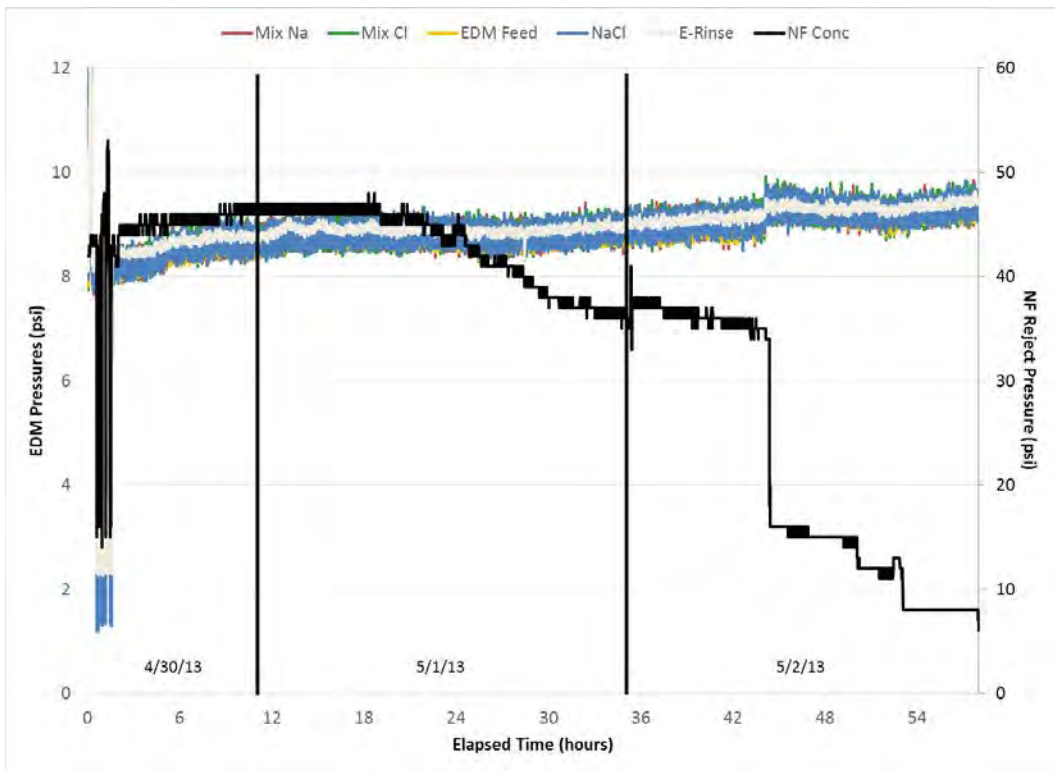


Figure B2-44.—EDM and NF reject stream pressures (experiment 2).

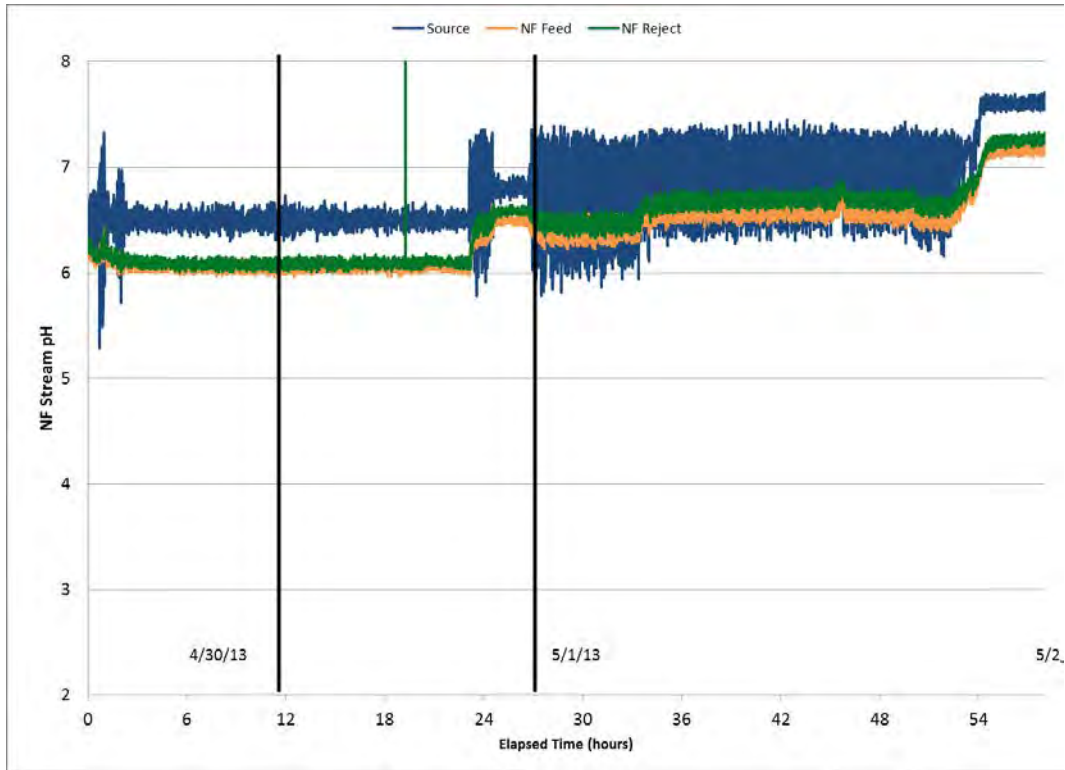


Figure B2-45.—NF stream pH (experiment 2).

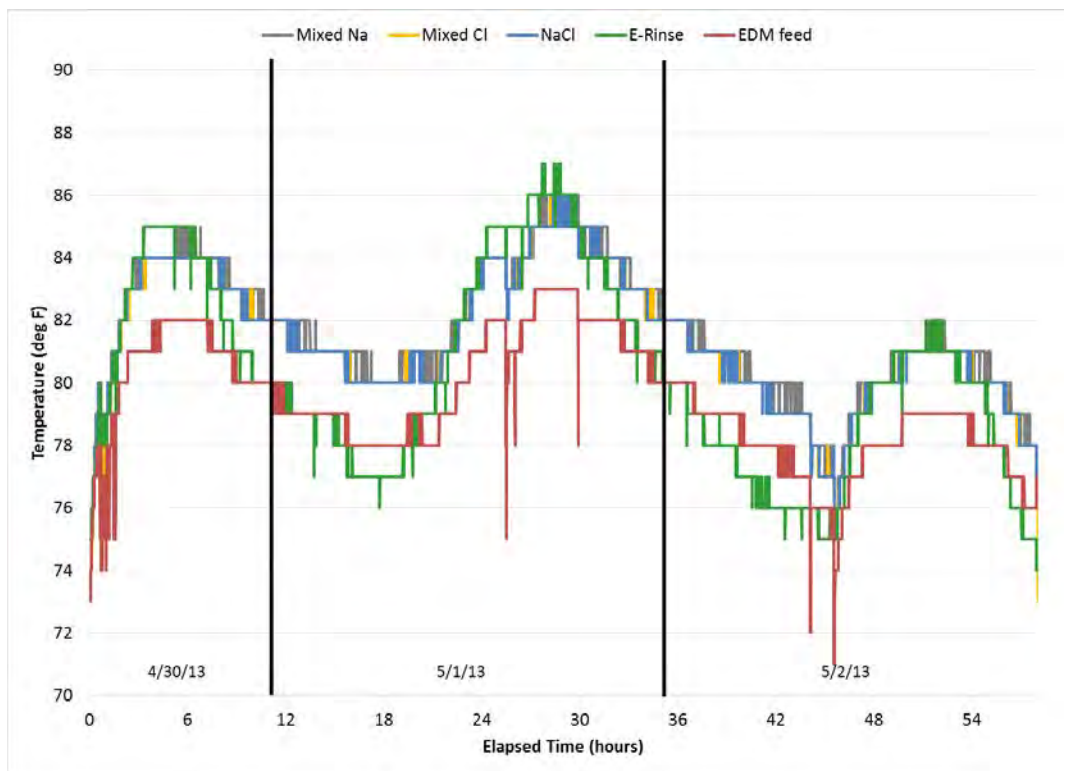


Figure B2-46.—EDM stream temperature (experiment 2)

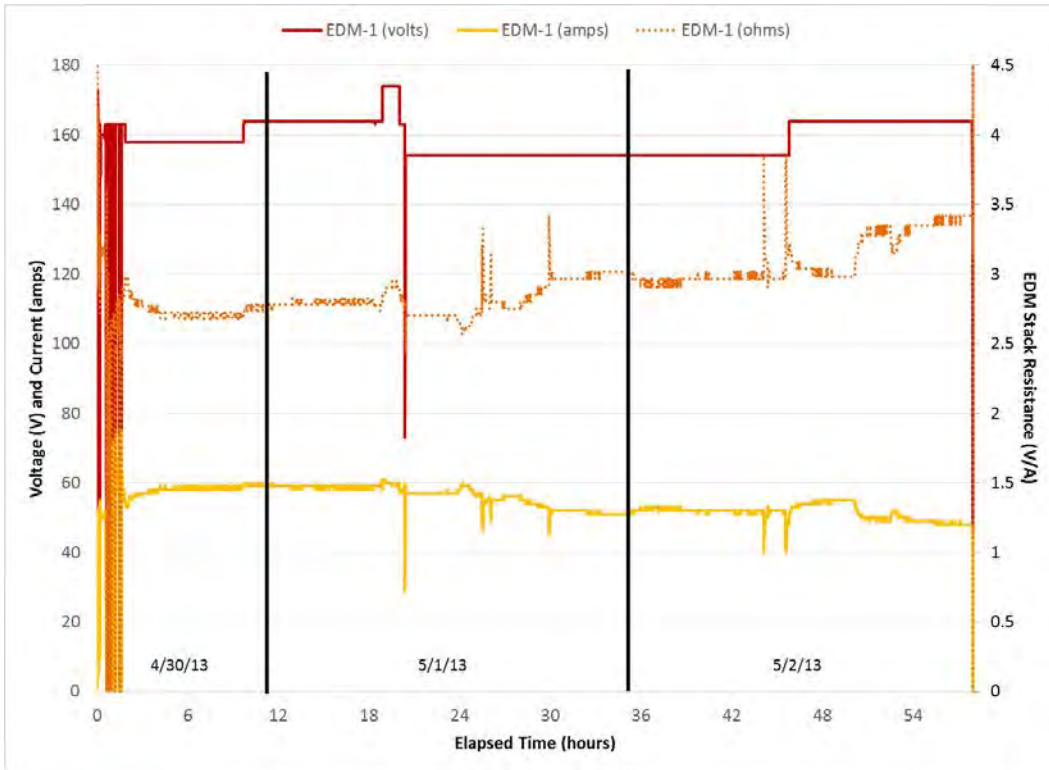


Figure B2-47.—EDM voltage, current, stack resistance (experiment 2).

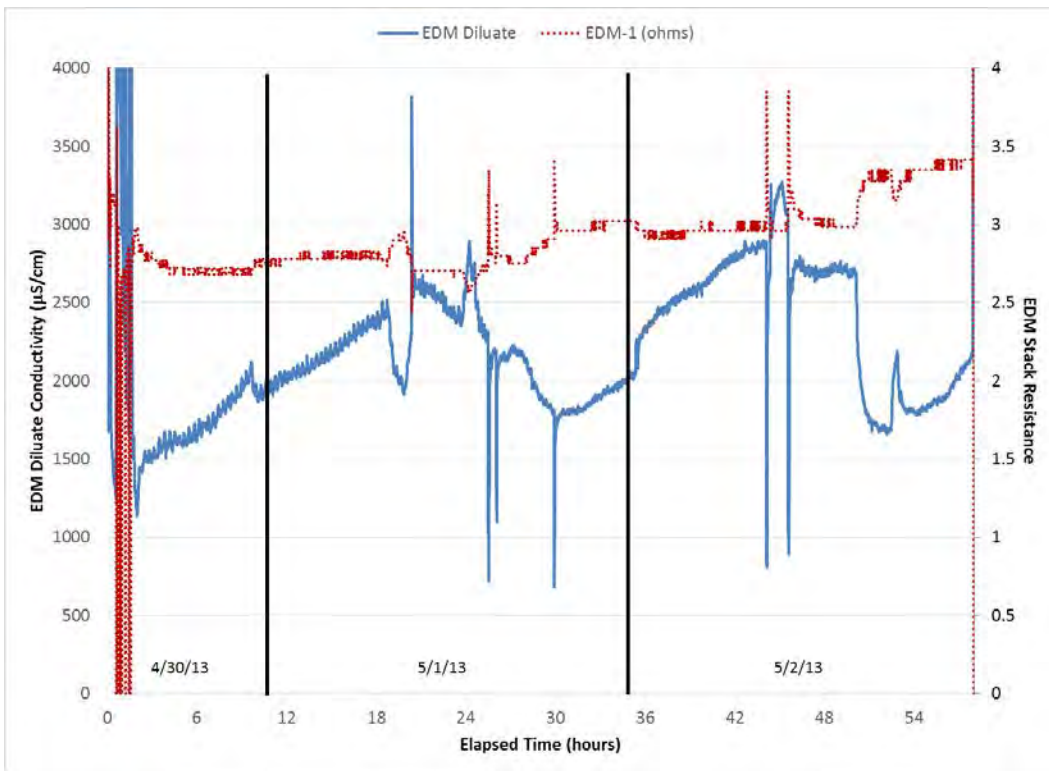


Figure B2-48.—EDM diluate conductivity and stack resistance (experiment 2).

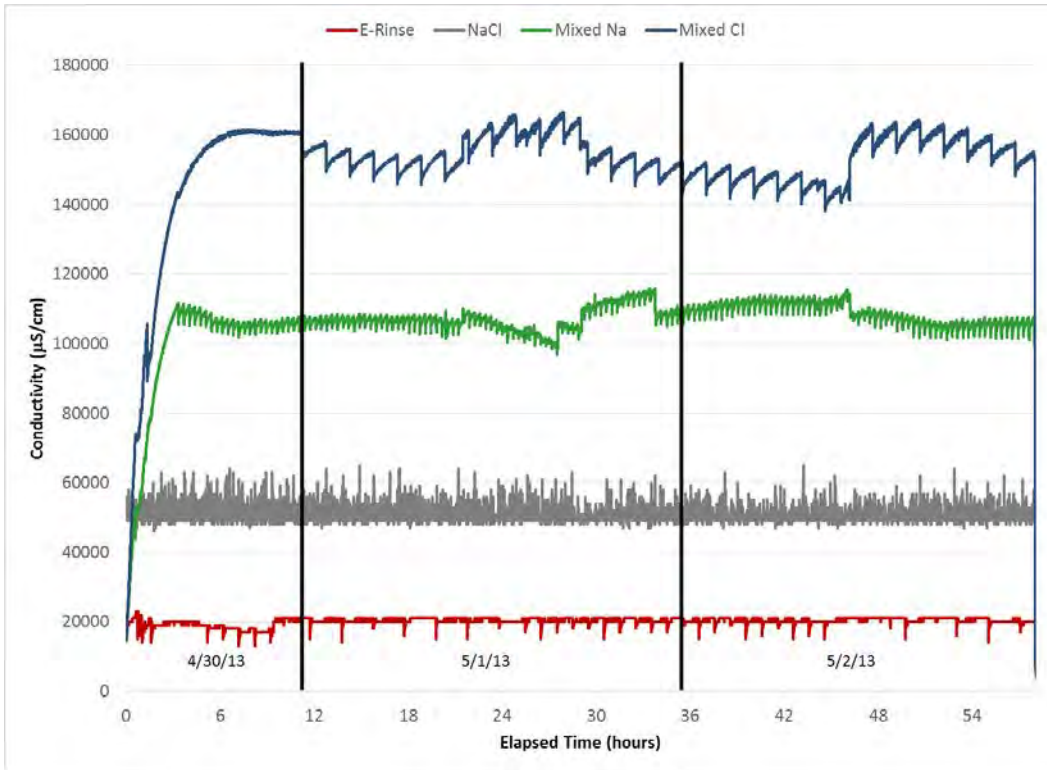


Figure B2-49.—High concentration EDM stream conductivity (experiment 2).

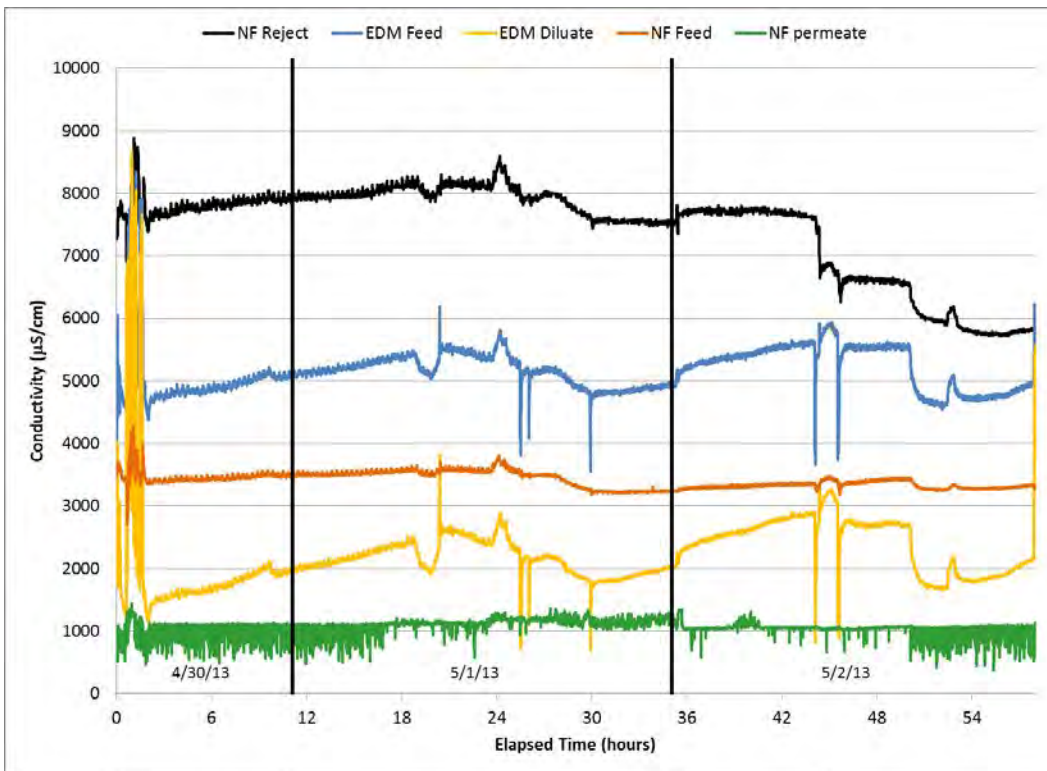


Figure B2-50.—Low concentration EDM stream conductivity (experiment 2).

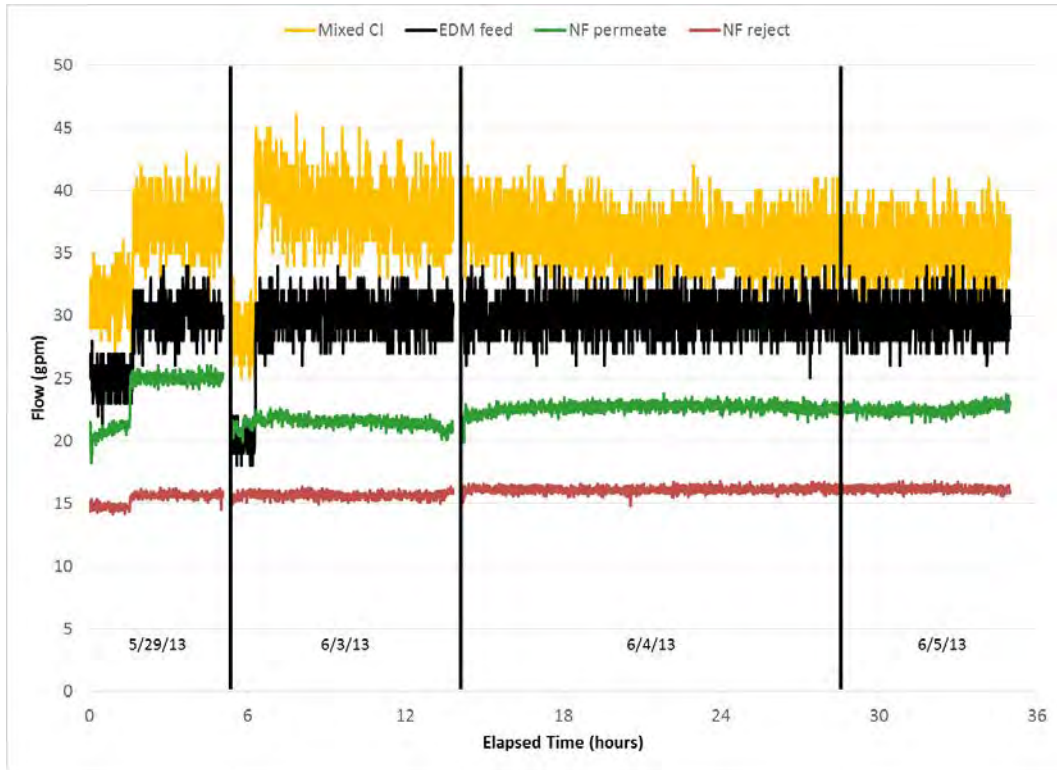


Figure B2-51.—EDM stream flows (experiment 3).

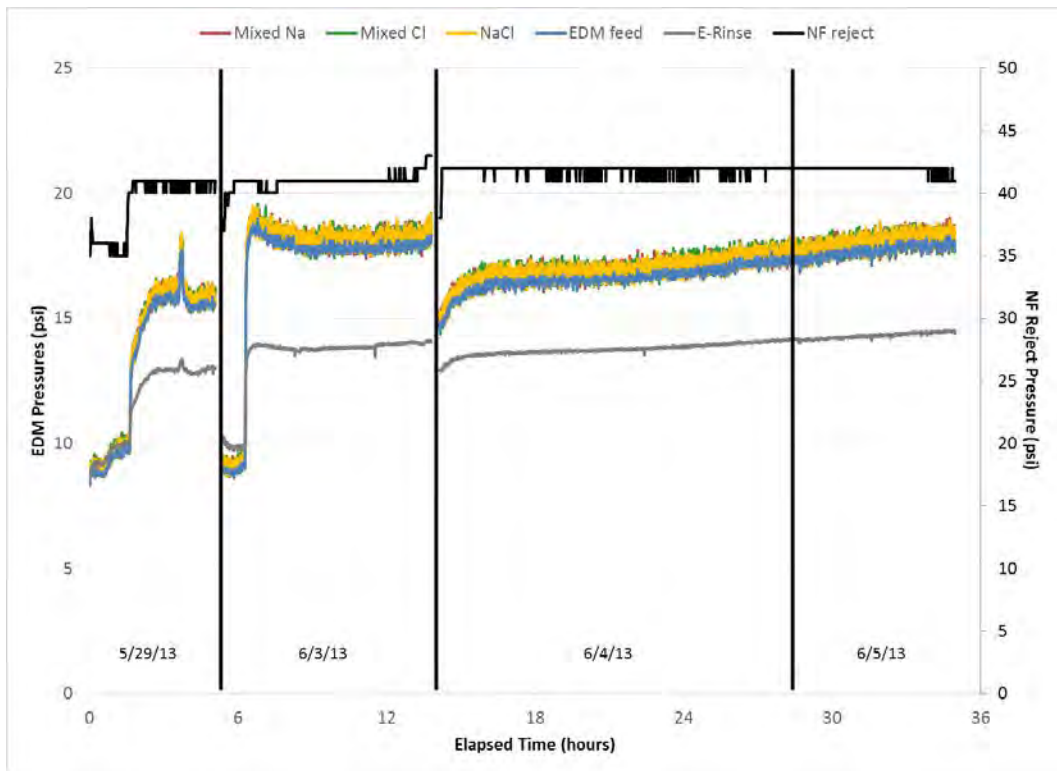


Figure B2-52.—EDM and NF reject stream pressures (experiment 3).

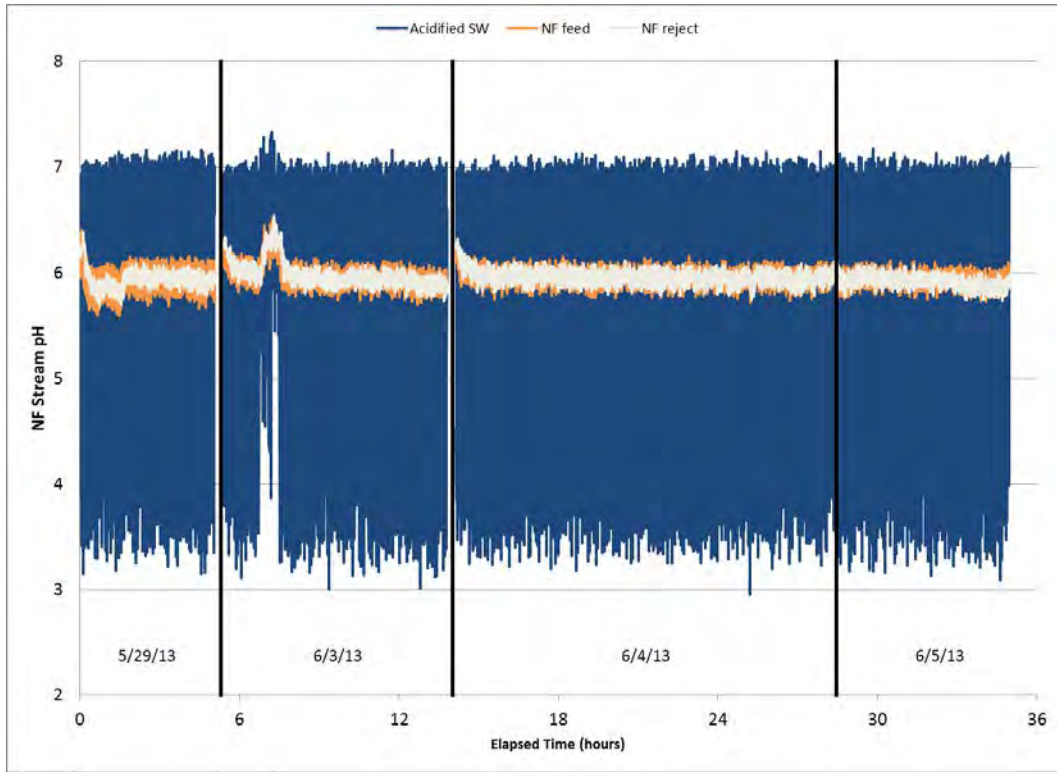


Figure B2-53.—NF stream pH experiment 3).
(note: 93% H₂SO₄ added at this point. pH control not modifiable in the field, which increased variation in pH.)

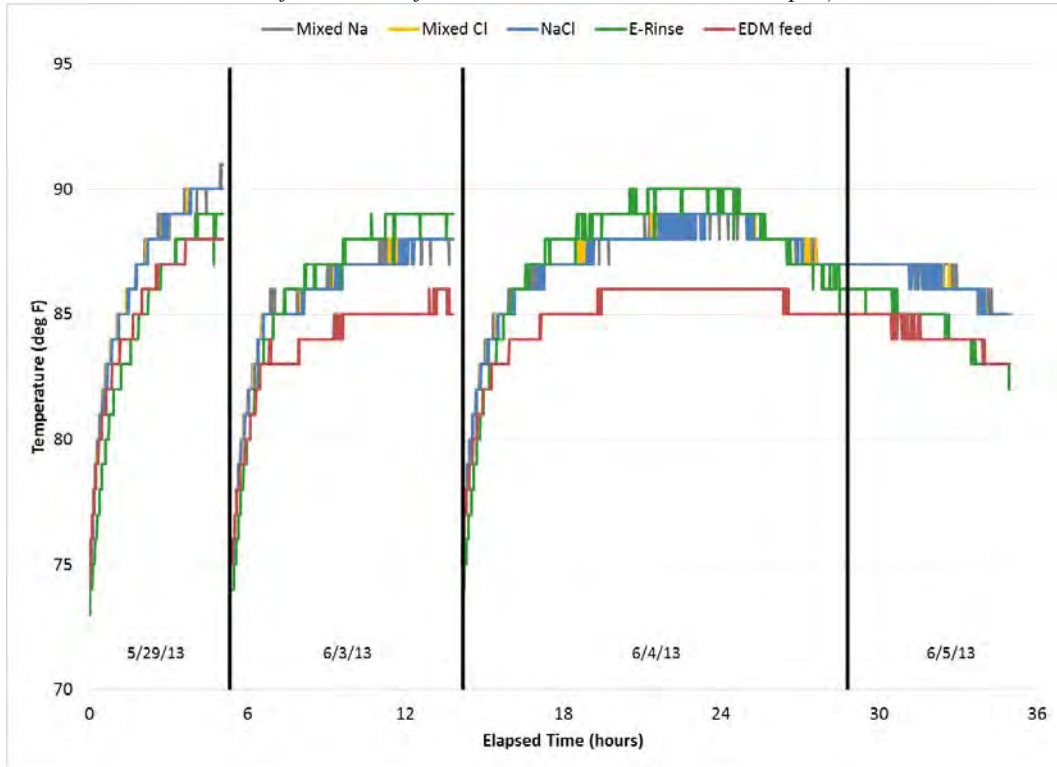


Figure B2-54.—EDM stream temperature (experiment 3).

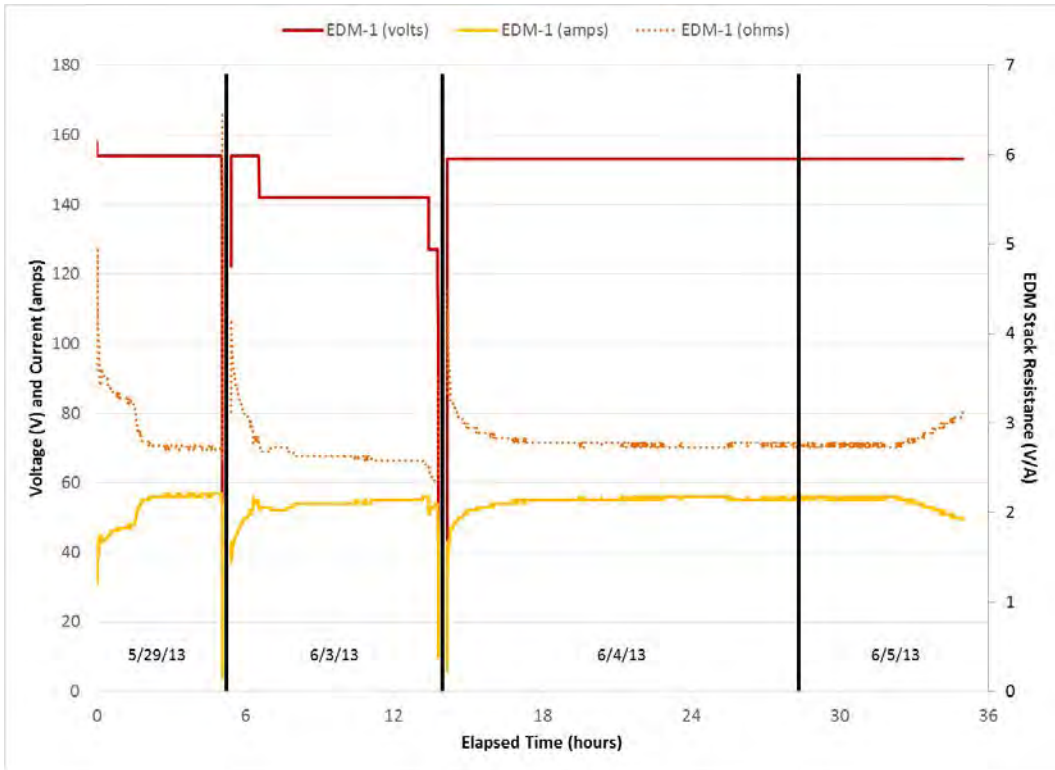


Figure B2-55.—EDM voltage, current, stack resistance (experiment 3).

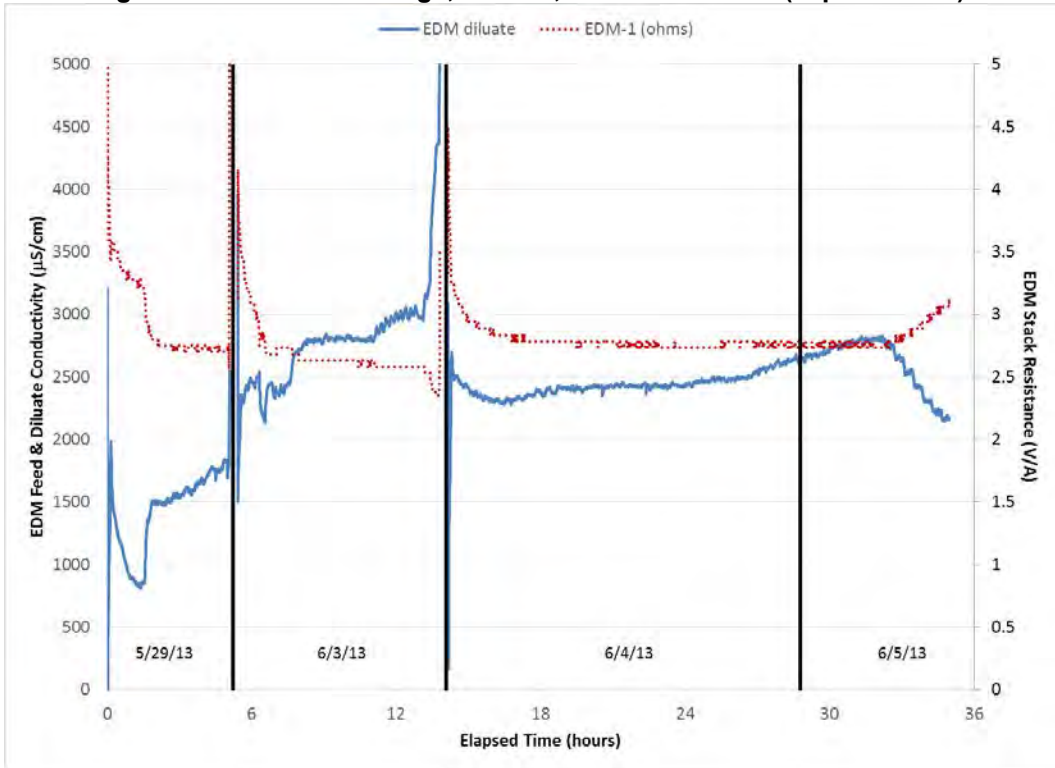


Figure B2-56.—EDM diluate conductivity and stack resistance (experiment 3).

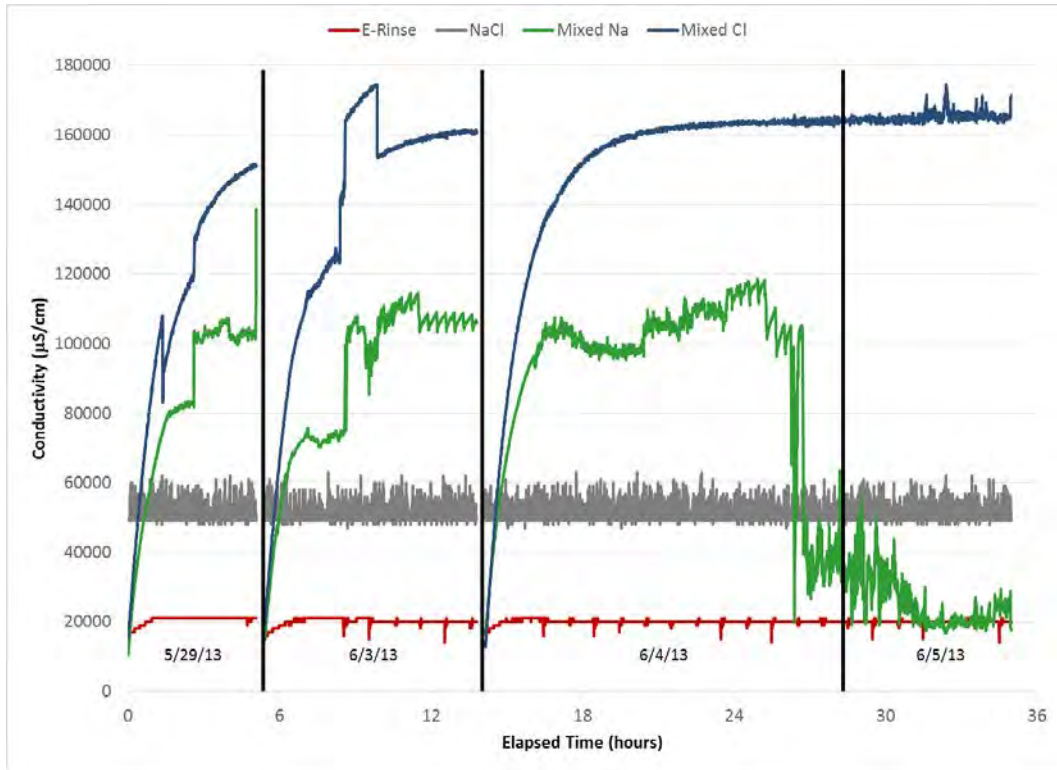


Figure B2-57.—High concentration EDM stream conductivity (experiment 3).
(note: Mixed Na not calibrated or scaled after 6/4/13).

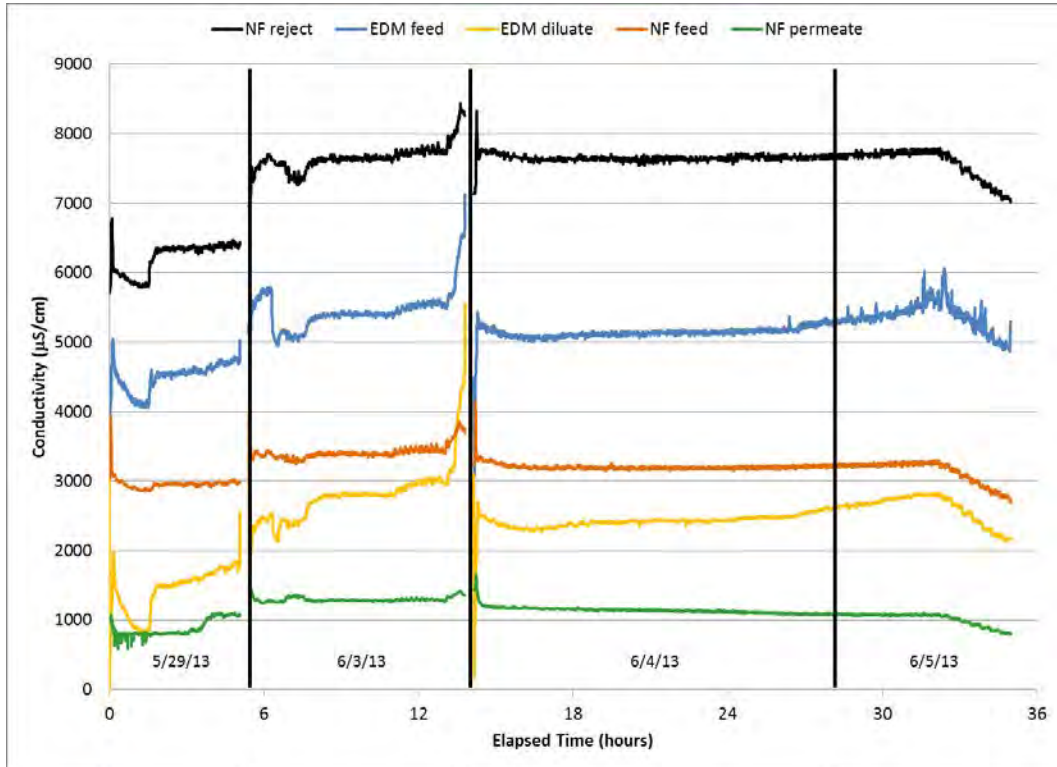


Figure B2-58.—Low concentration EDM stream conductivity (experiment 3).

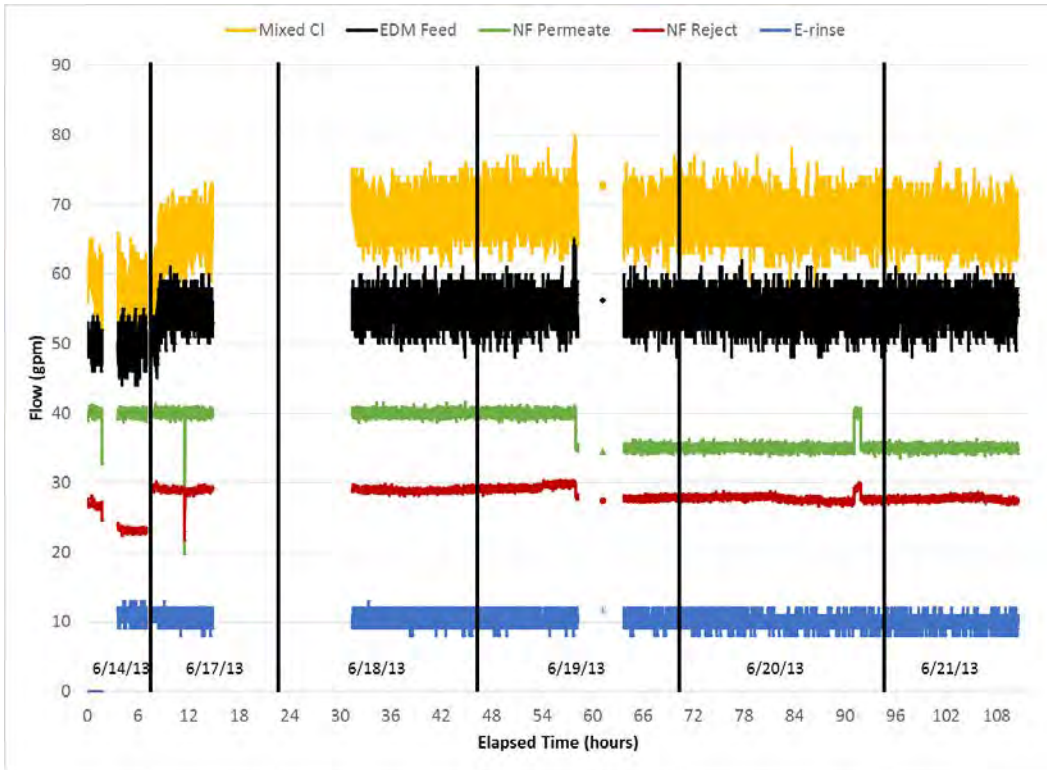


Figure B2-59.—EDM stream flows (experiment 4).

Datalogger failed on 6/19/13. Data was recorded on logsheets from online meters, probes, and gauges.

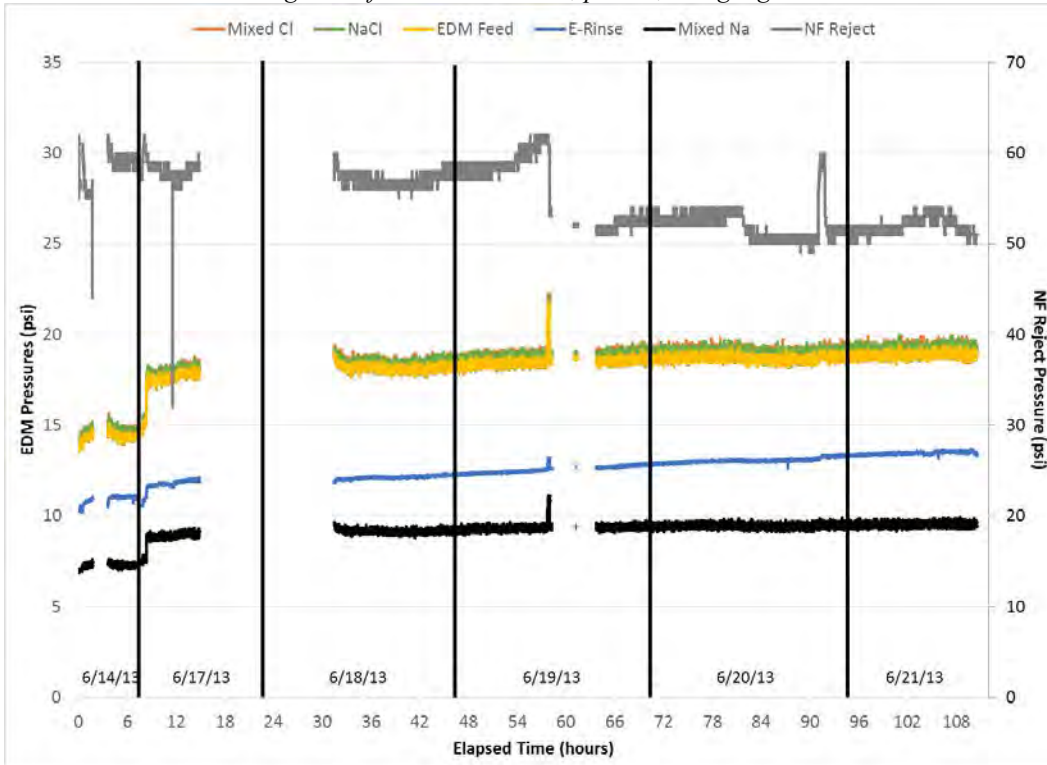


Figure B2-60.—EDM and NF reject stream pressures (experiment 4).

Datalogger failed on 6/19/13. Data was recorded on logsheets from online meters, probes, and gauges.

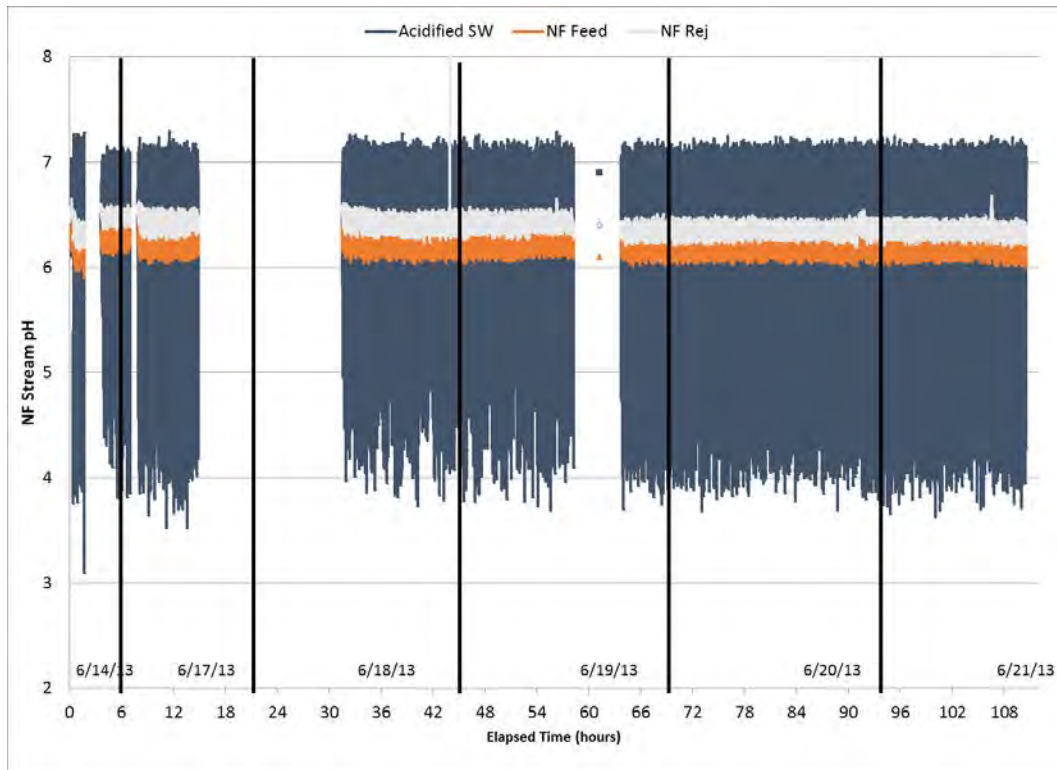


Figure B2-61.—NF stream pH (experiment 4).
*Datalogger failed on 6/19/13. Data was recorded on logsheets from online meters, probes, and gauges.
 Note: 93% H₂SO₄ added at this point. pH control not modifiable in the field, which increased variation in pH*

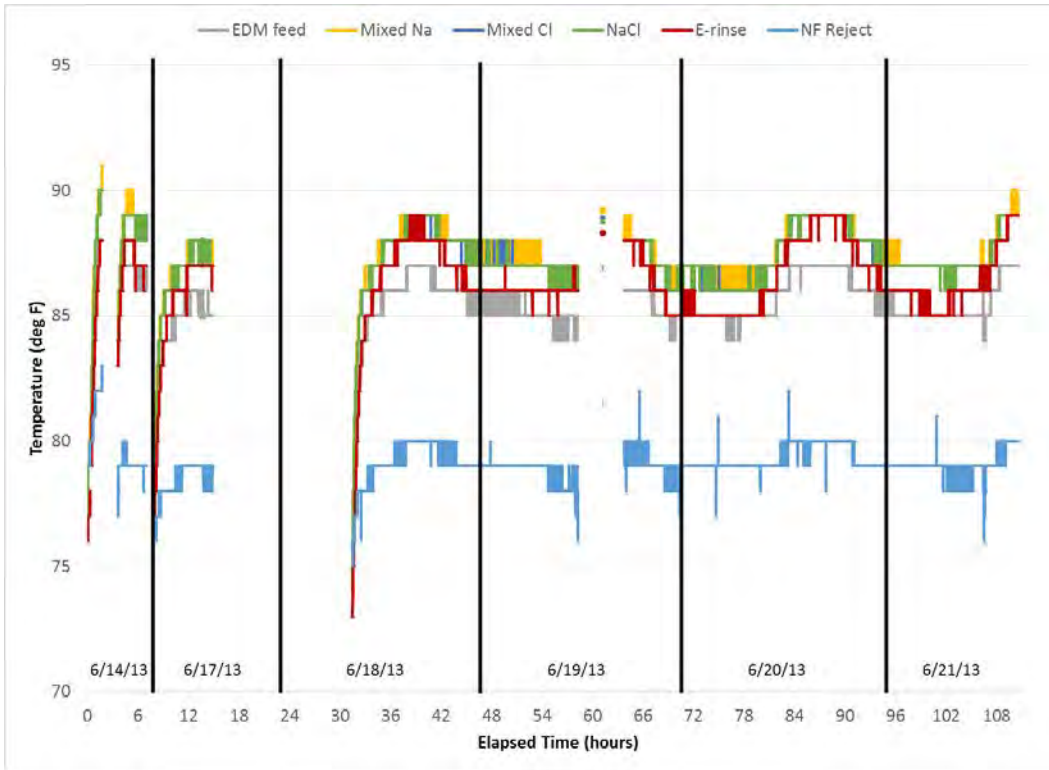


Figure B2-62.—EDM stream temperature (experiment 4).

Datalogger failed on 6/19/13. Data was recorded on logsheets from online meters, probes, and gauges.

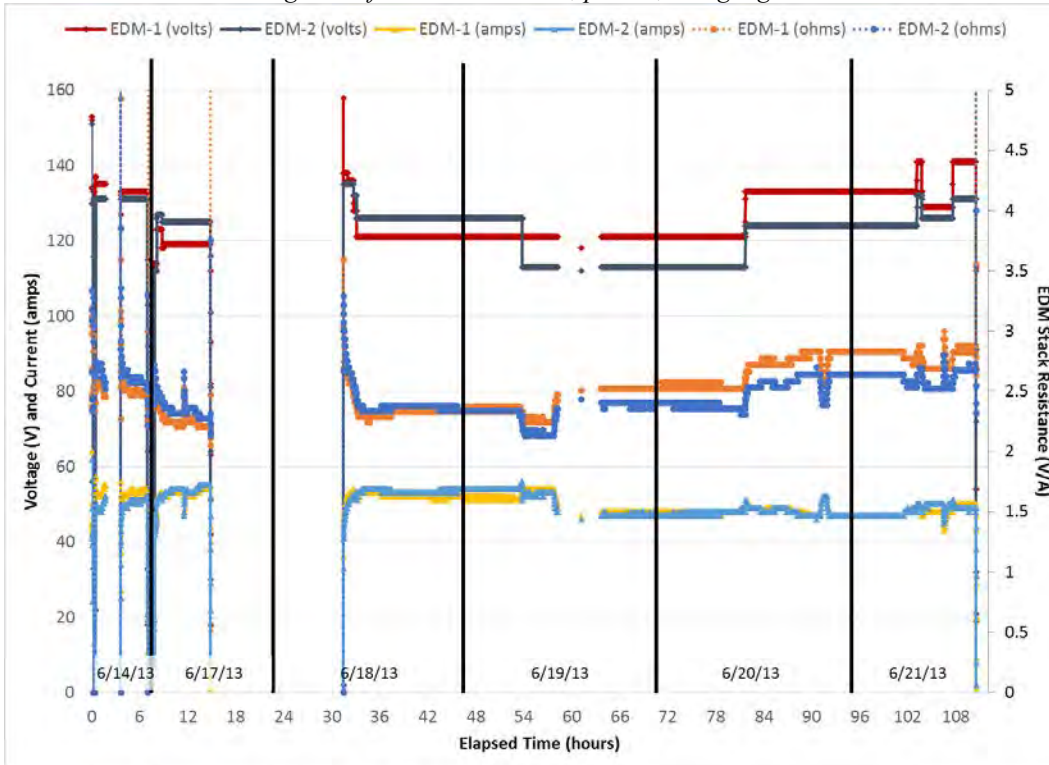


Figure B2-63.—EDM voltage, current, stack resistance (experiment 4).

Datalogger failed on 6/19/13. Data was recorded on logsheets from online meters, probes, and gauges.

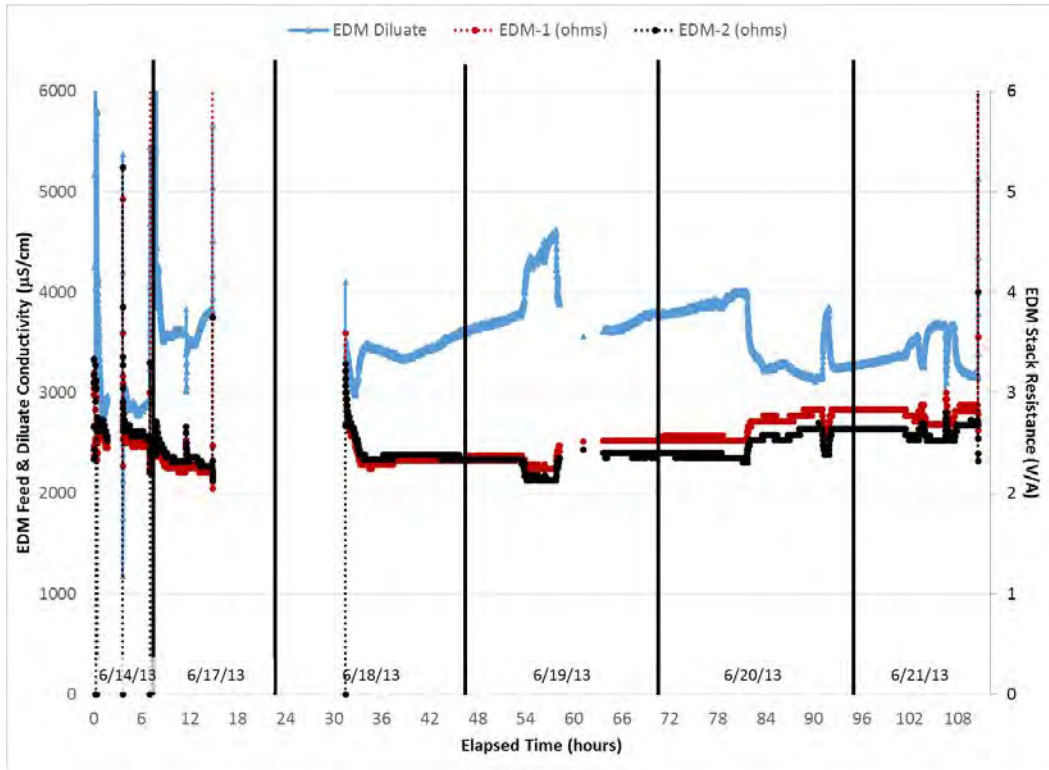


Figure B2-64.—EDM diluate conductivity and stack resistance (experiment 4).

Datalogger failed on 6/19/13. Data was recorded on logsheets from online meters, probes, and gauges.

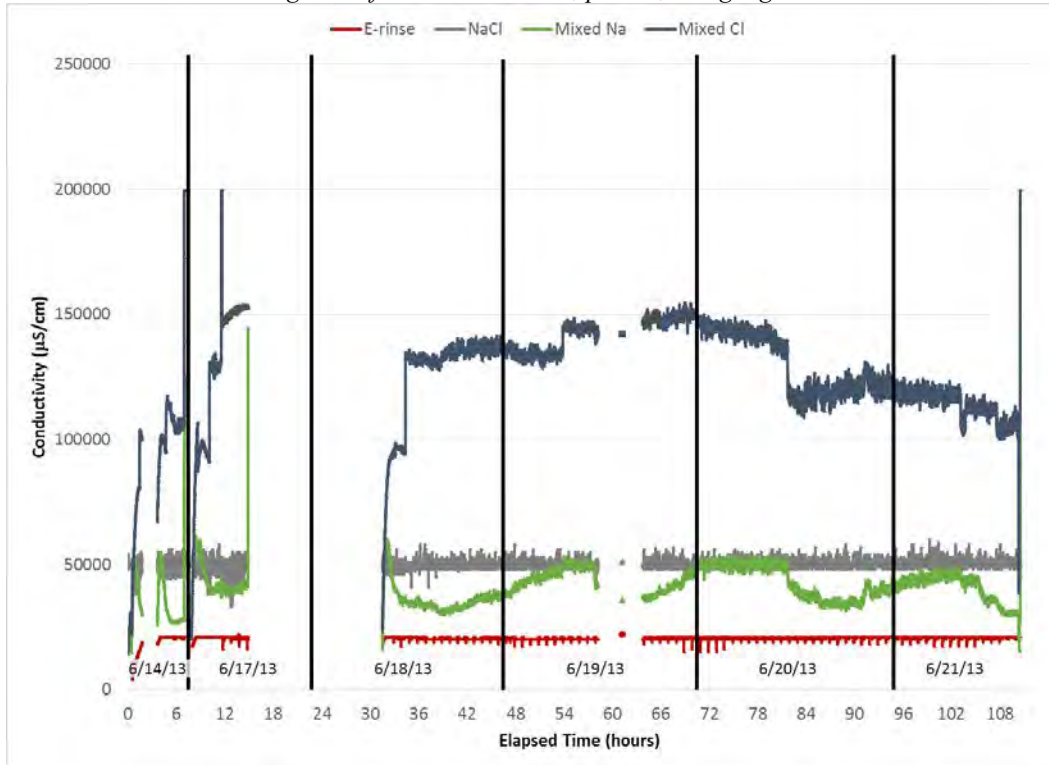


Figure B2-65.—High concentration EDM stream conductivity (experiment 4).

*(note: Mixed Cl & Mixed Na are not calibrated or scaled)
Datalogger failed on 6/19/13. Data was recorded on logsheets from online meters, probes, and gauges.*

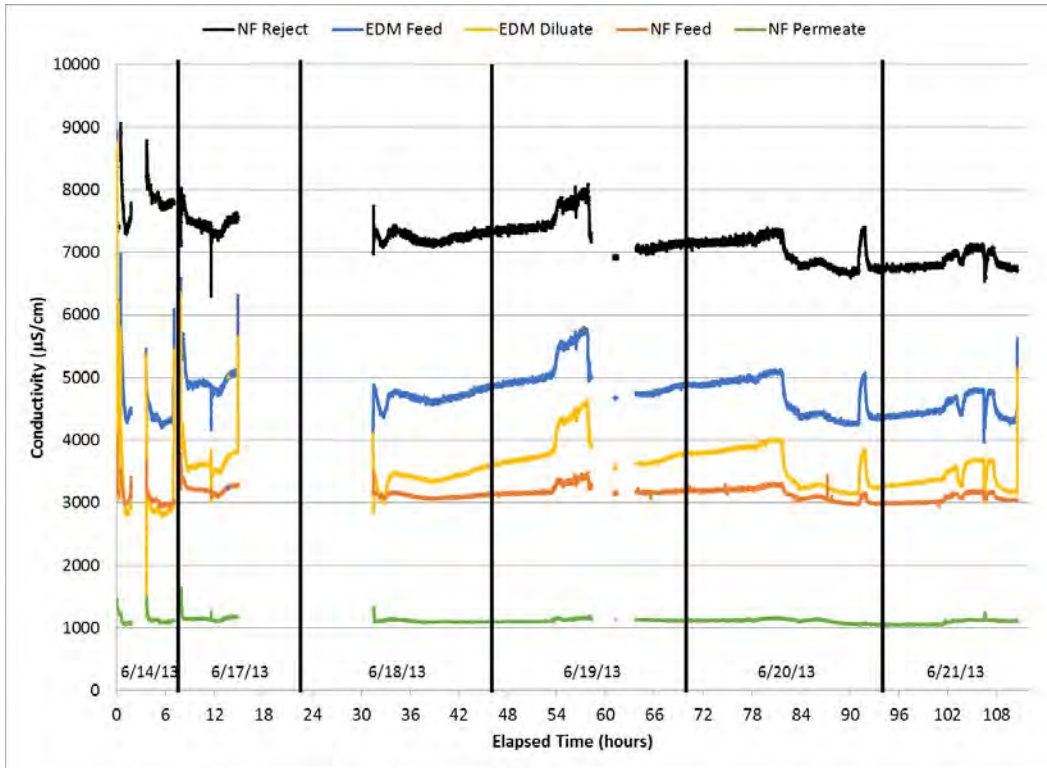


Figure B2-66.—Low concentration EDM stream conductivity (experiment 4).
Datalogger failed on 6/19/13. Data was recorded on logsheets from online meters, probes, and gauges.

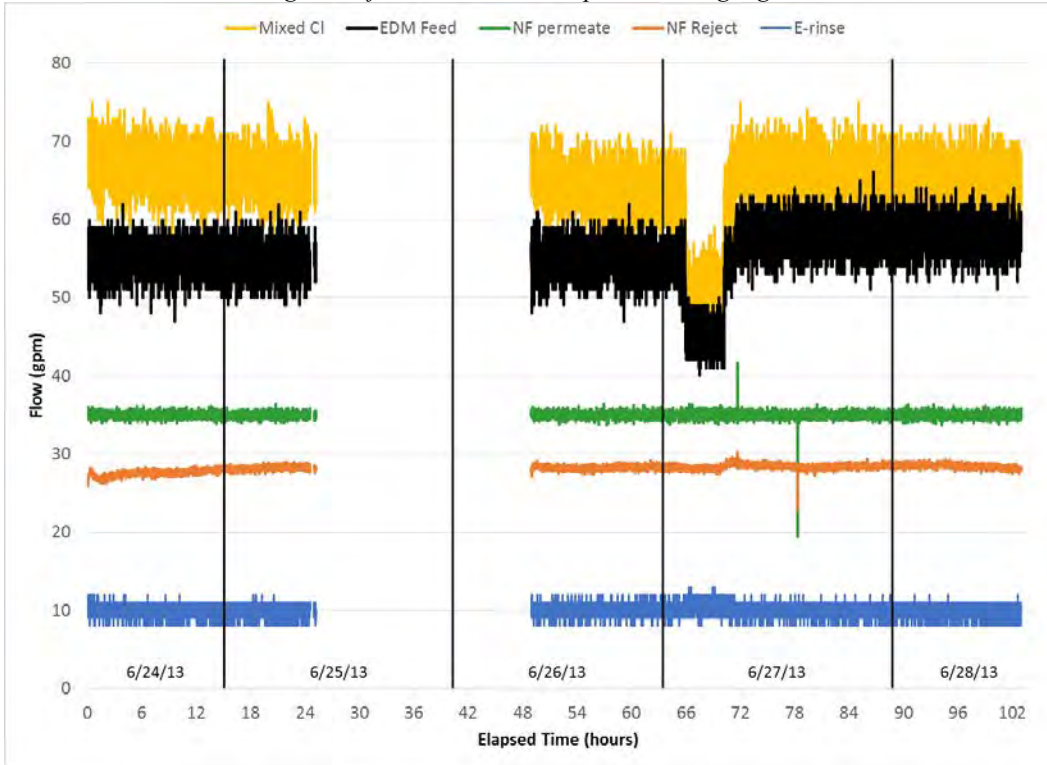


Figure B2-67.—EDM stream flows (experiment 5).

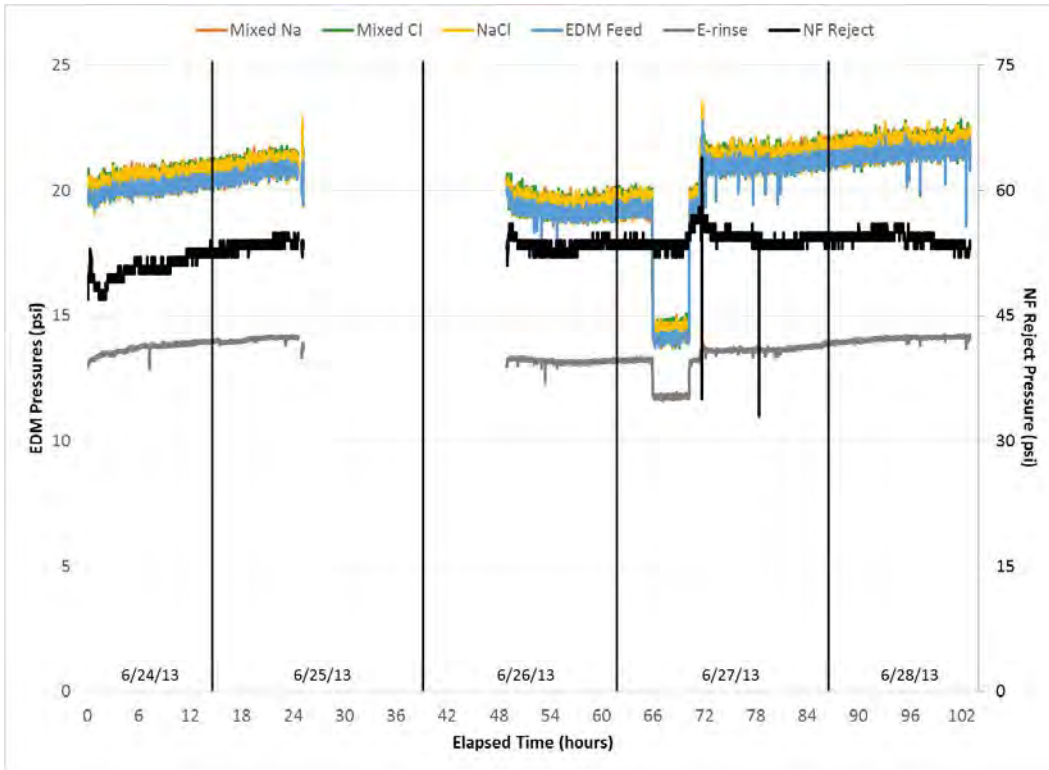


Figure B2-68.—EDM and NF reject stream pressures (experiment 5).

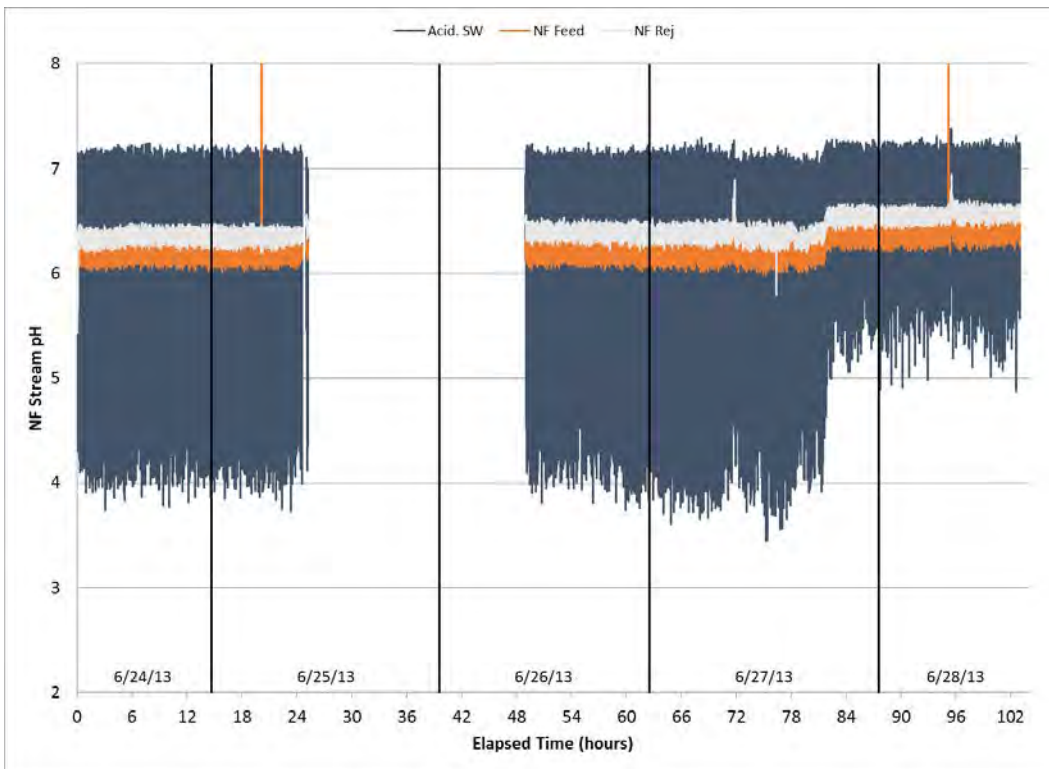


Figure B2-69.—NF stream pH (experiment 5).
(note: 93% H₂SO₄ added at this point. pH control not modifiable in the field, which increased variation in pH)

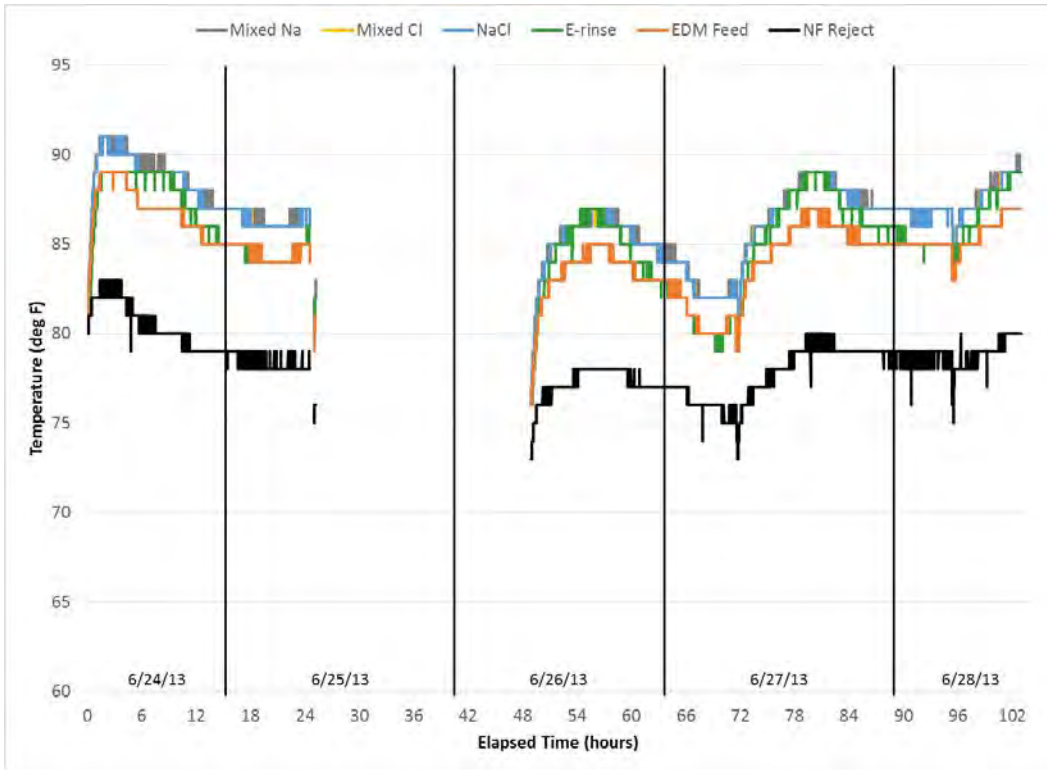


Figure B2-70.—EDM stream temperature (experiment 5).

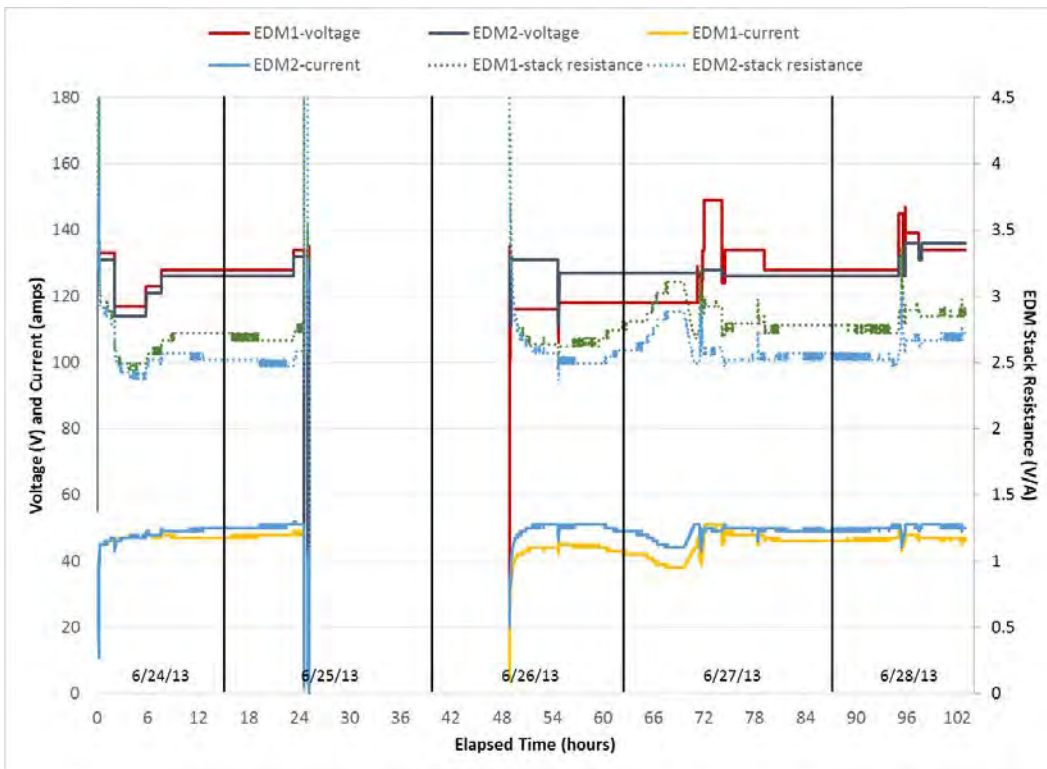


Figure B2-71.—EDM voltage, current, stack resistance (experiment 5).

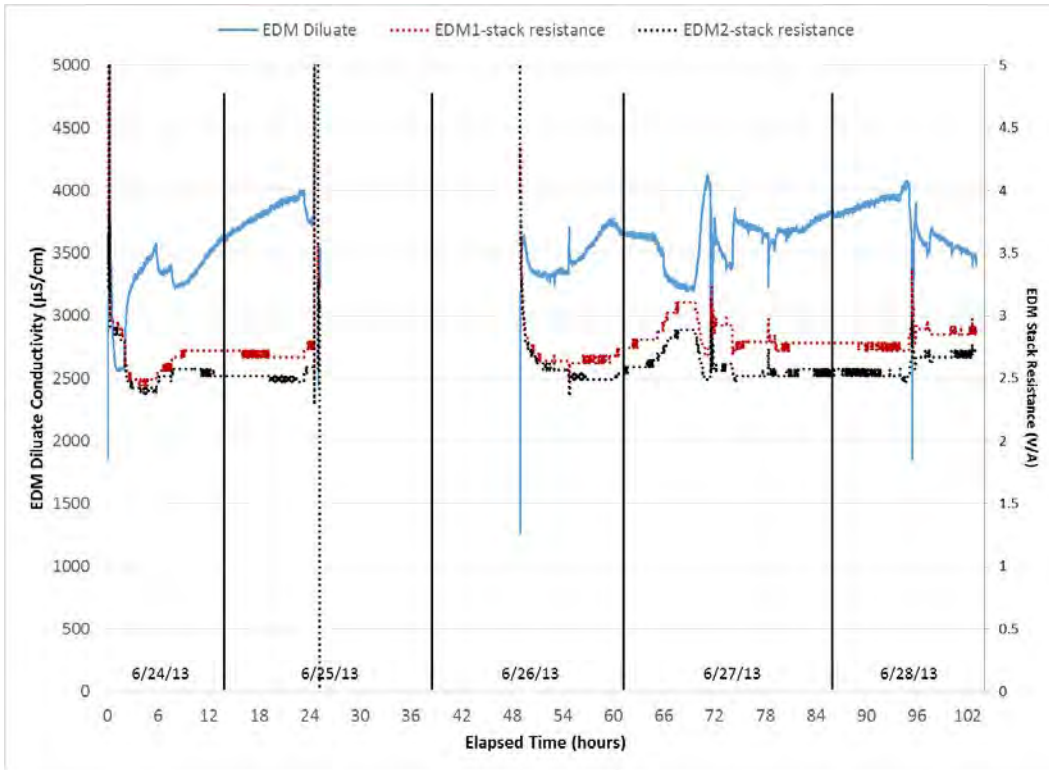


Figure B2-72.—EDM diluate conductivity and stack resistance (experiment 5).

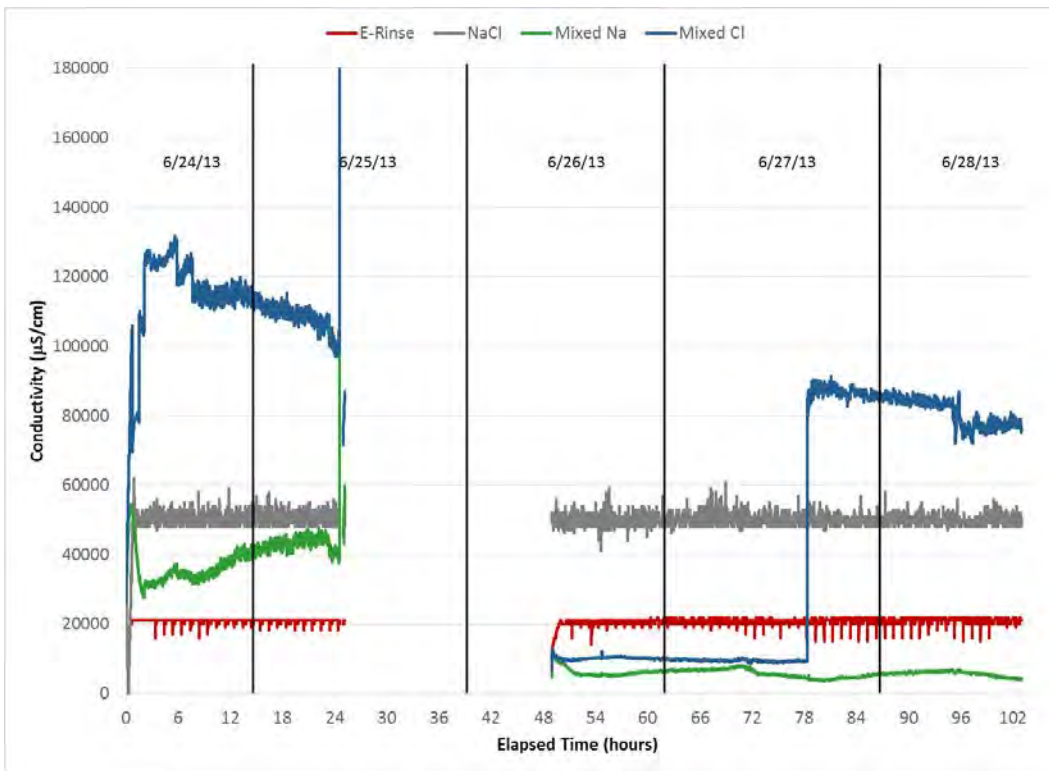


Figure B2-73.—High concentration EDM stream conductivity (experiment 5).
(note: Mixed Cl & Mixed Na are not calibrated or scaled)

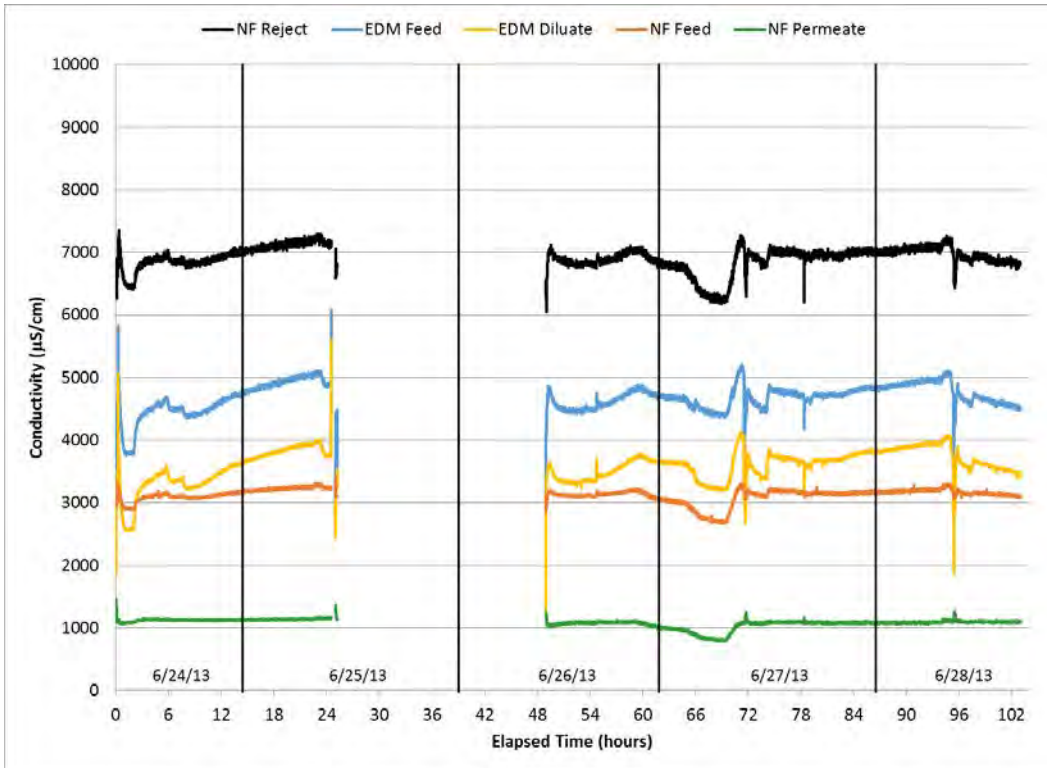


Figure B2-74.—Low concentration EDM stream conductivity (experiment 5).

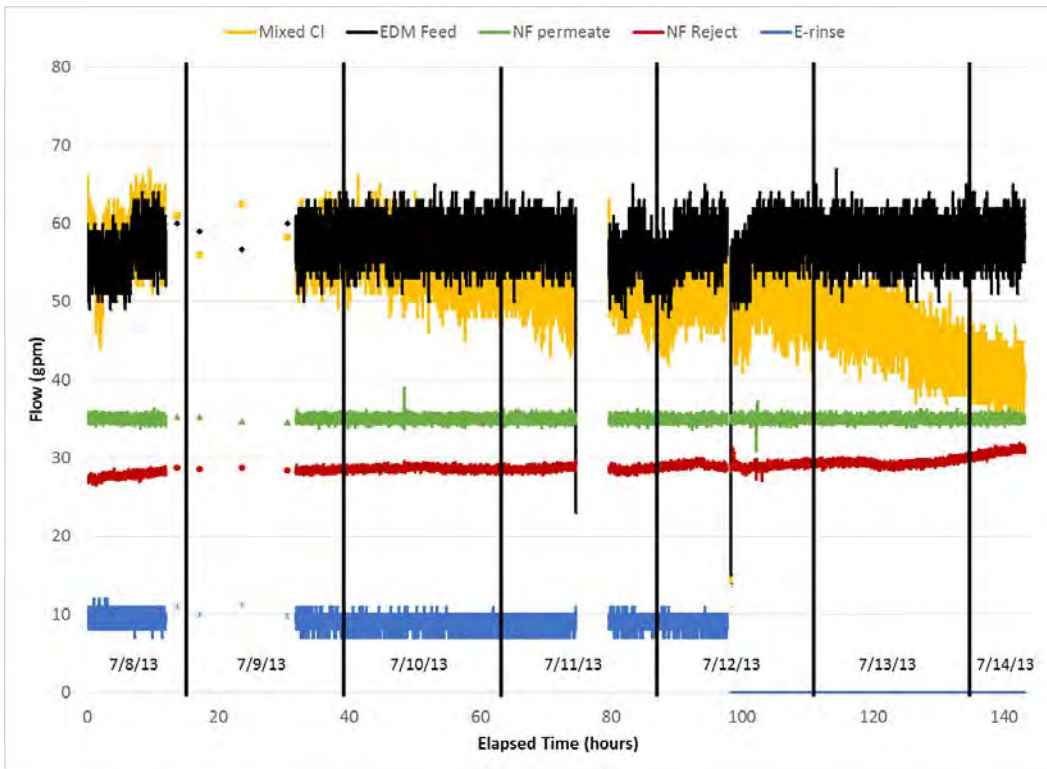


Figure B2-75.—EDM stream flows (experiment 6).

Datalogger failed on 7/8-9/13. Data was recorded on logsheets from online meters, probes, and gauges.

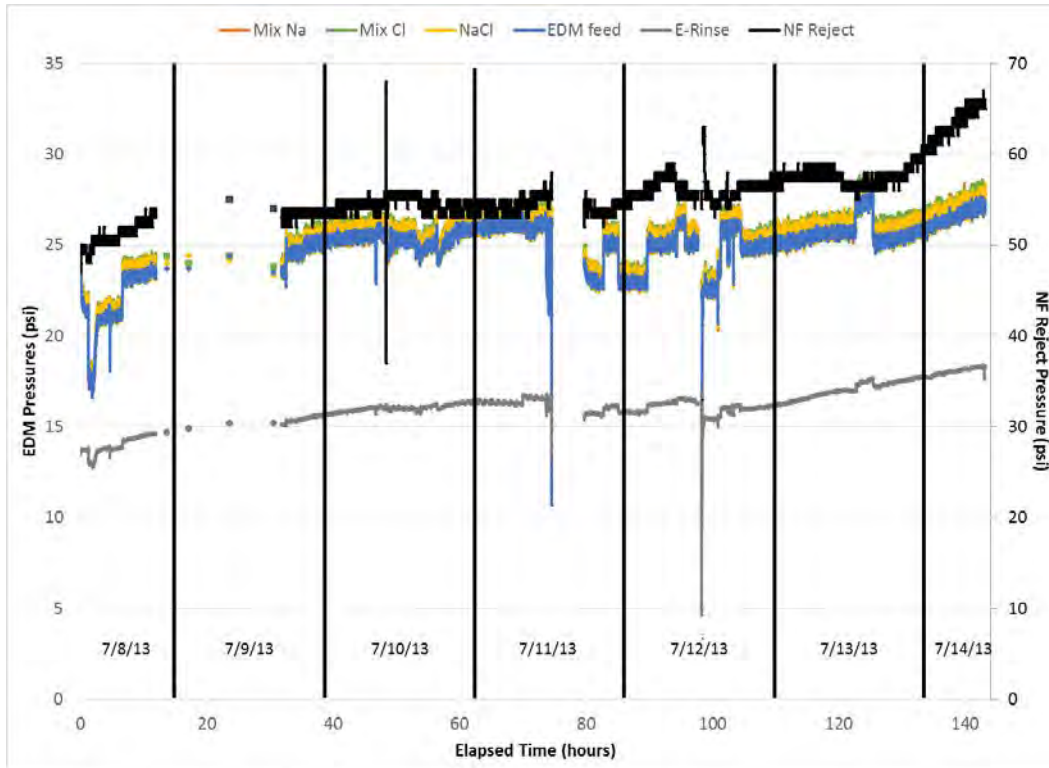


Figure B2-76.—EDM and NF reject stream pressures (experiment 6).
Datalogger failed on 7/8-9/13. Data was recorded on logsheets from online meters, probes, and gauges.

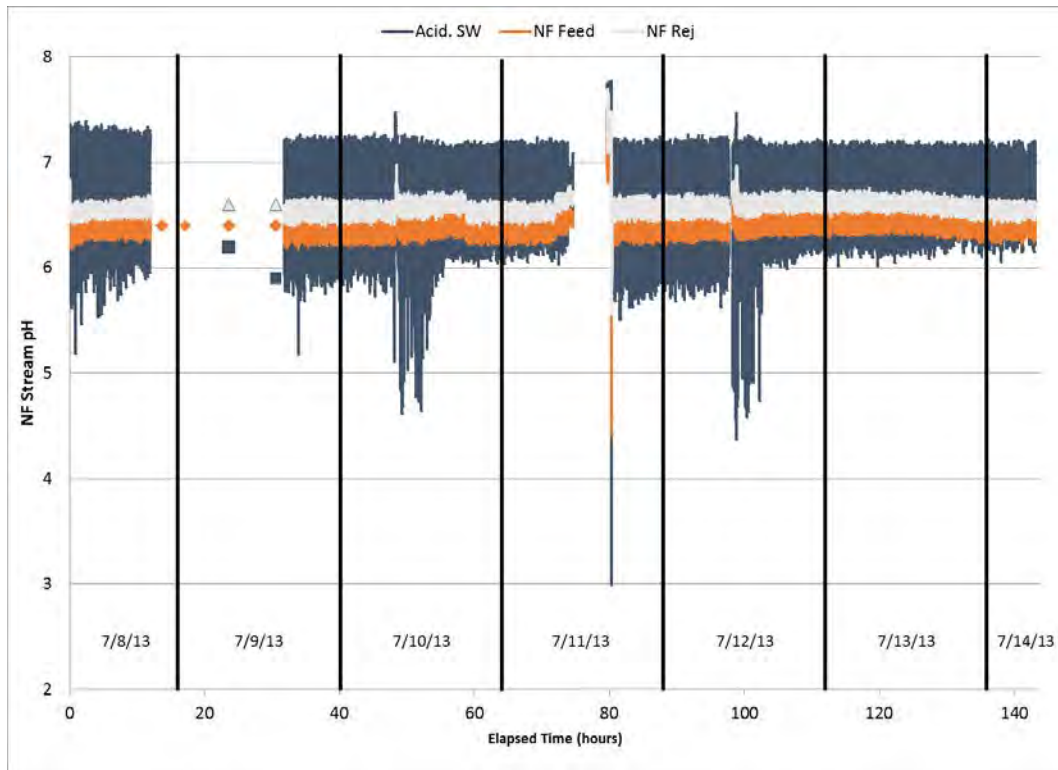


Figure B2-77.—NF stream pH (experiment 6).
Datalogger failed on 7/8-9/13. Data was recorded on logsheets from online meters, probes, and gauges. (note: 93% H₂SO₄ added at this point. pH control not modifiable in the field, which increased variation in pH)

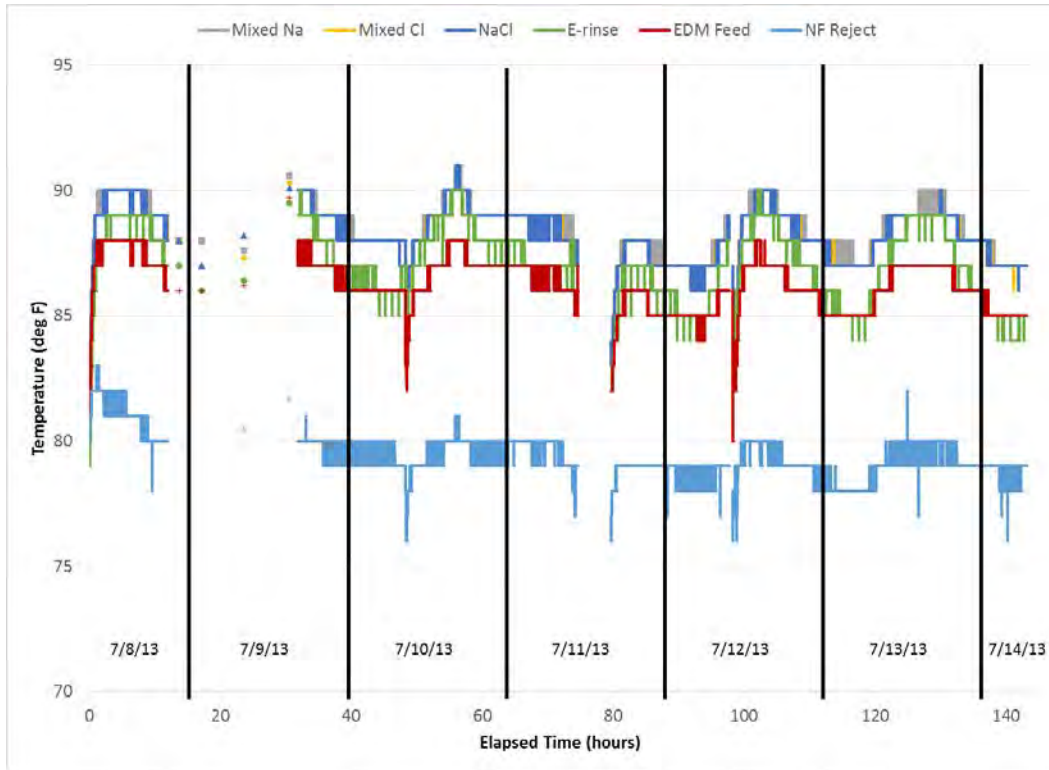


Figure B2-78.—EDM stream temperature (experiment 6).

(note: 93% H₂SO₄ added at this point. pH control not modifiable in the field, which increased variation in pH)

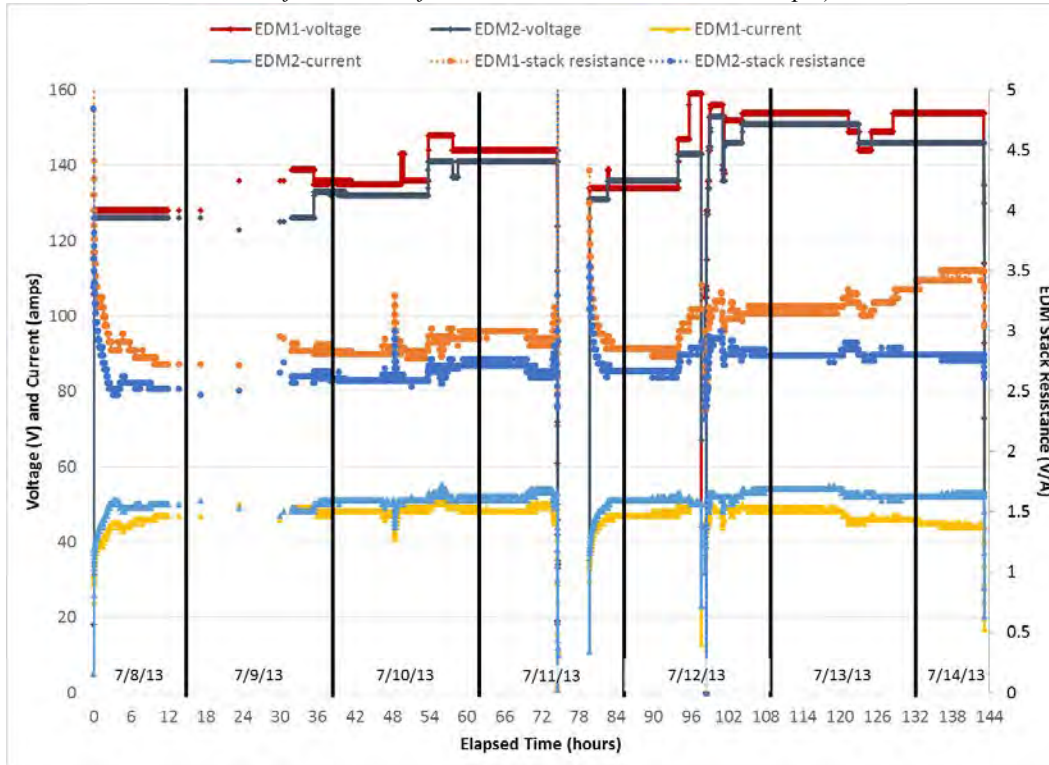


Figure B2-79.—EDM voltage, current, stack resistance (experiment 6).

(Datalogger failed on 7/8-9/13. Data was recorded on logsheets from online meters, probes, and gauges.)

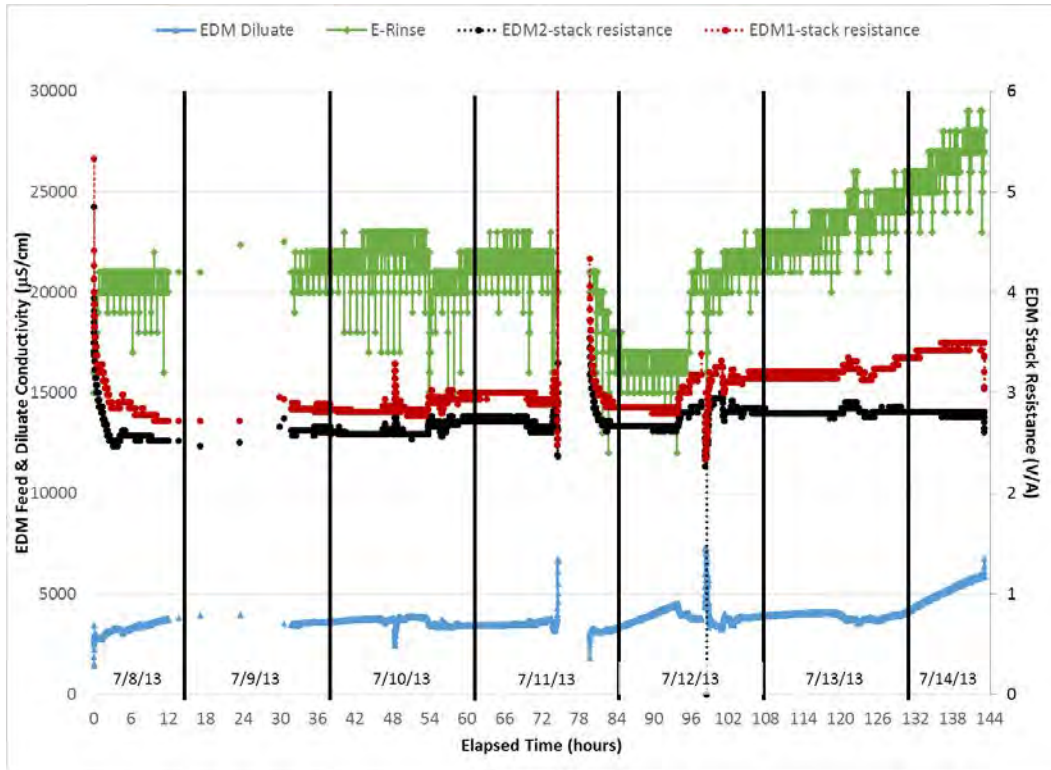


Figure B2-80.—EDM diluate conductivity and stack resistance (experiment 6).
(Datalogger failed on 7/8-9/13. Data was recorded on logsheets from online meters, probes, and gauges.)

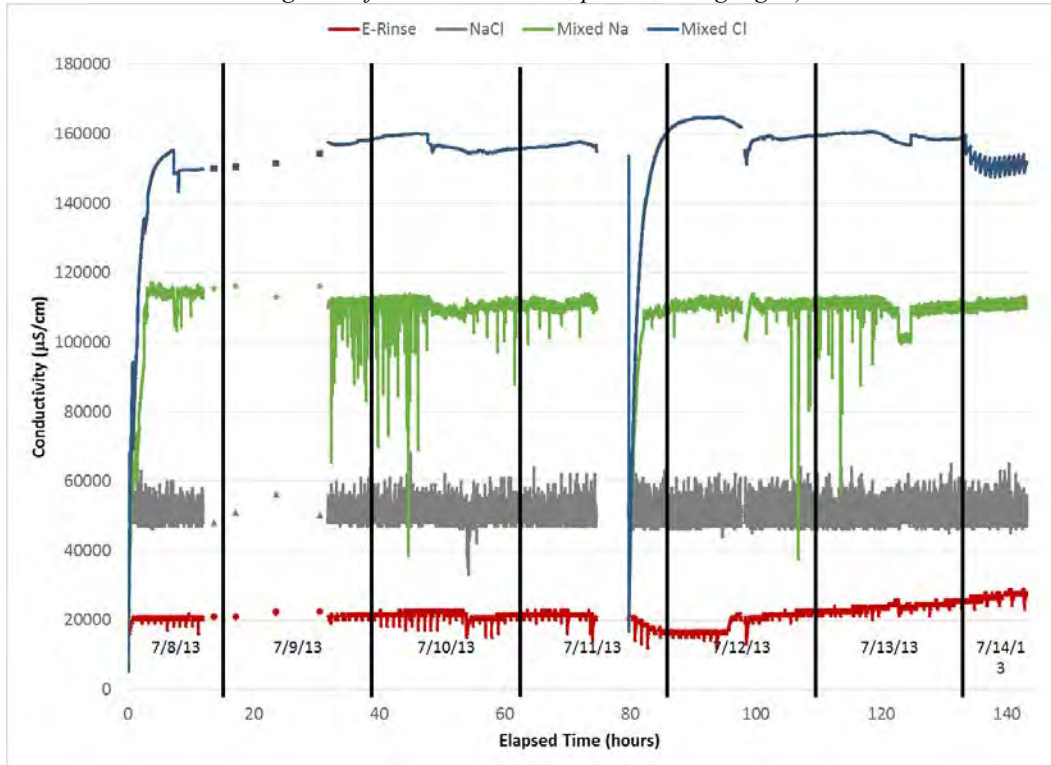


Figure B2-81.—High concentration EDM stream conductivity (experiment 6).
(Datalogger failed on 7/8-9/13. Data was recorded on logsheets from online meters, probes, and gauges.)

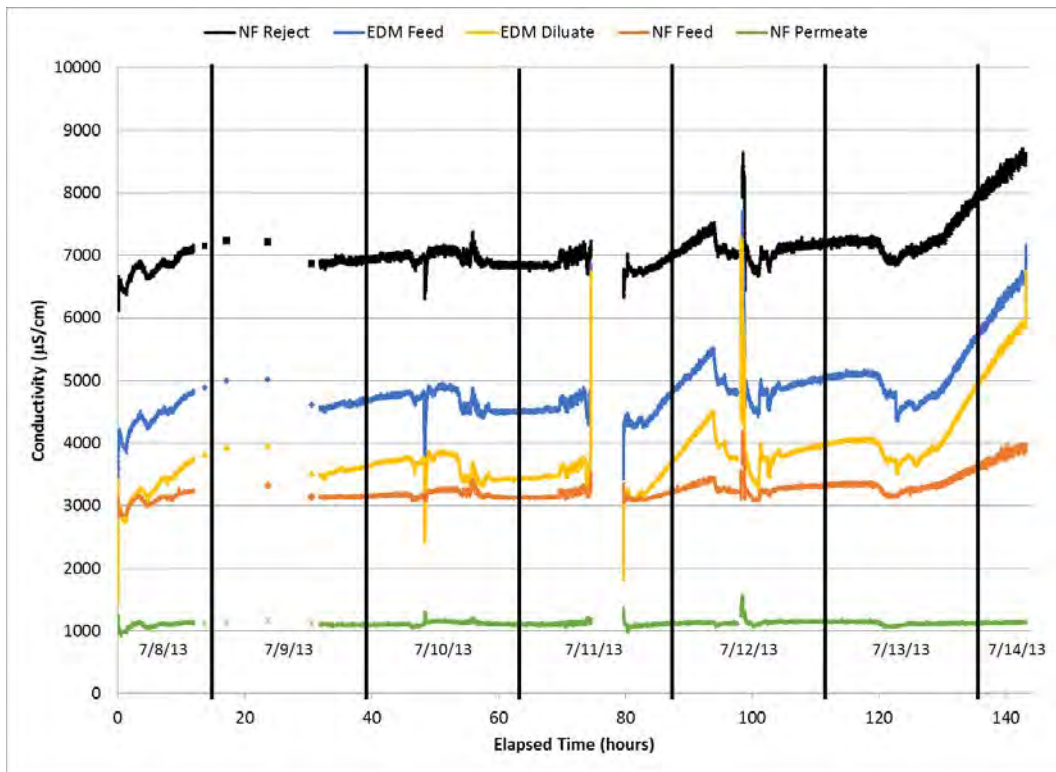


Figure B2-82.—Low concentration EDM stream conductivity (experiment 6).
(Datalogger failed on 7/8-9/13. Data was recorded on logsheets from online meters, probes, and gauges.)

APPENDIX B3
BRIGHTON ONLINE DATA

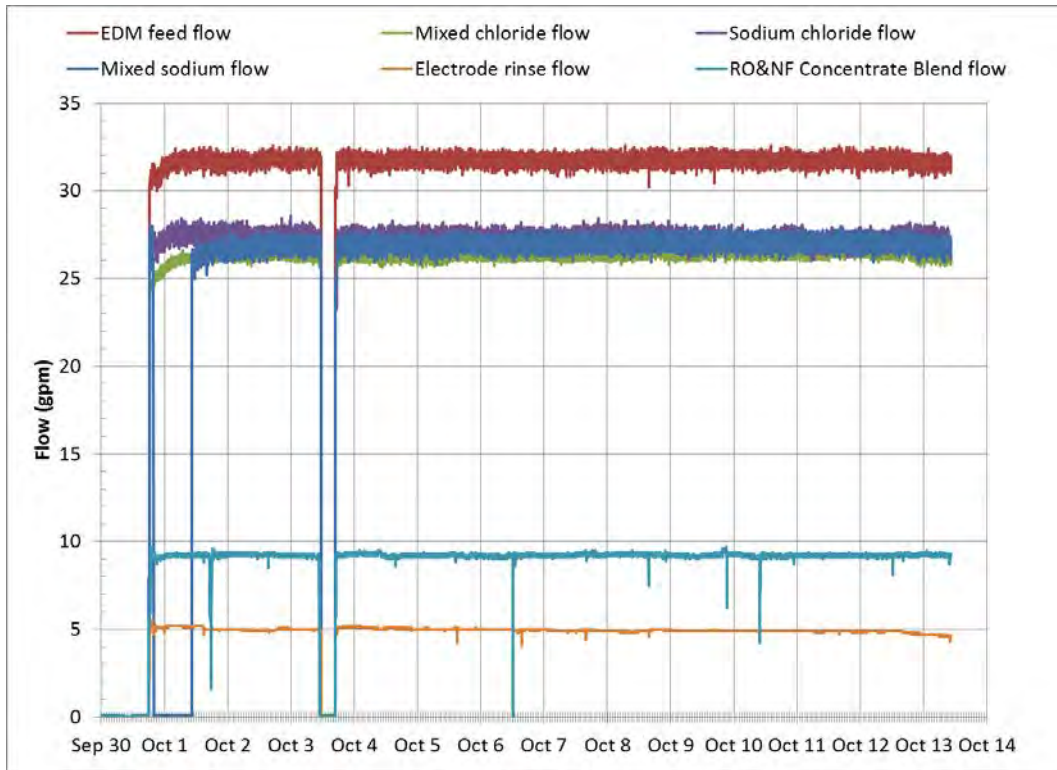


Figure B3-83.—EDM and NF stream flows (experiment 1).

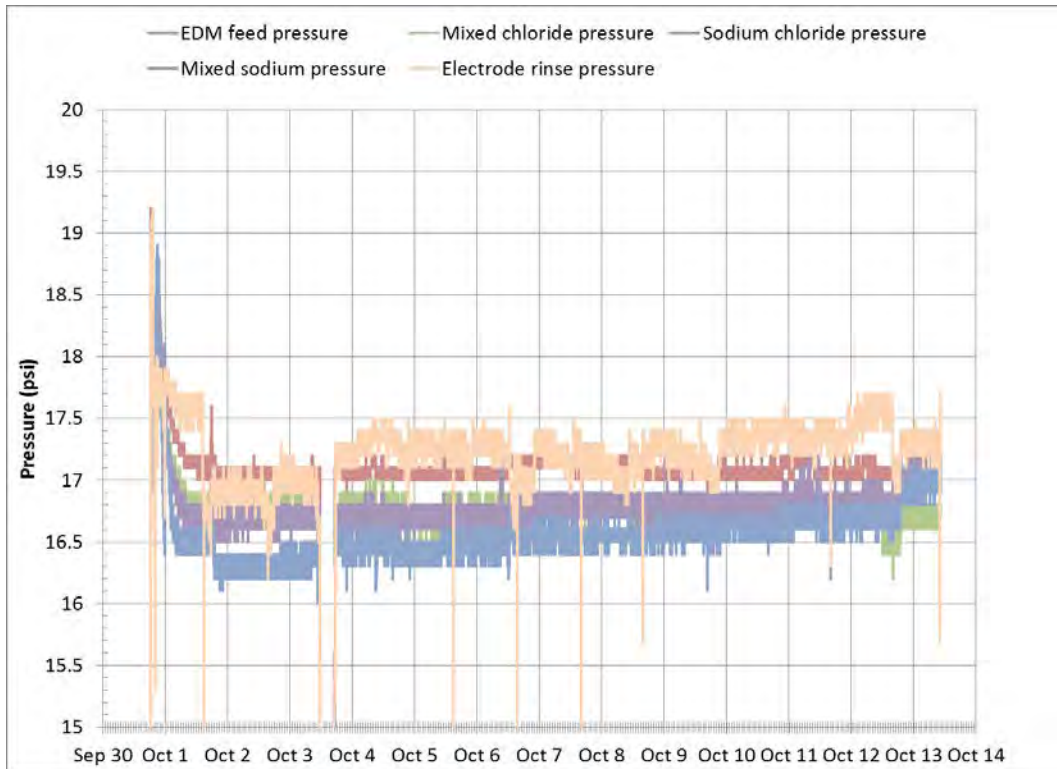


Figure B3-84.—EDM stream pressures (experiment 1).

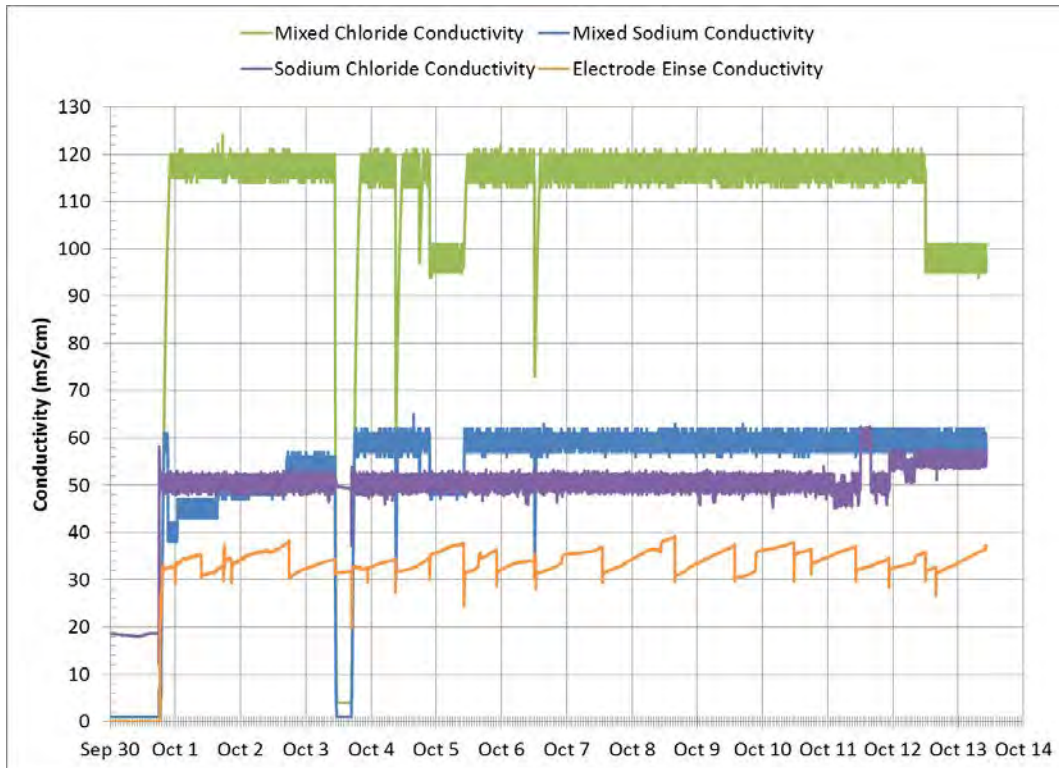


Figure B3-85.—High concentration EDM stream conductivity (experiment 1).

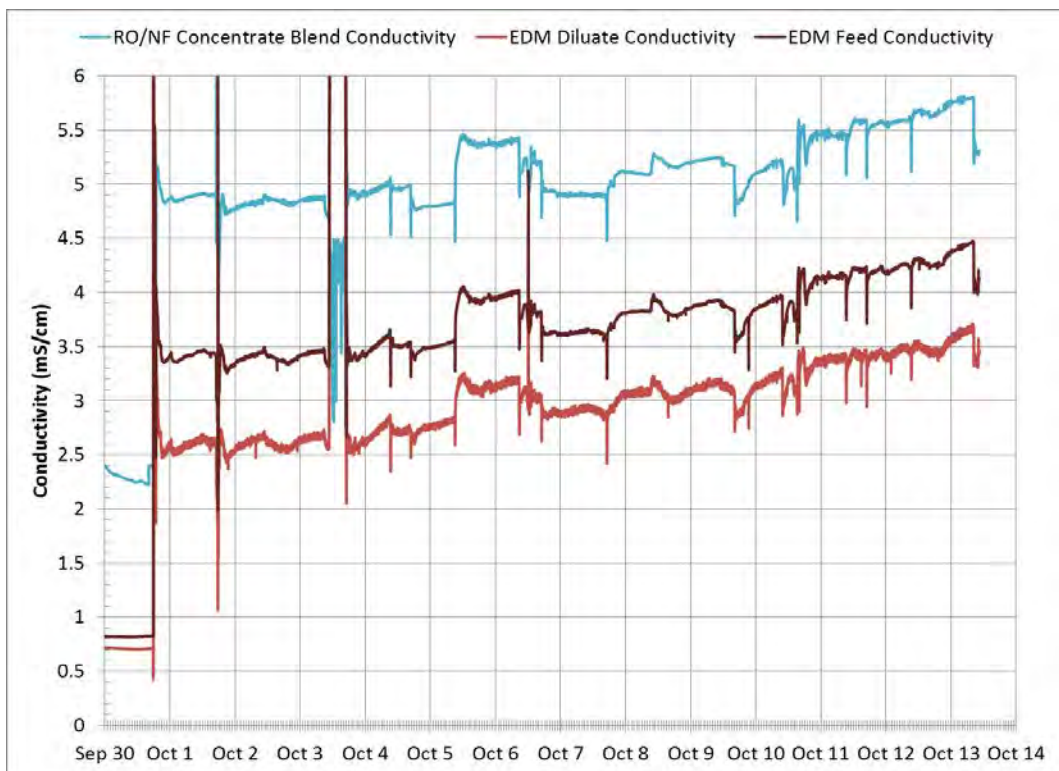


Figure B3-86.—Low concentration EDM stream conductivity (experiment 1).

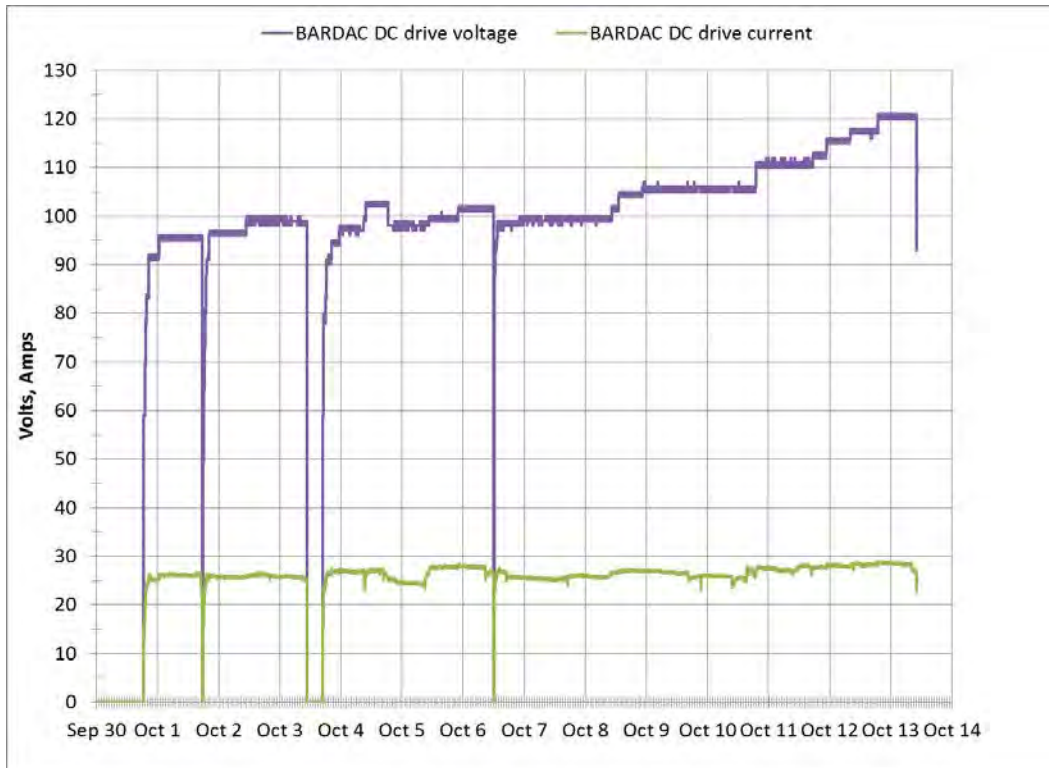


Figure B3-87.—EDM voltage, current (experiment 1).

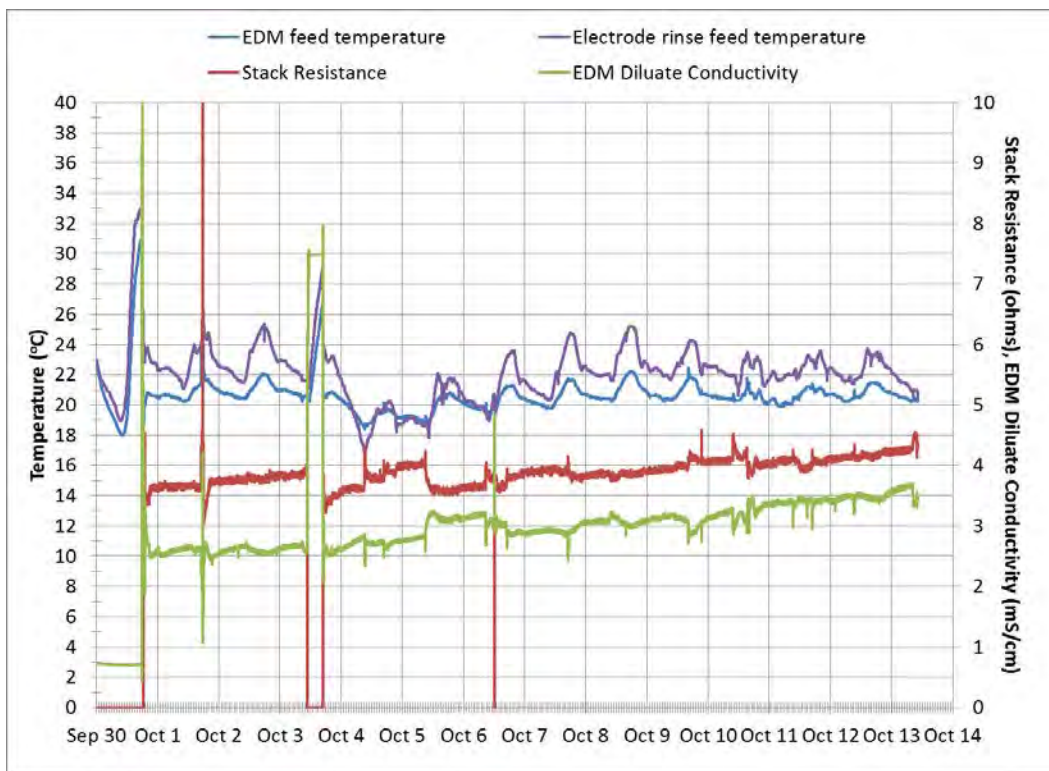


Figure B3-88.—EDM temperatures and stack resistance (experiment 1).

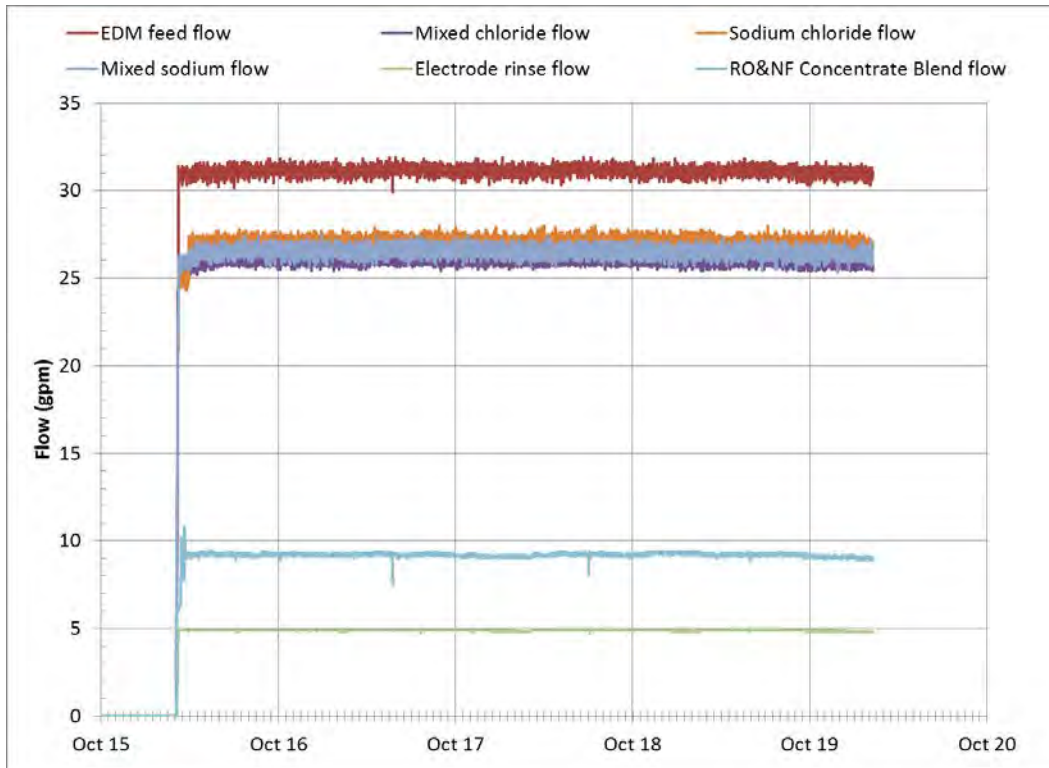


Figure B3-89.—EDM and NF stream flows (experiment 2).

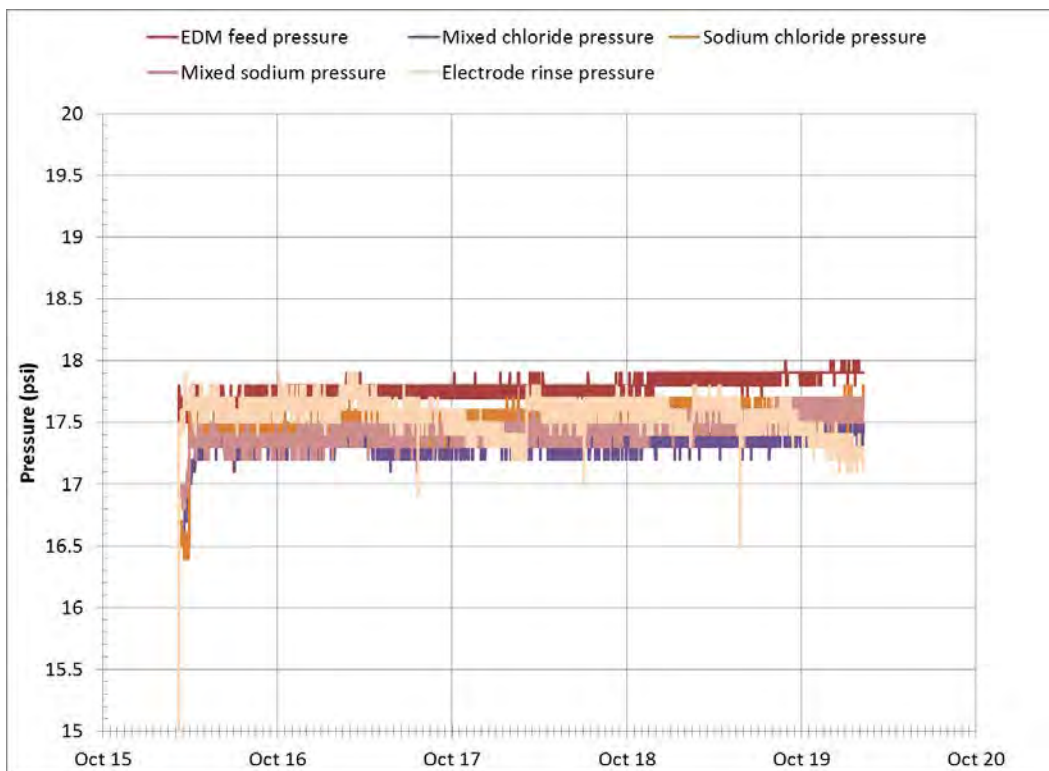


Figure B3-90.—EDM stream pressures (experiment 2).

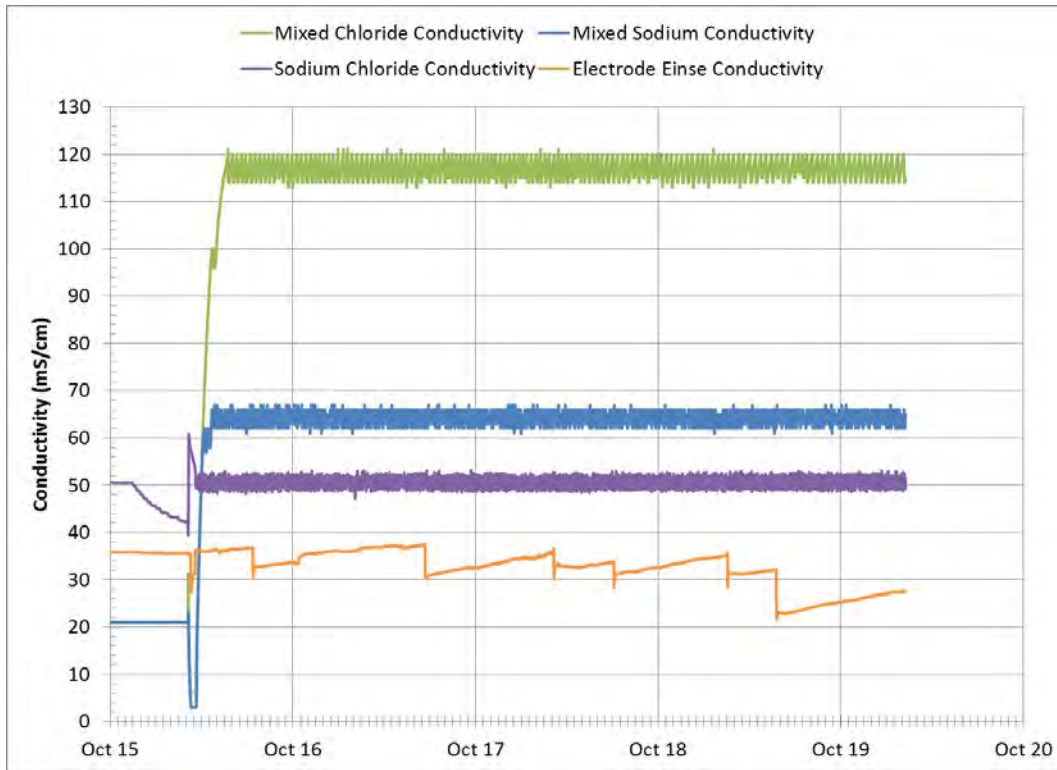


Figure B3-91.—High concentration EDM stream conductivity (experiment 2).

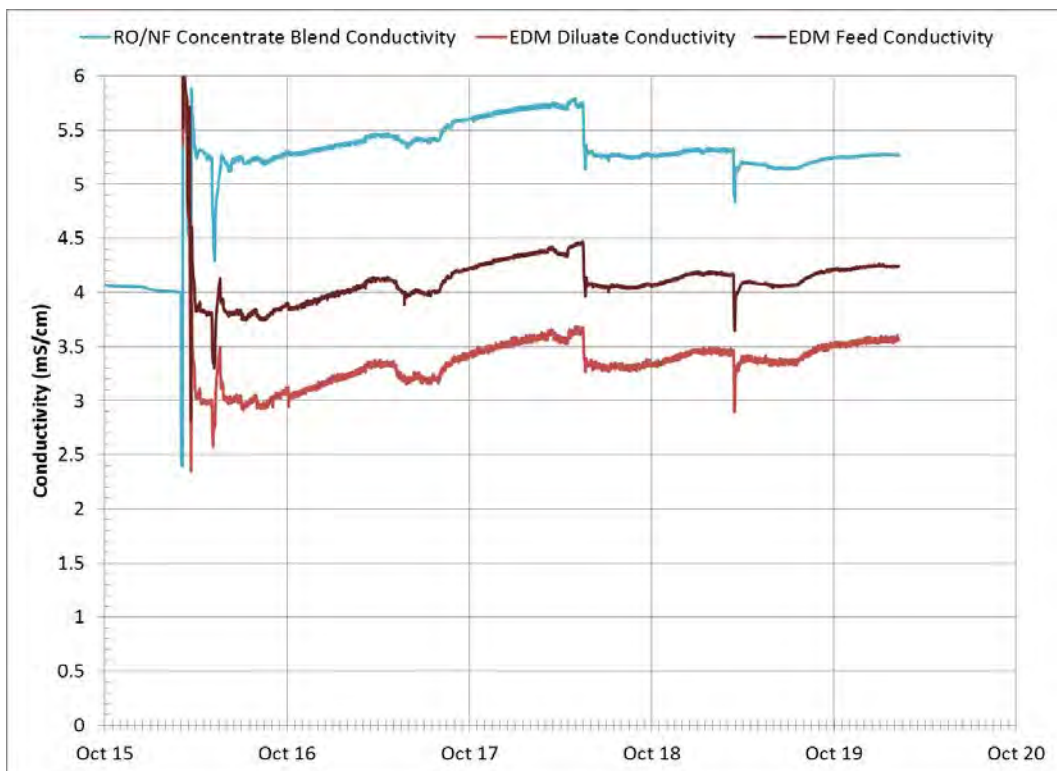


Figure B3-92.—Low concentration EDM stream conductivity (experiment 2).

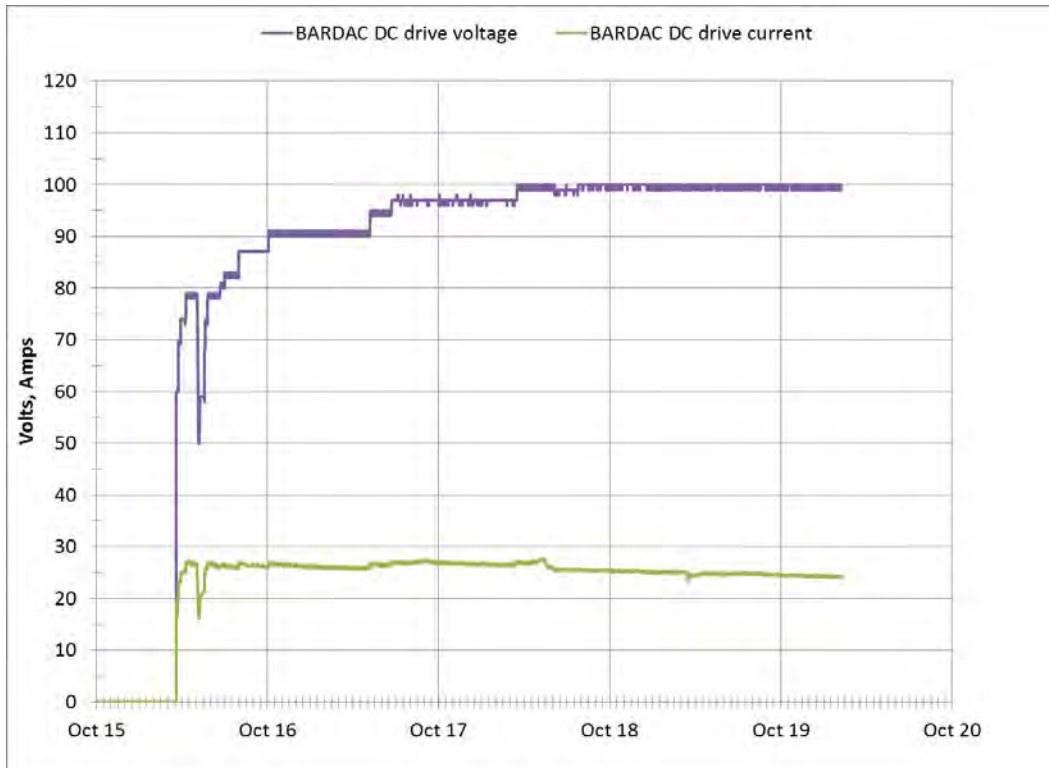


Figure B3-93.—EDM voltage, current (experiment 2).

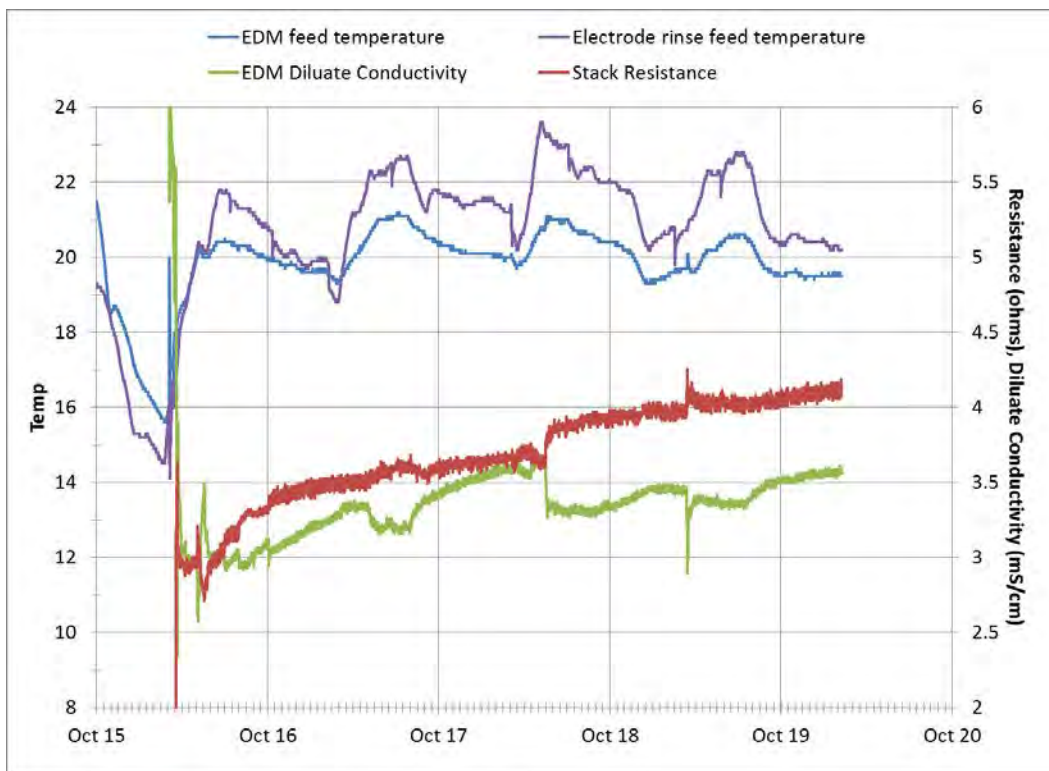


Figure B3-94.—EDM temperatures and stack resistance (experiment 2).

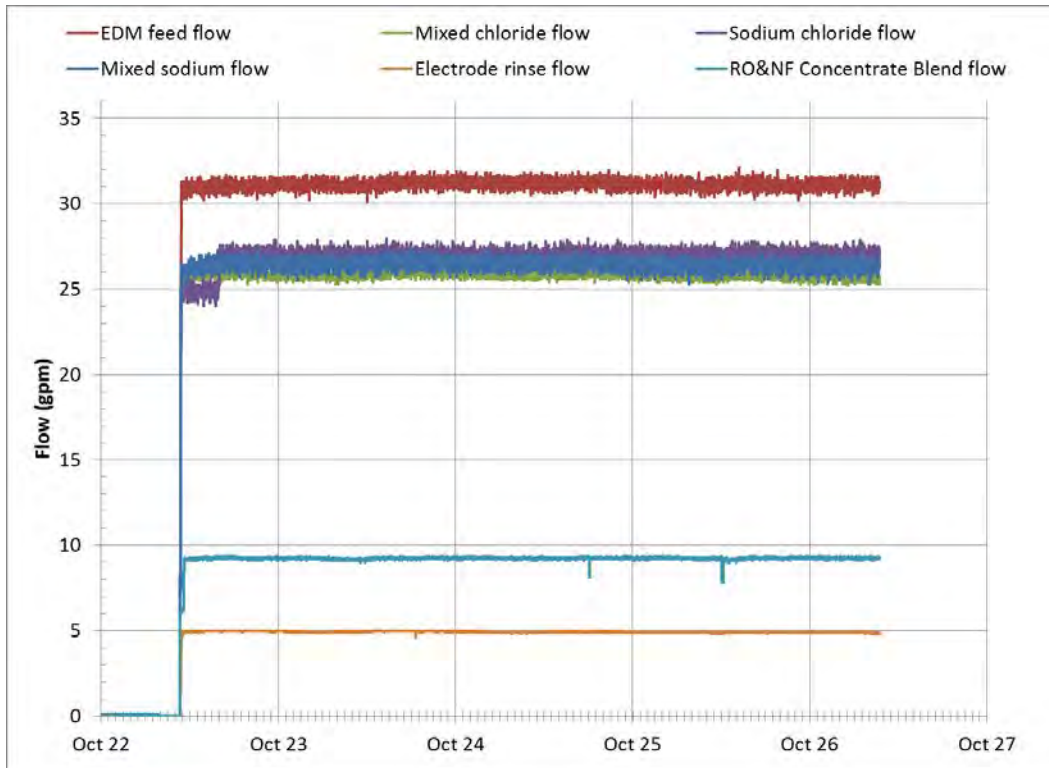


Figure B3-95.—EDM and NF stream flows (experiment 2).

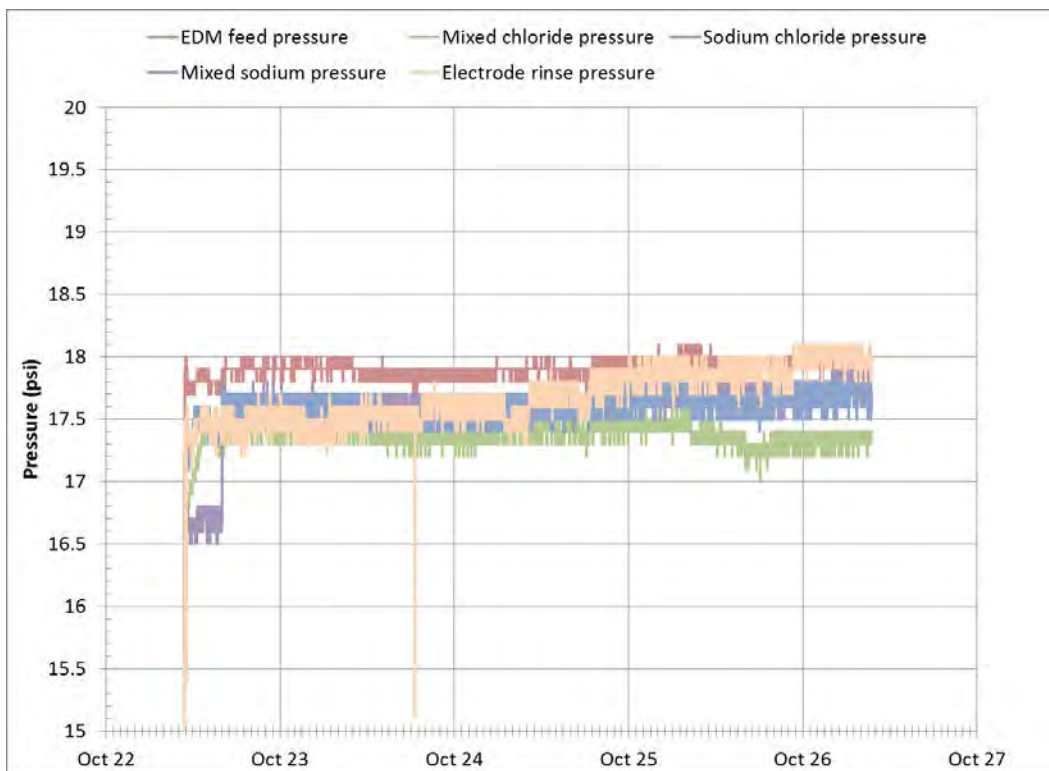


Figure B3-96.—EDM stream pressures (experiment 2).

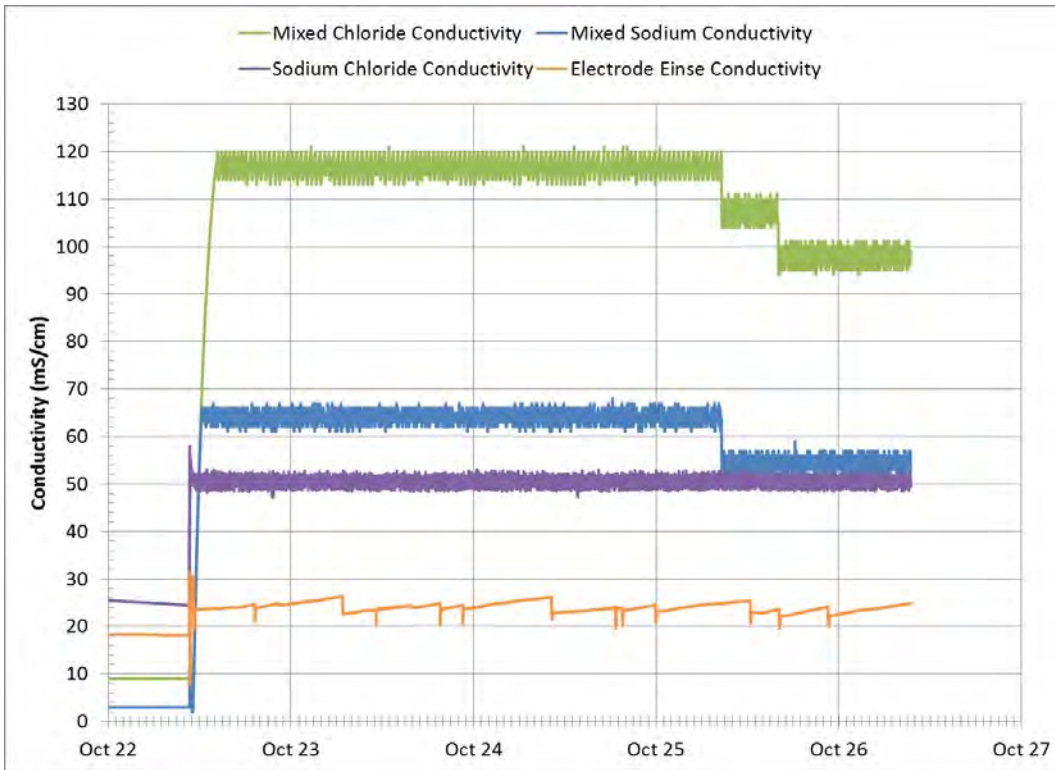


Figure B3-97.—High concentration EDM stream conductivity (experiment 2).

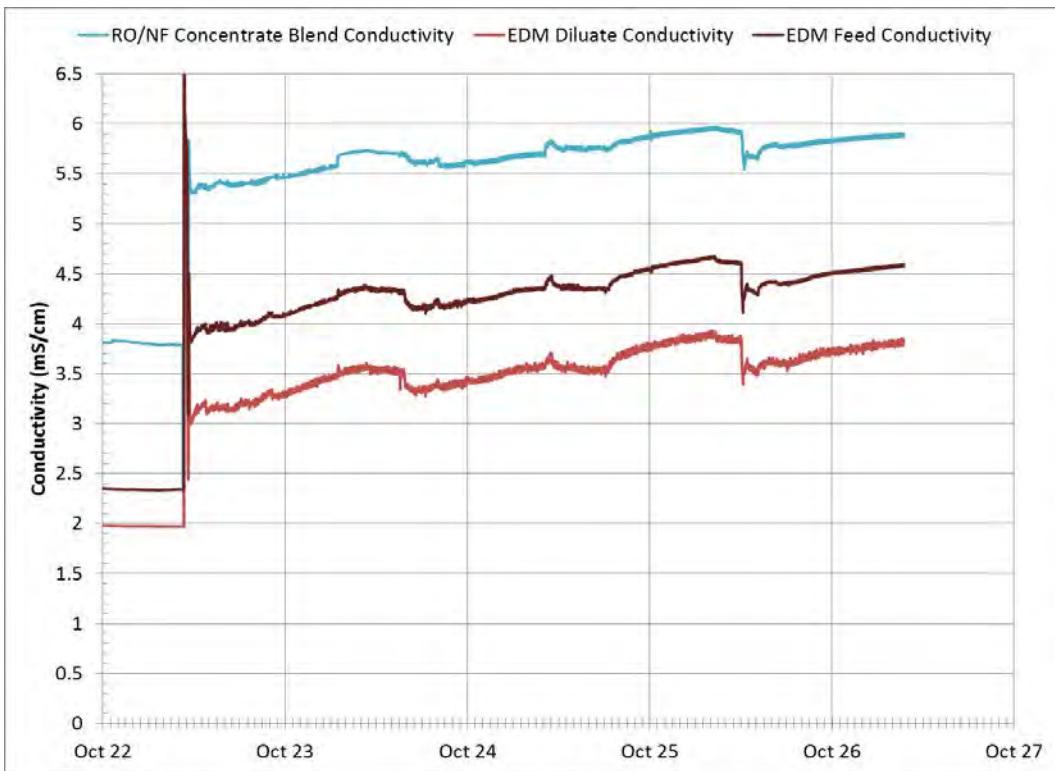


Figure B3-98.—Low concentration EDM stream conductivity (experiment 2).

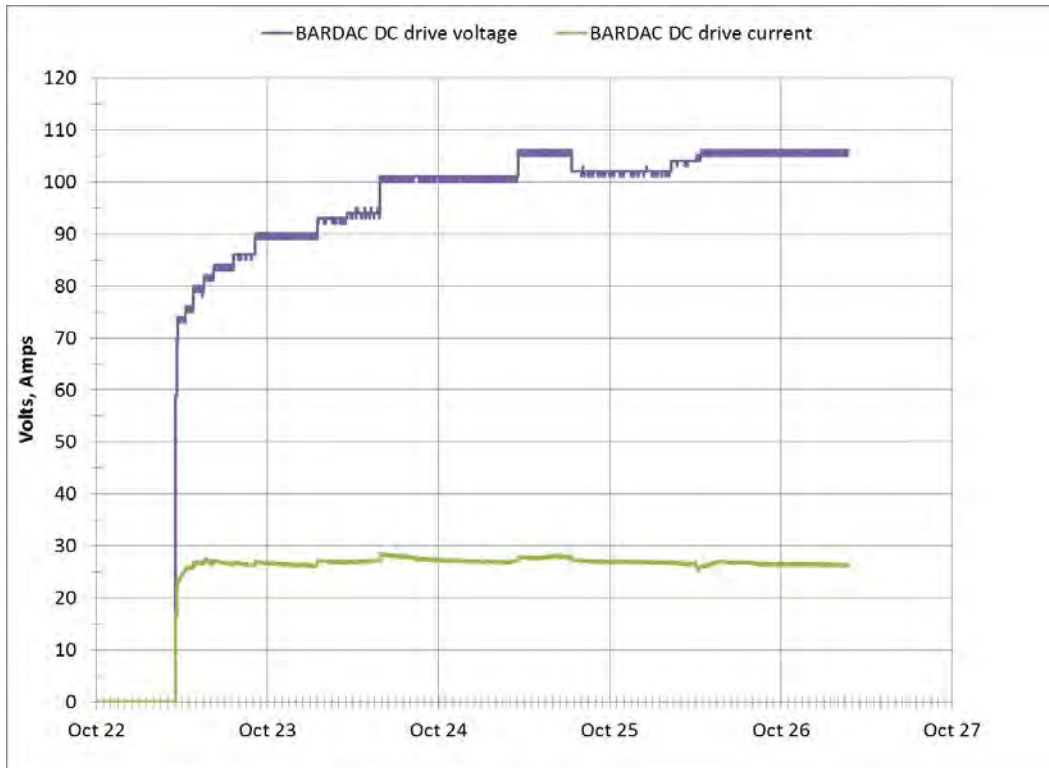


Figure B3-99.—EDM voltage, current (experiment 2).

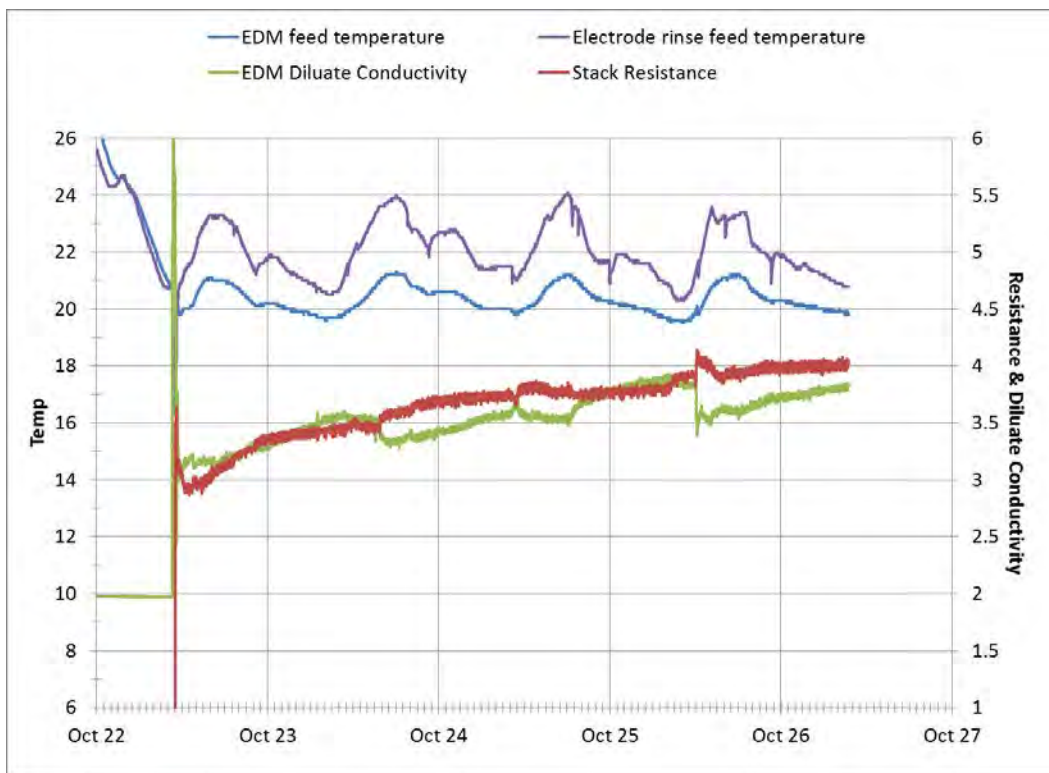


Figure B3-100.—EDM temperatures and stack resistance (experiment 2).

APPENDIX C

WATER QUALITY ANALYSES

Appendix C1 – Alamogordo, Year 1
Appendix C2 – Alamogordo, Year 2
Appendix C3 – Brighton, Colorado

APPENDIX C1
ALAMOGORDO, YEAR 1

Table C-1.—Water Quality Analyses by Stream

		pH	Conductivity	TDS*	HCO ₃	Ca	Mg	Na	Cl	SO ₄	SiO ₂
		-	(µS/cm)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)
3/22/2013	Edm dil	6.2	3390	2590	51.9	ND	ND	ND	ND	ND	26.2
3/25/2013	Edm dil	6.4	4370	3555	62.2	249.4	433.4	ND	104.2	2179.0	18.0
3/26/2013	Edm dil	7.0	4130	4090	112.2	265.0	199.6	467.1	110.9	ND	28.0
3/27/2013	Edm dil	6.0	3780	2790	62.2	278.4	160.7	455.8	102.3	1880.4	ND
3/28/2013	Edm dil	6.6	3630	3420	87.2	274.3	92.4	459.2	100.7	ND	27.0
3/29/2013	Edm dil	6.8	3300	3060	89.1	235.5	195.2	387.7	92.4	1552.1	26.0
4/5/2013	Edm dil	6.5	3270	2710	122.0	214.2	118.2	344.2	82.9	1531.3	22.2
4/8/2013	Edm dil	6.2	3710	3075	183.0	244.5	133.0	410.2	95.8	1759.2	25.0
4/9/2013	Edm dil	6.6	4140	3315	185.4	ND	ND	ND	ND	ND	23.0
4/10/2013	Edm dil	6.6	3950	3320	80.5	ND	ND	ND	ND	ND	23.2
4/11/2013	Edm dil	6.3	3740	3295	78.1	317.7	147.9	447.6	110.7	1977.1	23.8
4/17/2013	Edm dil	6.8	3530	2700	53.7	245.3	121.9	382.0	86.6	1664.8	21.0
4/18/2013	Edm dil	6.7	3930	3480	67.7	271.3	134.4	429.1	99.9	1858.9	24.0
4/20/2013	Edm dil	6.6	4410	3840	50.0	343.8	163.2	485.6	120.0	2257.6	25.5
4/22/2013	Edm dil	6.3	3940	3175	81.7	297.5	136.9	673.4	124.1	2252.8	25.0
4/25/2013	Edm dil	6.3	3690	3050	38.8	283.2	132.5	390.7	112.8	1839.6	25.0
4/26/2013	Edm dil	7.2	3780	3055	39.0	304.7	142.8	438.2	95.5	1945.5	23.5
4/30/2013	Edm dil	5.7	3620	3280	ND	290.2	144.7	395.4	90.4	1844.7	23.0
5/29/2013	Edm dil	5.5	3570	2975	34.2	223.6	102.7	496.8	85.2	1952.9	23.4
5/31/2013	Edm dil	5.7	5050	4590	48.8	353.5	181.7	714.5	133.2	2864.7	26.0
6/3/2013	Edm dil	5.9	4450	3955	61.0	328.5	172.6	639.9	125.0	2485.1	25.0
6/4/2013	Edm dil	5.9	3890	3420	42.7	267.3	129.6	541.8	100.4	2035.2	27.0
6/14/2013	Edm dil	6.7	3140	2395	48.8	174.6	87.6	370.7	94.3	1385.8	29.0
6/18/2013	Edm dil	6.8	3500	2770	54.9	199.7	99.4	452.2	90.3	1622.8	29.0
6/19/2013	Edm dil	6.5	3790	3070	54.9	211.4	105.3	512.3	92.8	1771.5	27.0
6/20/2013	Edm dil	6.3	3380	2665	54.9	190.5	92.8	437.2	90.1	1540.0	27.0
6/21/2013	Edm dil	6.6	3330	2650	122.0	192.4	102.2	434.2	97.5	1537.3	29.0

		pH	Conductivity	TDS*	HCO ₃	Ca	Mg	Na	Cl	SO ₄	SiO ₂
		-	(µS/cm)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)
6/24/2013	Edm dil	6.4	3520	2815	61.0	202.0	99.8	454.7	92.4	1625.3	27.0
6/26/2013	Edm dil	6.5	3600	2855	73.2	204.0	100.0	473.0	90.6	1679.4	28.0
6/27/2013	Edm dil	6.2	3840	3080	91.5	202.0	100.6	538.7	95.5	1805.7	27.0
7/8/2013	EDM Dil	6.4	3580	2810	79.3	177.5	95.9	506.5	82.5	1648.9	26.0
7/9/2013	EDM Dil	6.3	3620	2860	91.5	180.2	98.0	507.7	86.8	1660.0	27.0
7/10/2013	EDM Dil	6.5	3720	2965	97.6	196.3	113.0	511.4	89.9	1754.1	33.0
7/11/2013	Edm dil	6.4	3900	3105	183.0	191.8	109.1	535.5	89.1	1824.2	28.0
7/12/2013	EDM Dil 3:45pm	6.5	4070	3095	36.6	188.5	104.9	536.2	93.9	1818.9	28.0
7/12/2013	EDM Dil 5:20pm	6.5	4150	3140	42.7	189.2	106.8	546.2	93.4	1850.8	29.0
7/13/2013	EDM Dil 5:15pm	6.4	4530	3445	97.6	207.8	114.4	618.9	89.0	2042.5	32.0
7/14/2013	Edm dil	7.3	6280	5735	54.9	363.5	206.5	992.5	103.2	3349.3	27.0
7/20/2013	Edm dil	6.0	3780	3090	6.1	168.5	91.6	586.8	75.9	1803.9	25.0
3/22/2013	Edm feed	6.5	5360	26195	108.0	ND	ND	ND	ND	ND	35.0
3/25/2013	Edm feed	6.3	6380	5735	111.0	458.1	624.8	ND	149.8	3428.6	27.1
3/26/2013	Edm feed	7.1	5740	5555	189.1	533.9	305.4	584.8	156.8	3183.6	27.0
3/27/2013	Edm feed	6.3	5340	4840	111.0	ND	ND	ND	ND	ND	27.2
3/28/2013	Edm feed	6.9	5250	5335	163.5	480.0	347.2	533.1	138.0	ND	25.0
3/29/2013	Edm feed	6.9	4840	4895	200.1	434.4	268.4	502.2	124.2	2526.8	25.0
4/5/2013	Edm feed	6.4	4820	4315	126.9	419.1	213.0	475.9	117.4	2507.6	22.3
4/8/2013	Edm feed	6.4	5220	4825	126.9	441.9	224.5	526.4	126.1	2688.5	25.0
4/9/2013	Edm feed	6.5	5580	4965	142.7	ND	ND	ND	ND	ND	23.6
4/10/2013	Edm feed	6.7	5340	5090	141.5	ND	ND	ND	ND	ND	24.1
4/11/2013	Edm feed	6.7	5180	4975	263.5	547.8	238.0	578.7	147.3	3029.6	25.2
4/17/2013	Edm feed	7.1	4980	4475	103.1	437.3	203.9	498.3	119.0	2569.8	29.6
4/18/2013	Edm feed	6.8	5360	5030	117.1	476.0	221.8	565.5	133.1	2862.5	25.1
4/20/2013	Edm feed	6.7	5870	5490	97.6	548.1	244.5	603.4	152.4	3222.3	22.1

		pH	Conductivity	TDS*	HCO ₃	Ca	Mg	Na	Cl	SO ₄	SiO ₂
		-	(µS/cm)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)
4/22/2013	Edm feed	6.5	5300	4700	25.6	494.4	208.9	596.2	138.9	2901.9	26.0
4/25/2013	Edm feed	6.8	5240	4765	67.7	474.5	209.2	508.4	113.1	2808.6	23.4
4/26/2013	Edm feed	6.5	5250	4645	67.1	518.8	223.5	567.4	120.9	3005.1	22.5
4/30/2013	Edm feed	5.9	5290	4905	ND	527.6	241.1	549.0	114.1	3015.2	23.0
5/29/2013	Edm feed	5.8	4890	4440	67.1	395.9	169.8	646.8	90.8	2935.3	22.7
5/31/2013	Edm feed	5.9	6160	5920	67.1	523.4	252.8	830.2	128.6	3751.3	26.0
6/3/2013	Edm feed	6.2	5540	5350	24.4	482.3	236.9	720.4	120.0	ND	25.0
6/4/2013	Edm feed	6.1	5060	4830	61.0	433.3	199.0	668.6	102.2	ND	26.0
6/14/2013	Edm feed	6.8	4690	3945	91.5	350.5	165.3	509.6	113.7	2281.0	28.0
6/18/2013	Edm feed	6.8	4800	4195	79.3	361.8	169.9	576.0	108.0	2435.0	29.0
6/19/2013	Edm feed	6.5	4950	4360	79.3	363.0	171.1	620.7	107.1	2494.7	27.0
6/20/2013	Edm feed	6.6	4560	3970	91.5	337.1	155.8	552.2	103.3	2282.7	27.0
6/21/2013	Edm feed	6.8	4500	3935	67.1	338.3	169.7	541.7	109.7	2258.2	29.0
6/24/2013	Edm feed	6.7	4670	4070	91.5	347.4	162.8	561.4	108.4	2340.0	27.0
6/26/2013	Edm feed	6.5	4780	4190	91.5	351.0	163.9	582.3	108.1	2407.6	28.0
6/27/2013	Edm feed	6.4	4920	4250	79.3	342.8	161.4	639.1	108.6	2494.0	27.0
6/28/2013	Edm feed 4pm	6.6	4720	3775	91.5	329.2	154.2	604.4	107.7	2366.9	29.0
7/8/2013	EDM Feed	6.5	4680	4030	122.0	317.6	156.4	608.0	95.7	2333.4	25.0
7/9/2013	EDM Feed	6.5	4740	4075	146.4	329.7	162.4	614.4	103.0	2372.9	27.0
7/10/2013	EDM Feed	6.6	4930	4320	146.4	357.6	186.0	622.3	110.6	2510.9	27.0
7/11/2013	Edm feed	6.5	4970	4290	91.5	334.6	170.2	633.2	105.3	2502.7	27.0
7/12/2013	EDM Feed 3:45pm	6.6	5230	4295	91.5	330.6	165.7	635.0	109.6	2524.9	29.0
7/12/2013	EDM Feed Post Side filter 5:20pm	6.8	5330	4445	85.4	344.5	171.0	650.9	111.0	2593.9	26.0
7/12/2013	EDM Feed Pre Side filter 5:20pm	6.7	5300	4405	67.1	339.1	169.5	648.7	111.5	2574.7	28.0

		pH	Conductivity	TDS*	HCO ₃	Ca	Mg	Na	Cl	SO ₄	SiO ₂
		-	(µS/cm)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)
7/13/2013	EDM Feed 5:15pm	6.6	5680	4715	115.9	361.7	177.1	723.7	103.8	2771.4	31.0
7/14/2013	Edm feed	6.8	7200	6930	97.6	513.9	271.0	1075.3	110.2	4004.4	25.0
7/20/2013	Edm feed	6.5	4740	4155	115.9	295.7	145.7	682.9	87.9	2429.7	27.0
3/22/2013	E-rinse	10.6	19700	15850	106.1	ND	ND	ND	ND	ND	2.8
3/26/2013	E-rinse	11.0	22860	19025	139.1	11.0	ND	6788.9	130.7	13898.9	2.0
3/27/2013	E-rinse	11.0	22100	17875	183.0	12.8	14.5	6682.7	124.7	13111.8	20.5
3/28/2013	E-rinse	10.9	24510	20640	179.3	20.0	ND	ND	144.1	ND	3.0
3/29/2013	E-rinse	10.9	24290	20880	262.3	28.1	12.5	ND	149.9	13023.8	3.0
4/5/2013	E-rinse	11.5	24590	20760	315.4	18.2	1.7	6547.7	93.8	12974.1	0.0
4/8/2013	E-rinse	11.6	22920	18765	449.0	37.0	1.2	5913.7	120.9	11518.2	2.0
4/9/2013	E-rinse	11.9	22630	18405	788.1	ND	ND	ND	ND	ND	0.0
4/10/2013	E-rinse	12.2	20800	18905	899.1	ND	ND	ND	ND	ND	0.5
4/11/2013	E-rinse	11.7	20840	18930	561.2	55.0	1.0	6330.6	171.4	11578.1	0.3
4/17/2013	E-rinse	11.8	22170	18505	507.5	39.3	1.9	5849.9	149.8	11047.1	0.8
4/18/2013	E-rinse	12.0	22730	18685	677.1	55.3	ND	5722.2	157.1	10692.7	1.4
4/20/2013	E-rinse	12.3	24540	17320	2293.6	34.5	1.6	5287.3	404.3	8366.2	3.7
4/22/2013	E-rinse	12.5	25070	17495	39.0	39.5	ND	5844.7	392.4	9439.5	2.0
4/25/2013	E-rinse	12.1	26920	23425	1006.5	21.4	0.5	7299.3	146.4	14303.6	2.1
4/26/2013	E-rinse	10.2	26820	24060	433.1	40.9	8.4	8193.3	120.8	16085.1	1.8
4/30/2013	E-rinse	11.8	25710	22020	ND	17.4	1.5	7515.0	175.1	13724.0	2.0
5/1/2013	E-rinse	10.6	23010	19755	597.8	5.4	4.0	6396.7	96.1	8177.8	3.0
5/2/2013	E-rinse	11.2	23210	19755	658.8	4.2	0.6	6560.5	55.9	8178.6	4.0
5/29/2013	E-rinse	9.8	22330	20130	390.4	24.3	11.1	6461.0	232.3	13288.1	1.1
5/31/2013	E-rinse	9.8	21460	19660	384.3	24.4	11.4	6012.4	164.5	11985.1	4.0
6/3/2013	E-rinse	10.0	20890	19335	134.2	28.7	12.2	6022.6	152.3	11408.5	4.0
6/14/2013	E-rinse	11.3	22370	19230	305.0	23.1	1.9	6019.3	153.3	11682.6	10.0
6/18/2013	E-rinse	11.1	22400	19495	ND	26.6	2.6	6047.6	165.5	11479.8	4.0
6/19/2013	E-rinse	11.0	22550	19840	170.8	21.6	3.6	6144.6	175.9	11270.8	1.0

		pH	Conductivity	TDS*	HCO ₃	Ca	Mg	Na	Cl	SO ₄	SiO ₂
		-	(µS/cm)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)
6/20/2013	E-rinse	11.1	22500	19750	195.2	19.3	3.5	6099.5	237.5	11029.7	2.0
6/21/2013	E-rinse	11.0	22800	20100	164.7	15.7	2.8	6182.2	193.7	11089.2	3.0
6/24/2013	E-rinse	10.9	22660	19785	250.1	24.1	3.6	6090.9	268.7	11081.9	3.0
6/26/2013	E-rinse	11.0	23200	20285	201.3	26.3	3.1	6344.2	124.8	12361.3	3.0
6/27/2013	E-rinse	10.8	24090	21450	280.6	21.3	3.7	6840.0	188.3	12777.6	2.0
6/28/2013	E-rinse 4pm	10.8	23840	21275	305.0	22.3	3.9	6885.6	221.1	12481.9	6.0
7/11/2013	E-rinse	7.7	24770	22480	256.2	38.2	7.5	6737.9	348.3	11895.5	3.0
7/12/2013	E-Rinse 3:45pm	7.9	26070	21710	146.4	47.9	ND	6903.8	335.1	12896.5	3.0
7/14/2013	E-rinse	3.7	32200	31535	ND	82.5	14.0	9408.1	456.7	16055.6	4.0
7/20/2013	E-rinse	10.1	23840	21470	170.8	28.6	4.3	6651.3	263.7	12765.5	3.0
3/22/2013	Mix Na	7.9	110800	144730	3523.4	ND	ND	ND	ND	ND	10.9
3/26/2013	Mix Na	7.8	123700	179780	5827.9	429.8	321.4	66561.2	11834.5	113899.9	2.7
3/27/2013	Mix Na	7.9	125600	186020	4912.9	516.0	340.6	69412.6	12294.5	114761.8	7.7
3/28/2013	Mix Na	7.7	134900	224640	6425.0	611.7	373.8	74624.2	15236.4	131499.1	2.0
3/29/2013	Mix Na	7.6	126900	193655	5620.0	490.4	447.8	65855.9	13326.6	111874.7	6.0
4/5/2013	Mix Na	8.1	104700	129590	4392.0	193.6	120.0	41239.3	8378.6	71578.8	3.9
4/8/2013	Mix Na	7.9	118700	164365	4538.4	382.9	303.2	52196.4	11415.9	91183.5	9.0
4/9/2013	Mix Na	7.7	114900	152920	4514.0	ND	ND	ND	ND	ND	
4/10/2013	Mix Na	7.9	111100	163935	5172.8	ND	ND	ND	ND	ND	7.3
4/11/2013	Mix Na	7.8	107800	154080	4544.5	664.0	297.9	55603.9	13500.7	91153.4	6.8
4/17/2013	Mix Na	7.6	114700	143850	3672.2	668.0	249.5	51058.0	11331.0	87719.8	5.0
4/18/2013	Mix Na	7.6	118400	164285	4428.6	746.0	298.4	53441.1	11571.6	93002.5	8.1
4/20/2013	Mix Na	7.7	116900	160290	3775.9	694.2	209.1	59274.3	12756.9	97633.9	6.2
4/22/2013	Mix Na	7.8	120700	173135	ND	748.8	274.9	59865.3	14669.0	100324.8	7.0
4/25/2013	Mix Na	8.2	114900	161990	2807.2	650.2	229.7	53244.5	10238.6	98126.8	6.9
4/26/2013	Mix Na	8.3	11700	151825	3379.4	814.5	228.7	53477.4	9532.0	90613.1	9.7
4/30/2013	Mix Na	8.2	116100	163070	ND	907.5	248.6	52591.4	9476.5	96569.0	9.0
5/1/2013	Mix Na	8.3	116000	160500	ND	519.0	193.9	50293.8	10213.7	87932.3	8.0

		pH	Conductivity	TDS*	HCO ₃	Ca	Mg	Na	Cl	SO ₄	SiO ₂
		-	(µS/cm)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)
5/2/2013	Mix Na	8.1	112100	150346	ND	216.1	85.2	52606.0	9764.8	88642.4	9.0
5/29/2013	Mix Na	8.2	111500	167535	1421.3	884.2	316.2	56083.2	10383.5	112802.3	7.7
5/31/2013	Mix Na	7.5	108300	164280	1128.5	910.9	295.2	54198.0	9437.8	103814.8	8.0
6/3/2013	Mix Na	7.5	106700	161935	597.8	1056.3	406.9	50966.7	9285.3	91857.9	9.0
6/4/2013	Mix Na	7.4	105400	155070	854.0	1169.2	507.4	59856.7	9621.5	107103.6	10.0
6/14/2013	Mix Na	8.0	119300	177680	3233.0	702.5	172.9	57315.6	10066.4	100496.3	12.0
6/18/2013	Mix Na	7.9	119000	174560	3135.4	716.2	210.4	56184.2	11420.0	97511.5	20.0
6/19/2013	Mix Na	7.7	114900	162465	2696.2	533.5	168.8	52567.5	11023.0	89800.1	7.0
6/20/2013	Mix Na	7.9	115800	162755	2842.6	418.0	136.4	53606.3	11610.5	90813.6	8.0
6/21/2013	Mix Na	8.0	120700	177400	2281.4	294.0	141.5	58853.9	12393.8	99671.8	8.0
6/26/2013	Mix Na	7.8	115600	163260	3184.2	575.7	152.3	52207.8	10828.6	90115.8	27.0
6/27/2013	Mix Na	7.8	114600	159795	2806.0	417.8	143.1	53582.5	11766.5	91412.5	8.0
6/28/2013	Mix Na 4pm	8.0	116000	163795	3294.0	416.0	132.1	55882.5	12409.3	94414.7	11.0
7/8/2013	Mix Na	7.8	115200	163525	3294.0	625.0	141.8	53842.5	11327.3	91689.6	11.0
7/11/2013	Mix Na	8.2	114800	163035	3538.0	672.4	325.0	49713.1	11022.8	86620.7	5.0
7/13/2013	Mix Na 5:15pm	7.8	120900	164505	3538.0	816.6	497.2	49942.5	10952.0	88928.1	8.0
7/14/2013	Mix Na	8.3	109300	159990	2745.0	1051.7	531.6	65882.8	84716.6	ND	9.0
7/20/2013	Mix Na	8.3	111800	167340	3660.0	1361.2	260.0	54019.8	11258.3	94552.9	17.0
3/22/2013	Mix Cl	2.5	138600	122800	0.0	ND	ND	ND	ND	ND	3.0
3/25/2013	Mix Cl	4.5	145100	134565	41.5	10042.4	11288.7	ND	70676.2	1731.9	6.0
3/26/2013	Mix Cl	5.4	131700	117925	42.7	15481.4	9767.5	10479.1	54718.6	881.5	2.5
3/27/2013	Mix Cl	5.2	151900	133305	41.5	18574.8	9490.1	13432.3	82716.3	975.7	4.2
3/28/2013	Mix Cl	5.5	151000	195610	97.6	18100.9	6605.9	13108.2	ND	1073.2	4.0
3/29/2013	Mix Cl	5.5	150600	202925	67.1	17210.7	ND	14223.2	162830.5	ND	4.0
4/5/2013	Mix Cl	2.2	166300	163125	0.0	20130.6	9205.8	14762.7	85554.8	801.4	2.0
4/8/2013	Mix Cl	2.2	176300	184045	0.0	22372.1	10560.5	16032.1	95674.0	553.6	5.0
4/9/2013	Mix Cl	2.1	177600	185610	0.0	ND	ND	ND	ND	ND	3.1
4/10/2013	Mix Cl	2.1	166400	182330	0.0	ND	ND	ND	ND	ND	2.8

		pH	Conductivity	TDS*	HCO ₃	Ca	Mg	Na	Cl	SO ₄	SiO ₂
		-	(µS/cm)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)
4/11/2013	Mix Cl	1.8	167100	185970	0.0	25902.1	10906.1	17675.0	102639.4	687.2	5.3
4/17/2013	Mix Cl	2.4	160100	157990	0.0	20176.2	8661.7	14609.3	84681.1	738.0	ND
4/18/2013	Mix Cl	2.1	177000	214110	0.0	23617.7	10257.7	17220.4	100542.9	599.4	3.9
4/22/2013	Mix Cl	1.9	175400	165435	0.0	23662.9	9757.0	17339.0	98621.2	888.5	6.0
4/25/2013	Mix Cl	2.0	171500	181750	0.0	21937.3	9058.1	17036.8	100648.2	1273.3	5.2
4/26/2013	Mix Cl	2.0	172900	179395	0.0	24838.9	10644.0	18448.3	102558.7	1040.0	7.1
4/30/2013	Mix Cl	1.9	160200	167725	0.0	22035.8	9422.0	16780.0	93362.9	1756.9	8.0
5/2/2013	Mix Cl	1.9	170300	178655	0.0	21810.8	9103.3	17393.0	93827.2	748.2	5.0
5/29/2013	Mix Cl	1.6	161500	162945	0.0	21821.0	8972.7	20064.4	108610.0	1463.0	6.4
5/31/2013	Mix Cl	1.9	164600	185585	0.0	24422.4	10169.2	20368.6	111472.7	922.3	8.0
6/3/2013	Mix Cl	1.6	161400	170560	0.0	19838.5	8649.8	16272.8	86346.2	612.1	6.0
6/4/2013	Mix Cl	1.6	164500	182460	0.0	19707.6	8365.4	17744.1	86663.3	895.7	5.0
6/14/2013	Mix Cl	3.3	162100	163900	0.0	19565.1	8621.6	17593.3	87485.8	981.7	6.0
6/18/2013	Mix Cl	6.2	172000	182675	97.6	21066.8	9216.3	19178.0	94827.5	888.8	5.0
6/19/2013	Mix Cl	6.0	172400	187285	115.9	21862.6	9548.2	19890.5	97863.3	573.2	5.0
6/20/2013	Mix Cl	5.8	173700	187080	48.8	21677.8	9486.9	19286.9	96753.5	419.1	5.0
6/21/2013	Mix Cl	5.8	175100	229990	42.7	16333.0	7682.0	14986.5	74263.0	365.9	5.0
6/24/2013	Mix Cl	6.0	170600	186200	115.9	20614.2	8994.1	19071.6	93193.9	806.9	5.0
6/26/2013	Mix Cl	3.9	168300	183335	0.0	21030.2	8944.3	18699.0	93553.4	1350.6	5.0
6/27/2013	Mix Cl	5.8	172300	190615	128.1	25921.1	11310.8	23294.2	116668.5	766.1	5.0
6/28/2013	Mix Cl 4pm	6.0	171700	185835	134.2	22192.2	9642.9	19634.4	99239.8	669.7	10.0
7/8/2013	Mix Cl	3.6	167800	185150	0.0	21553.0	9489.9	18665.0	94122.8	814.9	5.0
7/11/2013	Mix Cl	5.1	169600	179140	36.6	20784.5	9499.7	17368.7	91691.4	780.8	5.0
7/13/2013	Mix Cl 5:15pm	5.9	180900	180500	85.4	21523.0	9600.5	17199.8	93050.7	1380.8	5.0
7/14/2013	Mix Cl	5.8	154500	166840	36.6	20577.5	9146.1	14606.2	84550.0	1646.8	5.0
7/20/2013	Mix Cl	4.2	140900	141025	6.1	17407.4	7143.4	15695.0	78528.5	1896.8	6.0
3/22/2013	NaCl	5.5	46700	26120	67.1	ND	ND	ND	ND	ND	11.0
3/26/2013	NaCl	6.0	50300	29115	101.3	39.0	22.2	11921.7	18353.3	270.3	8.5

		pH	Conductivity	TDS*	HCO ₃	Ca	Mg	Na	Cl	SO ₄	SiO ₂
		-	(µS/cm)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)
3/27/2013	NaCl	6.0	48300	29685	36.6	1.2	0.0	10760.6	16399.9	229.6	7.0
3/28/2013	NaCl	6.2	48700	34055	110.4	47.4	44.4	8823.7	13321.4	233.2	5.0
3/29/2013	NaCl	6.3	51800	33080	107.4	50.8	42.5	10305.4	15972.2	262.2	8.0
4/8/2013	NaCl	4.0	52100	30500	0.0	61.6	27.2	11642.2	17662.2	611.3	
4/8/2013	NaCl	3.7	51600	30180	0.0	65.2	39.0	11360.9	17354.3	645.7	6.0
4/9/2013	NaCl	3.3	52300	30395	0.0	ND	ND	ND	ND	ND	4.2
4/10/2013	NaCl	3.5	45700	26150	0.0	ND	ND	ND	ND	ND	5.2
4/11/2013	NaCl	3.4	42500	26205	0.0	58.1	20.1	10358.5	12174.9	265.0	4.7
4/17/2013	NaCl	3.7	48600	27570	0.0	60.3	20.7	10878.6	16693.2	463.6	2.1
4/18/2013	NaCl	3.6	50700	29925	0.0	93.4	40.7	11445.2	17547.2	702.7	5.5
4/20/2013	NaCl	3.6	47500	27480	0.0	ND	ND	ND	17912.2	666.0	2.6
4/25/2013	NaCl	3.7	56000	33425	0.0	58.7	55.8	14951.7	23316.6	1962.5	12.5
4/26/2013	NaCl	3.7	55300	33010	0.0	163.6	53.8	14166.2	20482.5	797.4	10.0
4/30/2013	NaCl	3.2	55800	32825	0.0	135.4	50.2	14533.7	21277.9	310.4	9.0
5/1/2013	NaCl	3.4	57300	34060	0.0	53.7	18.5	13771.2	21298.3	909.4	7.0
5/2/2013	NaCl	3.3	56500	33440	0.0	61.6	22.3	7847.8	11409.3	151.8	7.0
5/29/2013	NaCl	2.9	52800	34010	0.0	52.6	38.0	13078.1	21649.4	2212.7	10.3
5/31/2013	NaCl	3.2	52200	33455	0.0	61.9	36.6	12764.9	20533.3	1073.9	9.0
6/3/2013	NaCl	2.9	50200	32550	0.0	78.7	56.0	12111.3	18207.4	1080.3	8.0
6/14/2013	NaCl	6.2	54400	32745	109.8	51.6	24.0	12417.8	18938.8	752.7	8.0
6/18/2013	NaCl	6.4	54700	33340	97.6	52.4	26.0	12475.0	18978.1	830.0	5.0
6/19/2013	NaCl	6.3	54400	33285	91.5	46.3	23.6	12429.2	18544.7	938.7	5.0
6/20/2013	NaCl	6.3	54800	33765	103.7	38.2	21.1	12586.3	18804.9	1033.0	4.0
6/21/2013	NaCl	6.5	54400	36190	213.5	31.9	22.1	12753.9	18675.9	1523.3	6.0
6/24/2013	NaCl	6.4	55500	33825	164.7	39.3	23.8	12671.0	19116.2	985.7	5.0
6/26/2013	NaCl	6.1	55400	33365	134.2	48.8	23.9	13201.3	19942.9	1007.3	5.0
6/27/2013	NaCl	6.3	56000	34335	128.1	39.8	22.8	13631.6	20380.8	1227.0	5.0
6/28/2013	NaCl 4pm	6.4	54000	32880	140.3	33.8	17.6	12647.5	18886.7	1179.1	4.0

		pH	Conductivity	TDS*	HCO ₃	Ca	Mg	Na	Cl	SO ₄	SiO ₂
		-	(µS/cm)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)
7/11/2013	NaCl	6.4	53400	33345	195.2	73.2	51.3	12450.4	18481.4	1570.3	5.0
7/12/2013	NaCl 3:45pm	6.5	56100	33320	140.3	98.7	83.1	12772.1	18400.1	2301.1	5.0
7/14/2013	NaCl	7.0	53000	35530	219.6	171.2	81.3	13723.7	17518.0	3580.3	6.0
7/20/2013	NaCl stand pipe	6.7	56300	35065	103.7	49.7	24.0	12990.6	19597.4	1229.5	8.0
3/22/2013	NF conc	6.6	7880	7880	226.9	ND	ND	ND	ND	ND	27.2
3/25/2013	NF conc	6.5	8400	8350	244.0	699.8	801.0	ND	217.5	4309.8	25.0
3/26/2013	NF conc	7.4	4130	9045	380.6	472.5	208.8	924.0	257.1	4693.3	23.3
3/27/2013	NF conc	6.5	8520	8070	244.0	416.4	624.9	933.5	242.4	4494.4	20.1
3/28/2013	NF conc	7.4	8040	8960	327.0	427.6	864.5	871.4	254.6	4803.6	26.0
4/5/2013	NF conc	6.9	8020	8070	289.1	466.6	485.9	844.9	215.5	4213.2	25.2
4/8/2013	NF conc	6.9	9110	9550	248.9	475.8	508.9	895.8	206.9	4455.2	26.0
4/9/2013	NF conc	6.9	19380	9820	307.4	ND	ND	ND	ND	ND	22.0
4/10/2013	NF conc	6.8	8520	8120	301.3	ND	ND	ND	ND	ND	23.4
4/11/2013	NF conc	6.9	7840	8070	280.6	547.6	515.0	933.0	242.2	3929.9	25.7
4/17/2013	NF conc	7.5	7850	7445	291.6	539.9	456.5	864.3	217.5	4291.6	29.3
4/18/2013	NF conc	6.9	7880	8015	236.7	465.3	424.3	836.1	187.4	4017.1	27.3
4/20/2013	NF conc	6.9	8170	8300	263.5	512.0	478.0	917.2	234.1	4556.8	25.8
4/22/2013	NF conc	6.9	8400	7830	201.3	679.1	439.0	941.1	198.9	4681.5	25.0
4/25/2013	NF conc	7.0	77900	7745	152.5	488.6	431.0	862.0	167.2	4429.1	20.5
4/26/2013	NF conc	6.9	7640	7060	126.9	534.7	471.9	960.0	184.1	4168.6	23.4
4/30/2013	NF conc	6.2	7890	7960		505.7	488.9	917.6	174.1	4357.5	23.0
5/29/2013	NF conc	6.1	8500	8900	95.2	916.7	363.3	1108.2	84.2	6049.3	22.2
5/31/2013	NF conc	6.0	8130	8385	329.4	452.7	441.0	1111.8	113.0	5068.1	25.0
6/3/2013	NF conc	6.4	9000	10300	54.9	1043.9	487.1	1095.8	122.8	6130.1	24.0
6/14/2013	NF conc	7.1	9060	9465	189.1	977.5	432.0	991.8	163.0	5573.4	28.0
6/18/2013	NF conc	7.1	8350	8535	146.4	841.0	369.7	946.0	147.0	5002.8	29.0
6/19/2013	NF conc	6.7	8210	8350	146.4	820.6	357.0	953.8	154.8	4882.6	27.0

		pH	Conductivity	TDS*	HCO ₃	Ca	Mg	Na	Cl	SO ₄	SiO ₂
		-	(µS/cm)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)
6/20/2013	NF conc	6.8	7860	7940	146.4	794.5	343.2	903.3	156.0	4673.5	26.0
6/21/2013	NF conc	6.9	7830	7920	134.2	769.5	364.1	857.1	152.9	4445.2	27.0
6/24/2013	NF conc	6.8	7990	8080	134.2	797.0	345.7	912.8	148.2	4808.8	26.0
6/26/2013	NF conc	6.8	7990	8155	195.2	781.3	335.7	898.4	157.6	4718.3	27.0
6/27/2013	NF conc	6.6	8140	8235	146.4	794.8	346.1	986.1	148.4	4946.6	27.0
6/28/2013	NF Conc. 4pm	6.8	7920	8025	176.9	780.9	340.0	936.7	161.9	4693.6	26.0
7/11/2013	NF Perm	6.8	8190	8335	176.9	831.3	ND	943.5	3181.5	ND	26.0
7/12/2013	NF Conc 3:45pm	7.0	8570	8180	176.9	784.4	359.1	936.3	151.0	4826.6	25.0
7/14/2013	NF conc	7.3	8230	8440	225.7	862.9	397.5	961.7	197.2	4913.1	26.0
7/20/2013	NF conc	7.0	7780	7830	176.9	725.3	329.2	1017.9	124.1	4647.1	26.0
3/22/2013	NF Feed	6.7	3920	4800	144.6	ND	ND	ND	ND	ND	22.0
3/25/2013	NF Feed	6.4	4150	4435	152.5	211.9	365.5	ND	300.2	1657.0	22.3
3/26/2013	NF Feed	7.4	8880	3355	211.1	340.1	197.7	421.5	341.9	1609.1	25.7
3/27/2013	NF Feed	6.5	4040	2870	152.5	328.8	202.9	407.0	332.8	1582.0	26.7
3/28/2013	NF Feed	7.3	3980	3530	200.1	331.2	95.9	437.7	330.7	ND	
4/5/2013	NF Feed	6.8	3870	3020	179.3	290.5	133.7	367.5	301.9	1428.6	18.4
4/8/2013	NF Feed	6.7	4000	3180	173.2	304.5	140.5	387.8	311.1	1504.8	21.0
4/9/2013	NF Feed	6.7	4030	3060	172.0	ND	ND	ND	ND	ND	17.0
4/10/2013	NF Feed	7.0	3890	3150	187.9	ND	ND	ND	ND	ND	26.3
4/11/2013	NF Feed	7.1	3870	3215	219.0	358.4	133.9	415.6	324.7	1628.9	19.3
4/17/2013	NF Feed	7.4	3860	2960	176.3	303.3	130.3	381.7	301.4	1446.9	19.7
4/18/2013	NF Feed	7.0	3910	3210	169.6	311.4	131.9	400.6	298.0	1522.2	21.5
4/20/2013	NF Feed	6.8	4160	3415	127.5	343.6	135.4	406.8	337.2	1617.7	18.6
4/22/2013	NF Feed	7.6	3880	2960	158.6	315.9	117.1	408.9	319.5	1570.4	21.0
4/25/2013	NF Feed	6.7	3900	3005	ND	312.2	140.3	379.3	317.3	1529.4	26.1
4/26/2013	NF Feed	6.7	3900	2990	123.2	330.1	124.7	406.0	315.3	742.2	42.3

		pH	Conductivity	TDS*	HCO ₃	Ca	Mg	Na	Cl	SO ₄	SiO ₂
		-	(µS/cm)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)
4/30/2013	NF Feed	6.2	3920	3185	ND	334.2	128.0	403.2	307.0	1570.0	21.0
5/1/2013	NF Feed	6.7	3880	6740	158.6	300.4	124.2	399.7	309.8	1520.1	22.0
5/2/2013	NF Feed	6.9	3770	2920	148.2	149.7	117.6	398.6	145.3	1394.8	22.0
5/29/2013	NF Feed	6.0	3250	2570	67.1	233.4	91.9	395.7	228.7	1441.5	19.7
5/31/2013	NF Feed	5.8	3920	3230	79.3	303.0	130.8	471.2	278.1	1821.3	22.0
6/3/2013	NF Feed	6.5	3860	3290	97.6	303.2	138.6	450.5	280.6	1683.1	23.0
6/14/2013	NF Feed	7.1	3410	2545	122.0	232.9	100.3	360.0	253.3	1266.8	27.0
6/18/2013	NF Feed	7.0	3450	2635	109.8	239.7	102.8	377.9	244.0	1328.3	24.0
6/19/2013	NF Feed	6.7	3560	2750	97.6	247.6	105.9	401.3	246.5	1399.3	22.0
6/20/2013	NF Feed	6.6	3440	2640	97.6	236.8	100.2	376.6	240.4	1323.2	22.0
6/21/2013	NF Feed	6.9	3410	2635	97.6	239.2	110.8	378.5	152.6	4592.4	23.0
6/24/2013	NF Feed	6.7	3480	2650	115.9	242.6	104.1	385.6	243.7	1343.0	23.0
6/26/2013	NF Feed	6.7	3530	2700	122.0	245.5	103.9	390.6	247.6	1376.3	23.0
6/27/2013	NF Feed	6.7	3550	2720	146.4	240.2	102.3	406.5	243.8	1389.1	23.0
6/28/2013	NF Feed 4pm	6.7	3490	2655	164.7	235.7	101.9	396.9	241.1	1341.1	22.0
7/11/2013	NF Feed	6.7	3700	2855	256.2	244.9	114.7	414.4	251.6	1468.3	23.0
7/12/2013	NF Feed 3:45pm	6.8	3810	2800	122.0	239.4	109.2	407.6	251.0	1420.7	21.0
7/14/2013	NF Feed	7.1	3620	2940	140.3	262.0	118.3	412.8	275.8	1407.5	22.0
7/20/2013	NF Feed	6.9	3360	2655	146.4	216.2	97.6	418.1	213.2	1346.6	23.0
3/22/2013	NF Perm	6.9	1438	200	122.6	ND	ND	ND	ND	ND	18.7
3/26/2013	NF Perm	7.2	1539	790	156.2	43.8	17.9	221.8	359.5	29.3	19.5
3/27/2013	NF Perm	6.5	1496	960	119.3	45.2	15.8	217.6	352.5	29.6	27.5
3/28/2013	NF Perm	7.2	1539	1070	147.6	50.2	19.6	ND	ND	34.0	20.0
3/29/2013	NF Perm	7.3	1463	860	170.8	0.5	ND	208.9	327.1	28.5	20.0
4/5/2013	NF Perm	6.9	1496	820	142.7	46.2	16.1	209.6	339.4	48.8	17.1
4/8/2013	NF Perm	6.7	1593	965	276.9	46.5	16.1	217.7	354.1	43.3	19.0
4/9/2013	NF Perm	7.0	1565	875	137.9	ND	ND	ND	ND	ND	16.9

		pH	Conductivity	TDS*	HCO ₃	Ca	Mg	Na	Cl	SO ₄	SiO ₂
		-	(µS/cm)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)
4/10/2013	NF Perm	6.6	1501	1980	280.6	ND	ND	ND	ND	ND	17.1
4/11/2013	NF Perm	6.8	1517	785	144.0	53.2	15.3	230.8	370.1	43.0	16.8
4/17/2013	NF Perm	8.2	1571	780	140.3	50.1	16.7	211.7	333.1	70.3	17.9
4/18/2013	NF Perm	7.0	1544	945	126.9	47.5	15.7	214.6	336.7	70.7	18.6
4/20/2013	NF Perm	6.9	1602	910	142.7	52.9	15.4	220.6	353.9	77.6	17.6
4/22/2013	NF Perm	8.1	1516	780	134.2	49.3	15.6	223.3	337.7	80.5	20.0
4/25/2013	NF Perm	6.6	1658	3140	100.7	55.9	20.0	217.8	354.6	104.4	19.8
4/26/2013	NF Perm	6.6	1609	900	103.1	56.9	19.6	223.1	360.6	93.4	17.5
4/30/2013	NF Perm	6.2	1682	915		61.1	21.8	226.9	370.4	109.1	20.0
5/1/2013	NF Perm	6.8	1642	2540	134.2	52.6	20.0	224.0	339.2	100.6	21.0
5/2/2013	NF Perm	7.0	1601	865	138.2	54.3	19.4	221.5	325.7	101.3	21.0
5/29/2013	NF Perm	6.5	1145	605	63.4	23.2	8.5	184.0	289.7	62.5	18.8
5/31/2013	NF Perm	5.6	1373	715	109.8	30.8	11.7	220.9	343.1	82.0	22.0
6/3/2013	NF Perm	6.4	1398	770	97.6	34.0	13.0	223.5	338.3	72.2	21.0
6/14/2013	NF Perm	7.1	1338	615	109.8	26.1	8.3	182.4	289.5	38.1	21.0
6/18/2013	NF Perm	7.1	1237	600	97.6	24.3	8.0	183.7	281.2	39.8	21.0
6/19/2013	NF Perm	6.7	1226	645	79.3	24.8	8.1	187.8	282.6	35.8	21.0
6/20/2013	NF Perm	6.7	1203	605	91.5	25.4	8.1	182.7	278.6	33.2	21.0
6/21/2013	NF Perm	6.8	1224	625	79.3	25.6	8.9	181.9	284.5	34.2	21.0
6/24/2013	NF Perm	6.8	1217	660	103.7	25.8	8.4	185.1	280.0	34.0	21.0
6/26/2013	NF Perm	6.6	1226	650	85.4	25.7	8.1	187.1	285.0	36.5	21.0
6/27/2013	NF Perm	6.7	1219	625	85.4	23.6	7.8	188.7	280.0	36.3	21.0
6/28/2013	NF Perm 4pm	6.6	1213	580	109.8	23.9	7.9	186.2	275.2	34.1	20.0
7/8/2013	NF Perm	6.7	1230	635	109.8	24.7	9.0	191.9	286.7	32.7	22.0
7/9/2013	NF Perm	6.7	1221	625	109.8	24.4	9.1	191.3	283.1	33.0	22.0
7/10/2013	NF Perm	6.8	1329	680	146.4	27.8	10.5	204.7	308.0	34.4	22.0
7/11/2013	NF conc	6.7	1285	650	103.7	26.9	8.6	187.7	281.4	170.8	22.0
7/12/2013	NF Perm	6.9	1340	640	91.5	24.5	9.0	196.6	295.4	31.4	20.0

		pH	Conductivity	TDS*	HCO ₃	Ca	Mg	Na	Cl	SO ₄	SiO ₂
		-	(µS/cm)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)
	3:45pm										
7/14/2013	NF Perm	7.2	1312	695	109.8	29.9	9.4	209.3	318.4	25.2	21.0
7/20/2013	NF Perm	7.2	1097	590	103.7	17.7	7.0	181.9	260.4	30.3	21.0
3/22/2013	raw	7.7	3970	2710	280.6	ND	ND	ND	ND	ND	17.0
3/25/2013	Raw	7.4	4060	5550	263.5	202.3	379.1	ND	1103.8	ND	16.0
3/26/2013	RAW	7.4	4120	3375	305.6	342.8	123.5	439.5	421.2	1472.0	19.2
3/27/2013	raw	7.4	4120	3255	263.5	330.7	169.4	384.2	395.9	1380.5	20.3
3/28/2013	raw	7.5	4130	3465	245.2						18.0
3/29/2013	raw	7.5	3910	3205	229.4	297.5	ND	379.7	396.2	1412.9	20.0
4/5/2013	raw	7.4	4050	3150	354.4	304.5	138.4	376.6	378.7	1372.8	17.2
4/8/2013	raw	7.3	4120	3155	406.3	308.8	141.3	376.8	384.5	1383.6	20.0
4/9/2013	raw	7.4	4020	2940	252.5	ND	ND	ND	ND	ND	17.7
4/10/2013	raw	7.3	3840	3070	337.9	ND	ND	ND	ND	ND	16.1
4/11/2013	raw	7.7	4000	2875	136.0	370.4	144.9	411.6	423.9	1455.3	17.6
4/17/2013	raw	7.8	3990	2980	247.1	324.0	131.7	379.7	376.1	1345.3	19.2
4/18/2013	raw	7.5	3960	3210	250.1	318.6	128.3	373.8	368.2	1336.7	19.5
4/20/2013	raw	7.6	4070	3245	213.5	337.2	132.6	397.1	410.8	1405.6	19.4
4/22/2013	raw	7.4	3850	2850	ND	324.0	125.6	391.4	373.5	1333.5	20.0
4/25/2013	raw	7.8	3920	2870	205.0	316.1	123.0	380.2	381.0	1325.8	17.6
4/26/2013	raw	7.6	3950	2960	238.5	343.6	134.4	400.7	398.2	1356.1	19.9
4/30/2013	raw	7.4	4030	3065	ND	356.5	143.0	412.6	404.0	1429.6	19.0
5/1/2013	raw	7.6	3750	16280	242.2	290.9	116.0	374.3	343.2	1278.4	20.0
5/2/2013	raw	7.6	3780	2910	241.6	320.7	127.0	398.4	355.7	1307.3	20.0
5/29/2013	raw	7.7	3200	2435	203.7	248.5	94.6	372.0	296.4	1265.3	18.7
5/31/2013	raw	7.6	3400	2670	201.3	276.2	107.4	383.1	316.3	1312.4	23.0
6/3/2013	raw	7.7	3750	3045	305.0	326.3	146.1	428.0	356.1	1485.4	21.0
6/18/2013	raw	7.9	3630	2515	244.0	250.4	103.9	380.0	327.6	1187.0	21.0
6/19/2013	raw	7.7	3470	2610	244.0	258.1	105.2	360.0	298.2	1161.6	20.0

		pH	Conductivity	TDS*	HCO ₃	Ca	Mg	Na	Cl	SO ₄	SiO ₂
		-	(µS/cm)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)
6/20/2013	raw	7.8	3450	2555	207.4	255.4	103.1	357.3	296.2	1150.6	20.0
6/21/2013	raw	7.9	3430	2580	183.0	255.7	114.1	359.5	298.2	1161.7	22.0
6/24/2013	raw	7.9	3460	2585	237.9	258.4	105.7	359.3	300.5	1169.2	21.0
6/26/2013	raw	7.9	3480	2575	262.3	262.0	105.9	363.7	309.0	1193.4	21.0
6/27/2013	raw	7.8	3430	2535	256.2	254.4	102.8	359.4	299.1	1166.8	22.0
6/28/2013	raw 4pm	7.8	3420	2535	244.0	255.6	104.9	359.8	301.4	1159.0	21.0
7/8/2013	RAW	7.7	3470	2610	237.9	259.9	110.8	366.1	299.6	1193.1	20.0
7/9/2013	RAW	7.7	3440	2565	231.8	256.2	110.3	361.0	295.8	1165.5	20.0
7/10/2013	RAW	7.6	3840	2925	231.8	291.2	136.6	395.5	330.7	1344.9	21.0
7/11/2013	raw	7.7	3570	2675	231.8	262.2	116.4	365.5	309.1	1212.2	22.0
7/12/2013	RAW 3:45pm	7.7	3700	2670	225.7	253.9	111.6	360.7	316.9	1208.2	20.0
7/13/2013	RAW 5:45pm	7.7	3650	2635	219.6	262.5	108.3	360.3	304.4	1166.2	21.0
7/14/2013	raw	8.0	3230	2575	231.8	255.1	110.0	357.1	293.2	1153.7	21.0
7/20/2013	RAW	7.7	3170	2430	256.2	237.8	100.9	351.5	271.4	1097.5	21.0
3/22/2013	RO (DI)	6.6	86	360	55.5	ND	ND	ND	ND	ND	1.0
3/25/2013	RO (DI)	6.0	123	30	51.9	ND	18.2	ND	19.8	1.1	1.0
3/26/2013	RO (DI)	7.5	71	170	43.9	0.0	0.0	11.3	13.9	0.7	0.9
3/27/2013	RO (DI)	5.6	65	810	51.9	0.0	0.0	11.2	12.3	4.1	0.4
3/28/2013	RO (DI)	6.7	149	465	48.8	0.0	0.0	25.2	29.0	2.2	2.0
3/29/2013	RO (DI)	6.4	59	240	48.8	1.1	0.4	10.3	12.7	1.3	2.0
4/5/2013	RO (DI)	5.6	40	265	36.6	0.9	0.3	5.3	6.3	0.9	0.0
4/8/2013	RO (DI)	6.2	116	535	43.9	1.7	0.6	19.4	19.1	1.8	2.0
4/9/2013	RO (DI)	6.3	88	190	22.0	ND	ND	ND	ND	ND	0.0
4/10/2013	RO (DI)	5.9	93	340	22.0	ND	ND	ND	ND	ND	0.0
4/11/2013	RO (DI)	6.0	76	305	90.3	0.1	ND	0.1	0.1	0.1	0.1
4/17/2013	RO (DI)	7.2	104	340	ND	1.9	0.6	17.3	16.5	2.3	0.0
4/18/2013	RO (DI)	6.6	82500	425	18.9	1.8	0.5	13.6	13.6	2.0	0.0

		pH	Conductivity	TDS*	HCO ₃	Ca	Mg	Na	Cl	SO ₄	SiO ₂
		-	(µS/cm)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)
4/20/2013	RO (DI)	6.6	107	355	25.0	1.8	0.5	18.5	16.5	1.9	0.0
4/22/2013	RO (DI)	5.8	57800	385	242.2	1.5	0.4	9.2	9.4	1.7	2.0
4/25/2013	RO (DI)	7.0	173	110	51.0	3.3	1.1	29.8	29.6	3.7	3.0
4/26/2013	RO (DI)	6.5	105	215	34.8	1.9	0.6	17.9	15.3	2.7	0.4
4/30/2013	RO (DI)	5.7	88	490	ND	2.3	0.7	13.5	14.1	3.5	2.0
5/1/2013	RO (DI)	6.1	81	150	25.6	3.3	0.8	17.1	13.6	7.7	2.0
5/2/2013	RO (DI)	6.4	83	460	24.4	1.4	0.4	14.6	11.9	1.9	4.0
5/29/2013	RO (DI)	5.8	64	510	46.4	1.3	0.4	10.7	11.0	3.1	2.0
5/31/2013	RO (DI)	6.2	174	225	48.8	1.5	0.5	34.8	27.1	3.1	3.0
6/3/2013	RO (DI)	6.3	59	360	ND	0.7	0.2	11.9	8.6	1.3	2.0
6/18/2013	RO (DI)	6.3	463	515	48.8	3.9	1.7	39.5	74.2	41.2	3.0
6/19/2013	RO (DI)	6.2	54	635	36.6	0.6	0.1	10.1	7.4	1.0	1.0
6/20/2013	RO (DI)	6.7	117	570	54.9	0.9	0.2	23.0	14.4	1.2	2.0
6/21/2013	RO (DI)	6.9	786	385	91.5	12.1	3.7	126.7	182.6	5.5	16.0
6/24/2013	RO (DI)	6.4	104	455	36.6	0.7	0.2	18.9	18.8	4.3	1.0
6/26/2013	RO (DI)	6.7	124	560	61.0	1.0	0.3	24.1	15.7	1.1	2.0
6/27/2013	RO (DI)	6.3	63	645	54.9	0.6	0.2	12.1	7.6	0.8	2.0
6/28/2013	RO (DI) 4pm	6.7	138	115	79.3	1.1	0.3	26.2	18.6	1.3	3.0
7/20/2013	RO	6.5	41000	20	61.0	0.4	0.1	7.1	5.3	0.7	1.0

APPENDIX C2
ALAMOGORDO, YEAR 2
WATER QUALITY ANALYSES

Table C-2.—Water Quality Analyses Excluded from Statistical Analyses

		pH	Conductivity	TDS*	HCO ₃	Ca	Mg	Na	Cl	SO ₄	SiO ₂
Date	Stream	-	(µS/cm)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)
4/19/2013	Edm dil	6.6	4010	3385	78.7	616.1	347.7	850.9	227.5	3876.3	27.1
4/23/2013	Edm dil	6.3	4200	3565	ND	648.4	372.5	864.7	241.8	4096.1	
5/1/2013	Edm dil	6.0	3890	3210	39.0	133.3	72.1	221.4	47.8	1870.2	23.0
5/2/2013	Edm dil	6.3	3610	3035	40.3	132.8	143.2	214.2	99.7	1101.1	26.0
6/28/2013	Edm dil 4pm	6.4	3640	2895	128.1	15.4	6.9	27.8	91.9	1697.7	29.0
4/19/2013	Edm feed	6.8	5510	4940	86.6	1014.9	547.8	1083.7	309.7	5666.1	25.1
4/23/2013	Edm feed	6.6	5820	5375	ND	1092.7	612.7	1117.8	310.4	6130.8	ND
5/1/2013	Edm feed	6.4	5380	4775	81.7	231.7	117.8	287.3	58.1	2814.5	23.0
5/2/2013	Edm feed	6.6	4980	4685	83.0	221.9	102.0	272.3	54.5	1659.8	25.0
6/28/2013	EDM Feed 8:45am	6.4	4910	4330	146.4	ND	ND	ND	ND	ND	25.0
3/25/2013	E-rinse	10.9	19220	14600	183.0	ND	5092.8	ND	107.8	9757.0	3.0
4/19/2013	E-rinse	12.4	25970	20195	1372.5	208.1	1.6	12477.5	610.9	22851.5	1.2
4/23/2013	E-rinse	12.7	25640	16110	ND	51.2	30.8	9172.5	814.2	12017.6	
4/19/2013	Mix Cl	2.1	177400	193360	0.0	54113.4	25462.8	35662.7	215550.6	1629.7	7.4
4/20/2013	Mix Cl	2.0	172200	198520	0.0	18.9	ND	ND	12.9	ND	3.4
4/23/2013	Mix Cl	1.6	176000	169045	ND	47868.2	23668.2	32671.9	199267.4	1639.2	ND
5/1/2013	Mix Cl	1.9	166700	169960	0.0	12013.1	3465.7	8933.4	48738.0	459.6	6.0
3/25/2013	Mix Na	7.8	132800	202690	4912.9	216.9	91829.3	ND	12600.6	118851.0	9.0
4/19/2013	Mix Na	8.0	117100	161145	4026.0	1322.3	568.8	103507.8	23748.4	176571.9	8.1
4/23/2013	Mix Na	7.7	117200	163180	ND	1330.7	614.3	102915.9	23556.4	185453.7	
6/24/2013	Mix Na	7.8	113700	159935	2989.0	835.8	269.3	75460.4	12981.8	131245.2	8.0
3/25/2013	NaCl	6.0	36400	19825	36.6	ND	8384.0	0.0	11848.9	287.1	10.4
4/19/2013	NaCl	3.6	48200	28315	0.0	204.8	68.6	22292.4	18617.4	1209.0	6.0
4/22/2013	NaCl	3.4	50200	29230	0.0	3660.9	1671.0	534108.2	751208.3	34125.4	7.0
4/23/2013	NaCl	3.4	47400	27830	0.0	266.7	44.5	19641.0	26373.8	1792.0	ND
3/29/2013	NF conc	7.2	8120	8735	345.3	647.4	8451.6	8697.9	228.5	47961.4	26.0

		pH	Conductivity	TDS*	HCO ₃	Ca	Mg	Na	Cl	SO ₄	SiO ₂
Date	Stream	-	(µS/cm)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)	(mg/L)
4/19/2013	NF conc	7.0	8110	8355	200.7	1061.1	1124.6	1787.9	471.1	8356.3	26.2
4/23/2013	NF conc	7.1	8010	7645	186.1	999.4	1094.0	1725.9	479.8	8182.6	
5/1/2013	NF conc	6.6	8410	8095	201.3	272.1	94.0	1054.5	2247.6	597.1	24.0
5/2/2013	NF conc	7.1	7750	7975	213.5	276.6	209.5	907.5	157.6	4026.8	25.0
7/8/2013	NF Conc	6.8	7940	7990	183.0	21.9	10.1	149.5	221.4	14.7	26.0
3/29/2013	NF Feed	7.2	3760	3275	201.3	0.5	0.0	0.0	311.9	0.0	23.0
4/19/2013	NF Feed	6.8	4090	3210	115.9	666.6	321.5	800.8	642.4	3240.9	20.8
4/23/2013	NF Feed	7.5	4120	3105	122.6	679.6	328.1	766.1	651.9	3098.0	
3/25/2013	NF Perm	6.3	1442	135	119.3	13.5	216.3	ND	340.9	27.5	18.7
4/19/2013	NF Perm	7.0	1588	835	90.9	102.4	36.7	430.4	704.7	137.1	17.4
4/23/2013	NF Perm	7.4	1715	830	96.4	120.1	47.3	431.9	780.7	141.8	
4/19/2013	raw	7.7	4100	2986	213.5	702.4	331.9	780.0	814.1	2790.2	18.5
4/23/2013	raw	7.9	4090	3000	251.3	685.1	330.8	754.2	812.4	2730.3	ND
4/19/2013	RO (DI)	6.2	55	300	12.8	2.7	0.9	17.3	18.0	2.8	0.0
4/23/2013	RO (DI)	6.3	49	315	ND	2.2	0.7	14.7	14.5	2.6	

APPENDIX C3
BRIGHTON, COLORADO

Appendix C3 - Brighton Water Quality Data

Date	Location	Ca mg/L	K mg/L	Mg mg/L	Na mg/L	Cl mg/L	F mg/L	NO ₃ mg/L
10/1/2013	EDM Dil	133	5.9	34.1	430	110	3.6	19
10/2/2013	EDM Dil	120	5.7	31.3	415	103	3.4	19
10/3/2013	EDM Dil	120	5.7	30.3	422	99	2.0	18
10/4/2013	EDM Dil	138	5.7	34.9	447	110	3.6	20
10/6/2013	EDM Dil	145	6.7	38.6	507	110	3.3	19
10/7/2013	EDM Dil	128	6.3	33.7	469	104	3.4	19
10/8/2013	EDM Dil	140	6.7	37.0	515	110	3.4	20
10/9/2013	EDM Dil	130	6.1	34.5	495	102	3.3	19
10/10/2013	EDM Dil	168	7.4	45.9	569	120	3.1	21
10/11/2013	EDM Dil	161	8.2	50.1	581	113	3.3	21
10/12/2013	EDM Dil	158	8.0	48.5	583	114	3.1	21
10/13/2013	EDM Dil	142	7.1	43.8	545	104	2.7	21
10/15/2013	EDM Dil	136	7.5	41.3	484	131	3.5	23
10/16/2013	EDM Dil	140	8.2	41.0	545	129	3.0	21
10/17/2013	EDM Dil	148	8.0	44.6	564	126	3.0	23
10/18/2013	EDM Dil	153	7.9	46.2	569	125	2.7	23
10/19/2013	EDM Dil	175	8.9	54.4	621	133	2.7	25
10/22/2013	EDM Dil	167	6.0	37.5	498	126	4.3	26
10/23/2013	EDM Dil	186	6.6	41.2	564	129	4.4	25
10/24/2013	EDM Dil	191	7.0	44.5	594	130	4.4	25
10/25/2013	EDM Dil	175	9.6	32.8	614	122	0.9	25
10/26/2013	EDM Dil	195	10.4	41.8	651	130	1.0	27
10/1/2013	EDM Feed	234	7.7	53.8	504	194	3.7	51
10/2/2013	EDM Feed	217	7.6	50.6	484	183	3.7	49
10/3/2013	EDM Feed	211	7.4	48.5	488	175	2.2	46
10/4/2013	EDM Feed	233	7.7	54.5	509	188	3.7	50
10/6/2013	EDM Feed	245	8.3	59.7	572	188	3.7	48
10/7/2013	EDM Feed	222	8.1	52.7	538	181	3.7	49
10/8/2013	EDM Feed	234	8.3	55.7	573	186	3.7	50
10/9/2013	EDM Feed	217	7.9	52.6	549	172	3.3	46
10/10/2013	EDM Feed	267	9.2	66.2	631	198	3.6	50
10/11/2013	EDM Feed	259	9.8	71.1	647	197	3.6	50
10/12/2013	EDM Feed	256	9.8	67.7	647	196	3.6	50
10/13/2013	EDM Feed	234	9.0	65.1	608	177	3.4	49
10/15/2013	EDM Feed	234	9.7	62.9	555	218	3.9	52
10/16/2013	EDM Feed	233	10.2	60.9	610	209	3.3	48
10/17/2013	EDM Feed	237	10.1	63.7	619	198	3.2	51
10/18/2013	EDM Feed	239	9.8	64.5	620	191	2.9	49
10/19/2013	EDM Feed	266	10.3	73.7	676	204	2.8	53
10/22/2013	EDM Feed	293	7.7	57.2	581	217	4.5	61
10/23/2013	EDM Feed	292	7.9	58.0	603	197	4.1	54

Appendix C3 - Brighton Water Quality Data

Date	Location	SO ₄ mg/L	HCO ₃ mg/L	SiO ₂ mg/L	TDS mg/L	pH	Conductivity µS/cm
10/1/2013	EDM Dil	1097	98	112	2205	5.17	2889
10/2/2013	EDM Dil	1055	55	108	2160	5.11	2828
10/3/2013	EDM Dil	1061	61	104	2190	4.96	2836
10/4/2013	EDM Dil	1157	73	112	2335	5.1	3010
10/6/2013	EDM Dil	1304	183	112	2570	5.22	3360
10/7/2013	EDM Dil	1213	49	104	2430	4.71	3200
10/8/2013	EDM Dil	1305	110	116	2625	5.27	3390
10/9/2013	EDM Dil	1273	ND	100	2515	4.41	3270
10/10/2013	EDM Dil	1499	92	108	2965	5.01	3780
10/11/2013	EDM Dil	1519	110	116	3015	4.99	3850
10/12/2013	EDM Dil	1524	ND	112	3005	5.05	3930
10/13/2013	EDM Dil	1448	0	108	2870	3.71	3810
10/15/2013	EDM Dil	1202	104	116	2455	5.55	3360
10/16/2013	EDM Dil	1360	79	116	2625	5.36	3620
10/17/2013	EDM Dil	1424	85	104	2745	5.07	3680
10/18/2013	EDM Dil	1462	55	100	2880	5.12	3410
10/19/2013	EDM Dil	1616	61	100	3220	4.93	3690
10/22/2013	EDM Dil	1135	98	116	2710	5.77	3210
10/23/2013	EDM Dil	1353	79	116	2910	5.57	3410
10/24/2013	EDM Dil	1487	73	112	3095	5.2	3600
10/25/2013	EDM Dil	1550	67	120	3150	5.02	3690
10/26/2013	EDM Dil	1676	61	116	3285	4.76	3890
10/1/2013	EDM Feed	1333	128	112	2885	5.47	3670
10/2/2013	EDM Feed	1315	85	108	2815	5.49	3620
10/3/2013	EDM Feed	1303	73	104	2810	5.3	3580
10/4/2013	EDM Feed	1398	85	108	2960	5.42	3760
10/6/2013	EDM Feed	1565	140	108	3200	5.53	4100
10/7/2013	EDM Feed	1489	92	104	2995	5.18	3900
10/8/2013	EDM Feed	1547	122	108	3195	5.65	4110
10/9/2013	EDM Feed	1499	98	100	3060	5.13	3930
10/10/2013	EDM Feed	1747	140	108	3535	5.47	4510
10/11/2013	EDM Feed	1779	85	116	3625	5.48	4550
10/12/2013	EDM Feed	1772	85	120	3645	5.51	4570
10/13/2013	EDM Feed	1707	ND	108	3470	4.38	4460
10/15/2013	EDM Feed	1469	140	116	3105	5.72	4160
10/16/2013	EDM Feed	1614	116	116	3300	5.7	4380
10/17/2013	EDM Feed	1654	128	100	3320	5.42	4360
10/18/2013	EDM Feed	1687	61	100	3440	5.6	3990
10/19/2013	EDM Feed	1860	73	100	3740	5.41	4280
10/22/2013	EDM Feed	1492	140	116	3360	6	3920
10/23/2013	EDM Feed	1513	146	116	3565	5.83	4120

Appendix C3 - Brighton Water Quality Data

Date	Location	Ca mg/L	K mg/L	Mg mg/L	Na mg/L	Cl mg/L	F mg/L	NO ₃ mg/L
10/24/2013	EDM Feed	308	8.6	63.2	663	203	4.7	55
10/25/2013	EDM Feed	264	12.9	55.7	681	202	1.3	56
10/26/2013	EDM Feed	286	13.1	67.2	711	211	1.3	58
10/1/2013	E-Rinse	26	11.2	0.0	9098	252	0.0	58
10/6/2013	E-Rinse	38	11.3	0.0	9659	248	26.7	110
10/13/2013	E-Rinse	28	5.2	4.2	4977	166	28.5	139
10/16/2013	E-Rinse	127	23.2	11.6	13913	505	48.3	328
10/25/2013	E-Rinse	43	6.9	3.4	5752	244	1.7	1060
10/26/2013	E-Rinse	456	12.7	ND	5487	216	1.6	979
10/1/2013	Mix Cl	18207	300	3674	13910	64642	133.5	126
10/2/2013	Mix Cl	17071	277	3596	13060	62060	131.1	201
10/3/2013	Mix Cl	15900	277	3227	12698	56325	121.1	178
10/4/2013	Mix Cl	15854	283	3259	12951	57535	140.7	226
10/6/2013	Mix Cl	16431	288	3370	12681	58847	129.6	220
10/7/2013	Mix Cl	17661	307	3706	13690	62424	94.0	260
10/8/2013	Mix Cl	16046	277	3410	12693	57331	101.9	235
10/9/2013	Mix Cl	16481	283	3486	13271	59341	96.0	234
10/10/2013	Mix Cl	16872	281	3631	12608	58708	79.5	244
10/11/2013	Mix Cl	12783	259	2966	10595	48354	104.8	179
10/12/2013	Mix Cl	11193	231	2541	9535	43358	102.5	160
10/13/2013	Mix Cl	11443	216	2579	9648	43307	74.5	160
10/15/2013	Mix Cl	14438	279	3271	11166	51881	38.7	221
10/16/2013	Mix Cl	15370	341	3427	13186	59144	113.9	223
10/17/2013	Mix Cl	14925	287	3403	12151	55471	94.6	229
10/18/2013	Mix Cl	15506	274	3565	11919	56255	84.4	257
10/19/2013	Mix Cl	15401	255	3552	11666	54978	56.3	242
10/22/2013	Mix Cl	19708	213	3091	11771	54244	64.8	280
10/23/2013	Mix Cl	19667	223	3246	12606	55914	64.3	238
10/24/2013	Mix Cl	19388	219	3156	12663	54972	66.5	231
10/25/2013	Mix Cl	11780	321	2202	9640	45320	2.4	162
10/26/2013	Mix Cl	11859	318	2382	9325	45079	2.3	159
10/1/2013	Mix Na	134	36.3	15.3	14833	7953	176.6	1836
10/2/2013	Mix Na	86	53.3	8.1	15636	8651	176.5	1917
10/3/2013	Mix Na	170	34.8	24.9	18573	9505	98.5	2245
10/4/2013	Mix Na	287	48.9	39.8	20160	10328	160.0	2538
10/6/2013	Mix Na	235	43.7	47.9	19711	10163	187.7	2428
10/7/2013	Mix Na	384	46.0	65.7	22491	11351	158.6	2804
10/8/2013	Mix Na	291	46.5	45.5	19894	10215	85.8	2487
10/9/2013	Mix Na	358	47.2	63.7	20033	10027	80.8	2451

Appendix C3 - Brighton Water Quality Data

Date	Location	SO ₄ mg/L	HCO ₃ mg/L	SiO ₂ mg/L	TDS mg/L	pH	Conductivity µS/cm
10/24/2013	EDM Feed	1715	134	116	3765	5.69	4270
10/25/2013	EDM Feed	1796	140	116	3770	5.6	4350
10/26/2013	EDM Feed	1924	85	112	3890	5.37	4540
10/1/2013	E-Rinse	17501	195	1	27735	10.52	31100
10/6/2013	E-Rinse	18150	262	1	30530	10.01	33700
10/13/2013	E-Rinse	8728	256	3	38290	8.26	excl
10/16/2013	E-Rinse	24717	268	3	46595	8.16	47400
10/25/2013	E-Rinse	10200	262	3	20640	9.58	22000
10/26/2013	E-Rinse	9081	220	3	18055	9.34	19850
10/1/2013	Mix Cl	756	49	12	123670	4.87	145600
10/2/2013	Mix Cl	743	0	12	122640	2.87	142600
10/3/2013	Mix Cl	642	0	12	114095	2.69	133800
10/4/2013	Mix Cl	649	0	10	113295	2.86	134000
10/6/2013	Mix Cl	714	0	10	108465	2.81	136500
10/7/2013	Mix Cl	761	0	10	113335	2.39	141100
10/8/2013	Mix Cl	705	0	9	103255	2.87	132000
10/9/2013	Mix Cl	724	0	10	108220	2.11	136600
10/10/2013	Mix Cl	716	0	11	110080	2.6	137900
10/11/2013	Mix Cl	691	0	11	106765	2.63	133800
10/12/2013	Mix Cl	612	0	9	90185	2.66	116700
10/13/2013	Mix Cl	661	0	8	87905	2.13	115800
10/15/2013	Mix Cl	767	0	10	107680	3.44	134900
10/16/2013	Mix Cl	733	43	11	116100	5.98	139400
10/17/2013	Mix Cl	697	0	10	106355	2.75	131400
10/18/2013	Mix Cl	732	0	10	108595	2.65	120100
10/19/2013	Mix Cl	689	0	9	101815	2.59	117600
10/22/2013	Mix Cl	1033	79	11	105230	5.33	119100
10/23/2013	Mix Cl	660	0	11	109745	3.87	122000
10/24/2013	Mix Cl	635	0	10	107780	2.71	125600
10/25/2013	Mix Cl	311	0	8	82650	2.63	102200
10/26/2013	Mix Cl	301	0	9	85945	2.64	104700
10/1/2013	Mix Na	16277	2464	12	44755	7.36	53900
10/2/2013	Mix Na	16899	2525	11	50385	7.36	58400
10/3/2013	Mix Na	20847	2330	10	55380	7.34	63400
10/4/2013	Mix Na	22911	2684	11	62375	7.35	69000
10/6/2013	Mix Na	22395	2928	9	61120	7.58	68500
10/7/2013	Mix Na	26346	2257	11	63705	7.34	70700
10/8/2013	Mix Na	22699	3294	15	60225	7.72	67900
10/9/2013	Mix Na	23384	2806	11	61035	7.78	68400

Appendix C3 - Brighton Water Quality Data

Date	Location	Ca mg/L	K mg/L	Mg mg/L	Na mg/L	Cl mg/L	F mg/L	NO ₃ mg/L
10/10/2013	Mix Na	299	45.9	53.7	17825	9138	77.9	2167
10/11/2013	Mix Na	299	42.8	54.0	19028	9369	82.1	2258
10/12/2013	Mix Na	239	41.1	49.5	19231	9351	82.7	2262
10/13/2013	Mix Na	258	38.7	57.5	17659	8345	76.6	2055
10/15/2013	Mix Na	437	54.4	80.1	19671	10326	80.5	2381
10/16/2013	Mix Na	362	67.1	71.9	22172	11382	73.4	2491
10/17/2013	Mix Na	413	60.7	65.3	21043	10603	63.8	2591
10/18/2013	Mix Na	467	57.1	85.6	20306	10170	64.8	2490
10/19/2013	Mix Na	482	57.5	87.8	20800	10387	64.8	2529
10/22/2013	Mix Na	492	39.9	0.0	19690	9644	91.0	2588
10/23/2013	Mix Na	427	44.3	62.1	19950	9712	94.2	2522
10/24/2013	Mix Na	508	52.4	66.8	22138	10606	95.3	2755
10/25/2013	Mix Na	1059	45.4	ND	17595	8592	28.3	2253
10/26/2013	Mix Na	1058	43.6	ND	17170	8313	27.2	2238
10/1/2013	NaCl	29	15.8	9.3	11848	17706	11.9	26
10/6/2013	NaCl	60	17.5	15.2	12084	18307	13.4	49
10/13/2013	NaCl	45	15.3	15.8	12542	19035	54.5	51
10/16/2013	NaCl	0	18.8	15.8	11678	17647	44.9	47
10/25/2013	NaCl	ND	26.1	ND	12501	19088	2.0	55
10/26/2013	NaCl	914	26.6	ND	11646	17761	2.4	54
10/1/2013	NF Conc	307	9.6	78.8	772	52	2.7	5
10/6/2013	NF Conc	314	12.0	83.7	920	51	5.8	5
10/13/2013	NF Conc	306	12.8	95.4	992	39	5.1	6
10/16/2013	NF Conc	296	14.7	88.5	975	61	4.6	6
10/25/2013	NF Conc	367	18.5	71.9	1112	31	0.4	5
10/26/2013	NF Conc	391	19.0	84.9	1136	38	0.4	6
10/1/2013	NF Perm	6	2.5	1.3	150	159	1.7	34
10/2/2013	NF Perm	4	2.3	1.0	140	153	1.5	33
10/3/2013	NF Perm	4	2.2	0.7	136	excl	0.4	4
10/4/2013	NF Perm	5	2.2	1.0	145	161	1.7	34
10/6/2013	NF Perm	5	2.5	0.9	155	169	1.8	34
10/7/2013	NF Perm	3	2.4	0.6	146	166	1.6	34
10/8/2013	NF Perm	4	2.4	0.8	151	165	1.7	35
10/9/2013	NF Perm	3	2.2	0.5	145	161	1.5	32
10/10/2013	NF Perm	4	2.6	0.8	163	185	1.8	37
10/11/2013	NF Perm	4	2.9	1.0	163	180	1.9	37
10/12/2013	NF Perm	4	2.9	0.9	163	179	1.8	37
10/13/2013	NF Perm	2	3.0	0.5	176	161	1.5	33
41562	NF Perm	7	3.3	1.7	174	187	2.1	37

Appendix C3 - Brighton Water Quality Data

Date	Location	SO ₄ mg/L	HCO ₃ mg/L	SiO ₂ mg/L	TDS mg/L	pH	Conductivity µS/cm
10/10/2013	Mix Na	20275	2928	9	55765	7.88	63500
10/11/2013	Mix Na	21189	3294	9	59165	7.93	65700
10/12/2013	Mix Na	21255	3611	8	59260	8.18	65000
10/13/2013	Mix Na	20722	2916	9	53945	8.32	63200
10/15/2013	Mix Na	22547	2855	9	60620	7.52	69900
10/16/2013	Mix Na	25830	3209	13	68665	7.72	76100
10/17/2013	Mix Na	24330	2879	10	64875	7.9	72500
10/18/2013	Mix Na	23757	2684	10	62800	7.88	64100
10/19/2013	Mix Na	24811	2635	13	63660	7.78	64900
10/22/2013	Mix Na	22241	2586	15	61195	7.6	62800
10/23/2013	Mix Na	21438	3416	15	61790	7.9	63800
10/24/2013	Mix Na	24263	3477	12	68835	8.04	68700
10/25/2013	Mix Na	19574	2806	13	54955	8	57400
10/26/2013	Mix Na	19450	2684	12	52640	8.01	55800
10/1/2013	NaCl	518	122	6	30015	5.86	52700
10/6/2013	NaCl	445	159	7	31885	5.64	54700
10/13/2013	NaCl	531	0	5	33225	3.03	58800
10/16/2013	NaCl	478	146	6	30850	6.08	54200
10/25/2013	NaCl	103	0	3	32990	3.7	52500
10/26/2013	NaCl	97	0	4	31465	3.55	50100
10/1/2013	NF Conc	2384	92	124	4235	4.85	4890
10/6/2013	NF Conc	2805	67	120	4980	4.79	5760
10/13/2013	NF Conc	3035	0	120	5590	3.5	6630
10/16/2013	NF Conc	2892	55	128	5075	4.83	6210
10/25/2013	NF Conc	3258	6	144	5965	4.4	6310
10/26/2013	NF Conc	3382	6	128	6125	4.2	6520
10/1/2013	NF Perm	26	104	104	540	5.34	838
10/2/2013	NF Perm	24	73	100	535	5.43	804
10/3/2013	NF Perm	excl	61	96	510	5.28	777
10/4/2013	NF Perm	24	98	100	530	5.37	828
10/6/2013	NF Perm	29	116	100	540	5.46	882
10/7/2013	NF Perm	39	73	96	525	5.07	852
10/8/2013	NF Perm	31	104	96	550	5.56	875
10/9/2013	NF Perm	50	61	96	510	4.95	850
10/10/2013	NF Perm	36	116	96	585	5.35	954
10/11/2013	NF Perm	36	55	104	575	5.32	982
10/12/2013	NF Perm	36	ND	108	605	5.4	974
10/13/2013	NF Perm	excl	0	96	645	4.38	1048
41562	NF Perm	22	140	108	615	5.78	1022

Appendix C3 - Brighton Water Quality Data

Date	Location	Ca mg/L	K mg/L	Mg mg/L	Na mg/L	Cl mg/L	F mg/L	NO ₃ mg/L
	41563 NF Perm	4	3.2	1.1	170	192	1.8	35
	41564 NF Perm	5	3.0	1.0	168	190	1.7	39
	41565 NF Perm	4	2.8	0.9	165	192	1.7	38
	41566 NF Perm	4	3.1	0.9	176	203	1.7	41
	41569 NF Perm	9	2.5	1.4	170	181	1.5	41
	41570 NF Perm	7	2.6	1.2	181	193	1.5	42
	41571 NF Perm	6	2.6	1.0	180	200	1.5	42
	41572 NF Perm	5	2.8	0.7	185	5	0.1	3
	41573 NF Perm	20	3.1	ND	188	216	1.6	49
10/1/2013	NF/RO Conc	463	13.7	102.1	674	401	7.8	127
10/2/2013	NF/RO Conc	432	12.4	96.2	647	374	7.5	122
10/3/2013	NF/RO Conc	406	12.1	89.1	632	348	7.6	113
10/4/2013	NF/RO Conc	447	12.4	99.1	669	376	7.4	122
10/6/2013	NF/RO Conc	458	13.1	103.4	721	370	7.7	114
10/13/2013	NF/RO Conc	427	13.9	111.7	751	346	7.2	113
10/16/2013	NF/RO Conc	431	16.0	107.2	757	390	6.8	108
10/25/2013	NF/RO Conc	480	18.7	89.6	821	380	2.1	125
10/26/2013	NF/RO Conc	501	19.3	99.3	844	387	2.2	127
10/6/2013	Raw	108	3.6	22.3	123	137	3.8	48
10/13/2013	Raw	114	3.1	25.5	121	134	1.2	45
10/17/2013	Raw	106	3.6	23.9	118	134	1.7	40
10/26/2013	Raw	106	4.0	17.4	113	129	0.7	45
10/1/2013	RO Conc	593	15.0	119.8	589	666	9.6	221
10/2/2013	RO Conc	554	14.4	114.9	559	635	9.6	213
10/3/2013	RO Conc	533	14.0	106.7	536	601	9.2	202
10/4/2013	RO Conc	516	14.7	116.1	570	648	9.3	217
10/6/2013	RO Conc	579	14.7	120.1	567	650	9.2	208
10/7/2013	RO Conc	529	14.1	108.4	539	614	9.6	207
10/8/2013	RO Conc	562	14.3	115.7	555	639	9.6	214
10/9/2013	RO Conc	528	13.3	110.5	513	596	9.0	199
10/10/2013	RO Conc	527	14.2	114.5	537	626	8.9	197
10/11/2013	RO Conc	559	15.3	127.1	565	657	9.3	205
10/12/2013	RO Conc	582	16.2	131.4	589	683	9.2	213
10/13/2013	RO Conc	513	14.3	116.7	524	601	8.8	201
10/15/2013	RO Conc	539	17.3	127.1	570	674	9.5	202
10/16/2013	RO Conc	523	16.4	116.5	556	650	7.8	189
10/17/2013	RO Conc	507	14.6	112.6	522	601	7.6	199
10/18/2013	RO Conc	502	14.1	116.1	508	595	7.4	194
10/19/2013	RO Conc	523	13.2	121.2	511	591	4.9	191

Appendix C3 - Brighton Water Quality Data

Date	Location	SO ₄ mg/L	HCO ₃ mg/L	SiO ₂ mg/L	TDS mg/L	pH	Conductivity µS/cm
41563	NF Perm	28	85	104	515	5.52	995
41564	NF Perm	32	122	92	520	5.37	972
41565	NF Perm	35	61	92	580	5.5	885
41566	NF Perm	40	61	88	625	5.23	936
41569	NF Perm	21	122	100	660	5.82	988
41570	NF Perm	27	98	100	640	5.64	948
41571	NF Perm	34	122	100	630	5.53	958
41572	NF Perm	169	110	104	0	7.34	951
41573	NF Perm	42	61	112	600	5.1	1002
10/1/2013	NF/RO Conc	2031	171	112	4380	5.62	5380
10/2/2013	NF/RO Conc	1935	195	104	4310	5.69	5280
10/3/2013	NF/RO Conc	1883	110	100	4135	5.6	5120
10/4/2013	NF/RO Conc	2020	134	104	4445	5.62	5400
10/6/2013	NF/RO Conc	2177	201	108	4655	5.73	5760
10/13/2013	NF/RO Conc	2305	146	108	4860	5.45	6010
10/16/2013	NF/RO Conc	2216	183	116	4725	5.84	6020
10/25/2013	NF/RO Conc	2393	195	116	5325	5.85	5840
10/26/2013	NF/RO Conc	2515	171	120	5340	5.67	5980
10/6/2013	Raw	186	317	20	855	7.28	1434
10/13/2013	Raw	181	299	21	865	7.33	1478
10/17/2013	Raw	175	281	21	745	7.34	1405
10/26/2013	Raw	166	281	21	805	7.54	1321
10/1/2013	RO Conc	925	1098	100	4140	7.48	5810
10/2/2013	RO Conc	872	1147	100	4075	7.57	5720
10/3/2013	RO Conc	830	1049	96	3890	7.53	5480
10/4/2013	RO Conc	891	1086	100	4130	7.63	5800
10/6/2013	RO Conc	905	1183	96	4100	7.55	5890
10/7/2013	RO Conc	840	1110	100	3870	7.6	5590
10/8/2013	RO Conc	882	1171	100	4055	7.68	5780
10/9/2013	RO Conc	833	1098	92	3750	7.63	5450
10/10/2013	RO Conc	860	1147	96	3955	7.66	5690
10/11/2013	RO Conc	912	1220	104	4240	7.65	6060
10/12/2013	RO Conc	951	ND	104	4370	7.64	6350
10/13/2013	RO Conc	839	1110	96	3870	7.59	5720
10/15/2013	RO Conc	879	1196	96	4050	7.63	5940
10/16/2013	RO Conc	862	1098	104	3915	7.63	5820
10/17/2013	RO Conc	823	976	92	3740	7.61	5520
10/18/2013	RO Conc	820	1086	92	3790	7.64	4960
10/19/2013	RO Conc	820	1098	88	3875	7.63	4960

Appendix C3 - Brighton Water Quality Data

Date	Location	Ca mg/L	K mg/L	Mg mg/L	Na mg/L	Cl mg/L	F mg/L	NO₃ mg/L
10/22/2013	RO Conc	732	12.1	115.8	582	641	6.1	228
10/23/2013	RO Conc	685	12.0	110.7	549	654	6.3	225
10/24/2013	RO Conc	748	13.1	123.5	595	664	6.5	228
10/25/2013	RO Conc	433	19.8	101.6	586	677	2.9	230
10/26/2013	RO Conc	390	19.9	106.6	586	682	2.6	231
10/1/2013	RO Perm	0.7	0.3	0.2	9.0	5.7	0.7	5.8
10/6/2013	RO Perm	0.7	0.3	0.2	10.3	6.4	0.6	8.5
10/13/2013	RO Perm	1.0	0.4	0.2	12.4	8.2	0.7	8.3
10/17/2013	RO Perm	0.8	0.3	0.2	7.8	5.7	0.5	4.6
10/26/2013	RO Perm	39.5	1.8	8.8	113.2	3.6	0.0	0.6

Appendix C3 - Brighton Water Quality Data

Date	Location	SO₄ mg/L	HCO₃ mg/L	SiO₂ mg/L	TDS mg/L	pH	Conductivity µS/cm
10/22/2013	RO Conc	857	1196	104	4345	7.63	5630
10/23/2013	RO Conc	876	1226	104	4370	7.64	5630
10/24/2013	RO Conc	891	1208	104	4360	7.64	5650
10/25/2013	RO Conc	886	1190	108	4330	7.6	5610
10/26/2013	RO Conc	892	1196	104	4200	7.54	5590
10/1/2013	RO Perm	3.4	79.3	1.0	355.0	5.8	57.7
10/6/2013	RO Perm	2.8	91.5	1.0	40.0	5.9	65.8
10/13/2013	RO Perm	4.2	103.7	2.0	75.0	6.2	84.4
10/17/2013	RO Perm	2.8	85.4	2.0	495.0	6.0	53.6
10/26/2013	RO Perm	331.5	54.9	1.0	360.0	5.9	53.1

APPENDIX D

MEGA EVALUATIONS

Appendix D1 – Mega Evaluations of Membranes from La Junta Trials

**Appendix D2 – Mega Evaluations of Stack 3 Membranes
(swollen membranes, resistance testing)**

**APPENDIX D1
MEGA EVALUATIONS
OF MEMBRANES FROM
LA JUNTA TRIALS**

Evaluation of membranes from La Junta trials

Goal

Make analyses of membranes from La Junta trials. A lot of precipitants were observed on membrane surface. There was a broken membrane too.

Precipitants

There were observed a lot of precipitant on membrane surface. A diluted hydrochloric acid (HCl) was used for cleaning. Bubbles of gaseous carbon dioxide (CO₂) were seen (Fig. 1). A piece of membrane (like in Fig. 2) was cut and put in diluted HCl. Bubbles of gas were seen too. That means precipitants were inside the membranes and they were carbonates.

We met the similar situation after piloting where the saturation of calcium fluoride (CaF₂) was overstepped (Fig. 4).

Broken membrane

A broken membrane was found (Fig. 3). Fitting fabric (textile) did not damage, it means that the destruction was not caused by pH. Ion exchange mixture was very fragile, it was seen under microscope. It could be caused either oxidant reagents or precipitants inside membrane.

Conclusion

We recommend you to dose hydrochloric acid (HCl) into both concentrate streams during operation (working) and keep pH 5 to 6. It could protect concentrates against precipitation of carbonates.

Elaborated by: Lubomir Machuca; lubomir.machuca@membrain.cz; August 15th, 2012



Fig. 1: Precipitants on membrane surface, bubbles in red circuit mean gaseous CO₂



Fig. 2: Precipitants inside membrane

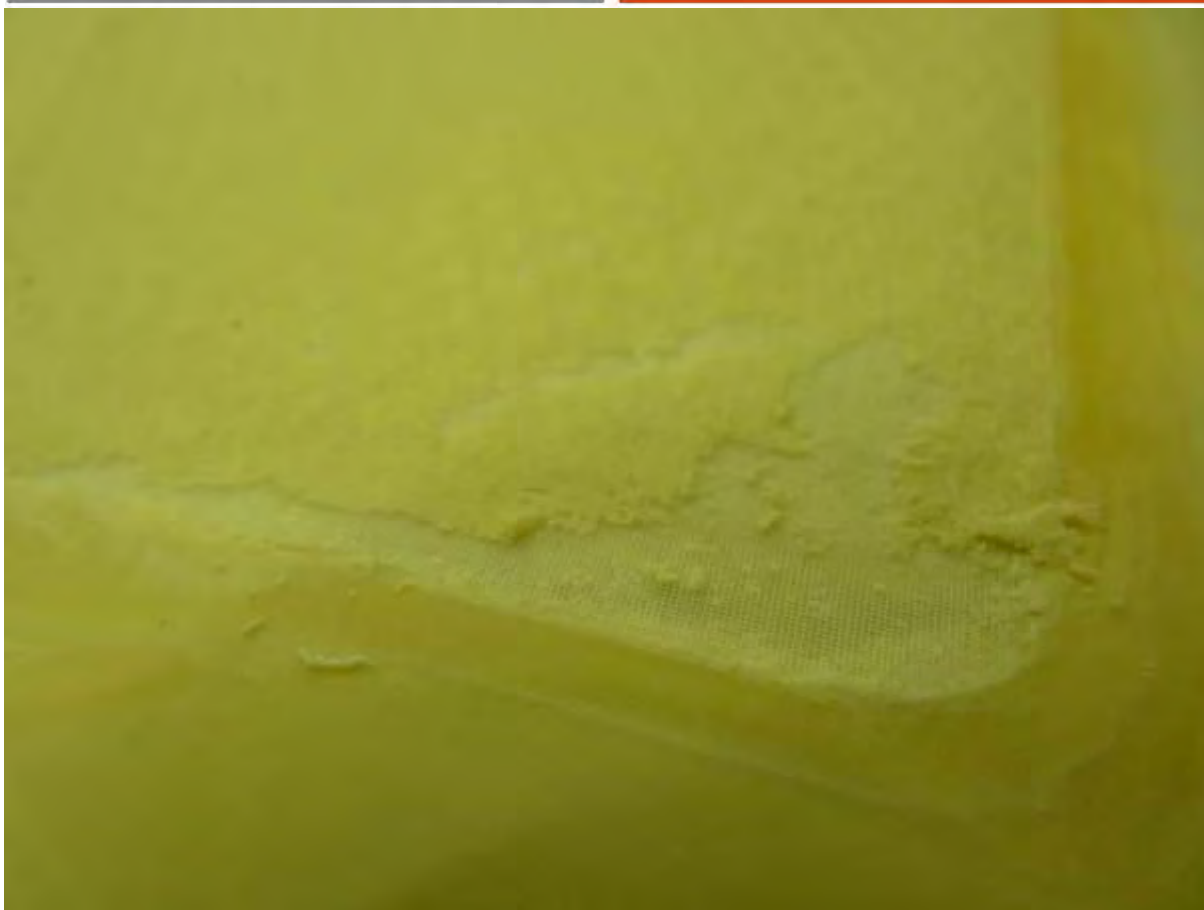


Fig. 3: Damaged membrane




Fig. 4: Precipitants of CaF₂ inside membrane

**APPENDIX D2 – MEGA
EVALUATIONS OF
STACK 3 MEMBRANES
(SWOLLEN MEMBRANES, RESISTANCE TESTING)**

EVALUATION OF STACK 3 MEMBRANES

Authors: **Lubomir Machuca**

Straz pod Ralskem, January 7th, 2014

MemBrain s.r.o. is part of a group 

Goal

Two quads from Stack 3 were sent to analysis because there were “fluffy” anion exchange membranes between NaCl and Mixed Cl. Analysis consists of:

- Measurement of electrochemical properties
- Determination of ionic content in fluffy AM
- Determination of precipitants on both sides of AM between EDM feed and Mixed Na

The Stack 3 operated at several different sites: Alamogordo New Mexico, La Junta Colorado and Brighton Colorado.

Membranes description

Two quads were checked and described. Description is shown in Table 1 and a scheme is in Figure 1.

Table 1: Membranes description

Membrane	Production batch	Production number	Position	Note
C1	13-32	62918	Mixed Na – NaCl	
A1	11-82	98755	NaCl – Mixed Cl	fluffy, weight of 695 g
C2	11-18	95213	Mixed Cl – EDM feed	
A2	11-82	98805	EDM feed – Mixed Na	red colour to EDM feed, precipitants on both side
C1'	11-18	95279	Mixed Na – NaCl	
A1'	11-82	98757	NaCl – Mixed Cl	fluffy, weight of 695 g
C2'	11-18	95216	Mixed Cl – EDM feed	
A2'	11-82	98813	EDM feed – Mixed Na	red colour to EDM feed, precipitants on both side

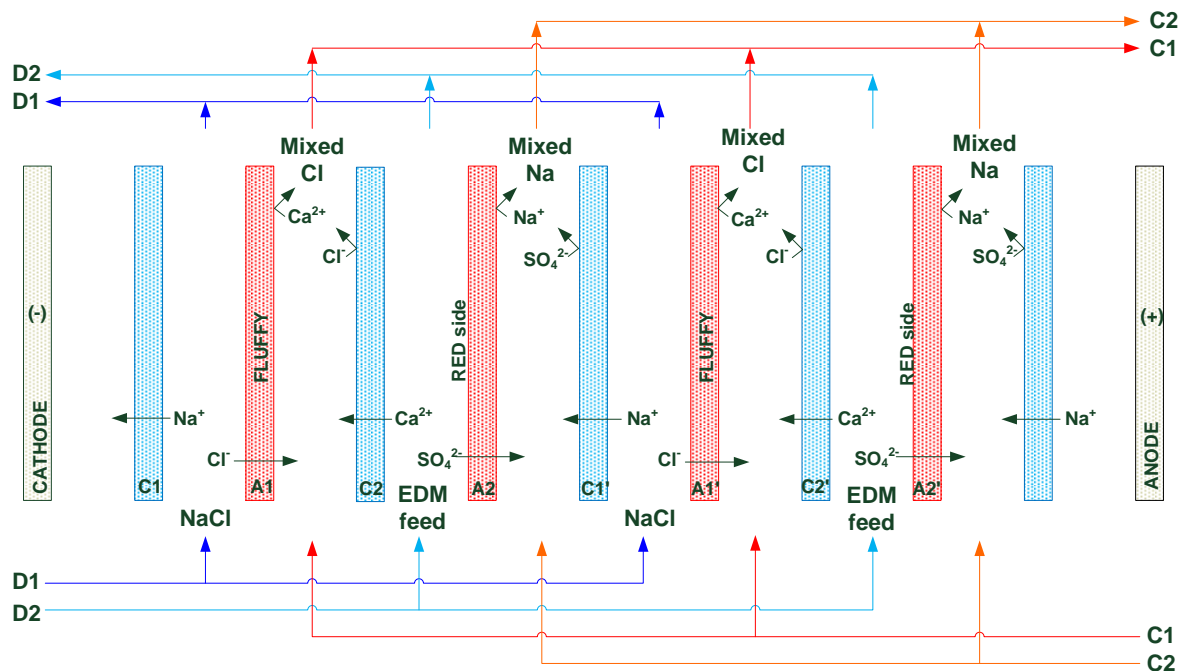


Figure 1: EDM scheme

Electrochemical properties

Surface resistance and specific resistance¹ both in 0.5M NaCl, and transport number and permselectivity both in 0.1/0.5M KCl were measured as electrochemical properties for each membrane. Ion exchange capacity (IEC) was determined too.

Electrochemical properties were measured by two ways, without cleaning and with cleaning. The cleaning means the washing in 1M HCl and 1M NaOH.

C1, C1' results

The cleaning was helpful and the resistances got better (go down). Lower resistance of C1 than C1' was due to C1 was produced in 2013 and it worked shorter time than C1'. Ion exchange capacity had common value, there was not any blockage or damage of ion exchange centres in the membrane.

A1, A1' results

Electrochemical properties and thickness got down after cleaning. Low resistance and low permselectivity after cleaning indicated huge channels (pores) inside the membrane. The cleaning flushed out some ions from membrane and it caused loosening the pores. In this case cations could pass through anion exchange membranes during the permselectivity measurement. We recommend you to replace all A1 membranes. Ion exchange capacity had common value, there was not any blockage or damage of ion exchange centres in the membrane.

The used membranes were thicker and heavier than new ones. New ones have thickness ~ 0.6 mm and weight ~ 500 g. Reasons of thickness and weight are discussed further.

Table 2: Electrochemical properties of membranes

membrane		C1	A1	C2	A2	C1'	A1'	C2'	A2'
batch		13-32	11-82	11-18	11-82	11-18	11-82	11-18	11-82
membrane number		62918	98755	95213	98805	95279	98757	95216	98813
solution		Mixed Na - NaCl	NaCl - Mixed Cl	Mixed Cl - EDM Feed	EDM Feed - Mixed Na	Mixed Na - NaCl	NaCl - Mixed Cl	Mixed Cl - EDM Feed	EDM Feed - Mixed Na
without cleaning									
thickness	[mm]	0.64	0.93	0.62	0.59	0.62	0.95	0.62	0.59
surface resistance	[$\Omega \cdot \text{cm}^2$]	7.69	9.25	10.24	8.73	9.48	9.27	9.58	7.66
specific resistance	[$\Omega \cdot \text{cm}$]	119.5	99.6	165.5	147.1	152.8	97.6	153.8	130.5
transport number		0.9575	0.9255	0.9616	0.9528	0.9611	0.9251	0.9659	0.9506
permselectivity	[%]	91.50	85.10	92.31	90.57	92.22	85.01	93.18	90.12
with cleaning									
thickness	[mm]	0.66	0.73	0.63	0.60	0.64	0.75	0.64	0.59
surface resistance	[$\Omega \cdot \text{cm}^2$]	5.82	4.43	7.01	7.10	7.21	4.71	7.01	7.17
specific resistance	[$\Omega \cdot \text{cm}$]	88.6	60.6	110.9	119.0	112.9	62.5	109.0	121.4
transport number		0.9419	0.9200	0.9592	0.9405	0.9607	0.9339	0.9618	0.9446
permselectivity	[%]	88.38	84.00	91.85	88.10	92.13	86.79	92.36	88.92
ion exchange capacity	[meq/g _{dry}]	2.44	1.91	2.38	1.80	2.42	1.94	2.40	1.80

¹ Resistance of membrane is measured in a special cell which is divided by membrane in two chambers. Solution of 0.5M NaCl is in each chamber. Two electrodes measure voltage under 10 mA of DC and resistance is calculated from voltage and current by Ohm's law.

C2, C2' results

The cleaning was helpful and resistance got better. Ion exchange capacity had common value. That seems only dirt was on the membrane surface. The membranes were alright.

A2, A2' results

The cleaning was helpful and resistance got better. Permselectivity after cleaning was rather low, it could cause that cations went through anion exchange membrane. Ion exchange capacity was 10 % lower in comparison with new membrane.

Analysis of A1, A1' membrane

The A1 membranes were tested by two methods. The first one represented burning of membrane and analysis of ash. Inorganic ions were determined by this method. The second one was infrared (IR) spectroscopy of surface and incision of membrane. Organic pollutants were determined by this method.

Very high content of calcium (Table 3) was found in A1 in spite of this membrane was anion exchange. We suppose that $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$ (and $\text{Ca}_3(\text{PO}_4)_2$) precipitated inside the membrane and it probably caused “fluffy” membrane. We recommend you to replace all A1 membranes. IR spectra of A1 and new anion exchange membrane were the same thus there were not any organic pollutants on/in A1 membrane.

Table 3: Content of inorganic ions in A1 membrane

Ion	Unit	Value
Al	mg/kg of dry membrane	29.8
Ba	mg/kg of dry membrane	1
Ca	mg/kg of dry membrane	113 000
Cu	mg/kg of dry membrane	5.72
Fe	mg/kg of dry membrane	26.3
Mg	mg/kg of dry membrane	22.6
Mn	mg/kg of dry membrane	2.92
Ni	mg/kg of dry membrane	1.68
Sr	mg/kg of dry membrane	563
CO_3^{2-}	mg/kg of dry membrane	< 500
SO_4^{2-}	mg/kg of dry membrane	283 000
PO_4^{3-}	mg/kg of dry membrane	1 972
F^-	mg/kg of dry membrane	< 150
SiO_2	mg/kg of dry membrane	2 900

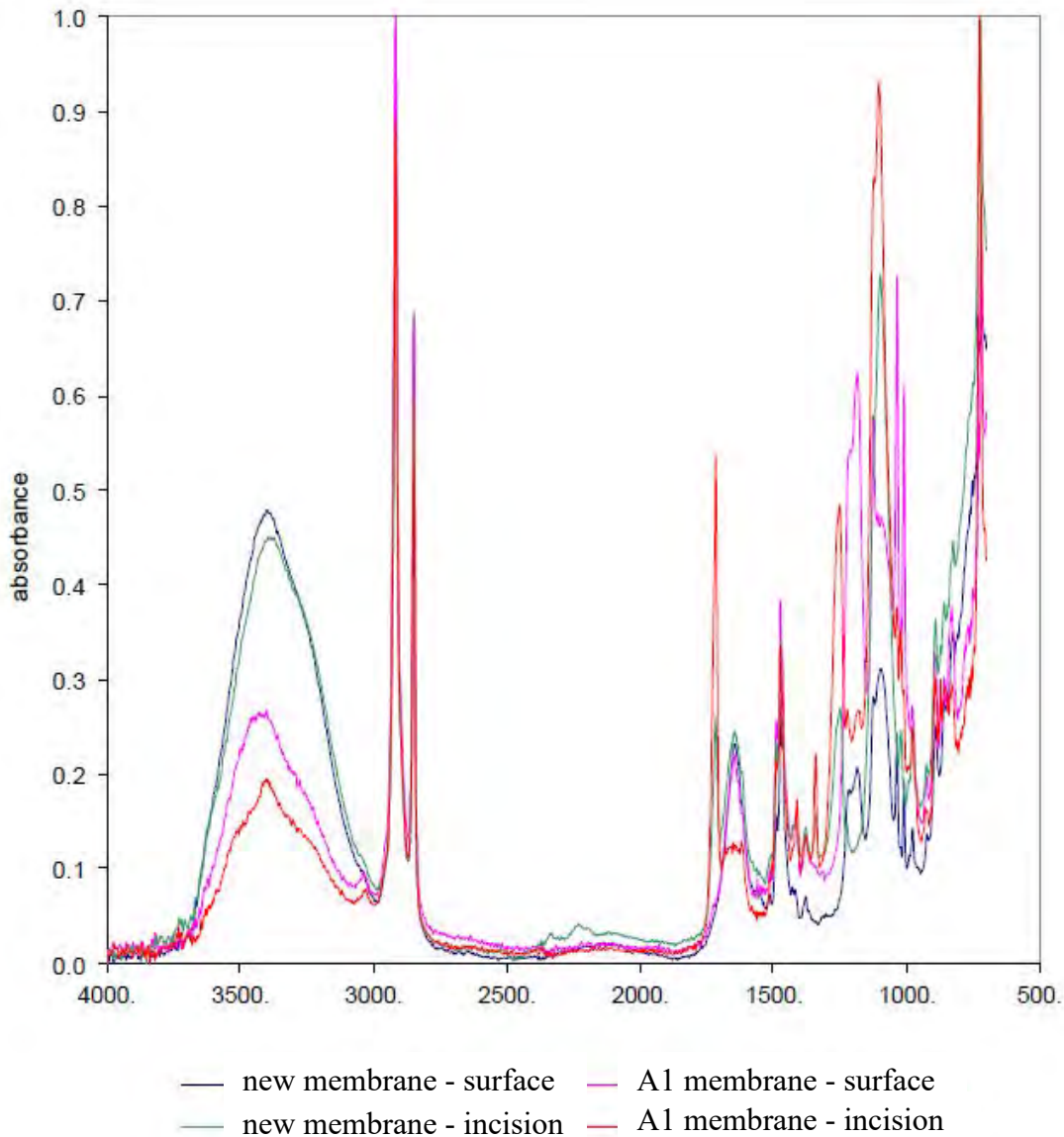


Figure 2: IR spectra of A1 and new anion exchange membrane

Precipitants on A2, A2' surface

Precipitants on A2 membrane surface which was directed at C2 membrane are shown in Figure 3. EDM feed flowed along this side. The membrane side and precipitants were brown coloured. Precipitants contained $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$ based on microscopic and X-ray analysis. Precipitants on A2 membrane surface which was directed at C1' membrane are shown in Figure 4. Mixed Na solution flowed along this side. Precipitants contained CaCO_3 and $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$ based on microscopic and X-ray analysis.



Figure 3: Precipitants on A2 membrane surface directed at C2

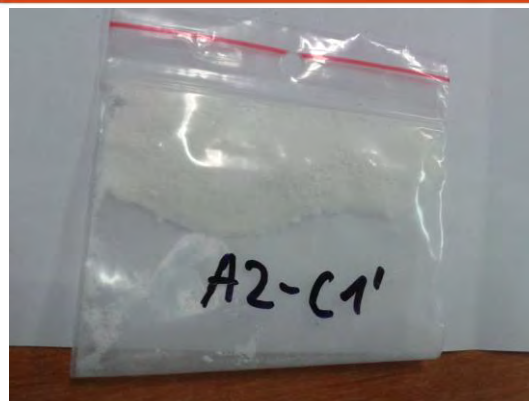


Figure 4: Precipitants on A2 membrane surface directed at C1'

Conclusion

Two quads from Stack 3 were analysed.

- The Stack 3 was produced in 2011 and FAT (Factory Acceptance Test) was carried out. The Stack 3 passed this test successfully without any mistake by assembling.
- The membranes were probably replaced several times on-site because membranes A1 – C2' were produced in 2011 and C1 membrane in 2013.
- “Fluffy” anion exchange membranes were caused probably by $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$ (and $\text{Ca}_3(\text{PO}_4)_2$) precipitation. Any organic matter was not found inside “fluffy” membranes. We recommend you to replace all A1 membranes.
- The second kind of anion exchange membranes had precipitants on surface. Precipitants of $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$ were on EDM feed side, mixture of CaCO_3 and $\text{CaSO}_4 \cdot 2\text{H}_2\text{O}$ were on Mixed Na side. We are not able to determine the reason of precipitation because we do not know all operation conditions.
- We recommend you to make cleaning by acid and alkaline periodically because it is helpful for getting better electrochemical properties. It necessary to find the optimum cleaning frequency and cleaning solution for the stacks used in ZDD process, based on pilot experience.
- Permselectivity of each ion exchange membrane depends on concentration ratio of solution on both sides. The higher concentration ratio, the lower permselectivity. In RALEX membrane case we usually measure permselectivity in 0.1/0.5 M KCl and value for new anion exchange membrane is min. 90 %. If we measured anion exchange membrane in 0.05/1.5 M KCl, permselectivity was only 86 %. That is necessary to consider when using RALEX membranes for ZDD application.

APPENDIX E

COST CALCULATIONS

Additional Cost Calculations

Equations and sample calculations used for capital and operational cost estimations are described here and supplemented with a 4-MGD design for Alamogordo. Summary information for Alamogordo and La Junta are provided as well.

Adjustment to common year for cost estimates:

All cost estimates are made with 2013 as the cost year. The Producer Price Index for commodities was used to adjust prices. If sources provided the year for their cost basis, it was used. Otherwise, the year of publication was utilized.

Calculating installed cost of equipment

The equipment capital cost was estimated as described in the report, and subsequently multiplied by 1.4 to estimate total installed cost.

The capital cost for concentrate management, both evaporation pond and deep well injection, was each estimated using four different cost models / methods. System flows, including those to the concentrate disposal systems, were calculated using Veolia's design software (called Pearl), and edited in Excel to match actual pilot operation in the field.

Evaporation Pond Estimations

Evaporation pond size was calculated using the evaporation rate for the region (14.8 cm/month for Alamogordo and 6.9 cm/month for La Junta, CO).

$$\text{Evaporative Area} = (\text{flow to Evap Pond}) / (\text{Evap Rate})$$

The 4-MGD Alamogordo ZDD system:

$$\begin{aligned} \text{Evap Area} &= (57\text{gpm}) * (1440\text{min/day}) * (365\text{ days/yr}) * (1\text{m}^3/264.17\text{gal}) \\ &* (1/14.8\text{ cm/mos}) * (1\text{yr}/12\text{mos}) * (100\text{cm/m}) * (1\text{acre}/4046.856\text{m}^2) = \\ &15.7\text{ acres} \end{aligned}$$

DWPR Report #123 includes an estimation method for estimating the total size of the evaporation pond (including dikes and 20% contingency). This method was used to estimate the pond size and used for all cost estimations.

$$\text{Total area (plus 20\% contingency)} = 1.2 \times \text{evaporative area} \times (1 + 0.155 * \text{dike height} / \text{sqrt evap area})$$

The 4-MGD Alamogordo ZDD with a 4-ft dike height:

$$\text{Total area} = 1.2 \times 15.7\text{ acres} \times (1 + 0.155 \times 4\text{ft} / \text{sqrt}(15.7\text{acres})) = \underline{21.8\text{ acres}}$$

Evaporation pond capital costs were estimated using four methods, adjusted to 2013 values (PPI=203.0), then averaged:

- (1) Desalination Engineering, Planning & Design textbook (Voutchkov 2013, page 571).
Capex = evap pond install (Figure 16.47 in Voutchkov 2013) + \$0 land purchase + \$8500/acre for Leak Detection.
- (2) BGNDRF, Alamogordo, NM Evaporation pond installation cost of \$130,000/acre, adjusted from 2007 (PPI = 175.1)
- (3) El Paso KBH Plant cost estimate of \$52.3 M for 3-MGD of passive evaporation, and scaled to the Alamogordo or La Junta flowrate to Evaporation, adjusted from 2012 (PPI = 202.1)
- (4) Bureau of Reclamation Report by Mickley (2006), described below, and adjusted from 2001 (PPI = 135.7), plus \$8500/acre for Leak Detection.

Evaporation pond capital cost method (4) is from DWPR Report #123 (Mickley, 2006), and suggests an Evaporation Pond Regression Model to estimate the total unit area capital cost (\$/acres) as follows:

$$\text{Total unit area capital cost (\$/acre)} = 5406 + 465 \times \text{liner thickness} + 1.07 \times \text{land cost} + 0.931 \times \text{land clearing cost} + 217.5 \times \text{dike height}$$

The 4-MGD Alamogordo ZDD system (assuming no land cost):

$$\text{Total unit area capital cost (\$/acre)} = 5406 + 465 * 50\text{mil} + 1.07 * (\$0) + 0.931 * \$1000/\text{acre} + 217.5 * 4\text{ft} = \underline{\$30,457/\text{acre}}$$

$$\text{Total Pond Cost} = \text{Total area} * \text{Total unit area capital cost} = \$30,457 / \text{acre} * 21.8 \text{ acres} = \underline{\$663,271}$$

The DWPR Report 123 (Mickley, 2006) suggests a more detailed analysis can be obtained by specifying a range of evaporation pond-specific variables (Table 10.1 “Worksheet for Evaporation Pond Disposal Capital Costs, Preliminary Level Costs Only”). Table D-1 includes this methodology for a full scale/full flow Alamogordo (4-MGD) and La Junta (6.6 MGD) plants and also includes the reported values for La Junta (1.6 MGD flow) from the WERF 5T10 report. It is important to note the following differences between this Report’s analysis and the WERF analysis:

- the WERF study sized the Evaporation Pond for average plant production of 1.6-MGD, while we sized the Evaporation Pond for design plant production of 6.6-MGD,
- the WERF report assumed six ponds of 3 acres each, while this analysis assumed one large pond,

Table E-1.—Example Evaporation Pond Estimation Calculations

Item	Variables	Range	Alamogordo (4MGD ZDD)	La Junta (6.6MGD ZDD)	WERF 510T (1.6MGD)
A	Evaporation Surface (Acres)	0 to 100	16	56	3
B	Dike height (feet)	4, 8, 12	4	4	4
C	Total liner thickness (mm)	20 to 120	50	60	60
D	Land Unit Cost (\$/acre)	0-10,000	\$0	\$10,000	\$10,000
E	Land Type	1, 2, 3, 4	1 (Brush)	1 (Brush)	1 (Brush)
Item	Calculation of Total Acreage	Remarks			
F	Ratio: Total Acreage to Evaporative Acreage	Reference	1.17	1.08	1.3
G	Total Acreage	A*F	18.72	60.2	6.5
Item	Unit Area Costs Using Total Acreage	Remarks			
H	Land, \$/acre	Same as D	0	\$ 10,000	\$10,000
I	Land clearing, \$/acre	Reference	1000	1000	\$1,500
J	Dike, \$/acre	Reference	3000	1600	\$7,400
K	Nominal liner, \$/acre	Reference	26600	28600	\$35,300
L	Liner, \$/acre	K*C/60	22167	28600	\$35,300
M	Fence, \$/acre	Reference	3750	2050	\$8,300
N	Road, \$/acre	Reference	700	360	\$1,400
	Total Unit Cost (per acre)		\$30,617	\$43,610	\$67,000
	Subtotal		\$573,144	\$2,623,789	\$416,000
	Engineering and Admin (20%)		\$114,629	\$524,758	\$83,200
	Contingency (15%)		\$85,972	\$393,568	\$62,400
	TOTAL (1 pond)		\$773,744	\$3,542,115	\$570,000 (1 pond)
	TOTAL (all ponds)		\$773,744	\$3,542,115	\$3,500,000 (6 ponds)

- the WERF report added 20% for Engineering and Administration costs, and 15% for Contingency; while the Regression Model seems to include these Fees already.

For La Junta 6.6 MGD ZDD, the Regression Model estimated \$3.3M, while the Detailed Analysis in the Table below estimated \$3.5M installed, so the two models are quite similar. The Regression Model is an equation and requires no job-specific interpolation on plots published in the USBR 2006 publication, and therefore the Regression Model is preferred by the authors of this paper.

Deep Well Injection Costs

For Alamogordo, Deep Well Injection was also considered as a disposal method. Five methods were calculated in this Report's cost model, and averaged. As with Evaporation Costs, these are adjusted to 2013 dollar values using PPI. The five models are described briefly below.

- (1) Cost from Alamogordo EIS for conventional BWRO, which assumes 1 well for BWRO disposal for a BWRO producing 1.8MGD, is \$2.6M, in 2010 dollars (PPI = 187)
- (2) Cost for each KBH (El Paso, TX) injection well is \$3.6M (2007, PPI = 175)
- (3) AWWA M46 equation

DWI Capital Cost = $(-0.0001 * (\text{flow to well, gpm})^2 + 1.0178 * (\text{flow to well, gpm}) + 1734.2) * 1000$, in 2004 dollars (PPI = 147.6)

- (4) Cost from Desalination Engineering, Planning & Design by (Voutchkov, 2013). Pretreatment cost \$20-\$50/m³/day of plant capacity. Monitoring well \$600-\$800/well depth, m.

DWI Capital Cost = $165 * (\text{Flow, m}^3/\text{day}) + 310 * (\text{Well depth, m}) + 100000 + (\$50 \text{ Pretreatment cost}) * (\text{DW plant capacity, m}^3/\text{day}) + (\text{Well depth, metres}) * 800$.

- (5) Equation from DWPR Report 123 (Mickley, 2006), based upon wells in California, Florida, and Texas, and includes the cost for monitoring well, which was not required for KBH in El Paso, Texas. Last section of chapter defines Deep Well Injection. Cost basis is 2001 (PPI = 135.7).

DWI Capital Cost = $-288,000 + 145,900 * \text{Tube diameter} + 754 * \text{depth}$.

Equations provided somewhat similar values, as shown in table D-2.

Table E-2.—Deep Well Injection Cost Estimates According to Five Different Cost Models, Adjusted to 2013 Dollars Using PPI

Model and Variables	Deep Well Injection CapEx estimates (adjusted to 2013 dollars)			
	Alamogordo 4-mgd BWRO	Alamogordo 4-mgd ZDD	Alamogordo 1-mgd BWRO	Alamogordo 1-mgd ZDD
<i>Flow to Well</i>	697	57	176	14
<i>Tubing Diameter (in)</i>	6	4	4	4
<i>Depth (ft)</i>	3,000	3,000	3,000	3,000
(1) Alamogordo EIS	\$2,849,599	\$2,849,599	\$2,849,599	\$2,849,599
(2) El Paso KBH Plant	\$4,173,615	\$4,173,615	\$4,173,615	\$4,173,615
(3) AWWA M46	\$3,294,271	\$2,464,018	\$2,627,893	\$2,404,922
(4) Voutchkov	\$2,499,241	\$1,165,967	\$1,462,878	\$1,127,730
(5) DWPR #123	\$4,222,160	\$3,785,643	\$3,785,643	\$3,785,643
AVERAGE	\$3,407,777	\$2,887,768	\$2,979,926	\$2,868,302

Operating Costs

Operating costs include power, chemicals, membrane replacement, expendables, and concentrate management. Annual operating costs calculated assuming an interest rate of 6% for 20 years.

Power Consumption was calculated based on 365day/yr, 24hr/day operation, using the following equation

$$Pump\ BHP = (GPM * PSI * Sp.Gr) / (1713 * Pump.eff).$$

The NF feed pump operates at 3,800 gpm and 90 psig (Since the booster pumps are outputting 30 psi pressure, the NF feed pump needs to provide only an additional 60 psi to produce permeate).

$$Pump\ BHP = (GPM * PSI * Sp.Gr) / (1713 * Pump.eff) = (3800gpm * 60\ psi * 1SG) / (1714 * 70\% \text{ assumed pump efficiency}) = 190$$

Unit Motor Horse Power (MHP) is a Lookup function in Excel, comparing BHP to MHP by calculating $BHP * (1 + BHP:MHP\ Ratio)$, where $BHP:MHP\ Ratio = 15\%$. The look-up table defines 400MHP for a 350 BHP. Subsequently, Unit motor BKW is calculated.

$$Unit\ motor\ BKW = BHP * 746 / (Motor\ efficiency * 1000) = 350BHP * 746 / (85\% * 1000) = 307$$

Motor efficiency is assumed to be 80% for motors <50HP, and 85% for motors >85HP. The Unit Motor BKW is then multiplied by a Power Factor of 0.90 to

calculate the Unit Operating BKW, and multiplied by the number of operating units to calculate the Total Operating BKW.

$$\text{Unit Operating BKW} = \text{Unit Motor BKW} * \text{Power Factor} = 102.5 * 0.90 = 114$$

$$\text{Total Operating BKW} = \text{Unit Operating BKW} * \text{Number of Units} = 114 * 1 = 114$$

The power consumed per day is calculated by multiplying the Total Operating BKW by the number of hours operating per day; and usage per year is similarly calculated.

$$\text{Electrical usage (kWhr/day)} = \text{BKW} * \text{Hrs/day} = 114 * 24\text{hrs/day} = 8,198 \text{ kWhr/day.}$$

$$\text{Electrical usage (kWhr/yr)} = 8198 * 365 \text{ days/yr} = 2,992,120 \text{ kWhr/yr}$$

$$\text{Electrical usage (\$/yr)} = \text{kWhr/yr} * \$0.08/\text{kWhr} = \$239,370/\text{yr}$$

The EDM stack power is calculated from the voltage and current needed (based on the piloting results and full scale design).

$$\text{EDM Stack Power (kW)} = (\text{Voltage} * \text{Current}) / (\text{Motor Efficiency} * \text{AC-to-DC-Efficiency} * 1000) = (110\text{V} * 37\text{Amps}) / (85\% * 97\% * 1000) = 4.9 \text{ kW}$$

For Alamogordo, each of the 93 EDM stacks has its own DC drive which operates at 140 Volts and 46 amps. We assume an AC to DC efficiency of 97% and motor efficiency of 85%, which equates to an EDM Stack Power of 7.8 kW.

For La Junta, each of the 90 EDM stack has its own DC drive which operates at 110 volts and 37 amps. We assume an AC to DC efficiency of 97% and motor efficiency of 85%, which equates to an EDM Stack Power of 4.9 kW.

Similar to the pump power equations described previously, the Unit Operating BKW and Total Operating BKW are calculated, and subsequently the electrical usage in kWhr/day can be calculated (for La Junta):

$$\text{Unit operating BKW} = \text{EDM Stack Power (kW)} / \text{Power Factor} = 4.9 / 1 = 4.9$$

$$\text{Total Operating BKW} = \text{Unit Operating BKW} * \text{Number of Units} = 4.9 * 90 \text{ units}$$

$$\text{Power consumption per day (kWhr/day)} = \text{Total Operating BKW} * \text{Operating hrs/day} = 444 \text{ BKW} * 24 \text{ hrs/day} = 10,662 \text{ kWhr/day}$$

$$\text{Power consumption per year (kWhr/yr)} = \text{Daily electrical usage} * \text{days per year} = 10,662 * 365 \text{ days} = 3,891,799 \text{ kWhr/yr}$$

$$\text{Yearly power cost} = 3,891,799 \text{ kWhr/yr} * \$0.08/\text{kWhr} = \$311,344/\text{yr from the 90 EDM stacks.}$$

The total annual power costs for Alamogordo and La Junta ZDD designs are summarized in Table D-3. This total power includes all pumps (feed, booster, chemical, transfer), clean in place systems, and other ancillary equipment.

Table E-3.—Annual Power Cost by System and Location

	NF	EDM	Other	Deep Well Injection
Alamogordo 1MGD	\$63,100	\$195,885	\$2,092	\$515
Alamogordo 4MGD	\$225,986	\$772,490	\$7079	\$2,059
La Junta 6.6MGD	\$397,682	\$644,874	\$10,894	--

Chemical consumption was estimated based both on dosage and flowrate (sulfuric acid and antiscalant) and by total pounds per day calculated in Veolia’s design software called Pearl (NaCl). An example equation for calculating sulfuric acid for pH adjust in the ZDD feed is as follows.

ZDD Chemical, NF Sulfuric Acid Daily Consumption

- Design flow = 2,263gpm
- Dosage = 4mg/L
- Chemical dosage frequency = 24hrs/day, 365 days/yr
- Chemical quantity (lbs/day) = (flowrate gpm) * (60min/hr) * (Dosage, mg/L) * (Hrs/day) * (8.34/1000000) = 109 lbs/day
- Chemical quantity (lbs/yr) = lbs/day * 365days/year = 39,676 lbs/yr
- Chemical cost per year = Quantity per year * Cost per lb (\$0.20) = \$7,935

NF CIP chemical quantities are calculated by first knowing the vessel volume. An example calculation is below.

Vessel Volume Calculation

- Bank quantity = 2 units
- Vessels per bank = 16
- Elements per vessel = 8
- Volume per vessel, gallons = $((8)/(2*12))^2*((40/12)*Elements/vessel)*7.480519$
- CIP volume for vessels, gallons = (Vessels/bank * Volume/vessel * Bank quantity) = (16*69.7*4)=4,458 gallons

Pipe Volume Calculation

- Vessels / stage, units = 16
- CIP pump capacity, gpm = # vessels/stage * 40gpm = 640-gpm
- Diameter of pipe, inches = 8”

- Length of pipe, ft = 150ft
- Volume of pipe, gal =(((Diameter, inches)/(2*12))^2)*LengthOfPipe, ft*7.48)*BankQuantity = 499gallons

*Total CIP volume required = (CIP volume required per flush, gallons) * (Number of cycles per CIP) * (1+SafetyFactor) = (4957gal per flush) * (3 cycles per CIP) * (1 + 0.20 safety factor) = 17,845gallons*

Caustic requirement

- Concentration = 0.1%
- Quantity of NaOH required per CIP =(17,845 gal per CIP * 0.1% *100*10000*8.34)/1000000 = 149 lbs at 100%
- Quantity of NaOH required per year = (NaOH per CIP) * (# CIP / year) = (149lbs/CIP) * (2CIP/yr) = 298 lbs at 100%

The CIP chemicals required are <\$0.01/1000gal.

The biggest contributor to ZDD chemical operating costs is the NaCl required for the EDM. The NaCl required is calculated by the Veolia design software, Pearl, based upon the fact that 1 meq of NaCl is required for every meq removed from the EDM feed water. NaCl consumption for the ZDD designs evaluated in this report are listed below. NaCl is assumed \$0.075 /lbs delivered.

Table E-4.—Annual NaCl Consumption and Cost for Each Site

	NaCl per day (lbs)	NaCl per year (lbs)	\$/yr
Alamogordo 1MGD	12,204	4.45M	\$334,104
Alamogordo 4MGD	48,350	17.6M	\$1.32M
La Junta 6.6MGD	42,327	15.5M	\$1.16M

The Expendables budget for ZDD and BWRO consists of Membranes, Cartridge filters, and EDM Anodes and Cathodes. Replacement costs are annualized by multiplying the average lifespan (5yrs for NF, 10yrs for EDM) by the ratio of Total units divided by the Design/Duty units. The estimated Annualized replacement costs for these items are listed below.

Table E-5.—Annual Expendables Budget for Each Site

	NF Membranes	EDM Membranes & Spacers	Cartridge filters	EDM Anodes & Cathodes
Alamogordo 1MGD	\$61,440	\$101,837	\$38,720	\$19,392
Alamogordo 4MGD	\$63,160	\$394,618	\$153,908	\$75,144
La Junta 6.6MGD	\$100,000	\$381,888	\$248,626	\$72,720

Salt Recovery Estimation

Conceptual costs for NaCl recovery techniques described in Chapter 11 were made using estimates for typical equipment used for lime softening, solids dewatering, and electrodialysis systems common in Japan for recovering NaCl from seawater.

The first step is to determine how much NaCl is required by the EDM. This is done by calculating the difference in anion and cations, measured in equivalents per day, between the NF reject and EDM diluate. The amount of Na required is equal to the change in cations and the amount of Cl required is equal to the change in anions. Each of these values (in eq/day) is converted to mg/day and summed. Finally, the amount of NaCl is converted to lb/day.

Table E-6.—Calculation of Daily NaCl needed for EDM (Alamogordo)

		NF Concentrate	EDM Feed	EDM Diluate	Recycle
Flow rate	gpm	1,521.0		1,513.4	
TDS	ppm	6207.2		3144.0	
Conductivity	□S/cm	4323.7		2598.9	
in meq/L					
Calcium		55.86		27.93	
Magnesium		20.05		10.02	
Sodium		15.68		7.84	
Potassium		0.26		0.13	
Bicarbonate		0.00		0.00	
Chloride		3.34		1.67	
Sulfate		86.25		43.13	
Σ cations		91.85		45.93	
Σ anions		89.60		44.80	
NaCl required by EDM					
		mg/day		meq/day	
Na		8584968863		373259516	
Cl		13392374033		382639258	
NaCl		48350		lb/day	

To maximize NaCl recovery and minimize waste, lime is added to the Mixed Cl stream to remove Mg(OH)₂. This allows the calcium and sulfate to be nearly balanced. There will be excess Mixed Na because there is still more sulfate than calcium. The streams are blended at a ratio of the treated Mixed Cl calcium/Mixed Na sulfate to ensure a stoichiometric mix of calcium and sulfate.

The remainder of the Mixed Na stream is sent to a separate evaporation pond. The amount of NaCl recoverable is found by summing the concentration of Na and Cl in eq/day in each of the streams then converting to lb/day.

$$\text{Available Na} \left(\frac{\text{eq}}{\text{day}} \right) = Q_{M.Cl} C_{M.Cl}(Na) + Q_{M.Na} C_{M.Na}(Na)$$

$$\text{Available Cl} \left(\frac{\text{eq}}{\text{day}} \right) = Q_{M.Cl} C_{M.Cl}(Cl) + Q_{M.Na} C_{M.Na}(Cl)$$

Table E-7.—Calculation of Daily NaCl recoverable (Alamogordo)

		NF Conc	Mixed Cl	for NaCl Production		to 2nd Evap Pond
				Treated Mixed Cl	Mixed Na	Mixed Na
Flow rate	gpm	1521.01	35.49	35.49	35.49	6.11
TDS	ppm	6207	107364	107364	140007	140007
Conductivity	□S/cm	4324	262428	262428	255122	255122
Calcium	meq/L	56	1203	1635	0	0
Magnesium	meq/L	20	432	0	0	0
Sodium	meq/L	16	338	338	1975	1975
Potassium	meq/L	0	5	5	0	0
Bicarbonate	meq/L	0	0	0	0	0
Carbonate	meq/L	0	0	0	2	2
Chloride	meq/L	3	2002	2002	72	72
Sulfate	meq/L	86	0	0	1858	1858
Nitrate	meq/L					
Silica	meq/L	92	1978	1978	1975	1975
CO2	meq/L	90	2002	2002	1931	1931
		NaCl Recoverable				
			eq/day			
		Available Na	381,576			
		Available Cl	398,665			
			39,480	lb/day		
			82%	of required NaCl		

The next steps are shown in tabular form.

3.26	mgd NF feed
1433.00	mg/L SO ₄ in feed
48	g/eq sulfate
20	g/eq calcium
18	g/eq water in dihydrate
86	g/eq of dihydrate
2.57	g of dihydrate in liter of feed (assuming all SO ₄ is converted to dihydrate)
12,331,530	L/day of feed
31.66	mT/day of dihydrate in feed
1000.00	kg/m ³ bulk density of dihydrite at bottom of pond
31.66	m ³ /day of dihydrate generated
4047.00	m ² /acre
0.01	fraction of feed evaporated to raise salt content of supernatant from 1.5M to 3M
151.68	m ³ /day evaporation rate
0.148	meters/month evaporation rate in Alamogordo
30.42	days/month
7.70	acres evaporation pond needed to raise salt content of supernatant from 1.5M to 3M
3.0	days/yr
0.25	fraction of solids in pond
11.4	gpm, excess Mixed Na flow to 2 nd evaporation pond (assume 2 x calculated 5.7 gpm)
3	Acres, evaporation pond
12308.019	O&M on evap pond (no disposal); assume 1% of capital cost
115000	\$/Ac evap pond cost
20	years of pond life
35.35	meq/L anions in BW feed
0.70	NF rejection
1230802	capital cost of ponds (10.7 acres total)
377949	eq/day NaCl required
309918	eq/day NaCl recoverable (90%)
14563609	lb/year NaCl recoverable
77228	Annual cost of evap pond (20 years, 6% interest)
Lime for Mg(OH)₂ Recovery	
88	mg/L Mg in sourcewater
1085	kg/day Mg in feed
2604	kg/day Mg(OH) ₂ in feed
800	\$/mT sales price of high purity Mg(OH) ₂
400	\$/mT sale price of Mg(OH) ₂
31242	\$/month potential market for Mg(OH) ₂

74	molecular weight of Ca(OH) ₂
24	atomic weight of Mg
3305	kg/day Ca(OH) ₂ consumption
106145	cost of Ca(OH) ₂ \$/year
(clarifier costs: http://www.ecodwellinternational.org/images/EC/ECvsLime%20Softening.pdf)	
Clarifier for Lime:	
14427	\$/gpm treated, includes 1.4 installation factor
15	kWh/1000 gal
28	flow rate of Mixed Cl stream, gpm
403944	Cost of clarifier
35218	annual cost of clarifier (6% at 20 years)
0.0024	\$/lb NaCl for lime clarifier Capex
17660	annual electricity
(clarifier costs: http://www.ecodwellinternational.org/images/EC/ECvsLime%20Softening.pdf)	
If a Clarifier Is Used for CaSO₄ Recovery	
14426.56	\$/gpm treated, includes 1.4 installation factor
15.00	kWh/1000 gal
56.00	flow rate of Mixed Cl stream, gpm
807887.60	Cost of clarifier
70435.32	annual cost of clarifier (6% at 20 years)
0.0048	\$/lb NaCl for CaSO ₄ clarifier
35320.32	annual electricity
(belt press costs: http://water.epa.gov/scitech/wastetech/upload/2002_06_28_mtb_belt_filter.pdf)	
Belt Press for CaSO₄	
384.71	dry lb/hr
480.88	1.25 x calculated amount for safety, lb/hr
47500.00	Cost of 0.5 meter belt (2000)
71531.90	Cost of 0.5 meter belt (2013)
6236.48	annual cost of CaSO ₄ belt press
if ED Is Used for Salt Recovery	
3.23	eq/sec NaCl produced
51.00	current density, UTEP experiments, mA/cm ²
0.90	current efficiency
678.61	m ² required
1221507	From Daniel Bar, for very large plants: ~\$1,800/m ² for ED with ACS/CMX
1832260	1.5xDaniel's estimate
159745	Annual CapEx cost of ED
2.38	Ø/lb NaCl Produced, Power cost for ED, Average of two high current density UTEP experiments
280,757	\$/yr, ED Power

Capital costs for each salt recovery method are calculated by summing the reddish highlighted annual capital expenses for the equipment used:

- Method A (CapEx): Evaporation ponds + Lime Clarifier + CaSO₄ clarifier + CaSO₄ Belt Press
- Method B (CapEx): Evaporation ponds + Lime Clarifier
- Method C (CapEx): Method B (CapEx) + ED with monovalent-selective membranes
- Method D (CapEx): Method A (CapEx) + ED with monovalent-selective membranes

Operating costs are found by summing the blue highlighted annual operating costs for the process:

- Method A: Lime + CaSO₄ clarifier power + Lime clarifier power + Evap Pond OpEx
- Method B: Lime + Lime clarifier power + Evap Pond OpEx
- Method C: Method B (OpEx) + ED power
- Method D: Method A (OpEx) + ED power

Sources of information for capital and operating cost:

- Lime softener:
<http://www.ecodwellinternational.org/images/EC/ECvsLime%20Softening.pdf>
- Belt press:
http://water.epa.gov/scitech/wastetech/upload/2002_06_28_mtb_belt_filter.pdf
- ED capital cost (typical): Daniel Bar, of Ameridia
- ED electricity was calculated using experimental data (see chapter 9)